

Research report

# Ionic selectivity of the $\text{Ca}^{2+}/\text{H}^{+}$ antiport in synaptic vesicles of sheep brain cortex

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## Abstract

As we previously reported, synaptic vesicles isolated from sheep brain cortex contain a  $\text{Ca}^{2+}/\text{H}^{+}$  antiport that permits  $\text{Ca}^{2+}$  accumulation inside the vesicles ( $\sim 5$  nmol/mg protein) at expenses of the pH gradient generated by the  $\text{H}^{+}$ -pumping ATPase. We observed that the system associates  $\text{Ca}^{2+}$  influx to  $\text{H}^{+}$  release and operates with low affinity for  $\text{Ca}^{2+}$ . In the present work, we found that  $\text{Ca}^{2+}/\text{H}^{+}$  antiport mediates exchange of protons with other cations such as  $\text{Zn}^{2+}$  and  $\text{Cd}^{2+}$ , suggesting that these cations and  $\text{Ca}^{2+}$  share the same transporter molecules to enter the intravesicular space.  $\text{Zn}^{2+}$  and  $\text{Cd}^{2+}$  induce  $\text{H}^{+}$  release in a concentration-dependent manner (fluorimetrically evaluated) and they inhibit the antiport-mediated  $\text{Ca}^{2+}$  uptake by the vesicles (isotopically measured). In contrast, large cations such as  $\text{Ba}^{2+}$  and  $\text{Cs}^{+}$  do not alter  $\text{Ca}^{2+}$  influx and they are unable to induce proton release from the vesicles. With respect to  $\text{Sr}^{2+}$ , which has an intermediary size relatively to the other groups of cations, we found that it does not induce  $\text{H}^{+}$  liberation from the vesicles, but it has a concentration-dependent inhibitory effect on the  $\text{Ca}^{2+}$ -induced  $\text{H}^{+}$  release and  $\text{Ca}^{2+}$  uptake by the vesicles. These results indicate that the cation selectivity of the synaptic vesicles  $\text{Ca}^{2+}/\text{H}^{+}$  antiport is essentially determined by the size of the cation transported into the vesicles. © 1999 Elsevier Science B.V. All rights reserved.

**Keywords:** Cations transport;  $\text{Ca}^{2+}/\text{H}^{+}$  antiport; Synaptic vesicle; Brain cortex

## 1. Introduction

In neurons, synaptic vesicles function as reservoirs of neurotransmitters that are ready to deliver them by exocytosis after a stimulus-linked signal. However, there are evidences that synaptic vesicles also sequester  $\text{Ca}^{2+}$  [36–38], suggesting that they may function as internal  $\text{Ca}^{2+}$  stores required in localized regions of the exocytotic machinery [3–5,16,40,44,45] or as  $\text{Ca}^{2+}$  loaded bags useful for its exocytotic extrusion [16,37,40]. An ATP-dependent  $\text{Ca}^{2+}$  accumulation has been reported by several investigators [24,32] and, in previous experiments, we found a

$\Delta\text{pH}$ -dependent,  $\text{Ca}^{2+}/\text{H}^{+}$  antiport-mediated  $\text{Ca}^{2+}$  uptake by synaptic vesicles [18], which was recently distinguished from a  $\text{Ca}^{2+}$  transport system mediated by a  $\text{Ca}^{2+}$ -ATPase.

On the other hand,  $\text{Zn}^{2+}$  accumulation inside synaptic vesicles has been visualized with fluorescent probes, particularly in brain neurons [9,17] where release of  $\text{Zn}^{2+}$  to the synaptic cleft appears to modulate the activity of various receptors including ionotropic glutamate receptors and GABA receptors [1,11,14,30,51,52]. The mechanism of  $\text{Zn}^{2+}$  transport into synaptic vesicles is not known, but a putative ZnT-3 transporter has been suggested by Palmiter et al. [34] and Wenzel et al. [50]. They cloned ZnT-3 cDNA and, by in situ hybridization, they detected predominance of ZnT-3 mRNA in the hippocampus and cerebral cortex. Furthermore, they observed that, in brain, the antibody-staining pattern against ZnT-3 protein is very similar to the histochemical localization of zinc by the Timm's reaction [50].

In this work, we investigated the cation selectivity of the  $\Delta\text{pH}$ -dependent  $\text{Ca}^{2+}/\text{H}^{+}$  antiport of synaptic vesicles and we found that it has poor specificity for  $\text{Ca}^{2+}$ , since

*Abbreviations:* CCCP, carbonyl cyanide *m*-chlorophenylhydrazone; EGTA, ethyleneglycol-bis( $\beta$ -aminoethylether)-*N,N,N',N'*-tetraacetic acid; HEPES, *N*-2-hydroxyethylpiperazine-*N'*-2-ethane-sulfonic acid; Tris, 2-amino-2-hydroxymethylpropane-1,3-diol

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it is very efficient to exchange  $H^+$  with small ionic radius cations such as  $Zn^{2+}$  and  $Cd^{2+}$ .

## 2. Materials and methods

### 2.1. Materials

All reagents were analytical grade. CCCP, ATP and acridine orange were obtained from Sigma.  $^{45}CaCl_2$  (2 mCi/ml) was purchased from Amersham.

### 2.2. Isolation of synaptic vesicles

Synaptic vesicles were isolated from sheep brain cortex (obtained from a local slaughterhouse) according to the procedure described previously [20,21]. Briefly, the frozen and crushed sheep brain cortical tissue was homogenized in a solution containing 0.32 M sucrose, 10 mM HEPES-K (pH 7.3), 0.2 mM EGTA, 0.5  $\mu$ g/ml pepstatin and 1  $\mu$ g/ml leupeptin. After centrifugation during 10 min at  $47000 \times g$ , the supernatant was collected and centrifuged again for 40 min at  $120000 \times g$ . The supernatant obtained was layered onto a cushion of 0.65 mM sucrose containing 10 mM HEPES-K (pH 7.3) and centrifuged for 2 h at  $260000 \times g$ . The resulting pellet was resuspended in 0.32 M sucrose plus 10 mM HEPES-K (pH 7.4) and, after

centrifuging for 10 min at  $27000 \times g$ , the supernatant containing the purified synaptic vesicles was collected, frozen in liquid nitrogen and stored at  $-70^\circ C$ . This method allows preparation of a highly pure synaptic vesicles fraction evaluated by assays of immunoblotting and marker enzymes activity [20,21]. Analysis of protein was performed by the method of Gornall et al. [19].

### 2.3. Assay of $Ca^{2+}$ uptake

$Ca^{2+}$  accumulation by isolated synaptic vesicles was measured by rapid filtration and scintillation counting as described previously [18]. Briefly, the membrane vesicles (600  $\mu$ g protein/ml) were incubated at  $30^\circ C$  in a medium containing 60 mM sucrose, 2 mM  $MgCl_2$ , 150 mM KCl, 50  $\mu$ M EGTA, 10 mM Tris-HCl, pH 8.5 and 504  $\mu$ M ATP-Mg. The reactions were initiated by adding 607.7  $\mu$ M  $CaCl_2$  supplemented with  $^{45}CaCl_2$  (10  $\mu$ Ci/mmol) and they were carried out in the absence and in the presence of different cation specimens as indicated in the legends of the figures. At designated reaction times, the reactions were stopped by filtering 800  $\mu$ l aliquots under vacuum. After washing the filters (Millipore HAWP  $\varnothing$  0.45  $\mu$ m) with 3 ml of the reaction medium without  $Ca^{2+}$  and ATP-Mg, the retained radioactivity was measured by liquid scintillation spectrometry.

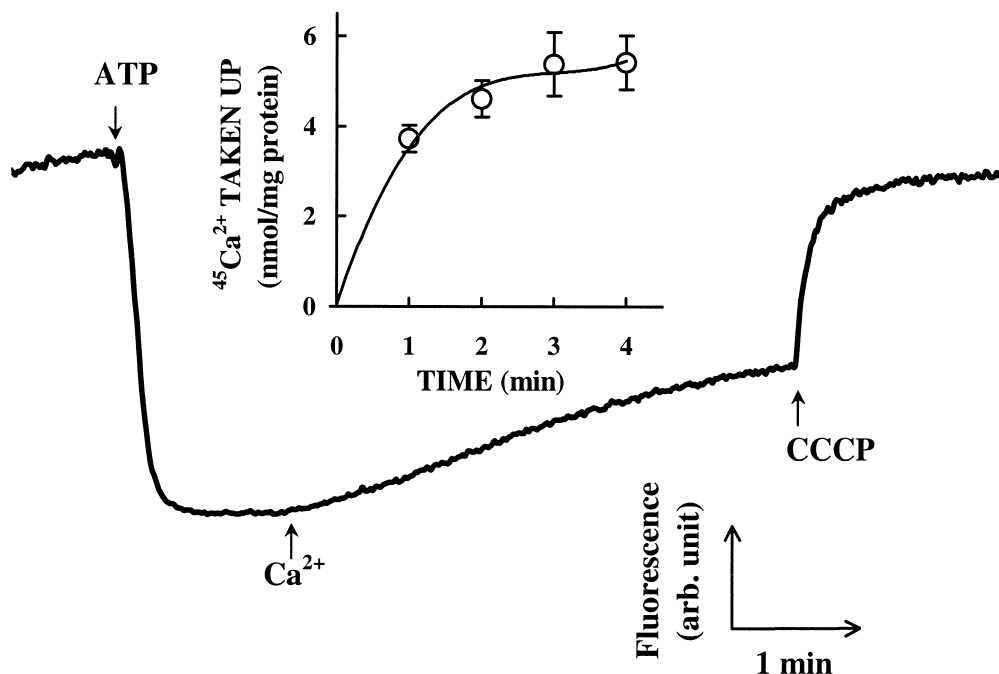


Fig. 1.  $Ca^{2+}/H^+$  antiport activity of synaptic vesicles isolated from sheep brain cortex. Synaptic vesicles (1.2 mg protein) were incubated, at  $30^\circ C$ , in 2 ml of a medium containing 60 mM sucrose, 2 mM  $MgCl_2$ , 150 mM KCl, 50  $\mu$ M EGTA and 10 mM Tris-HCl at pH 8.5. The reaction was started by adding 504  $\mu$ M Mg-ATP and the proton transport was visualized by quenching of acridine orange fluorescence as described in the text. At 1 min of reaction, the  $Ca^{2+}/H^+$  antiport was induced by adding 607.7  $\mu$ M  $CaCl_2$  (500  $\mu$ M  $Ca^{2+}$  free concentration) and its activity was evaluated by recording the fluorescence changes associated to the proton efflux from the vesicles. The ATP-dependent  $\Delta pH$  was checked by using the protonophore CCCP (10  $\mu$ M). The time-course of the  $Ca^{2+}$  uptake associated to  $H^+$  release (plot insert) was measured under the same experimental conditions, except that radioactive calcium ( $^{45}Ca$ ) was used as described in the text. Values represent  $\pm$ S.D. of four separate experiments.

#### 2.4. Assays of proton transport and cation-induced $H^+$ release

ATP-dependent proton transport was measured by following the fluorescence quenching of 2  $\mu\text{M}$  acridine orange [15] as previously described [18]. The membrane vesicles (600  $\mu\text{g}$  protein/ml) were incubated at 30°C in a medium (2 ml) containing 60 mM sucrose, 2 mM  $\text{MgCl}_2$ , 150 mM KCl, 50  $\mu\text{M}$  EGTA and 10 mM Tris-HCl, pH 8.5. Proton accumulation inside the vesicles was initiated by adding 504  $\mu\text{M}$  ATP-Mg and the pH gradient formed was checked by inducing  $H^+$  release with the protonophore CCCP (10  $\mu\text{M}$ ). In some experiments,  $H^+$ -release was induced by adding  $\text{CaCl}_2$  (final free concentration of 500  $\mu\text{M}$ ) or by adding the various cations as described in the

legends of the figures. Proton efflux was evaluated by the fluorescence increase after addition of the uncoupling agents. Controls in media without synaptic vesicles showed no alterations in the fluorescence signal.

The fluorescence changes associated to proton movements were followed by using a Perkin-Elmer computer-controlled spectrofluorometer, Model LS-50. The fluorescence emission was measured at 525 nm (slit width 3.5 nm) using an excitation wavelength of 495 nm (slit width 3.0 nm).

#### 2.5. Analysis of the data

Statistical differences were performed by the Student's *t*-test (two-tailed distribution; two-sample unequal vari-

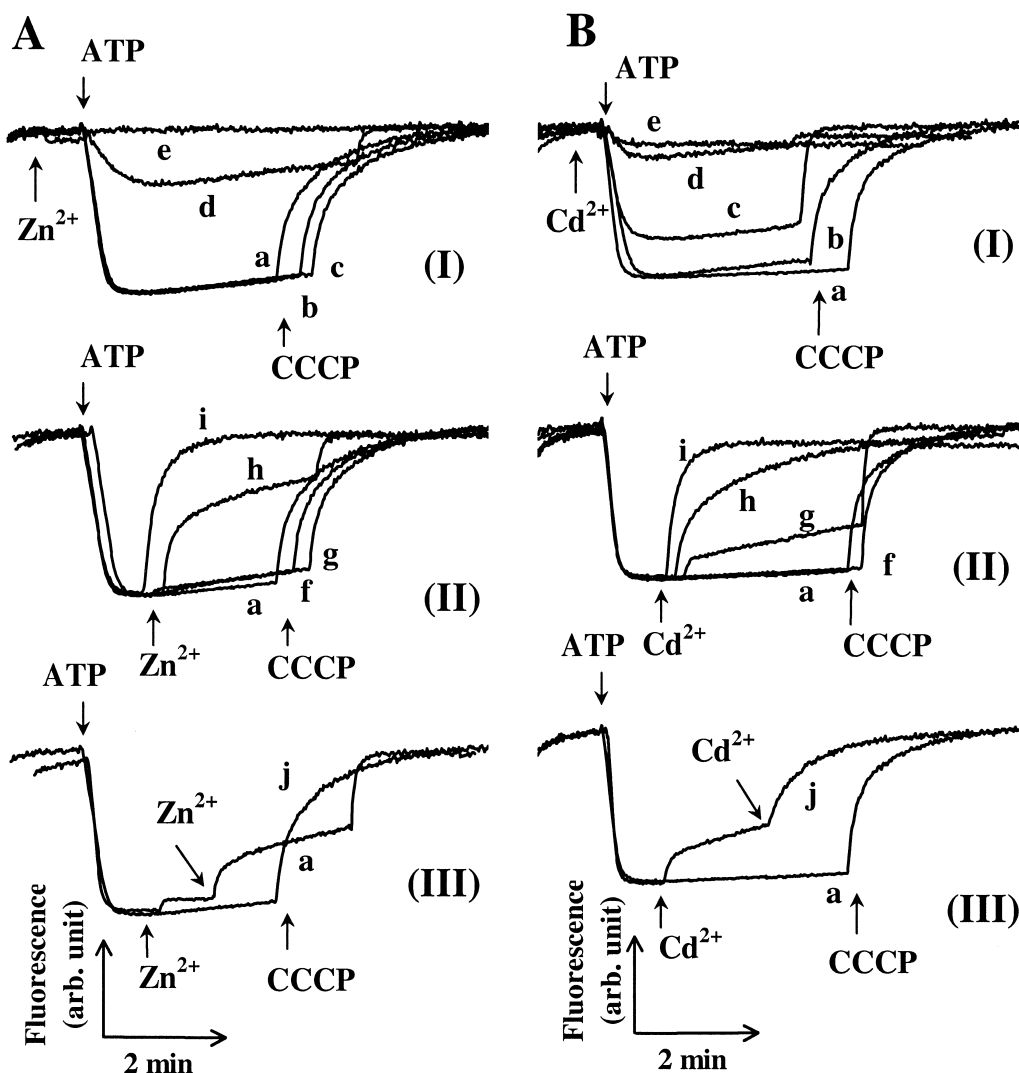


Fig. 2.  $\text{Zn}^{2+}$  (A) and  $\text{Cd}^{2+}$  (B)-induced  $H^+$  release from synaptic vesicle: decrease of the  $\Delta\text{pH}$  magnitude. The experiments were performed as described in the legend of Fig. 1. Panel A, Group I: control (a), and presence of 1  $\mu\text{M}$  (b), 10  $\mu\text{M}$  (c), 100  $\mu\text{M}$  (d) or 250  $\mu\text{M}$  (e) of  $\text{ZnCl}_2$ , before starting the reaction with 504  $\mu\text{M}$  ATP-Mg; Group II: control (a) and addition of 1  $\mu\text{M}$  (f), 10  $\mu\text{M}$  (g), 100  $\mu\text{M}$  (h) or 250  $\mu\text{M}$  (i) of  $\text{ZnCl}_2$ , after reaching maximal  $\Delta\text{pH}$ ; Group III: control (a) and effect of successive additions of 50  $\mu\text{M}$   $\text{ZnCl}_2$  to proton-loaded vesicles (j). Panel B, Group I: control (a) and presence of 20  $\mu\text{M}$  (b), 100  $\mu\text{M}$  (c), 250  $\mu\text{M}$  (d) or 500  $\mu\text{M}$  (e) of  $\text{CdCl}_2$ , before starting the reaction with 504  $\mu\text{M}$  ATP-Mg; Group II: control (a) and addition of 20  $\mu\text{M}$  (f), 100  $\mu\text{M}$  (g), 250  $\mu\text{M}$  (h) or 500  $\mu\text{M}$  (i) of  $\text{CdCl}_2$ , after reaching maximal  $\Delta\text{pH}$ ; Group III: control (a) and effect of successive additions of 150  $\mu\text{M}$   $\text{CdCl}_2$  (j) to proton-loaded vesicles. The protonophore CCCP (10  $\mu\text{M}$ ) was used to check  $\Delta\text{pH}$  generation at expenses of ATP.

ance) and the  $p$ -values are indicated in the legends of the figures.

### 3. Results

As reported before [18], we observed that, after formation of a pH gradient across synaptic vesicle membranes, addition of  $\text{Ca}^{2+}$  (500  $\mu\text{M}$  free concentration) to the reaction medium induces  $\text{H}^+$  release from the vesicles (Fig. 1). Indeed, the fluorescence quenching of acridine orange, observed during the proton accumulation by synaptic vesicles, was reversed by  $\text{Ca}^{2+}$  addition to the medium. We also observed that the  $\text{Ca}^{2+}$ -induced  $\text{H}^+$  release from the vesicles was accompanied by a  $\text{Ca}^{2+}$  movement into the vesicles, as judged by measurements of the  $\text{Ca}^{2+}$  taken up along the time under the same experimental conditions (Fig. 1, insert).  $\text{Ca}^{2+}$  uptake increased with time, reaching

a plateau ( $\sim 5$  nmol/mg protein) at 2 min of reaction, in good correlation with the time course of the  $\text{H}^+$  release induced by  $\text{Ca}^{2+}$  (Fig. 1).

The  $\Delta\text{pH}$  dissipation by  $\text{Ca}^{2+}$  is partial, since further addition of the protonophore CCCP promotes complete discharge of the ATP-dependent proton accumulated (Fig. 1).

Interestingly, we observed that  $\text{Zn}^{2+}$  and  $\text{Cd}^{2+}$  also promotes dissipation of the pH gradient across the synaptic vesicle membranes (Fig. 2A and B). The magnitude of this effect is  $\text{Zn}^{2+}$  or  $\text{Cd}^{2+}$  concentration-dependent, being maximal at about 250  $\mu\text{M}$   $\text{Zn}^{2+}$  or 500  $\mu\text{M}$   $\text{Cd}^{2+}$ , respectively (Fig. 2A and B, group II, curves 'i'). We also observed that, after partial  $\text{H}^+$  release induced by 100  $\mu\text{M}$   $\text{Zn}^{2+}$  or 150  $\mu\text{M}$   $\text{Cd}^{2+}$ , a new fraction of  $\text{H}^+$  was liberated by further addition of  $\text{Ca}^{2+}$  (500  $\mu\text{M}$  free concentration) (Fig. 3, curves 'b' and 'c') or of a new aliquot of  $\text{Zn}^{2+}$  or  $\text{Cd}^{2+}$  (Fig. 2A and B, group III, curves 'j'). It

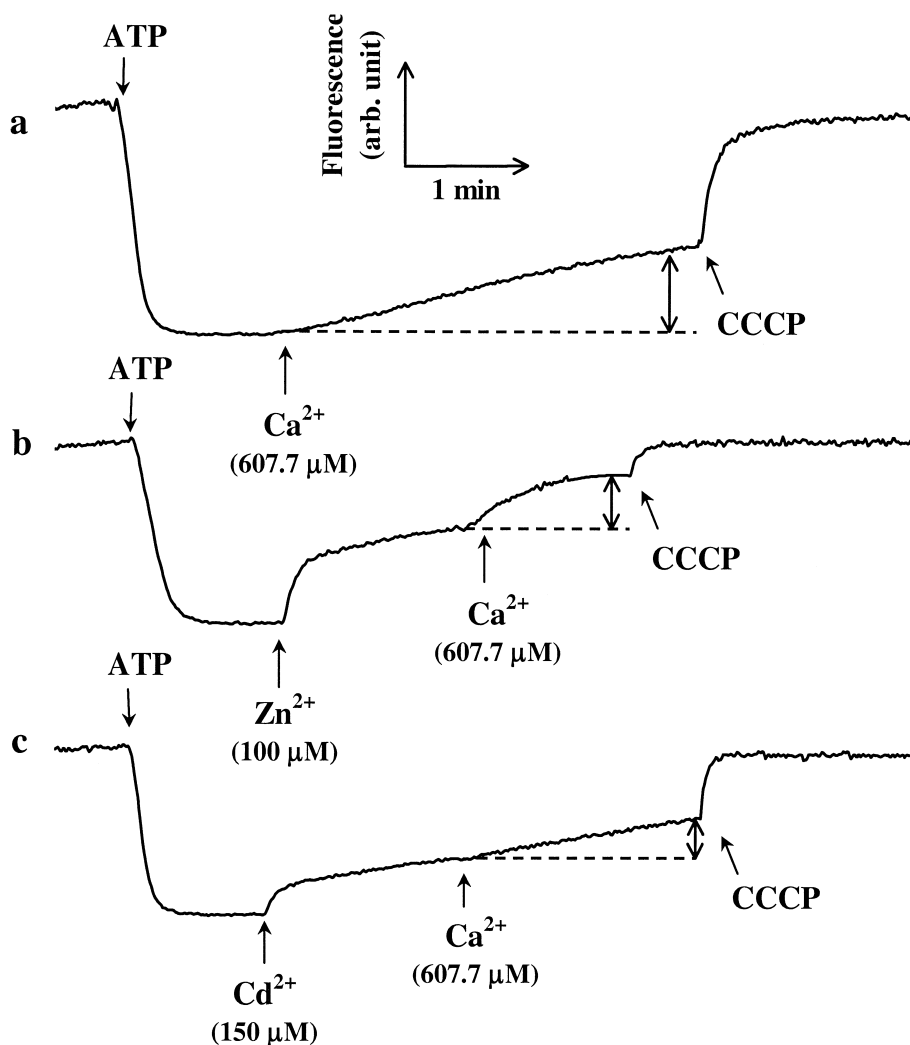


Fig. 3.  $\text{Zn}^{2+}$  and  $\text{Cd}^{2+}$  inhibition of the  $\text{Ca}^{2+}$ -induced  $\text{H}^+$  release from synaptic vesicle. The experiments were performed as described in the legend of Fig. 1. (a), control; (b), effect of successive additions of 100  $\mu\text{M}$   $\text{ZnCl}_2$  and 607.7  $\mu\text{M}$   $\text{CaCl}_2$  (500  $\mu\text{M}$  free concentration) to proton-loaded vesicles; (c), effect of successive additions of 150  $\mu\text{M}$   $\text{CdCl}_2$  and 607.7  $\mu\text{M}$   $\text{CaCl}_2$  (500  $\mu\text{M}$  free concentration) to proton-loaded vesicles. The protonophore CCCP (10  $\mu\text{M}$ ) was used to check  $\Delta\text{pH}$  generation at expenses of ATP.

appears that, like it happens with  $\text{Ca}^{2+}$ ,  $\text{Zn}^{2+}$  and  $\text{Cd}^{2+}$  enter the vesicles in exchange with protons. However, the antiport activity reaches a steady-state that is maintained by the activity of the proton-pump until new addition of cations reactivates the antiport by increasing the cation concentration in the medium.

The uncoupler properties of  $\text{Zn}^{2+}$  and  $\text{Cd}^{2+}$ , demonstrated by the cation-induced  $\text{H}^+$  release, were checked in experiments where the presence of these cations before ATP addition to the medium reduced the magnitude of the  $\Delta\text{pH}$  or, at critical concentrations (250  $\mu\text{M}$   $\text{Zn}^{2+}$  or 500  $\mu\text{M}$   $\text{Cd}^{2+}$ ), prevented its formation across the vesicle membranes (Fig. 2A and B, group I, curves 'e').

Studying the antiport with respect to the  $\text{Ca}^{2+}$  transport under conditions where  $\text{Ca}^{2+}$  was added to a  $\text{Zn}^{2+}$  or  $\text{Cd}^{2+}$ -containing medium, we observed that it was reduced as the concentration of those cations increased in the extravesicular space (Fig. 4A and B). This suggests that, indeed, these cations enter the vesicles at expenses of the  $\Delta\text{pH}$ , and promote its dissipation by inducing  $\text{H}^+$  release.

In contrast to the small cations  $\text{Zn}^{2+}$  and  $\text{Cd}^{2+}$  (0.74 and 0.97 Å dehydrated ionic radius, respectively), we observed that cations with large ionic radius, such as  $\text{Ba}^{2+}$  and  $\text{Cs}^+$  (1.35 and 1.69 Å dehydrated ionic radius, respectively) do not show uncoupling effects on the proton pump activity (Fig. 5) and do not inhibit significantly the  $\text{Ca}^{2+}/\text{H}^+$  antiport-mediated  $\text{Ca}^{2+}$  accumulation by the

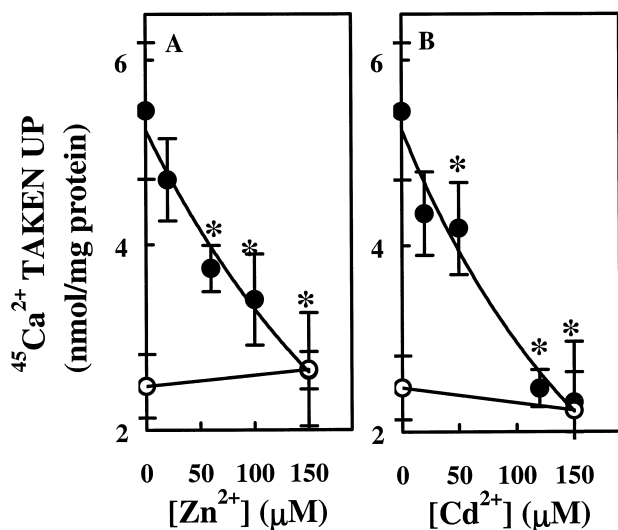


Fig. 4.  $\text{Zn}^{2+}$  (A) and  $\text{Cd}^{2+}$  (B) inhibition of the  $\text{Ca}^{2+}$  transport mediated by the synaptic vesicles  $\text{Ca}^{2+}/\text{H}^+$  antiport. The vesicles (600  $\mu\text{g}$  protein/ml) were incubated during 50 s in a medium containing 60 mM sucrose, 2 mM  $\text{MgCl}_2$ , 150 mM KCl, various concentrations of  $\text{ZnCl}_2$  or  $\text{CdCl}_2$  (if present), 50  $\mu\text{M}$  EGTA and 10 mM Tris-HCl (pH 8.5) in the absence (O-O) and in the presence (●-●) of 504  $\mu\text{M}$  Mg-ATP.  $\text{Ca}^{2+}$  uptake reaction was started by adding 500  $\mu\text{M}$  free  $\text{Ca}^{2+}$  (607.7  $\mu\text{M}$   $\text{CaCl}_2$  supplemented with 10  $\mu\text{Ci}/\text{mmol}$   $^{45}\text{CaCl}_2$ ) and carried out for 3 min. The  $\text{Ca}^{2+}$  taken up was determined as described in the text. Values represent  $\pm$ S.D. of four separate experiments. Asterisk indicates significant inhibition ( $*p < 0.01$ ) of  $\text{Ca}^{2+}$  uptake.

vesicles (Fig. 6). In fact, no  $\text{H}^+$  release was induced by  $\text{Ba}^{2+}$  or  $\text{Cs}^+$ , even at the high concentration of 500  $\mu\text{M}$  (Fig. 5A and B, group II, curves 'c'). Similarly, they do not prevent  $\Delta\text{pH}$  formation across the synaptic vesicle membranes as revealed by the identical magnitude of the acridine orange fluorescence quenching obtained when those cations are absent (control) or present in the reaction medium before starting  $\text{H}^+$  transport with addition of ATP (Fig. 5A and B, group I, curves 'b'). On the other hand,  $\text{Ba}^{2+}$  and  $\text{Cs}^+$  did not reduce significantly the secondary active  $\text{Ca}^{2+}$  transport ( $\text{Ca}^{2+}/\text{H}^+$  antiport-mediated) by the vesicles (Fig. 5, group III) as it happens with the other cations (Fig. 4).

Taking in consideration that large cations such as  $\text{Ba}^{2+}$  and  $\text{Cs}^+$  do not induce antiport activity like  $\text{Ca}^{2+}$  (0.99 Å dehydrated ionic radius) or the group of small cations ( $\text{Zn}^{2+}$  and  $\text{Cd}^{2+}$ ) does, we investigated whether  $\text{Sr}^{2+}$ , which has an intermediary size (1.13 Å dehydrated ionic radius), is able to induce exchange reaction. Interestingly, we observed that high concentrations of  $\text{Sr}^{2+}$  (500  $\mu\text{M}$ ) do not induce  $\text{H}^+$  release from proton loaded vesicles (Fig. 6, group II, curve 'c'), but it has a significant inhibitory effect (24%) on the  $\text{Ca}^{2+}$  uptake by synaptic vesicles (Fig. 7A) and on the  $\text{Ca}^{2+}$ -induced  $\text{H}^+$  release from the vesicles (Fig. 7B, curve 'a') when both cations, at similar concentrations (500  $\mu\text{M}$ ), are present in the reaction medium. It appears, therefore, that  $\text{Sr}^{2+}$  blocks the cation/ $\text{H}^+$  antiport, preventing proton efflux in exchange with  $\text{Ca}^{2+}$  influx. However, when  $\text{Ca}^{2+}$  is further added to a  $\text{Sr}^{2+}$ -containing medium, a slight  $\text{Ca}^{2+}/\text{H}^+$  exchange occurs. These results indicate that  $\text{Sr}^{2+}$  inhibits synaptic vesicles  $\text{Ca}^{2+}$  uptake, by blocking the antiport with consequent decrease in the number of its available molecules.

#### 4. Discussion

In previous investigations, we observed that  $\text{Ca}^{2+}$  (100–500  $\mu\text{M}$ ) partially dissipates the pH gradient generated by synaptic vesicles  $\text{H}^+$ -ATPase and, simultaneously, it is accumulated inside the vesicles at expenses of the  $\Delta\text{pH}$  energy [18]. Recently, we have observed that synaptic vesicles also accumulate  $\text{Ca}^{2+}$  by a process which requires low  $\text{Ca}^{2+}$  concentrations ( $< 100$   $\mu\text{M}$ ) and it is  $\Delta\text{pH}$ -independent and vanadate-inhibited. It appears, therefore, that synaptic vesicles can take up  $\text{Ca}^{2+}$  by two distinct mechanisms: a low  $\text{Ca}^{2+}$  affinity,  $\text{Ca}^{2+}/\text{H}^+$  antiport-mediated system and a high  $\text{Ca}^{2+}$  affinity,  $\text{Ca}^{2+}$ -ATPase-mediated system.

Interestingly, we observed in this work that the  $\text{Ca}^{2+}/\text{H}^+$  antiport system has poor specificity for  $\text{Ca}^{2+}$ , since it exchanges protons with various types of smaller ionic radius cations. We found that small cations such as  $\text{Zn}^{2+}$  and  $\text{Cd}^{2+}$  easily dissipate the pH gradient in a concentration-dependent manner as it happens with  $\text{Ca}^{2+}$ .

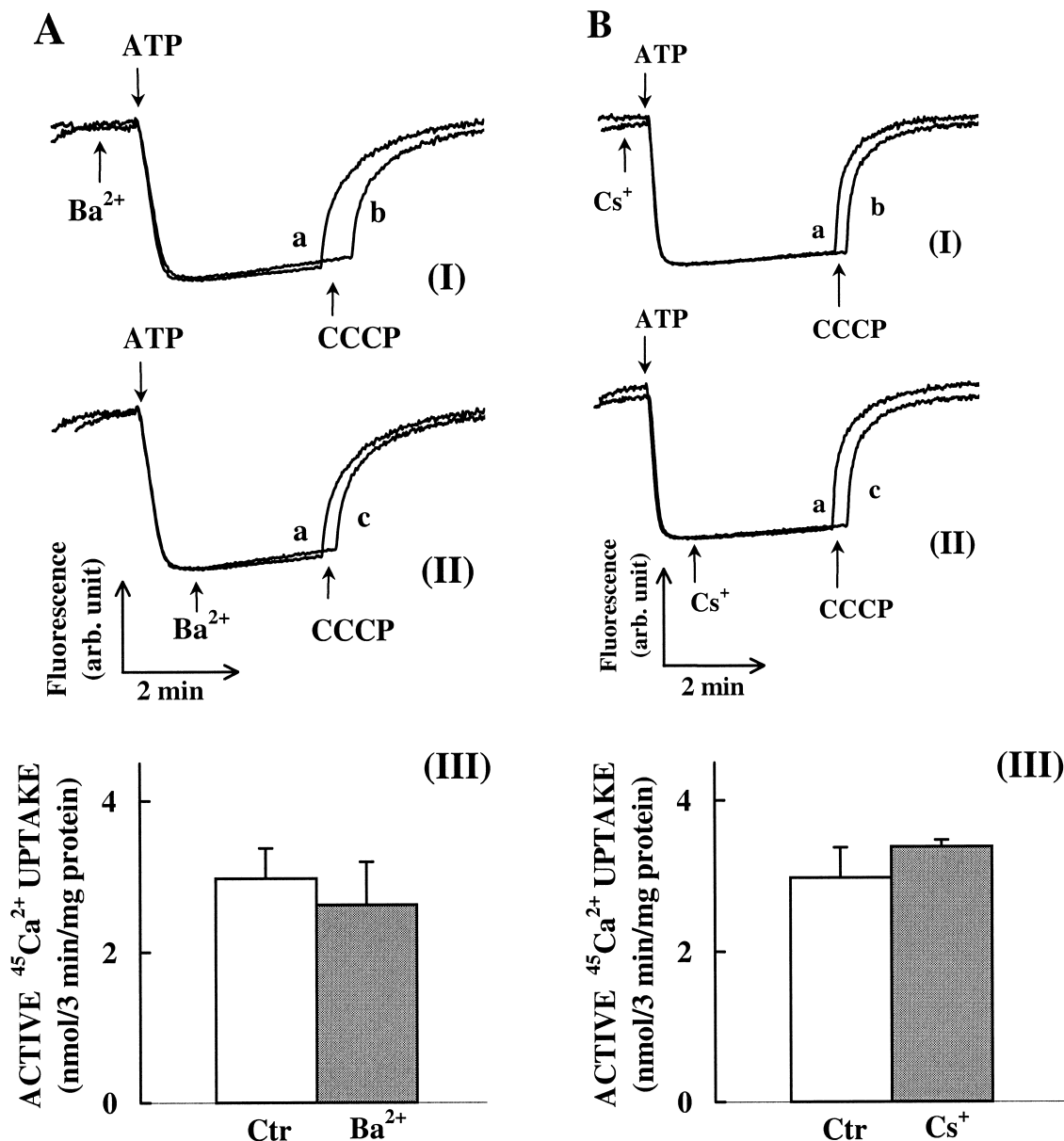


Fig. 5. Inability of Ba<sup>2+</sup> (A) and Cs<sup>+</sup> (B) to alter the  $\Delta$ pH and the antiport-mediated Ca<sup>2+</sup> uptake by synaptic vesicle membranes. The experiments were performed as described in the legend of Fig. 1. The protonophore CCCP (10  $\mu$ M) was used to check  $\Delta$ pH generation at expenses of ATP. Panel A, Group I: control (a) and addition of 500  $\mu$ M BaCl<sub>2</sub> before starting the reaction with 504  $\mu$ M ATP-Mg (b); Group II: control (a) and addition of 500  $\mu$ M BaCl<sub>2</sub> after reaching maximal  $\Delta$ pH (c). Panel B, Group I: control (a) and addition of 500  $\mu$ M CsCl before starting the reaction with 504  $\mu$ M ATP-Mg (b); Group II: control (a) and addition of 500  $\mu$ M CsCl after reaching maximal  $\Delta$ pH (c). The Ca<sup>2+</sup> uptake was measured as described in Fig. 4, except that 500  $\mu$ M BaCl<sub>2</sub> or CsCl were used instead of ZnCl<sub>2</sub> and CdCl<sub>2</sub> (Group III). Values represent  $\pm$ S.D. of four separate experiments.

However, the cation concentrations required for the effect decreased with the ionic radius of cations. About 500  $\mu$ M CaCl<sub>2</sub> is efficient to induce H<sup>+</sup> release (Fig. 1), whereas only 250  $\mu$ M CdCl<sub>2</sub> and about 100  $\mu$ M ZnCl<sub>2</sub> are high enough to induce strong liberation of protons (Fig. 2A and B). It appears that the affinity of the antiport is determined by the size of cations and, probably, competition exists between them, when they are simultaneously present in the reaction medium. Indeed, Ca<sup>2+</sup>-induced H<sup>+</sup> release was inhibited when 607.7  $\mu$ M CaCl<sub>2</sub> (500  $\mu$ M free concentration) was added in the presence of 150  $\mu$ M CdCl<sub>2</sub> (Fig. 3,

curve 'c') or 100  $\mu$ M ZnCl<sub>2</sub> (Fig. 3, curve 'b') as compared to the release observed when exclusively CaCl<sub>2</sub> existed in the reaction medium (Fig. 3, curve 'a'). This conclusion appears plausible if we take in consideration that, in contrast to Ca<sup>2+</sup>, a second addition of ZnCl<sub>2</sub> or CdCl<sub>2</sub> (Fig. 2A and B, group III) does not decrease H<sup>+</sup> liberation from the vesicles but, instead, it promotes H<sup>+</sup> release stimulation due to an increased concentration of these cations in the reaction medium (steep dose response).

On the other hand, Sr<sup>2+</sup>, which is larger than Ca<sup>2+</sup>, does not induce H<sup>+</sup> release from the vesicles (Fig. 6) and

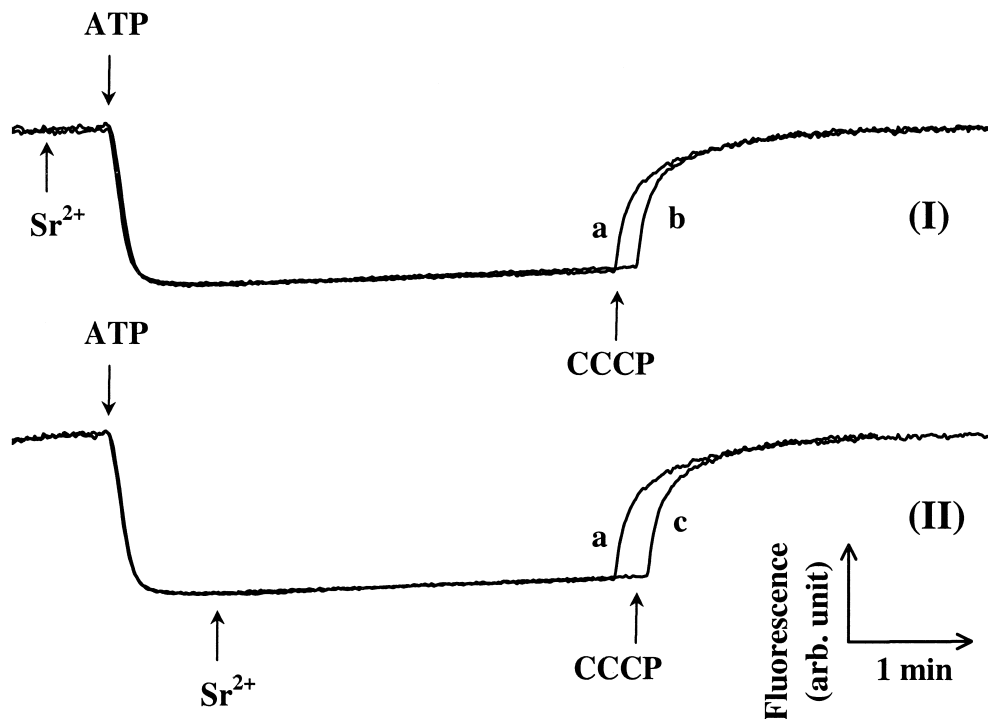


Fig. 6. Effect of Sr<sup>2+</sup> on the ATP-dependent ΔpH formation across synaptic vesicle membranes. The experiments were performed as described in the legend of the Fig. 1. The protonophore CCCP (10 μM) was used to check ΔpH generation at expenses of ATP. Group I: control (a) and addition of 500 μM SrCl<sub>2</sub> before starting the reaction with 504 μM ATP-Mg (b); Group II: control (a) and addition of 500 μM SrCl<sub>2</sub> after reaching maximal ΔpH (c).

it has a strong inhibitory effect on both Ca<sup>2+</sup> uptake (Fig. 7A) and Ca<sup>2+</sup>-induced H<sup>+</sup> release (Fig. 7B) mediated by

the antiporter. This indicates that Sr<sup>2+</sup> is not small enough to enter the vesicles in exchange with protons, but it blocks

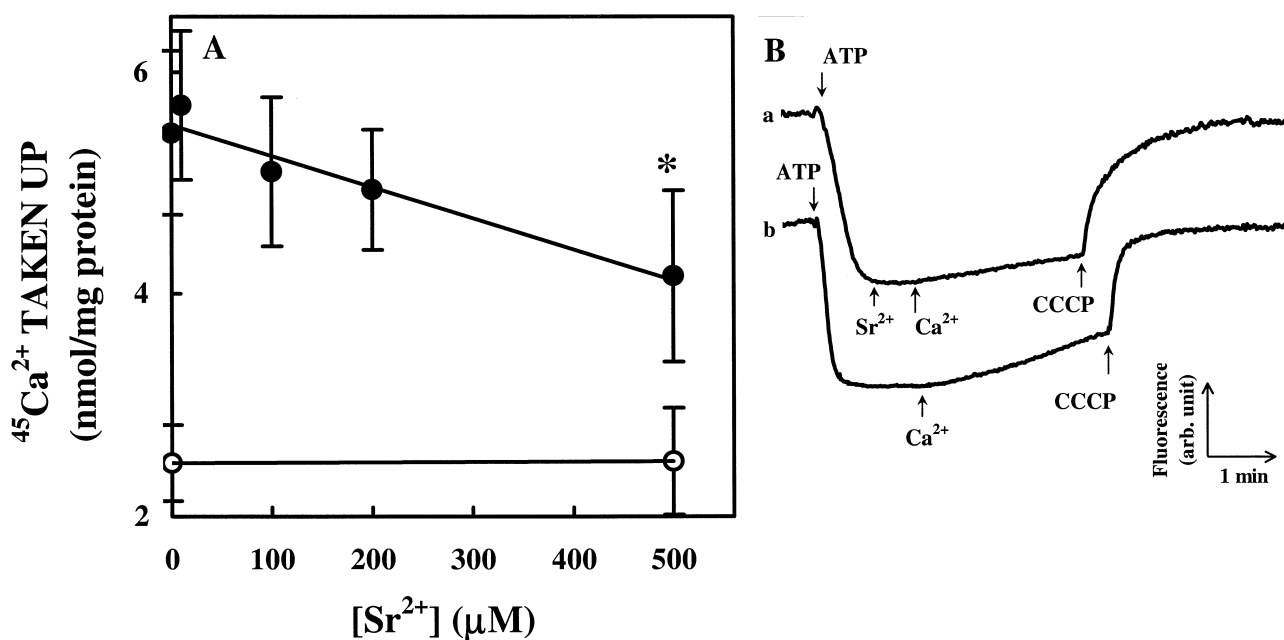


Fig. 7. Effect of Sr<sup>2+</sup> on the Ca<sup>2+</sup> uptake (A) and on the Ca<sup>2+</sup>-induced H<sup>+</sup> release (B) mediated by the synaptic vesicle Ca<sup>2+</sup>/H<sup>+</sup> antiport. Panel A: the vesicles (600 μg protein/ml) were incubated during 50 s in a medium containing 60 mM sucrose, 2 mM MgCl<sub>2</sub>, 150 mM KCl, 50 μM EGTA, various concentrations of SrCl<sub>2</sub> (if present) and 10 mM Tris-HCl (pH 8.5) in the absence (○-○) or in the presence (●-●) of 504 μM Mg-ATP. The reactions were started by adding 500 μM free Ca<sup>2+</sup> (607.7 μM CaCl<sub>2</sub> supplemented with 10 μCi/mmol <sup>45</sup>CaCl<sub>2</sub>) and carried out for 3 min. The Ca<sup>2+</sup> taken up was measured as described in the text. Values represent ±S.D. of four separate experiments. Asterisk indicates significant inhibition (\**p* < 0.01) of Ca<sup>2+</sup> uptake. Panel B: the experiments were performed as described in the legend of Fig. 1. (a), effect of successive additions of 500 μM SrCl<sub>2</sub> and 607.7 μM CaCl<sub>2</sub> (500 μM free concentration) to proton-loaded vesicles; (b), addition of 607.7 μM CaCl<sub>2</sub> (500 μM free concentration) to proton-loaded vesicles.

the antiporter by interacting with it, causing consequent loss of its availability for other cations ( $\text{Ca}^{2+}$ ).

In contrast, larger cations, such as  $\text{Ba}^{2+}$  and  $\text{Cs}^+$ , do not induce  $\text{H}^+$  release from the vesicles and do not inhibit  $\text{Ca}^{2+}$  uptake either (Fig. 5), suggesting that they are too large, not only to enter the vesicles through the antiport, but also to interact with it at the level of its cation binding site. It appears, therefore, that  $\text{Ba}^{2+}$  and  $\text{Cs}^+$  are neither blockers, nor competitive inhibitors of the cation/proton exchanger, as it happens with  $\text{Sr}^{2+}$  and with smaller cations ( $\text{Zn}^{2+}$  and  $\text{Cd}^{2+}$ ), respectively. It is interesting to note that all reactions were carried out in the presence of  $\sim 2$  mM  $\text{Mg}^{2+}$  (0.65 Å dehydrated ionic radius) which, under these conditions, functions as a coupling factor of the  $\text{H}^+$  pump activity necessary to generate the motive force for the antiport operativity. We do not know whether  $\text{Mg}^{2+}$  interferes with the antiport activity, however, it should not affect its response specificity to the other cations studied, as all reactions were performed under the same conditions ( $\text{Mg}^{2+}$  present).

An important question about the cation/proton exchanger in synaptic vesicles concerns its physiological function. As we report here, it is poorly selective for small divalent cations, so that its specificity is essentially determined by the ionic radius of the cations.

Although no important role has been reported for  $\text{Cd}^{2+}$  in the central nervous system,  $\text{Zn}^{2+}$  and  $\text{Ca}^{2+}$  appear to be involved in important neuronal functions [2–4,6,22,23,44,45,49,53].  $\text{Ca}^{2+}$  acts at various levels of the exocytotic machinery [5,28,46,48,54] and  $\text{Zn}^{2+}$  has been described as an efficient modulator of both excitatory and inhibitory synaptic transmission [11,14,30,51–53]. Both cations have been found in synaptic vesicles [9,17,24,31,32,36,38,39,43,50] which, essentially, serve as neurotransmitter reservoirs [25,29,33,41].

The  $\text{Ca}^{2+}$  sequestration properties of synaptic vesicles may be important to regulate cytoplasm  $\text{Ca}^{2+}$  concentration [7,16,36,40], particularly at the level of the high  $\text{Ca}^{2+}$  concentration microdomains previously suggested [27,47]. On the other hand, simultaneous exocytosis of  $\text{Zn}^{2+}$  and neurotransmitters may permit receptor modulation by this cation.

In spite of these important functions, both cations become toxic when their intracellular concentration exceed levels of tolerance [8,12,13,26,42], unless protective mechanisms of sequestration are induced in the cell. Our results show that  $\text{Ca}^{2+}$  and  $\text{Zn}^{2+}$  may be taken up by a low affinity cation/proton exchange system and, under conditions of cation concentration stress, they have a strong dissipater effect on the synaptic vesicle  $\Delta\text{pH}$ , due to accumulation of cations in exchange with protons release. Indeed,  $\text{Ca}^{2+}$  inhibition of neurotransmitter exocytosis has been reported [10] probably because the proton motive force responsible for transport of neurotransmitters into the vesicles is abolished by high  $\text{Ca}^{2+}$  concentrations. A similar inhibitory effect on neurotransmitter release was

reported for  $\text{Zn}^{2+}$  [35], which is not surprising since  $\text{Zn}^{2+}$  ( $> 100$   $\mu\text{M}$ ), like  $\text{Ca}^{2+}$  ( $> 300$   $\mu\text{M}$ ), uncouples the proton pump by inducing  $\text{H}^+$  release through the antiport reported here.

Our results indicate that  $\text{Zn}^{2+}$ ,  $\text{Ca}^{2+}$  and other small cations ( $\text{Cd}^{2+}$ ) share the same antiport system to enter the vesicles in exchange with protons. However, we do not know whether the antiport described here corresponds to the same transporter molecule ( $\text{ZnT-3}$ ) previously suggested by Palmiter et al. in cloning experiments [34] and ultrastructural observations [50].

The results reported here suggest that moderate concentrations of  $\text{Zn}^{2+}$  and  $\text{Ca}^{2+}$  permit their limited storing into the vesicles concomitantly with neurotransmitters, contributing altogether for the neurotransmission process, whereas high concentrations of those cations permit a great cation loading which may be of physiological importance as a mean of their exocytotic discharging under stress conditions.

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