

Ana Isabel Costa Escudeiro

## NANOCOMPOSITE ZrC/a-C(:H) COATINGS FOR POTENTIAL APPLICATIONS ONTO BIOMEDICAL IMPLANTS

Doctoral Thesis in Mechanical Engineering, branch Surface Engineering, supervised by Professor Doctor Albano Augusto Cavaleiro Rodrigues de Carvalho and Professor Doctor Tomas Polcar, submitted to the Department of Mechanical Engineering, Faculty of Sciences and Technology of the University of Coimbra.



UNIVERSIDADE DE COIMBRA



DEPARTAMENTO DE ENGENHARIA MECÂNICA

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**Supervisors** 

Professor Doctor Albano Cavaleiro Professor Doctor Tomas Polcar

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#### **Tecnical Sheet:**

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Cover: The cover images shows four different figures that briefly summarize the different steps undergo during the thesis:

Top left: Raman spectra of the a-C (left) and a-C\_Zr(4) (right) films. Top right: 3D profile of the as-deposited a-C\_Zr(4) film. Bottom left: Cross sectional TEM BF micrograph of a-C\_Zr(14) film. Bottom right: Hip prosthesis stylization. Centre: Cross-sectional SEM micrograph of a-C\_Zr(4) film.

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Aos meus pais e irmão



Peanuts by Schulz

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Obrigada por me fazerem acreditar!

#### RESUMO

A longevidade de implantes biomédicos tem vindo a ser associada à libertação de partículas para o organismo e à sua interação com todos os elementos do meio envolvente. Das mais problemáticas encontram-se os resíduos de polietileno provenientes das combinações metal-polímero e os iões metálicos das metal-metal causando reações inflamatórias agudas nos tecidos que suportam o implante levando à sua consequente falha. A utilização da engenharia de superfícies, nomeadamente através da deposição de revestimentos finos, para reduzir a produção e libertação de partículas de desgaste e iões é cada vez mais apontada como uma possível solução. No entanto, foram reportados problemas de adesão *in service* que limitam a sua implantação e uso clínico.

Esta tese incide na síntese e caracterização de revestimentos nanocompósitos de carbono amorfo incorporado com nanopartículas de ZrC com potencial para melhorarem o tempo de vida de implantes biomédicos. Os revestimentos foram depositados por co-pulverização catódica por magnetrão em atmosfera reativa (Ar+CH<sub>4</sub>) e inerte (Ar), permitindo o crescimento de filmes hidrogenados e não hidrogenados, respetivamente. O primeiro grande desafio que motivou a realização deste trabalho foi a necessidade de melhorar a adesão dos revestimentos o que foi possível com a interposição de uma intercamada em gradiente composta por Ti/TiN/TiNC entre o revestimento e o substrato.

A incorporação de diferentes teores de Zr na matriz de carbono amorfo foi possível variando o número de pastilhas de Zr colocadas no alvo de grafite. Assim, foram produzidos filmes com teores em Zr num intervalo de 0 a 14 %at. e de 0 a 10 %at. para os filmes sem e com H, respetivamente. Os filmes apresentaram uma estrutura nanocompósita consistindo em nanopartículas de ZrC (diâmetro entre I a 2.2 nm) distribuídas aleatoriamente numa matriz de carbono. A presença dos nanocristais permitiu a diminuição das tensões residuais dos filmes não-hidrogenados e hidrogenados melhorando a adesão dos revestimentos e influenciando fortemente as propriedades mecânicas dos revestimentos: maior dureza (H) e módulo de Young (E). Os revestimentos com cristalites mais pequenas revelaram um comportamento elástico mais favorável (maiores valores de H/E e  $H^3/E^2$ ) o que se revelou particularmente importante quando os revestimentos foram depositados em substratos com baixa capacidade de resistir a carga de contacto, tais como o Ti puro (CP-Ti) e a liga de Ti (Ti6Al4V). No que diz respeito às propriedades tribológicas dos revestimentos quando

ix

testados contra bolas de Ti6Al4V, os filmes dopados com Zr demonstraram um comportamento independente das condições ambientais do teste (ao ar e em soluções fisiológica e contendo proteínas). No entanto, o desgaste do contracorpo diminuiu.

Do ponto de vista biomédico, a presença de nanocristias de ZrC desempenhou um papel fundamental, alterando as propriedades físico-químicas dos revestimentos: aumentou o ângulo de contacto e, consequentemente, melhorou a adsorção de proteína. Na realidade, a capacidade de efetivamente adsorver proteínas demonstrou ser particularmente importante na proteção das superfícies ao deslizarem em contacto direto.

De modo a avaliar o desgaste dos revestimentos contra materiais poliméricos, os revestimentos foram testados num equipamento pino disco multidirecional que replica o movimento "cruzado" que ocorre entre os materiais constituintes das próteses articulares. Apesar da interposição de uma intercamada otimizada em gradiente e melhor adesão devido à incorporação de Zr nos filmes, os filmes dopados delaminaram após alguns milhares de ciclos. O efeito sinergético entre os diferentes mecanismos de corrosão devido à presença de fluidos corpóreos, e as elevadas tensões cíclicas de corte podem ter sido a causa da falha da interface e da consequente delaminação do filme.

**Palavras Chave:** Filmes nanocompósitos de ZrC; Carbono amorfo; ZrC/a-C(:H); Biotribologia; DLC

#### ABSTRACT

The release of particulate matter derived from biomedical implants has been considered determinant for their long-term survival. Polyethylene debris derived from polymer-on-metal coupling together with the metal ion release on metal-on-metal pairs appear to cause the most pronounced tissue reactions which may induce inflammatory reactions associated with the ultimate failure of the implant. The use of surface engineering, namely coating technology, offers an alternative approach in order to reduce the production of the wear debris. However, a critical feature for such an alternative technology is the adhesion of the coating on the metallic substrates after implantation and clinical use.

This thesis deals with the synthesis and characterization of nanocomposite ZrC amorphous carbon (a-C) coatings with potential to improve the life time of biomedical implants. The coatings were deposited by dc magnetron sputtering in reactive (Ar+CH<sub>4</sub>) and non-reactive (Ar) atmosphere in order to produce hydrogenated and non-hydrogenated coatings, respectively. The first challenge to motivate this work was the need for high adhesion of the coatings. The critical loads (and thus adhesion) of the coatings has strongly increased with the application of a Ti/TiN/TiNC gradient layer between coating and substrate.

Zr was incorporated into the C-matrix by varying the number of Zr pellets embedded in a graphite target. Different Zr contents, ranging from 0 to 14 at.% for nonhydrogenated coatings and 0 to 10 at.% for the hydrogenated ones, were obtained. Nanocomposite based films consisting of small ZrC nanoparticles (I to 2.2 nm) randomly distributed in the C-matrix for Zr content as low as 3 at.% were achieved. The presence of such nanocrystals relaxed the non-hydrogenated and hydrogenated films, thus improving the adhesion of the coatings and strongly influencing the overall mechanical properties of the coatings: high hardness (H) and Young modulus (E). Coatings with small crystallites revealed more favourable elastic-plastic response (lower H/E and  $H^{3}/E^{2}$ ) which was particularly important when depositing on low load carrying capacity substrates such as commercial pure titanium (CP-Ti) and titanium alloy (Ti6Al4V). Regarding the tribological properties of the coatings sliding against Ti6Al4V balls, the Zr-doped coatings behaviour was almost independent of the environmental conditions (ambient air, physiological and protein-containing solutions). Nevertheless, the counterbody wear was lower when compared to that of sliding pure carbon films. From the biomedical point of view, the presence of ZrC nanocrystals played an

important role by changing the physicochemical properties of the coatings: increasing contact angle and, consequently, enhancing albumin adsorption. This was particularly important, since the ability to effectively adsorb proteins protected both surfaces from rubbing in direct contact.

In order to evaluate the wear behaviour against polymeric materials, Zrcontaining films were tribologically tested in multidirectional pin-on-disk equipment, which replicated the "cross-shear" motion occurring between the mating components of a joint prosthesis. Despite the use of an optimized gradient interlayer scheme and improved adhesion by incorporation of Zr into the C-matrix, the Zr-containing film delaminated after several million of cycles. The synergetic effect of stress-induced corrosion through biological fluid and high cyclic shear stress may have caused interface fatigue and subsequent delamination of the coatings.

Keywords: Nanocomposite ZrC; Amorphous carbon; ZrC/a-C(:H); Biotribology; DLC

#### PAPERS INCLUDED IN THE THESIS

The thesis is based on 7 published/submitted papers listed below and which are referred to in the text by Roman numerals. Some results which have neither been published nor exist in manuscript are also included. Reprints of the papers can be found in the annexes.

I. DLC(H) Doped with Zr Coatings for Orthopaedics Applications: Adhesion Properties

<u>A. Escudeiro</u>, T. Polcar and A. Cavaleiro 55<sup>th</sup> Annual Technical Conference Proceedings, ed: Society of Vaccum Coaters, **2012** 

- II. a-C(:H) and a-C(:H)\_Zr coatings deposited on biomedical Tibased substrates: Tribological properties <u>A. Escudeiro</u>, T. Polcar and A. Cavaleiro Thin Solid Films 538 (2013) 89-96
- III. Indentation and scratch testing of DLC-Zr coatings on ultrafinegrained titanium processed by high-pressure torsion C.T. Wang, <u>A. Escudeiro</u>, T. Polcar, A. Cavaleiro, R.J.K. Wood, N. Gao and T.G. Langdon Wear, 306, 1-2 (2013) 304-310
- IV. Adsorption of bovine serum albumin on Zr co-sputtered a-C(:H) films: implication on wear behavior <u>A. Escudeiro</u>, T. Polcar and A. Cavaleiro Journal of Mechanical Behavior of Biomedical Materials 39 (2014) 316-327
- V. Sliding Properties of Zr-DLC Coatings: The effect of Tribolayer Formation
   T. Vitù, A. Escudeiro, T. Polcar and A. Cavaleiro

Surface and Coating Technology 258 (2014) 734-745

- VI. Tribological Behavior of uncoated and DLC-coated CoCr and Ti-alloy in contact with UHMWPE and PEEK counterbodies <u>A. Escudeiro</u>, M.A. Wimmer, T. Polcar and A. Cavaleiro Tribology International, under review
- VII. Structural and Mechanical properties of nanocrystalline Zr cosputtered a-C(:H) amorphous films <u>A. Escudeiro</u>, N.N. Figueiredo, T. Polcar and A. Cavaleiro Applied Surface Science 325 (2015) 64-72

#### Other publications to which the author has contributed:

#### Influence of Ag on mechanical and Tribological behavior of DLC coatings

N.K. Manninen, F. Ribeiro, <u>A. Escudeiro</u>, T. Polcar, S. Carvalho and A. Cavaleiro Surface and Coatings Technology 232 (2013) 290-300

## Tribological behaviour a-C and a-C:H films doped with Ti in biological solutions

<u>A. Escudeiro</u>, T. Polcar, and A. Cavaleiro Vacuum 85 **(2011)** 12: 1144-1148

# Carbon-based coatings doped by copper: Tribological and mechanical behavior in olive oil lubrication

R.M. Balestra, A.M.G. Castro, M. Evaristo, <u>A. Escudeiro</u>, P. Mutafov, T. Polcar and A. Cavaleiro

Surface and Coatings Technology 205 (2011) 85: 79-83

### LIST OF SYMBOLS AND ABREVIATIONS

θ	Position of the diffraction peak	[°]
$\theta_{water}$	Water contact angle	[°]
λ	Wavelength	[Å] or [nm]
v	Poisson ratio	
$\sigma_{RS}$	Residual stress	[GPa]
Ø	Diameter	[mm]
ASTM	American Society for Testing and Materials	
BF	Bright Field	
BM20	Bending magnet 20	
BSA	Bovine Serum Albumin	
СС	Crevice Corrosion	
CEMUC	Centro de Engenharia Mecânica da Universidade de Coimbra	
CEMUP	Centro de Engenharia Mecânica de Universidade do Porto	
CFR	Carbon-Fiber Reinforced	
CoC	Ceramic on Ceramic	
CoCrMo	Cobalt-chrome alloy	
CoP	Ceramic on Polymer	
COF	Coefficient of Friction	
СР	Commercial Pure	
CVD	Chemical Vapour Deposition	
DC	Direct Current	
DLC	Diamond-Like Carbon	
d <sub>s-t</sub>	Substrate-to-target distance	[nm]
E	Young Modulus	[GPa]
EBS	Non- Rutherford Backscattering Spectrometry	
EDS	Electron Dispersive X-ray Spectrometry	
EELS	Electron Energy Loss Spectroscopy	

EN	European Norm	
EPMA	Electron Probe Micro Analysis	
ERD	Elastic Recoil Detection	
ESRF	European Synchrotron Radiation Facility	
FAT	Fixed Analyser Transmission	
FEG	Field Emission Gun	
FIB	Focus Ion Beam	
FTT	Fourier Fast Transform	
FWHM	Full Width at Half Maximum	
GIF	Gatan Image Filter	
Н	Hardness	[GPa]
HPT	High Pressure Torsion	
HR	High resolution	
IBA	Total Ion Beam Analysis	
ICDD	International Centre of Diffraction Data	
Lc, Lc1, Lc2, Lc3	Critical Load	
Mc	Million cycles	
MoM	Metal on metal	
MoP	Metal on polymer	
NA	Not Analysed	
nc	Nanocrystalline	
NCD	Nano crystalline diamond	
Р	p value	
PE	Polyethylene	
PEEK	Polyetheretherketone	
PIXE	Particle-induced X-ray emission	
PS	Physiological Solution	
PTFE	Polytetrafluoroethylene	
PVD	Physical Vapour Deposition	
РуС	Pyrolytic carbon	

R <sub>a</sub>	Average surface roughness	[nm]
RBS	Rutherford Backscattering Spectrometry	
R-C	Rockwell-C indentation	
RH	Relative Humidity	
Rms	Roughness mean square	[nm]
ROBL-MRH	Material Research station of the Rossendorf Beam Line	
rpm	Rotation per min	
SAED	Selected Area Electron Diffraction	
SEM	Scanning Electron Microscopy	
SCC	Stress-corrosion cracking	
SNK	Student-Newman-Keuls	
SS	Stainless Steel	
TEM	Transmission Electron Microscopy	
THA	Total hip arthroplasty	
Ti6Al4V	Titanium alloy	
TJA	Total joint arthroplasty	
TJR	Total joint replacement	
ТКА	Total knee arthroplasty	
UFG	Ultrafine-grained	
V <sub>F</sub>	Volume Fraction	
UHMWPE	Ultra-high molecular weight polyethylene	
XPS	X-ray Photoelectron Spectroscopy	
XRD	X-ray Diffraction	
YTZ	Ytria stabilized zirconia	
ZTA	Zirconia tounghened alumina	

## TABLE OF CONTENTS

AGRADECIMENTOS/ACKNOLEGMENTS	vii
RESUMO	ix
ABSTRACT	xi
PAPERS INCLUDED IN THE THESIS	xiii
LIST OF SYMBOLS AND ABREVIATIONS	xv
TABLE OF CONTENTS	xix
LIST OF FIGURES	xxi
LIST OF TABELS:	xxv
	I
I. Hypothesis	2
2. Aims	3
3. Thesis structure	3
CHAPTER I	5
I. Joint Prosthesis	7
I.I. The natural synovial joint and its failure	7
I.2 A survey of artificial joints	8
1.3 Current and alternative materials for joint prosthesis	10
2. Diamond-like carbon (DLC)	16
3. The Main Goal of the thesis	20
CHAPTER II	23
I. Introduction	25
2. Coatings deposition and characterization	25
2.1 Deposition Process	25
2.2 Coating characterization methods	26
3. Coating optimization	
3.1 Adhesion and interlayer optimization	
3.2 Chemical composition	

3.3 Deposition rate	
4. Conclusions	40
CHAPTER III	41
I. Introduction	43
2. Characterization of DLC nanocomposite films	43
2.1 Coating characterization	43
2.2 Structure and chemical bonding	46
2.3 Mechanical properties of the coatings	50
2.4 Tribological properties	53
3. Conclusions	57
CHAPTER IV	59
I. Introduction	61
2. Surface Chemistry – Physicochemical Properties	61
3. Tribological properties under BSA	66
4. Conclusions	68
4. Conclusions CHAPTER V	
	71
CHAPTER V	71 73
CHAPTER V	71 73 74
CHAPTER V I. Introduction 2. Tribological behaviour against UHMWPE and PEEK counterbodies	71 73 74 81
CHAPTER V I. Introduction 2. Tribological behaviour against UHMWPE and PEEK counterbodies 3. Conclusions	71 73 74 81 83
CHAPTER V 1. Introduction 2. Tribological behaviour against UHMWPE and PEEK counterbodies 3. Conclusions OUTPUTS AND FUTURE RESEARCH	71 73 74 81 83 85
CHAPTER V I. Introduction 2. Tribological behaviour against UHMWPE and PEEK counterbodies 3. Conclusions OUTPUTS AND FUTURE RESEARCH REFERENCES	71 73 74 81 83 85 103
CHAPTER V 1. Introduction 2. Tribological behaviour against UHMWPE and PEEK counterbodies 3. Conclusions OUTPUTS AND FUTURE RESEARCH REFERENCES PAPER 1	71 73 74 81 83 85 
CHAPTER V I. Introduction 2. Tribological behaviour against UHMWPE and PEEK counterbodies 3. Conclusions OUTPUTS AND FUTURE RESEARCH REFERENCES PAPER I PAPER II	71 73 74 81 83 85 103 111 123
CHAPTER V  I. Introduction  2. Tribological behaviour against UHMWPE and PEEK counterbodies  3. Conclusions  OUTPUTS AND FUTURE RESEARCH  REFERENCES PAPER I PAPER II PAPER III	71 73 74 81 83 85 103 111 123 132
CHAPTER V I. Introduction 2. Tribological behaviour against UHMWPE and PEEK counterbodies 3. Conclusions OUTPUTS AND FUTURE RESEARCH REFERENCES PAPER I PAPER II PAPER III PAPER IV	71 73 74 81 83 85 103 111 123 132 147

#### **LIST OF FIGURES**

#### Figure I.I Representation of synovial joint (Adapted from [2])......7

Figure	II. I	Picture	and	schematic	representation	of	the	deposition	chamber	and
depositio	on pa	rameter	s		••••••••••••				••••••	26

Figure II.2 Scratch test at increasing load on a-C coating deposited without interlayer.

Figure II.8	Zr	content	as a	a function	of	relative	erosion	area	$(A_C/A_{Zr})$ :	(a)	non-
hydrogenate	d and	d (b) hyd	roge	nated coat	ings	. The ins	ets are t	he co	rresponde	nt n	ames
that will be u	sed	during th	is wo	rk	•••••	•••••	••••••	•••••		•••••	38

<b>Figure III.1</b> SEM micrographs of fractured cross-section of non-hydrogenated (a-C and a-C_Zr(7)) and hydrogenated coatings (a-C:H and a-C:H_Zr(8))44
Figure III.2 AFM micrographs showing the topography of (a,b) a-C, (c,d) a-C_Zr(13),
(e,f) a-C:H and (g,h) a-C:H_Zr(8) films deposited on Si wafers. The 3D in the right
column corresponds to the 2D in the left column
<b>Figure III.3</b> Grazing-angle X-Ray diffractograms of (a) non-hydrogenated and (b) hydrogenated series of coatings with different Zr content
<b>Figure III.4</b> Grazing-angle X-ray diffractogram of the coating a-C_Zr(14). The
measured data are displayed as black line and the fitted curves as red line
Figure III.5 XPS C1s spectra from the non-hydrogenated series of coatings with different Zr content
Figure III.6 (a) TEM BF images of the a-C_Zr(14) film and (b) FFT diffraction fringes
with phase identification
Figure III.7 $I_D/I_G$ ratio and G peak position as a function of Zr content
Figure III.8 Critical load of non-hydrogenated and hydrogenated coatings deposited on
steel substrates
Figure III.9 Tribological properties of non-hydrogenated and hydrogenated coatings
deposited on Ti grade 2 and 5, together with the uncoated values in ambient air. (NA
stands for not analysed)
Figure III.10 Tribological properties of non-hydrogenated and hydrogenated coatings
deposited on Ti grade 2 and 5, together with the uncoated values in PS
Figure III.11 SEM micrographic pictures from a-C_Zr(9) wear track when deposited
on Ti grade 2: (a) ambient air and (b) PS56
Figure III.12 Raman spectra and micrographs of the ball wear scar that slid against a-C
and a-C_Zr(9) films. The arrows show the acquisition zone
Figure IV.1 BSA adsorption vs contact angle measurements. The dash line represents
the Berg's limit <sup>1</sup> ( $\theta = 65^{\circ}$ )
Figure IV.2 XPS spectra of the uncoated and coatings after immersion in BSA for 24h:
(a) Ti6Al4V, (b) a-C, (c) a-C_Zr(4), (d) a-C_Zr(7) and (e) a-C_Zr(14). a-C_Zr(14)
before immersion was also add for comparison (f). The inset (*) shows a zoom in of the
200-400 eV region of the a-C_Zr(4) film XPS spectra revealing the S 2p photo-peak65
Figure IV.3 N Is core level spectra for coated and uncoated samples after immersion

<b>Figure IV.4</b> Average friction coefficient and ball wear rate of the coated and Ti6Al4V uncoated samples under BSA lubrication
<b>Figure IV.5</b> Schematic representation of the key mechanism of albumin-mediated lubrication on hydrophilic and hydrophobic DLC films
<b>Figure V.I</b> Average COF for all samples against UHMWPE and PEEK. (*) represents statistically significant differences between samples in the same group (p < 0.05, ANOVA test)
<b>Figure V.2</b> (a) UHMWPE and (b) PEEK volumetric wear loss and wear rate (in the insets) against the different surfaces
<b>Figure V.3</b> Images from the wear track of the disc after testing against (a,c,e,g) UHMWPE (a,c,e,g) and (b,d,f,h) PEEK
<b>Figure V.4</b> Optical micrographs of the Rockwell-C indentation of a-C and a-C_Zr(4) films before and after testing against UHMWPE and PEEK
<b>Figure V.5</b> Raman spectra for (a) a-C and (b) a-C_Zr(4) acquired before and after sliding against UHMWPE under BS. Since a-C_Zr(4) delaminate after 1.2 Mc cycles the acquisition was done near the delaminations zones. The $I_D/I_G$ ratio is also displayed in the insets
Figure V.6 Optical micrograph top-view of the Rockwell-C indentation test of the a-C

## LIST OF TABELS:

Table I.I Types of TJR (Adapted from [21])         9
Table I.2 Biomaterials for TJR and its properties and applications (Adapted from [21])
Table III.1 Chemical composition and density measured by IBA of the coatings
Table III.2 Mechanical properties of the coatings
Table III.3 Adhesive properties of the coatings on Ti6Al4V and CP-Ti
<b>Table IV.1</b> Surface roughness ( $R_a$ and $Rms$ ), water contact angle ( $\theta$ ) and surface energy ( $\gamma_s$ ) of the coatings
Table V.I Defects density on the coatings

# INTRODUCTION

Joint prosthesis is an artificial joint implanted into the human body to replace the natural articulating system, which has lost its functionality. The total joint arthroplasties decreasing pain and improving locomotion have gained increasing importance in orthopaedic surgery. The usage of joint prostheses entered the clinical practice mostly in the 70s; however, the design of endoprostheses dates back to the 14<sup>th</sup> century.

The materials used for artificial implants should fulfil several criteria, such as good mechanical properties, structural stability, corrosion resistance and suitable interaction with human cell environment. A wide range of materials is used as a bearing surface, from ceramics (e.g. alumina and zirconia) through metals (e.g. stainless steel, CoCr alloys and titanium and its alloys) to polymers (e.g. UHMVVPE and PEEK). However, it is estimated that the average survival rate of an implant is of 10 to 15 years. It is particularly limiting for young patients with expected intense physical activity. Revision surgeries are complicated and expensive with outcomes that are often not as successful as that of the original joint replacement.

Aseptic loosening has been identified as the major problem associated with implant failure. Demanding contact conditions coupled with the aggressive body environment causes the generation of the wear debris which, by acute host-tissue reactions, tend to aggravate and speed up the failure of the biomaterial. Wear is thus a critical factor for prostheses, implants and other medical devices and its reduction is still an ongoing scientific and technological challenge. It is evident that the reduction of wear and subsequent extension of the life span of a joint prosthesis would improve quality of life and have positive economic impact.

Numerous attempts have been done in the last years to improve the life time of the prosthesis ranging from changing the overall bulk material, surface modification of the bulk material or the use of coatings. Nevertheless, there is still no ideal solution that effectively prevents the failure of implants. Surface modification has been considered one of the most promising approaches to improve the mechanical and tribological properties together with biocompatibility of the prostheses. Therefore, hard coatings as diamondlike carbon (DLC), transition metal nitrides and carbides, and surface modification of metals have been have been widely studied for improving the wear and corrosion resistance. Unfortunately, the major problem associated with coatings is the poor adhesion to the substrate due to the high residual stress and poor chemical bonding between the film and the substrate. Consequently, the coatings are prone to delaminate which results in rapid failure of the coated implant. The presence of body fluids is an extremely important factor to be considered. The natural lubrication minimizes the shearing damage and decreases the friction energy loss. However, in artificial joint, the lubrication effect depends on many factors, such as surface treatment or coating properties, surface roughness, mechanical properties of the implant and many more, which might strongly limit the efficiency of natural lubricants and lead to catastrophic failure of the implant with partial or complete loss of functionality.

This thesis contributes to this area by the study of diamond-like carbon alloyed with zirconium as a potential coating for orthopaedic materials.

#### I. Hypothesis

The production of wear particles from the replacements of metal on metal (MoM) and metal on polymer (MoP) joints replacement and their detrimental effects on body have been a continual problem since they were first implanted. Surface modifications are believed to have the potential to reduce the wear of joint prosthesis. Many studies have been carried out focused mainly on the deposition of DLC; however, there is still limited success today due to adhesion problems of the coatings.

Nanocomposite films are known to show extremely low wear and good adhesion and have been successfully used in industry such as machine tooling. The deposition of nanocomposite structure is based on atoms with different chemical affinity that can form immiscible phases providing huge improvements in both mechanical and tribological properties. When incorporating small amounts of a metallic element in an amorphous carbon matrix a nanocomposite structure is likely to be achieved. In this case, a decrease of the residual stresses are expected with the consequent improvement of the adhesion of the coatings. Experimental and clinical studies support the general consensus that Zr compounds are biocompatible and exhibit low toxicity. Thus, amorphous carbon alloyed with Zr should fulfil the implant tribological and adhesive criteria without losing its biocompatiblility.

#### 2. Aims

This thesis seeks to contribute to wide efforts to solve the problems of inadequate artificial joint from a tribological perspective. The overall objective of this thesis is the development of nanocomposite coatings to improve the DLC performance under conditions similar to that of biomedical implants. The key idea was to combine the excellent mechanical and tribological properties of pure amorphous carbon (a-C) coatings with the specific characteristics of nanocomposite coatings. The aims of this thesis were could be then given as follows:

- To develop a-C(:H) coatings alloyed with with Zr (with different metal content and morphologies) in order to achieve a nanocomposite structured material by magnetron sputtering, using reactive (Ar+CH<sub>4</sub>) and non-reactive (Ar) atmospheres.
- To thoroughly understand the deposition process and the characterization of the chemical composition and structure of the coating as a function of the depositions conditions;
- To optimize the adhesion of the coatings onto Ti-based substrates by developing a functional graded interlayer.
- To study the functional properties of the coatings in relation to their physical and chemical properties as well as their mechanical and tribological properties.
- To evaluate the interaction between the films and simulated body fluids and to characterize the influence of exposure conditions on the surface state and physical properties.
- To select the best coating candidates in relation to their tribological properties (wear and friction) and long term stability in simulating body fluid environments.

#### 3. Thesis structure

Apart from this introductory chapter, the thesis consists of five main chapters. Chapter I presents the state of the art in the area of biomedical implants focused on protective coatings. It outlines the basis of knee anatomy and its failure, followed by the replacements that are available, the shortcoming of current replacements and the use of coatings as a method to improve their properties. Chapter II describes the deposition process and the parameters used for the coatings deposition. Special attention to the optimization of the coatings was paid, namely coatings adhesion, chemical composition and deposition rate. Chapter III is dedicated to fundamental characterization of the ZrC/a-C coatings including: chemical composition and bonding, morphology, microstructure, mechanical properties (hardness, residual stress, Young modulus and critical loads) and tribological properties in ambient air and in physiological solution. Chapter IV is focused on the physicochemical properties of the coatings by means of surface energy and protein adsorption assessment. The tribological properties of the coatings under simulated body fluid environments were also evaluated. Chapter V describes the evaluation of selected coatings sliding against ultra-high molecular weight polyethylene (UHMVVPE) and polyetheretherketone (PEEK) polymeric materials in a multidirectional pin-on-disc, which replicates the typical movement in joint prosthesis. Finally, the thesis concludes with a summary of the main outcomes and future research topics that should be taken into consideration when using coatings for biomedical implants.

# CHAPTER I

The gross anatomy of synovial joints is first described with emphasis on present failures and possible solutions. The history and development of joint prostheses is briefly referred in order to identify the deficiencies and the typical failure modes. The materials choice is probably the most important feature. A summary of the most common and alternative materials (bulk and surface engineered materials) used nowadays is presented. Moreover, a comprehensive overview on diamond-like carbon coatings, promising surface treatment of joint replacements, will be exposed as a possible way for improving their life span. Finally, the goal of this thesis is presented.

### I. Joint Prosthesis

#### I.I. The natural synovial joint and its failure

An articulation, or joint, is a place where two different bones come together. They are usually associated with movement although that is not always the case. Many joints allow only limited movement and others even allow no apparent movement [1]. The structure of a given joint is directly correlated with its degree of motion. This study will focus on the mobile synovial joints. Figure I.1 shows the structure of a typical synovial joint.

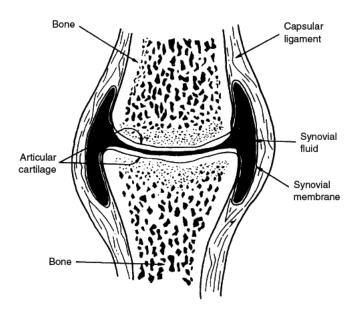


Figure I.I Representation of synovial joint (Adapted from [2]).

The synovial joint is a fluid containing joint (synovial fluid) that may look like a simple joint but it is anatomically one of the most complexes. Most of the joints that unite bones of the appendicular skeleton are synovial: hip, knee, ankle, shoulder, elbow and finger; reflecting their far greater mobility [1]. All synovial joints are covered with a thin layer of hyaline cartilage called *articular cartilage*, which promotes the sliding between bones and presents particular mechanical properties: resistance to compression (proteoglycans and water) and to shear stress (collagen) [2]. The articular surfaces are enclosed in a synovial joint cavity, surrounded by a joint capsule containing synovial fluid. This fluid is derived from the blood plasma but contains higher protein

content, hyaluronic acid, glycoproteins, proteinases, and collengases. Therefore, this fluid has non-Newtonian characteristics (viscosity and viscoelastic effects that are shear rate depending) providing a slippery weight-bearing film that reduces friction between the cartilage (typical healthy human joint coefficient of friction (COF) 0.005-0.04) [3, 4]. Although the main function of the synovial fluid is to act as a lubricant during the movement, it also gives nutrients to the cartilage by diffusion, it absorbs shocks and it rids the joint cavity of microbes and cellular debris, due to the presence of phagocytic cells [3-5]. For more information on biochemistry, structure and properties of articular synovial joint references [6-9] are suggested as further reading.

The joint function can be easily affected due to accident, heredity or disease (such as *arthritis*). All the structures composing the synovial joint are liable to assume a diseased action both individually and collectively, including deformity, instability and loss of motion, which is commonly associated with severe pain and functional disability [9]. Articular cartilage is the most affected one. Its function to absorb shocks by the elasticity can be lost, if the balance of growth factors and enzymes is changed. The treatments vary between pharmacological and, in worse cases, chirurgical replacement of total or part of the joint by a prosthetic device (*arthroplasty*) [10, 11]. Indeed, the number of total joint arthroplasties (TJAs) is increasing relatively fast. For instance, a total of approximately 1,324,000 primary and revision knee procedures are carried out annually worldwide (data from 18 countries, based on international arthroplasty registers reports) [12] and it is expected that the number of TJA in 2020 will rise by 98% [13, 14].

#### 1.2 A survey of artificial joints

The main reasons that lead to the implant of a joint prosthesis are unsustainable pain and difficulties in movements [15]. Joint implants are designed to replace biological materials that have been damaged (cartilage and/or bone) by articulating joint surfaces and synthetic materials.

The first total knee and hip arthroplasties (TKA and THA, respectively) were performed in 1890 by T. Gluck using cemented ivory. The few implants made had to be removed after a few months, mainly because of inadequate sterilization techniques [16]. Four years later, JE Pean substituted the proximal shoulder with platinum endoprosthesis [17]. During the World War II, materials originally manufactured for airplanes, automobiles, clocks and radios were taken "off-the-shelf" by surgeons and applied in the medical field (glass, titanium, stainless steel, metacrylates, Teflon<sup>®</sup>, nylon, polyurethanes, etc) but the real advance came in the 1950s with the principle of lowfriction arthroplasty, initially developed for the hip by Charnley [18,19]. After numerous material combinations, in 1962, Charnley successfully implanted ultra-high molecular weight polyethylene (UHMWPE) articulated against a metallic femoral head; it was used by over a million of patients and it significantly improved their mobility [18-20]. His developments remain the gold standard bearing surface combination and basic tribological design for total joint replacements (TJRs). Table I.I shows the main articular prosthesis and its components used nowadays. There have been several innovations regarding strength, shape, function and behaviour of the stem which holds the prosthesis into the bone, but surfaces have changed very little indeed since Charnley first replaced polytetrafluoroethylene (PTFE) by UHMWPE.

Joint	Туре
Нір	Ball and socket
Knee	Hinged, semiconstrained, surface resurfacing,
	unicompartment or bicompartment
Shoulder	Ball and socket
Ankle	Surface replacement
Elbow	Hinged, semiconstrained, surface resurfacing
Wrist	Ball and socket, space filler
Finger	Hinged, space filler

Table I.I	Types of	TJR	(Adapted	from	[21]).
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TJRs have been successfully implanted for over 30 years [20]. The success of the replacement is strongly related to the overall medical condition of the patient, the behaviour of the different materials used to produce the implants and the surgeon ability and experience [22]. Moreover, with a proper patient selection in terms of sex, age, bodyweight, profession and activity, good to excellent results can be achieved: 80% survivorship at 25 years for Charnley-type hip prosthesis [23] and 94% at 15 years for knee replacements [24]. However, the average survival rate of the implants is of 10-15 years [25, 26]. Unfortunately, with the increasing demand of prosthetic devices by younger people (< 65 years) this is not satisfactory. In fact, it is estimated that until 2030 these patients will represent the majority of the primary and revision TJR [27] due to more active lifestyles. This may also be synonymous of a growth in revision surgeries which are more costly, technically more difficult and also more complicated than primary arthroplasty, suggesting that this market will increase in the next few years. It is thus not surprising that the global market for joint prosthesis is worth US\$56 billion by 2017, spurred by a multitude of factors including a aging global population, the rising incidence of degenerative joint diseases, and improving healthcare infrastructures in

developing countries [28]. Besides, the development of durable and improved materials and implants is also expected to contribute to the market's growth.

#### I.2.1. Artificial joint failure and the role of wear

"All prostheses will fail sometime. It is a race between the life of the patient and the life of the prosthesis" [29]. Between 2000 and 2012 the top seven reasons for revision surgeries were, in order of proportion: aseptic loosening, infection, polyethylene (PE) wear, instability, pain/stiffness, osteolysis and malposition/malalignment [30, 31]. Failure due to instability and malposition is beyond the scope of this thesis but it lies on better surgeons training and the use of computer guided surgery [22]. On the other hand, implant loosening is mainly associated with biological potential of wear debris and consequently intimately related to the materials used [26]. Considering the nature of joint prosthesis, wear debris can be generated at any interface: at the articular surfaces, at the cement interfaces, with the bone and the implant, and at the junctions between any modular features in the main implant. Loosening and wear are then strongly related to each other by making the prosthetic failure a tribological issue.

Along with wear, friction is also another important issue. Excessive friction causes high torque loads that can be transmitted to the bone-implant interface and result in pain and instability [32]. It is therefore important to improve lubrication mechanisms and the use of low friction material combination in order to reduce friction. Even though wear will always be an intrinsic issue for biomaterials, it can be reduced having by taking the tribological aspects of the joint into consideration and proving materials that suit the natural synovial lubrication.

The presence of body fluids is also very important in the material degradation. Phagocytosis of wear debris by macrophages induces an inflammatory reaction associated with the release of cytokines and other inflammatory mediators. Once osteolysis has occurred, it tends to progress and may ultimately lead to implant failure [18, 33].

#### 1.3 Current and alternative materials for joint prosthesis

#### I.3.I State of the art

The implant design should fulfil the kinematic and dynamic load transfer characteristics of the natural joint [21] and should be patient- and pathology-related.

This section will only focus on material variations, especially materials used as bearing surfaces.

The implant components require desirable properties such as strength, high wear resistance, low Young modulus (close to the bone or cartilage), high ductility, high fracture toughness and high corrosion resistance [34]. Biocompatiblility is a vital parameter that must also be taken into account since the material should be implanted in the human body. In other words, as a foreign-body, the degree to which the homeostatic mechanisms is perturbed and the extent to which pathophysiologic conditions are created and undergo resolution is also an issue [18]. Decades of basic and clinical experimentation have resulted in a vast number of material combinations ranging from every class of materials. Generally though, there are only a few materials that have an optimal mix of those properties. The current standards include materials from all classes of materials: titanium and its alloys, cobalt-chrome alloys (CoCrMo), UHMWPE with or without cross-linking, alumina  $(Al_2O_3)$  or zirconia  $(ZrO_2)$  stabilized with ytria or toughened with alumina (YSZ and ZTA, respectively). These materials can be then combined by meanings of hard/soft (metal on polymer, MoP, and ceramic on polymer, CoP) and hard/hard (metal on metal, MoM, and ceramic on ceramic, CoC) combinations.

Table I.2 lists the relevant combination of materials based on the market leaders: DePuy, Zimmer, Biomet<sup>®</sup>, Smith & Nephew and Stryker<sup>®</sup> Corporation [25, 28].

Materials	Properties	Application	
CoCrMo alloy	Heavy, hard, stiff	Stem, head (ball)	
(cast or wrought)	High wear resistance	Cup, porous coating, Metal backing	
Ti alloys (Ti6Al6V)	Low stiffness	Stem, porous coating	
	Fatigue Resistance	Metal backing	
	Low wear resistance		
	Corrosion resistance		
Pure Ti	Excellent osseointegration	Porous coating	
Tantalum	Excellent osseointegration	Porous structure	
(Trabecular Metal)	Good mechanical strength		
Alumina (Al <sub>2</sub> O <sub>3</sub> )	Hard, Brittle	Ball, Cup	
	High wear resistance		
Zirconia	Heavy and high toughness	Ball	
(YSZ or ZTA)	High wear resistance		
UHMWPE (with or	Low friction, wear debris	Сир	
without crosslinking)	Low creep resistance		

Table I.2 Biomaterials for TJR and its properties and applications (Adapted from [21]).

It is worth noting that the way materials are processed, such as precision casting or forging (metal alloys), machining, grinding, grit blasting or other surfacing techniques, can also widely influence the properties of the prosthetic materials [35]. References [3, 9, 36-40] provide a deeper understanding of the fundamental properties of the materials and the influence of material processing for ensuring the desired performance of the implant devices. Besides extended research and testing, finding the optimal materials is a continuous effort of success and failures in search of perfection. Tribo-corrosive processes of metallic materials can increase the number of metallic ions in local tissues and systemically which may cause dramatic necrotic and inflammatory reactions in the surrounding tissues [33, 41]. Ceramics can be considered as the most preferred materials for joint endoprosthesis due to negligible body reaction to wear [26, 42, 43]. However, such materials can undergo brittle fracture [43, 44] and squeaking [45, 46], which can cause severe and complex problems. On the other hand, polymeric materials are soft and their wear resistance is strongly limited [38]. Searching for materials for a longer lasting implant is still an ongoing bioengineering challenge.

#### 1.3.2 New promising bulk materials

The most obvious option to change the tribological properties is to replace the overall bulk material with a new one that generates less wear with adverse biological interactions. Using this approach, the new material has to ensure lubrication under physiological condition together with appropriate mechanical strength, corrosion resistance and biocompatibility.

#### Non-oxide ceramics

Silicon nitride  $(Si_3N_4)$  has been considered as a reliable material for TJA due to higher strength and fracture toughness compared to alumina and zirconia and its excellent wear characteristics [42, 47, 48]. Long-term *in vivo* exposure did not reveal any adverse biological reaction making it a biocompatible and non-toxic material [49, 50]. However, some controversial results have been reported suggesting the presence of a nanometric oxidized layer (SiO<sub>2</sub>) that significantly improves third-body wear [51, 52]. Nevertheless, the first silicon nitride total hip joint was implanted in 2011 [53].

#### Sapphire

The purest aluminium oxide form with no porosity or grain boundaries has shown very low and stable friction coefficient (0.05-0.10) as well as an extraordinary high wear capacity [54]. Biocompatibility, biochemical inertness and low cost make the sapphire friction pair very attractive for endoprotheses. In fact, 5 sapphire heads had already been implanted with no complication reported within 5 year [53].

#### Vitamin E stabilized UHMWPE

Vitamin E is an antioxidant agent that improves the oxidation resistance of irradiated UHMWPE. When incorporated in the UHMWPE, it helps to eliminate the free radicals within the polymer which are partially responsible for the delamination type of wear, the primary cause of failure of UHMWPE [38, 55]. Moreover, it has been also shown that Vitamin E-doped polymers present superior crack and wear resistance and maintain its fatigue strength [56]. However, long-term studies to establish the effectiveness of vitamin E as an effective anti-oxidant *in vivo* are required.

#### Polyetheretherketone (PEEK) and carbon-fiber reinforced (CFR-PEEK) form

PEEK and its composites have been used for UHMWPE replacement due to their biocompatibility, high strength, stiffness, fatigue and wear resistance [57-59]. Moreover, the wear performance is strongly related to the presence or absence of CFR [59] and, when reinforced, the type [60], amount and orientation of incorporated carbon-fibers [61]. Moreover, PEEK wear particles were found to be less cytotoxic than UHMWPE [62]. In fact, in 2001, 30 alumina femoral heads against CFR-PEEK liners were implanted and 28 months after implantation only one liner had to be revised due to trauma [63]. The clinical study was expanded to a total of 121 patients by 2003. However, no published data is available yet. On the other hand, a recent study has found that two CFR-PEEK varieties increase cytokine expression that may suggest a proinflamatory potential *in vivo* [64].

### **1.3.3 Surface modification and coatings**

Implant surface is the most important issue in the modern TJR due to direct interaction with the body fluids. Wear and corrosion are then likely to occur on bearing surfaces rather than in the bulk material. Therefore, the use of surface modification can retain the desired mechanical properties of bulk material while modifying only the properties of the outmost surface. Furthermore, ceramic coating can improve wear and corrosion resistance and minimize metal ion release. The use of coatings can also help hard tissue compatibility by promoting bone growth. Goodman [65] has published an excellent review on biologic coatings for orthopaedic implants that facilitate osseointegration and mitigate possible adverse tissue responses including the foreign body reaction and implant infections. Here we list the most studied variations of thin films and coating prepared through different methods.

### Carbon-based coatings

Carbon-based coatings are one of the most used for a wide range of applications from the automotive industry to medical devices. From the biomedical point of view, the most common forms are:

- (i) Diamond-like carbon (DLC), which exhibits a peculiar amorphous structure that gives them properties between diamond and graphite. They are generally characterized by high hardness, low friction, high excellent environmental wear resistance, an behaviour and biocompatibility [66-69]. Unfortunately, retrieved femoral heads had shown numerous pits and spots of local coating delamination and crevice corrosion [70]. Thus, adhesion problems are still an ongoing issue due to an inadequate bonding at the interface between metallic substrate and coating and high residual stress [71]. Recently, research has been done to improve the wear properties and the adhesion strength of DLC coatings; it will be discussed more in detail in the next session.
- (ii) <u>Nano crystalline diamond (NCD)</u> is hard, smooth, corrosion and wear resistant, which makes it suitable to TJR application [72]. Despite the excellent results from *in vitro* testing in a mandibular simulator [73], the excessive residual stresses due to impurities at the grain boundaries affects their adhesion and may cause delamination [74]. No clinical studies are available so far.

(iii) <u>Pyrolytic carbon (PyC)</u> is a material with a good fatigue resistance, wettability and wear properties as well as a modulus of elasticity very close to the cortical bone [75, 76]. It has been applied to small joints, such as proximal interphalangeal [77] and metacarpophalangeal [78] joints with a survival rate of 81.4% within 10 year [79]. However, the use of PyC for load-bearing application still needs further *in vitro* consideration.

#### Surface modifications of metals

OXINIUM<sup>TM</sup> (Smith and Nephew, Memphis, TN) is the commercial name for surface oxidized zirconium. Technically this ceramic surface layer is not considered a coating but an intrinsically formed oxidation layer on the bulk metal surface 5 to 10  $\mu$ m thick [55]. The main principle is to oxidize a metallic zirconium alloy by means of thermal diffusion to create  $ZrO_2$  on the surface [55, 80]. This oxide layer offers good wear properties and adhesion to the metallic substrate avoiding the metal ion release and associated hypersensitivity [55]. However, retrieved femoral Oxinium<sup>TM</sup> heads have demonstrated loss of the Oxinium<sup>TM</sup> layer with exposure of the underlying substrate; moreover, an extensive damage to the polyethylene counterbody was also observed [67, 81].

#### Nitrides and Carbides

Metal nitrides, carbides and carbonitrides (TiN, WC, TiCN, TiAIN, etc.) has been considered the most appropriate as hard and wear resistant coatings [82]. Brilliant golden colour TiN coatings have been also used in several biomedical applications [53, 83]. Later, a variety of nitride coatings was prepared using chromium [84], silicon [85, 86] or zirconium [87] instead of titanium. Moreover, silver incorporation into such matrices as an antibacterial agent has also been subject of recent studies [88-90]. The mechanical and corrosion resistance can be improved or maintained allowing a limited Ag release to the environment to confer the aimed antibacterial properties. TiN coatings have been used in hip, knee, ankle and shoulder replacements and continue to be the main coating solution available on the market. However, Raimondi and Pietrabissa [91] retrieved four implants at two to eight years of follow-up and argued that TiN-coated titanium alloy femoral heads were inadequate in the task of resisting *in vivo* third-body wear. Moreover, a recent study [92] has shown a case of *in vivo* 

fracture of a TiN-coated femoral component after 3 year implantation. Debonding of the component may be the cause for stress fracture. The extreme hardness of the coatings and their adhesion resistance on the softer metal is apparently the major limitation of their use.

# 2. Diamond-like carbon (DLC)

DLC is an amorphous carbon (a-C) or hydrogenated amorphous carbon (a-C:H) thin film material containing both sp<sup>3</sup>- and sp<sup>2</sup>-hybridized carbon atoms, which give it structure and intermediate properties between diamond and graphite [68]. The abbreviation DLC can thus stand not for a specific coating but for a class of hard Cbased material [66]. DLC  $sp^2/sp^3$  ratio and hydrogen content are strongly influenced by the deposition technique; DLC coatings can be prepared by either chemical vapour deposition (CVD - only hydrogenated coatings) or physical vapour deposition (PVD both hydrogenated and non-hydrogenated films) technologies [66, 68]. The sp<sup>2</sup>/sp<sup>3</sup> ratio can be relatively easily varied and alter the whole properties of the coating. For instance, sp<sup>2</sup>-rich coatings commonly show low coefficient of friction in dry condition but a higher wear rate than sp<sup>3</sup>-rich ones [66]. Furthermore, the DLC films present a unique tribological behaviour. During sliding the outmost layers are transformed into graphite and transfer to the partner surface forming the so-called transfer layer, which protects the sliding surfaces from wear. The build-up of such layer strongly depends on the environment and testing conditions [66]. In addition, the graphitic layer acts as solid lubricant and low friction is then achieved. An introduction to the fundamental properties and applications of DLC coatings can be found in a number of review papers such as [66, 68, 82, 93-95]. In the next paragraphs, the possibilities and limitations of such coating for the joint replacement field will be highlighted.

Diamond-like carbon has emerged as a potential coating for joint replacements mainly due to its high hardness, low friction coefficient, high wear and corrosion resistance, bio- and hemo-compatibility and excellent smoothness. The use of such coatings in the orthopaedic field has been widely studied over the past years. A comprehensive overview of biomedical applications on DLC coatings can be found elsewhere [71, 96-103]. Early testing showed excellent tribo-corrosion performance, low toxicity to the body, and healthy interaction with the cells in the body; some of the coatings heve reached the market and been clinically implanted. In fact, Ching [104] has recently published a study on surface coatings used as orthopaedic implants showing that DLC coatings are the most frequently used coatings (61%) in artificial joints. Despite

outstanding tribological and chemical properties and good short-term results, all coated prostheses failed within 2 to 10 years *in vivo* due to delamination problems [105-107].

One of the main obstacles to the use of DLC coatings is that high internal stress (> 3 GPa) is developed during the deposition process [108]. As expected, also here the  $sp^{2}/sp^{3}$  ratio is important. Indeed, Dearnley [109] found that the higher the  $sp^{3}$ -C bonds the higher the internal compressive stress, which can be as high as 10 GPa. Unfortunately, high internal stress is commonly associated with debonding problems [66, 108] especially on soft metallic substrates [110]. In addition, the adhesion problems can also be related to the interface nature between the coating and substrate [71, 111]. Many attempts have been done to minimizes such issues: (i) deposition of a metallic interlayer, such Ti, Cr, Si, Zr or Ta [112-114] or a gradient transition layer (e.g. Me/MeC, Me/MeN, or Me/MeN/MeNC, where Me is the metallic element) to prevent distinct boundaries [115-117]; (ii) doping of carbon film with light (H, N, F, etc.) or metallic element (Si, Ti, Cr, Ti, etc) [66, 116, 118, 119]; (iii) changing the deposition parameters such as applying bias polarization [120]; and/or (iv) etching of the substrates [121]. Moreover, when implanted in vivo, it is expected that the corrosive body environment will further enhance coating premature failure [70, 71, 122]. In 2003, Taeger et al. [106] analysed a retrieved Ti6Al4V implant coated by DLC (no interlayer applied) and showed the presence of delaminated areas which might trigger adverse body reaction and consequent aseptic loosening. In 2007, Joyce [105] analysed a retrieved DLC coated CoCrMo metartarsophalangeal prostheses (Si interlayer) which also failed after 4 years in service, showing pinholes, imperfections, substrate flaws and scratches that allow the body fluids to penetrate underneath the coating (see Figure I.2a and b). Cracks can then propagate along the DLC/metal interface and enhance the delamination through stress-corrosion cracking (SCC) and crevice corrosion (CC) (see Figure I.2c and d). Hauert and co-workers have summarized the knowledge in the field in references [70, 71, 107, 114].

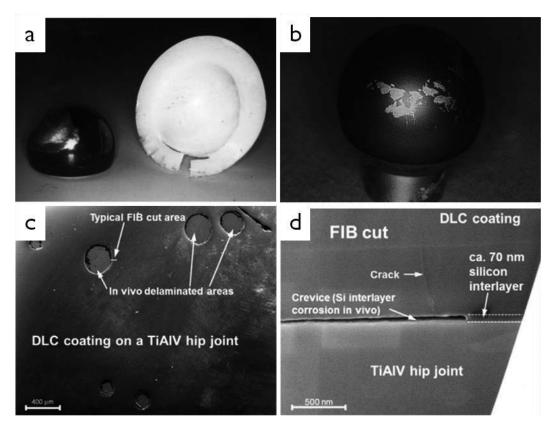


Figure I.2 (a) Retrieved DLC coated hip joint ball and correspondent UHMWPE acetabular component. (b) Numerous pits and confluent and irregular shaped larger defects in load bearing zones in the formerly articulating surface revealing the metallic substrate underneath. (c) Scanning electron microscopy (SEM) micrograph of the delaminated zones on DLC-coating and (d) the correspondent FIB cross-sectional cut near a delaminated area revealing the dissolution of the Si-based adhesionpromoting layer under the DLC due to crevice corrosion. (Adapted from [70, 106]).

Metal-alloying of DLC is another method to improve DLC properties. Even though metal co-sputtered carbon films differ from pure carbon films, it is possible to tune the properties for a desired application but still maintaining the typical amorphous structure. It was found that alloying the DLC coatings not only decreases the residual stress but also influences tribological, mechanical and electrical properties [66, 116, 123, 124], as well as the ion and protein adsorption and the cell reactions [66, 71, 125]. Alloying elements range from metallic (Ti, W, Si, Cr etc.) to non-metallic elements (H, N, P, etc.); an excellent review summarizing the results has been published by Sanchéz-López in Reference [66]. If carbide-forming elements are introduced above a certain concentration, a nanocomposite structure can be achieved: nanocrystalline (nc)-MeC embedded in an amorphous-carbon matrix [126-128]. In fact, Meng [126-127] reported a maximum concentration for Ti dissolution in the matrix as 1-2% and 4-8% in a-C:H and a-C matrix, respectively; for higher concentration nc-TiC nanocomposites are formed. The most common transition metals used as alloying materials are Ti, W, Cr and Nb [66]. Voevodin and co-workers pioneered nanostructured materials [119, 129]. For

instance, when alloying with Ti, the TiC/DLC systems exhibit a large amount of grain boundaries between the nc-TiC phase and the amorphous DLC matrix, which effectively enhances the hardness and toughness of the coatings, limiting crack initiation and/or terminating further crack growth and film delamination [115, 119, 120, 130-133]. Moreover, such nanocomposite structure could also lower the friction and wear and significantly improve load carrying capacity [129]. Apparently, encapsulation of finer grains of hard nc-TiC within the amorphous DLC matrix restricts dislocation activity, diverts and arrests macrocracks development, and maintains the high level of hardness [120, 132], see Figure I.3.

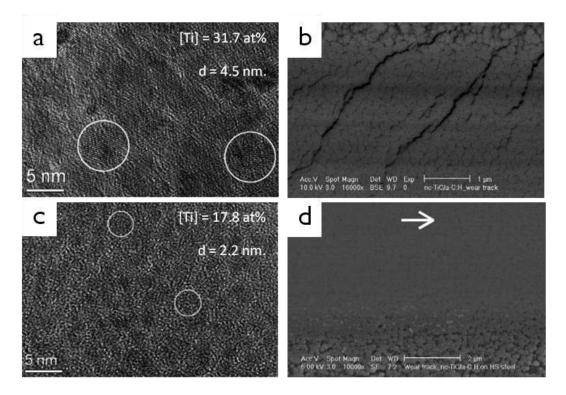


Figure I.3 (a) and (c) shows the HR-TEM micrographs of the evenly distributed and well separated TiC nanograins and (b) and (d) SEM micrographs of the corresponding the wear track under the same wear conditions. (Adapted from [132, 133]).

From the biomedical point of view, the introduction of alloying elements will modify the typical biological properties of DLC coatings. Ti [134], Si and F [135], Ca and P [71], N [136], Ag [137], Zr [138] and, more recently, Ta [139] are the most common doping materials which may enhance blood and cell compatibility due to surface chemistry modifications. Among these Ti has gained much interest owing to its biocompatible and tribological properties. Several authors [140, 141] found that proteins adsorbed on surface as a function of Ti content. Protein adsorption is the first event

occurring after implantation which signalizes the overall biological response. Moreover, the ability of naturally forming  $TiO_2$  by interaction with air improves corrosion resistance and may add affinity and biocompatibility, compared to pure a-C(:H) films. It was found that Ti-alloyed a-C(:H) stimulated osteoblast (bone producing cells) differentiation and inhibited osteoclast (bone resorbing cells) cell activity compared to a-C(:H) films [141, 142]. In the light of that Bruinink et al. [143] have studied the possible toxic effect of Ti-containing and pure a-C:H particles in the case of unexpected delamination of the coating *in service*: only pure films did not exhibit inflammatory induced response. Thus, Ti-containing DLC coatings may be suitable to bone related implants by improving fixation and enhanced osteo/osseointegration but not for loadbearing prostheses. It is still not clear how alloyed materials would perform *in vivo* due to a lack of adequate analysis of surface interacting with body fluids. However, it is expected that changes in surface chemistry could be beneficial for a successful *in vivo* performance if appropriate elements are used.

# 3. The Main Goal of the thesis

The use of alloyed DLC coating has revealed good biomedical potential together with an enhancement of the fundamental properties of the coating. Zirconium as a bulk material has shown low-toxicity, biocompatibility, high corrosion resistance and excellent tribological properties [144, 145]. Zr is a strong carbide forming element such as Ti but it differs slightly in the mechanical properties. The use of transition metals as doping elements is not new and has improved mechanical and tribological properties of the coatings due to the catalysis effect of the metal on DLC formation [66, 119, 146-149]. Moreover, the potential of protective high-strength material for biomechanical purposes is dependent on the average grain size due to the grain boundary effect typical of nanocomposite structures, as explained before. Thus, zirconium can be considered as a promising doping element competitive to Ti. In fact, Kao [150] has reported that an addition of 4-16 at.% Zr can considerable improve the tribological properties including the friction coefficient in dry environment. Kumar et al. [138, 151] and Wang et al. [116] have reported excellent pitting and crevice corrosion resistance of Zr-DLC coatings. Zr naturally forms a zirconium oxide layer considered as bioinert [152], which may enhance corrosion resistance [144]. Moreover, Tsutsumi et al. [153] and Manivasagam et al. [154] have found that such layer is more stable on Zr than on Ti when immersed in Hank's solution, which makes the Zr more passive and protective than Ti but also less bioreactive [155]. Thus, the incorporation of Zr is expected to decrease osseointegration compared to Ti-containing DLC coatings. In summary, the combination

of the nc-ZrC and amorphous-C phases with a functionally graded interface might provide excellent toughness, high resistance to adhesive failure and corrosive resistance under severe biotribological situations.

The main goal of this thesis was then to produce nanostructered Zr-containing DLC coatings with different Zr concentration and evaluate their possible use in loadbearing biomedical application such as articular prosthesis in order to minimize the wear and the prosthetic intrinsic problems.

# CHAPTER II

The deposition of hydrogenated and non-hydrogenated DLC and Zr-containing DLC coatings onto Ti grade 2 and 5 ELI is described in this chapter together with a brief summary of the characterization techniques. Since the adhesion of the coatings onto the metallic substrates has been used as one of the main optimization parameters, it will be exhaustively analysed as well.

# I. Introduction

This chapter is dedicated to the deposition and optimization of the hydrogenated and non-hydrogenated DLC and Zr-containing DLC coatings. The coatings were deposited by using reactive and non-reactive magnetron sputtering, respectively, on biomedical grade Ti substrates: Ti grade 2 (commercial pure (CP-) Ti) and Ti grade 5 ELI (Ti6Al4V). Ti-based materials are interesting candidates for biomedical applications because of the known milder tissue reactions of their degradation products compared with Co-Cr alloy and Ni-containing steels [33, 41]. In fact, a recent study [156] has shown a patient from New England with an apparent heart failure that underwent an orthotopic heart transplantation; after 13 months it was discovered that such problem came from a huge Co- and Cr-ion release from his bilateral DePuy metal on metal hip prosthesis and needed revision surgery. However, Ti-based materials are soft and their tribological properties are poor [9, 157, 158]. Therefore, the use of coating is a promising approach to improve their tribological properties. As described in Chapter I, a potential limitation of coatings is their adhesion to the substrate, which is directly related to the chemical bonds between the layers as well as the elastic properties of both bulk material and coating. The deposition of hard coating on a relatively soft substrate such as Ti is an enormous challenge. The high gradient in superficial hardness can result in coating chipping off due to different elastic behaviour under high punctual loads (thin ice effect) [26, 110, 159, 160]. Therefore, after summarizing the main information concerning the deposition of the films and their characterization, this Chapter is dedicated to the optimization of the coatings deposition; in particular, some aspects not covered in the published papers, which are the basis of the thesis, will be discussed: (i) an exhaustive study on the effect of the interlayers design on the adhesion of the coatings on Ti-based substrates in order to avoid in service delamination and fracture of the coating and (ii) the analysis of the influence of the wear of the Zr pellets on the final results of the chemical composition and deposition rate of the coatings.

# 2. Coatings deposition and characterization

#### 2.1 Deposition Process

DLC coatings can be produced by a variety of deposition processes. In general, the processes are very complex regarding the physical and chemical phenomena underlying the coating growth. In fact, the final coating structure strongly depends on the deposition technique. In this study, magnetron sputtering was the chosen technique to develop pure and Zr-containing amorphous carbon coatings. Figure II. I shows the deposition chamber and configuration as well as the schematic representation of the final deposition parameters. Two parallel magnetrons were in a close-field unbalanced configuration to enhance the magnetic flux density in the chamber and to provide more efficient electron trap.

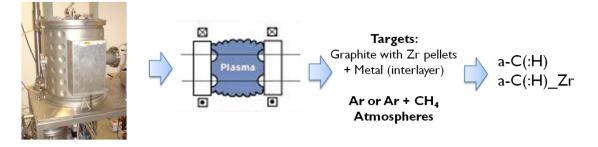


Figure II.1 Picture and schematic representation of the deposition chamber and deposition parameters.

Several parameters were kept constant for all depositions, such as the geometry of the deposition chamber, the magnetic field, a constant position of the inlets of the sputtering gas, the base pressure (<  $10^{-3}$  Pa), the substrate-to-target distance ( $d_{s,t}$ =170 mm), the substrate holder rotation (18 rpm) and the substrate thermal condition (no intentional heating was used). In addition, the C-target power density was also kept constant at 7.5 W.cm<sup>-2</sup>. The use of negative substrate bias was also invariable in all deposition. For that, a pulsed-DC power supply was connected to the substrate holder with a pulsed frequency of 250 kHz and a bias voltage of -50 V. Two main combinations of conventional process parameters were used for the development of both hydrogenated and non-hydrogenated coatings: reactive (Ar/CH<sub>4</sub>) and non-reactive atmospheres (Ar), respectively. The Ar flow was maintained at 45 sccm in order to reach the working pressure of ~0.4 Pa. For the deposition of hydrogenated coatings methane gas was introduced in the chamber to initiate the reactive process. The Ar gas flow was fixed at 40 sccm and the CH<sub>4</sub> gas flow at 5 sccm, in order to keep the working pressure constant.

#### 2.2 Coating characterization methods

To understand the influence of the deposition parameters on the functional properties of the coatings, several characterization methods were used. The properties

of the coatings were analysed in a first attempt with basic characterization: chemical composition, thickness and adhesion to the substrate. Si wafers and glass were generally used for microstructure and surface characterization and polished metallic substrates for mechanical and tribological properties assessment. Once proper deposition parameters had been determined, coatings were simultaneously deposited on appropriate substrates (i.e. CP-Ti and Ti6Al4V).

#### 2.2. I Chemical composition:

The chemical composition of the coatings was evaluated by electron probe microanalyses (EPMA), CAMECA Camebax SX 50 model. The applied voltage was 10 keV corresponding to a maximum penetration below 2 µm for C-based coatings. The values of the chemical composition were the average of 3-5 different measurements in different points of the sample. Unfortunately, this technique cannot evaluate hydrogen content. Therefore, in order to also evaluate the hydrogen content, the films were analysed by a full total ion beam analysis (IBA) – Rutherford backscattering (RBS), non-Rutherford elastic backscattering (EBS), elastic recoil detection (ERD) and particle-induced X-Ray emission (PIXE) self-consistently – used by the DataFurnace code (NDFv9.4f [161]). The experiments were conducted at Surrey Ion Beam Center, University of Surrey as part of a European grant under the SPIRIT<sup>1</sup> - project program. Detail experimental parameters can be found in Paper IV.

Chemical composition information of particular zones of the coated samples was also achieved during scanning electron microscopy (SEM) and transmission electron microscopy (TEM) through energy dispersive X-ray spectrometry (EDS) connected to both microscopes. During X-ray photoelectron spectroscopy (XPS) the chemical composition could also be evaluated through the analysis of the relative peak intensities, corrected by experimental determined sensitive factors.

# 2.2.2 Thickness, morphology and surface topography

The thickness of the films was evaluated by 2D optical profilometer (Perthometer S4P) using a substrate/coating step. A boron nitride drop was placed on a

<sup>&</sup>lt;sup>1</sup> **SPIRIT** – represents an Integrated Infrastructure Initiative (I3) funded by the European Commission. The main goal of SPIRIT is to grant public and industrial researchers from Europe free access to leading European ion beam facilities and R&D providers, which deliver ions in an energy range from ~10 keV up to 100 MeV for the modification and analysis of solid surfaces, interfaces, thin films and nanostructured systems. The main application areas are materials, biomedical and environmental research and technology [162].

Si substrate and removed with acetone after the deposition, leaving a step on the coated surface. This method was applied to measure the thickness of the coatings in order to optimize the deposition time. The thickness of the coating was also compared and checked by direct measurements in cross-section observation in SEM (FEI Quanta 400FEG E SEM, at CEMUP). The deposition rates were calculated from the film thickness and the sputtering time.

SEM was also used to get a general idea of the morphology of the coatings. Additional to plan-view, fractured cross-section of the coated samples were regularly studied as they often allow a straightforward observation of the coating morphology, cohesion and adhesion to the substrate. A quantitative complementary method to examine the surface morphology was the analysis of the surface roughness. The average surface roughness ( $R_a$ ) and the roughness mean square (Rms) were evaluated by atomic force microscopy (AFM – Buker Innova) - 5x5  $\mu$ m<sup>2</sup> line scan – and scanning white light interferometer 3D optical profilometer (Zygo New Vision). The AFM images depicted in Chapter III were performed in 3x3  $\mu$ m<sup>2</sup> line scan.

#### 2.2.3 Structure and microstructure

The phase identification was performed by using X-ray diffraction (XRD). A single circle diffractometer (Philips X'Pert) has been used with  $CoK_{\alpha}$  radiation ( $\lambda = 1.789$ Å) generator operated at 40keV. For more precise measurements, some samples were analysed in a six-circle goniometer (at the Materials Research station of the Rossendorf beam line (ROBL-MRH) located at the bending magnet 20 (BM20) at the European Synchrotron Radiation Facility (ESRF) in Grenoble) with an incident X-ray beam monochromatized to an energy of 11.5 keV ( $\lambda = 1.078$  Å). To increase the signal from the coating and to identify any crystalline phase, a grazing angle incidence XRD was performed, hereby the sample surface was tilted 2° in relation to the beam. The phase identification was done by comparing the position of the peaks with those of standard phases existing in the International Center for Diffraction Data (ICDD) database [163]. Even though the observed peaks were quite broad, due to small grain sizes typical of nanostructured materials, the XRD patterns were fitted using a Voigt function permitting the determination of several characteristics such as peak position (2 $\theta$ ), peak intensity and full width at half maximum (FWHM) that can be used for estimate an average size of the diffracting crystal through the Scherrer formula [164].

The structural analysis was complemented by TEM using a FEI Tecnai G2 T20 microscope with a  $LaB_6$  filament operated at 200 kV and a FEI Titan 80-300 microscope

equipped with field emission gun (FEG) and Gatan image filter (GIF) Tridium and operated at 300kV. Both bright field (BF) imaging and selected area electron diffraction (SAED) were carried out. SAED was performed with an aperture size of ~100 nm in diameter. Electron energy loss spectroscopy (EELS) was also used to reveal the bonding state of the elements present in the film. TEM specimens were prepared using an FEI Versa3D focus ion beam – scanning electron microscope (FIB-SEM). The experiments were carried out at Chalmers TH after a successful application to a European grant under the ESTEEM  $2^2$  project program.

Raman spectroscopy (Xplora, Horiba, DPSS laser,  $\lambda = 532$  nm) was used to characterize the amorphous carbon phases present on DLC and Zr-containing DLC coatings. The peaks were fitted using two Gaussian peaks following the procedure described elsewhere [166].

#### 2.2.4 Surface physicochemical characterization

Chemical bonding was investigated using X-ray photoelectron spectroscopy (XPS). The electron binding energy is specific for each element and orbital, thus enabling the elemental identification. The XPS analysis was performed using a Kratos AXIS Ultra with VISION software for data acquisition and CASAXPS software for data analysis. The analysis was carried out with a monochromatic AI K $\alpha$  X-ray source (1486.7 eV), operating at 15kV (90 W), in FAT mode (Fixed Analyser Transmission), with a pass energy of 40 eV for regions ROI and 80 eV for survey. Data acquisition was performed with a pressure lower than 10<sup>-6</sup> Pa, and a charge neutralisation system was used. The binding energy scale was charge referenced to the C Is at 285 eV. The spectra were acquired on as deposited surface and after sputter-cleaning with Ar<sup>+</sup> ion gun in order to remove any residual contamination. The deconvolution of the spectra was performed using the CasaXPS program, in which an adjustment of the peaks was performed by using peak fitting with Gaussian/Lorentzian peak shape and Shirley type background subtraction. The relevant peak positions, as found in literature, are given in Paper VII.

The contact angle measurements were performed through the "static"-sessile drop method. Drops were generated with a Krüs GmbH G-23 goniometer at  $\sim$ 20 °C

<sup>&</sup>lt;sup>2</sup> **ESTEEM 2** – represents a *European Netwotrk for Electon Microscopy* integrated infrastructure providing access for the academic and industrial research community in the physical sciences to some of the most powerful characterization techniques available at the nanoscale. Transnational access to ESTEEM2 centres is obtained through a transparent, simple peer review process based on merit and scientific priorities. Service to users is supported by a networking programme which addresses key issues such as specimen preparation, data interpretation through theory and simulation, and standardization of protocols and methodologies [165].

and room humidity (50%). The surface energy was calculated using the Owens equation [167] based on the contact angle measurements of the different liquids: water, glycerin  $(C_3H_8O_2)$ , formamide  $(CH_3NO)$  and diiodomethane  $(CH_3I_2)$  [167, 168]. The interaction with bovine serum albumin (BSA) was studied by immersion of the samples in a 2 ml BSA containing solution (4 mg.ml<sup>-1</sup>) diluted with a basic solution (NaCl: 9 (g/l); EDTA: 0.2 (g/l); Tris: 27 (g/l); sodium azide: 0.3%, pH 7.6) for 24 h at 4 °C. The amount of total immobilized protein was calculated using the Bradford reagent against a standard BSA calibration curve. After immersion, all samples were washed with water Mili-Q for eventual detachment of non-chemisorbed proteins. The detached proteins were also taken into account for the protein quantification using the same method described above. The samples were then air-dried for 24h before the XPS spectra were recorded. More info can be found in Paper IV.

#### 2.2.5 Mechanical Properties

The residual stress ( $\sigma_{RS}$ ) was calculated by the curvature or deflection method using the corrected Stoney equation [169]. The coatings were deposited on thin stainless steel (SS) 304 discs ( $\emptyset$  25 x 0.5 mm) and the curvature radius was evaluated by 2D profilometer before and after deposition.

The hardness (*H*) and Young modulus (E) of the coatings were evaluated by depth-sensing indentation (Micromaterials Nanotest) using a Berkovich indenter. The normal stylus load was 5 mN (indentation depth approx. 150 nm); To have reliable *H* values, 32 indentations from 2 distinct areas on the sample were performed. Furthermore, the reduced Young modulus was derived from the indentation measurements by standard Oliver and Pharr method [170]. The Young modulus was then calculated by using the following parameters:  $E_i = 1140$  GPa and  $v_i = 0.07$  for the diamond indenter [170] and v = 0.3 for the coating [66] (v is the Poisson ratio).

The adhesion of the coatings was evaluated by scratch test and the Rockwell-C (R-C) adhesion test, regulated by European standards (EN) 1071-3 [171] and 1071-8 [172], respectively. Both techniques use a Rockwell-C indenter: spherical diamond 120° tip with a diameter of 200  $\mu$ m (CSM Instruments). The main difference between these techniques is that in the scratch test the spherical end of the indenter slides against the surface, while in the R-C the indenter penetrates into the material. For scratch test, an increasing applied load with a loading speed of 10 N.mm<sup>-1</sup> and a range of 2-50N was used. During R-C testing, the normal load was dependent on the hardness of the substrate: for soft metallic substrates (< 54 HRC) a load of 981 N and for harder

metallic substrates (> 54 HRC) a load of 1471 N [172] was applied. These tests are strongly influenced by the elastic properties of both film and substrate, the thickness of the film, etc [66, 173]. Thus, M2 Steel was used as a reference material to study the adhesion of the coatings; then Ti-based substrates (CP-Ti and Ti6Al4V) were used to predict the adhesion *in service* of the coatings. The results of the tests were qualitatively evaluated by comparing the optical microscope images (100X magnifications). To determine the adhesion strength of coatings by scratch test, three critical loads were considered:  $Lc_1$ , the lower critical load corresponding to the first cracks that occured (cohesive failure);  $Lc_2$ , the upper critical load defined as the load where the first spallation/flaking of the film is observed; and  $Lc_3$ , the load where penetration of the coating is detected in the centre of the scratch [171]. On the other hand, the crack network and degree of delamination of the indentation, resulting from R-C test were analysed based on adhesion quality charts and classified into six categories from HF1 to HF6, where HF1-HF4 is considered as acceptable failure while HF5-HF6 corresponds to unacceptable failure [9]. For more details see Paper I.

#### 2.2.6 Tribological Properties

Wear testing was performed using a pin-on-disc CSM tribometer. Three different environments were tested: ambient air (relative humidity (RH) 30%); physiological solution (PS; 0.9 % NaCl water solution) and diluted foetal bovine serum (BSA, prepared according to ASTM F732 [174]). The tests were carried out against a Ti6Al4V ball with a diameter of 8 mm. The temperature was maintained constant at 37±2 °C for the lubricated tests. All measurements were provided with a normal load of IN and linear speed of 0.2 m.s<sup>-1</sup>. The maximum Hertzian pressures for ball-on-plate contact were almost identical close to 360 MPa (Ti6Al4V and Cp-Ti). The standard number of cycles was 10 000. Uncoated Ti-based substrates were also tested under the same conditions for comparison. Detail information can be found in Papers II and IV. A test closer to the in service application was also performed by using a multidirectional pin-on-disc (OrthoPod<sup>®</sup>, AMTI Inc.). These tests were carried out against UHMWPE and PEEK pins under a protein containing solution (BSA, protein concentration of 30g/mL). The temperature of the lubrication media was maintained at 37°C. The pins were subject to a 15 mm X 15 mm square motion pattern with a cycle frequency of 1 Hz and a constant applied load that generated a nominal contact pressure of 1.5 MPa. Paper VI presents a more detailed description of this particular experimental process.

The tribological behaviour was examined with respect to the friction coefficient and the wear rate. The surface and wear track topography was evaluated by using 3D profile measurements whereas the ball wear rate was calculated from optical inspection of the spherical wear cap (optical microscopy). On the other hand, gravimetric weight loss for the polymeric pins was used to calculate the volumetric wear loss (according to ASTM F2025 [175] and F732 [174]). The differences in the wear rate between UHMVVPE and PEEK articulating against coating and uncoated materials were evaluated statistically by using students' tests. One-way analysis of variance (ANOVA) followed by the Student-Newman-Keuls (SNK) method was used to determine pairwise differences between each polymer group. The level of significance was set to p = 0.05.

# 3. Coating optimization

#### 3.1 Adhesion and interlayer optimization

The biotribological performance of DLC coatings is strongly related to the adhesion of the coating to the metallic substrate. As referred to in Chapter I, it is indeed the main drawback for its use as load-bearing implants. Thus, one of the main requisites is that the coatings should adhere to the substrate not only after deposition but also *in service*.

To enhance adhesion, the substrate should be carefully cleaned with degreasing agents to remove surface contaminants and subsequently dried before being placed into the deposition chamber [176]. As a result, prior to be placed into the deposition chamber all substrates were ultrasonically cleaned with alcohol, acetone and distilled water for 15 min each. After the vacuum chamber was at the desired base pressure, substrate plasma cleaning was performed to remove surface oxides in Ar<sup>+</sup> atmosphere at a pressure of ~0.3 Pa for 1h with a bias voltage of -650 V allowing a self-current of ~1.0 A. A power of 600 W and 1000 W was also applied on the C and metal targets, respectively, during substrate cleaning to enhance the ionization of the glow discharge and to clean up the oxide contamination layer on top of the targets (note that the targets were covered by a shutter in order to avoid cross contamination of the targets and substrates). Even though the pre-cleaning and ion-cleaning steps are crucial for a good adhesion [176] the adhesion strength was insufficient when depositing pure DLC coatings directly on steel (see Figure II.2). It was then necessary to optimize the adhesion of the coatings onto metallic substrates by other methods.



Figure II.2 Scratch test at increasing load on a-C coating deposited without interlayer.

#### 3.1.1 Adhesion interlayer optimization

The deposition of a metallic and/or gradient transition layer between the functional coating and the substrate to improve the adhesion onto metallic substrates is not new and it has been considered as one of the most successful methods for improving adhesion, as mentioned in Chapter I. Thus, in this study the optimization of the adhesive layer was focused on the interface material between substrate and functional coating.

Many attempts to choose the best adhesive layer were carried out based on compositional gradient layer along the coating thickness using Zr, Cr and Ti as metallic elements. After etching, the bias voltage was switched to -50 V and a Me/MeN/MeCN(Zr) or Me/MeC (where Me stands for the metal element: Zr, Cr or Ti) composite gradient layer was deposited for 20 min, applying a constant power of 2000 W to the Me target. Me target was firstly sputtered in a pure Ar atmosphere (Ar flow 35 sccm); then, when necessary, nitrogen was introduced (17 sccm). The deposition of the functional coating (hydrogenated and non-hydrogenated) was then initiated by ramping up the power on the graphite target (without embedded Zr pellets) and ramping down the Me target power to zero and the nitrogen flux. Figure II.3 shows the different designed interlayers tested in this study.

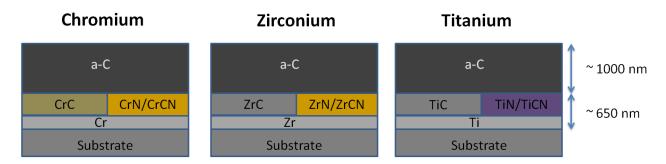


Figure II.3 Schematic representation of the different design of tested adhesion layers.

The adhesion strength was then measured by scratch test and it is shown in Figure II.4. Typically, the adhesion improving layer is usually based on the same alloying element [148, 177]. Unfortunately, when using Zr as metal element, the coating catastrophically delaminated due to the presence of ZrC brittle phase, especially when a ZrN layer was used. Cr/CrC, Cr/CrN/CrCN and Ti/TiN/TiNC were then found to be the best options for the development of a functional adhesive gradient layer. In fact, Cr has been the most widely used element for improving the adhesion of DLCs [66]. Although it has been reported that the use of Cr as interlayer might release less Cr-ion compared to a Co-Cr alloy typically used as an implant material [116, 178], it was decided to avoid chromium if possible. Ti/TiN/TiNC was then used as an interlayer material.

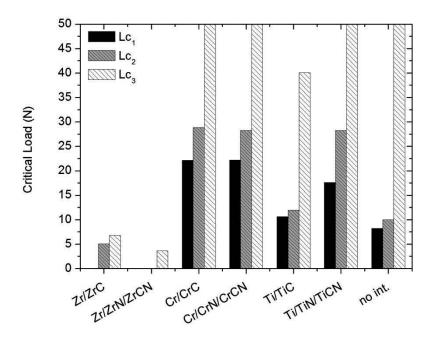


Figure II.4 Critical loads of a-C coated with the different designed interlayers on steel substrates.

It is also important to notice that using Ti as a metal element should also minimize the differences in the thermal expansion coefficients between the substrate and the coating when deposited on Ti-based substrates. The perfect match between Young modulus of the Ti-based substrate and the interlayer, as well as the subsequently gradation of the composition towards the coating side, may improve the compatibility between the film and substrate materials and, consequently, the adhesion [117]. Figure II.5 shows the cross-sectional TEM BF images showing the columnar morphology of TiN and the presence of an amorphous second gradient layer (TiCN(Zr)) between Ti/TiN and the DLC functional coating. By starting the coating growth from an amorphous

phase, fewer discontinuities between the interlayer and the top coating would be expected, thusly improving the overall adhesion of the coating.

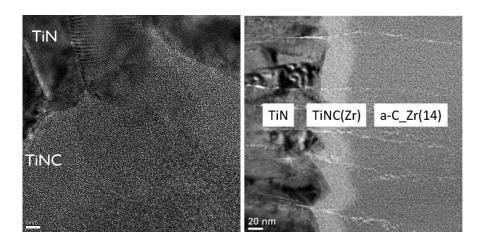


Figure II.5 Cross sectional TEM BF-images of a-C\_Zr(14).

#### 3.1.2 Adhesion onto Ti-based substrates

The adhesion of DLC films varies with the substrate and it is related to the chemical interaction between the film and substrate surfaces, the microstructural defects in the film-substrate interface and the compressive stress levels in the film [121]. The deposition of DLC coatings on soft biomedical substrates such as Ti-based materials is a continuous challenge due to substrate plastic deformation, which could cause fracture and delamination of the coating [110, 159, 160]. The coatings were deposited on CP-Ti and Ti6AI4V. It is worth noting that CP-Ti is softer than Ti6AI4V: 1.5 GPa and 3.1 GPa, both respectively [179] much softer than the M2 steel (8.2 GPa) originally used for the interlayer deposition optimization. Typical scratches produced by the adhesion test are shown in Figure II.6. As expected, the adhesion results were very sensitive to the mechanical properties of the substrate. Piling-up along the edges of the scratch track, which was clearly observed in both substrate materials, indicated plastic deformation (Figure II.6a and c). The plastic deformation was much more evident in the case of CP-Ti substrates, i.e. it was visible for lower loads. Figures II.6b and d show the scratches of a-C coatings deposited on CP-Ti and Ti6Al4V, respectively. The scratch tracks were much wider for the softer CP-Ti substrates when an identical load was applied leading to the destruction of a-C films; a harder substrate (Ti6Al4V/a-C) yielded improved coating adhesion and wear resistance.

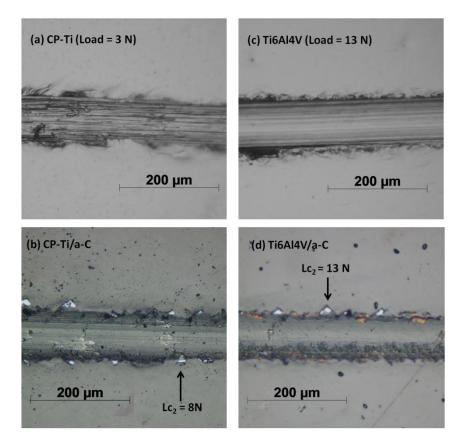


Figure II.6 Optical images of typical scratch tracks performed on (a) uncoated CP-Ti and (c) Ti6Al4V and a-C deposited on (b) CP-Ti and (d)Ti6Al4V. Scratch direction is from right to left. The arrows show the adhesive critical load Lc<sub>2</sub>.

#### 3.2 Chemical composition

For the deposition of Zr-containing films, Zr could be introduced either by the use of a Zr target or a Zr-C composite target (graphite target with Zr pellets). The former cannot be used due mainly to the impossibility of using very low target power to achieved nanocomposite films (very low Zr content) since the sputtering yield of Zr is almost 4 times higher than C [180]. Moreover, since only two target materials can be used (graphite and metal) with such magnetron sputtering configuration and Zr did not revealed sufficient adhesion when used as interlayer material, a Ti target should also be add to the deposition chamber. Thus, Zr Ø 5 mm pellets were embedded into the erosion area of the graphite target in order to deposit nanocomposite Zr alloyed DLC coatings. The targets (Ti and graphite) and the pellets were supplied by Testbourne Ltd with purity level of 99.99%. Figure II.7 shows the schematic representation of target and pellets.

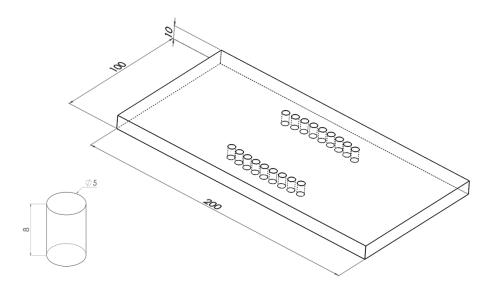
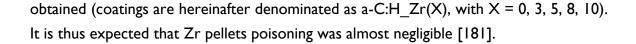


Figure II.7 Schematic representation of the mixed graphite target and the Zr pellets used to deposit Zr-containing DLC. The Zr pellets can be easily removed from the erosion zone and changed by C pellets to adjust the Zr content in the films. The target was glue to the Cu plate with a thermoplastic conductive silver ink (Sun Chemical<sup>®</sup>) to improve conductivity and then clamped by the target holder. All units in the Figure are in mm.

To synthesize coatings with the desired chemical compositions an interactive process of calibration and tuning of deposition parameters was used. The alloying of DLC coating with Zr was possible by incorporating different number of Zr pellets into the erosion zone of the C-target. However, co-sputtering with a mixed target is far from a simple process; it depends on the target / pellets aging, i.e., the erosion degree of both target and pellets. Thus, to achieve the desired compositions, the number of Zr pellets needed to be continuously adapted. Figure II.8 gathers all the results of chemical composition obtained by EPMAand shows how a different relative erosion area ( $A_{Zr}/A_C$ ) was needed to achieve the desired Zr content. As expected, the higher the number of Zr pellets (higher  $A_{Zr}/A_C$ ), the higher the Zr content.

For non-hydrogenated coatings, the Zr content was varied and set constant at values of approximately 0, 4, 7, 9 and 14 at.% (coatings are hereinafter denominated as  $a-C_Zr(X)$ , with X = 0, 4, 7, 9, 14). When introducing methane to the Ar atmosphere, the deposition process tends to be different due to both the incorporation of another carbon source and the poisoning effect on the Zr pellets. Under reactive atmosphere, the formation of compounds on top of the metallic target can occur and change the sputtering rates of the metallic element [176]. However, in this case, for the same number of Zr pellets, the Zr content was only slightly lower when compared to non-hydrogenated films. Thus, Zr contents of approximately 0, 3, 5, 8 and 10 at.% were



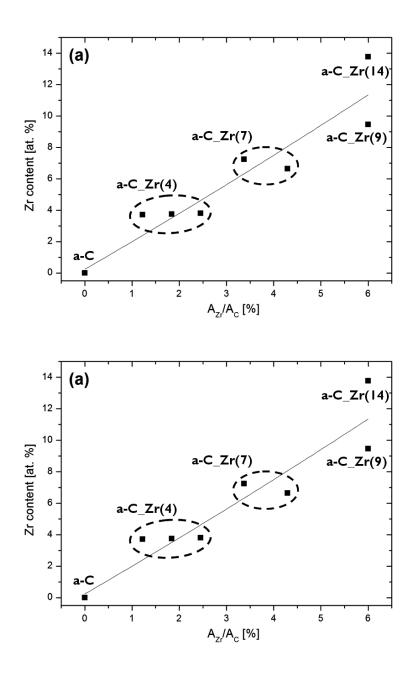


Figure II.8 Zr content as a function of relative erosion area  $(A_C/A_{Zr})$ : (a) non-hydrogenated and (b) hydrogenated coatings. The insets are the correspondent names that will be used during this work.

#### 3.3 Deposition rate

The greatest disadvantage of using magnetron sputtering to deposit DLC films is that carbon presents a very low sputtering yield compared to many other elements. Therefore, the deposition rate is relatively slow. The incorporation of a carbonaceous gas precursor (such as  $CH_4$ ) and/or the incorporation of the Zr pellets can provide coatings at higher deposition rates, since another carbon source is added and Zr has higher sputtering yield compared to C [180]. When depositing films for tribological applications, the thickness of the coating is one of the most important parameters to take into account [82] and it obviously depends on the deposition rate. Thus, the sputtering time needed to be adjusted in order to deposit the films with similar thickness  $1.5 - 1.7 \mu m$ .

Figure II.9 shows the deposition rate as a function of Zr content for both hydrogenated and non-hydrogenated coatings. As expected, the deposition rate was strongly related to the deposition conditions. In general, for both non-hydrogenated and hydrogenated coatings the deposition rate increases with the increase of Zr content, as a consequence of the higher sputtering rate of Zr in relation to C. In addition, when a carbonaceous gas was introduced, the deposition rate was almost the double when compared to H-free coatings. The reactive gas worked as a rich carbon source increasing the available carbon into the chamber [177]. Finally, the importance of the erosion status of the Zr pellets on the deposition rate should be remarked. As shown in the Figure II.8, to get approximately the same Zr content (e.g. 4 at.%), different  $A_{Zr}/A_C$  ratios have to be used, which also influenced the deposition rate. Therefore, this knowledge was extremely important for the optimization of the coating deposition to achieve the desired Zr content and thickness for a specific film.

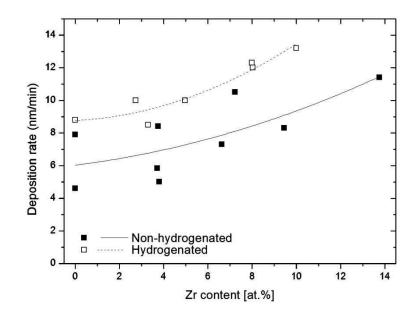


Figure II.9 Deposition rate as a function of the Zr content.

# 4. Conclusions

DLC and Zr-containing DLC coatings were successfully deposited onto steel and soft biomedical Ti-based substrates by magnetron sputtering in reactive and non-reactive atmosphere. Embedded pellets were used as Zr source; however, the erosion level of the target was found to be an issue. Therefore, it was important to study first the Zr content evolution as a function of  $A_C/A_{Zr}$ . When using reactive magnetron sputtering, the Zr content was slightly lower than that of non-hydrogenated coatings, since  $CH_4$  worked as a second carbon source; the poisoning effect of Zr pellets had a negligible effect on the deposition rate. The deposition rate was also influenced by the deposition conditions, in particular the erosion level of the target, the number of Zr pellets and the deposition in reactive mode. Thus, in order to maintain the thickness constant the sputtering time needed to be adjusted in accordance with the actual deposition rate.

The adhesion of the coatings was used as an optimization parameter since a-C coatings presented insufficient adhesion without an interlayer. Various attempts were performed based on a functional gradient layer. Ti/TiN/TiNC was chosen as the most suitable option with respect to the adhesion strength and the substrate material (Ti-based). Even though Cr/CrC presented good adhesion results, the use of such layer is not recommended for the final application due to possible toxicity. The adhesion of the coatings on Ti-based materials was found to be substrate related. Ti6Al4V offered better load-support compared to CP-Ti.

# **CHAPTER III**

This chapter deals with the basic physical and chemical characterization of Zr-containing a-C and a-C(:H) coatings, in particular their chemical composition, morphology and structure. The influence of both Zr and H on the mechanical and tribological properties is also discussed.

The main results of this chapter were presented in **Papers I**, **II**, **III**, **IV**, **V** and **VII**.

# I. Introduction

The central question of this Chapter is whether composition, microstructure morphology and properties of the material are correlated. The deposition of pure and Zr- and H-containing films has been evaluated in an exploratory manner in Chapter II. During this Chapter the task of correlating nanostructure and fundamental properties, and subsequently the tribological behaviour, is developed. The results were fully expanded and supported by the author's papers I-V and VII. Thus, the microstructure and phase composition have been studied in Paper VII. The adhesion of the coatings on Ti-based substrates was extensively analysed in Paper I and III. In Paper II and V, the correlation between mechanical and basic tribological properties (ambient air) with structure was evaluated and, finally, Paper IV reports to the H content of the films and their density, together with the study of the tribological behaviour of the coatings under a corrosive medium (physiological solution).

# 2. Characterization of DLC nanocomposite films

# 2.1 Coating characterization

Two series of coatings were investigated: hydrogenated and non-hydrogenated, with different Zr contents. Table III.I shows the chemical composition obtained by IBA.

	Chemical Composition						
Samples	C [at.%]	Ar [at.%]	Zr [at.%]	H [at.%]	<b>Density</b> [g/cm³]		
a-C_Zr(9)	90.3	0.6	9.0	0.2	3.1		
a-C_Zr(7)	88. I	3.0	7.4	١.5	3.9		
a-C_Zr(4)	91.4	3.8	3.8	0.9	3.3		
a-C	94.0	4.4	-	۱.6	2.6		
a-C:H_Zr(8)	67.6	1.9	5.7	24.8	2.8		
a-C:H_Zr(5)	71.3	1.0	4.8	22.9	2.3		
a-C:H_Zr(3)	64.I	1.1	3.3	31.5	3.1		
a-C:H	61.3	0.6	-	38.1	1.9		

Table III. I Chemical composition and density measured by IBA of the coatings.

It is important to notice that, as a result of the incorporation of H, for hydrogenated coatings the Zr content was slightly lower than that measured by EPMA (Chapter II, Figure II.8b). The use of a reactive atmosphere (Ar+CH<sub>4</sub>) led to H incorporation diminishing from 38 down to 23 at.% with increasing Zr content. The density was strongly related to H and Zr contents, although with antagonist effects. In general, the incorporation of H into the C-matrix decreased the density of the coatings compared to non-hydrogenated films. In contrast, Zr co-sputtered films were denser than pure films (Paper IV). Figure III.I shows the fractured cross-sectional SEM micrographs of selected hydrogenated and non-hydrogenated coatings. Pure a-C films presented a typical columnar growth. Although the incorporation of Zr did not reveal any significant morphological change, Zr, as a transition metal, can substitute the carbon atoms in the rigid C-C and C-H network distorting the electron density distribution [182] and, thus, decreasing the coordination defects and increasing the density of the coatings. On the other hand, the coating growth in a reactive atmosphere is influenced by the high mobility of carbonaceous species that may destroy the columnar boundaries leading to a glassy and compact morphology [120, 131].

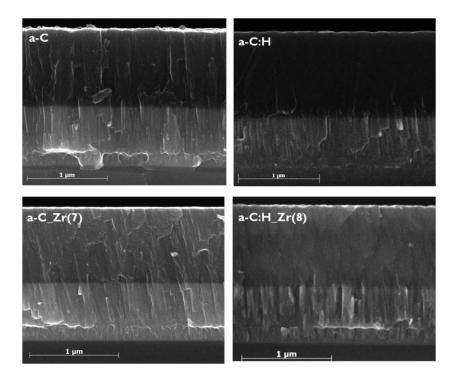


Figure III.1 SEM micrographs of fractured cross-section of non-hydrogenated (a-C and a-C\_Zr(7)) and hydrogenated coatings (a-C:H and a-C:H\_Zr(8)).

AFM contact-mode scans of the surface of the coatings are displayed in Figure III.2, together with the correspondent roughness values. Zr-containing films show very smooth surface regardless of hydrogen content, whereas pure hydrogenated and non-hydrogenated coatings showed dissimilar surface topographies. The a-C columnar

growth is associated with the cauliflower-like patterned surface, with the consequent increase in the roughness, different from the glassy/compact morphology of a-C:H films that leads to a smooth and featureless surface. A complete morphological and topographical analysis can be found in Paper VII.

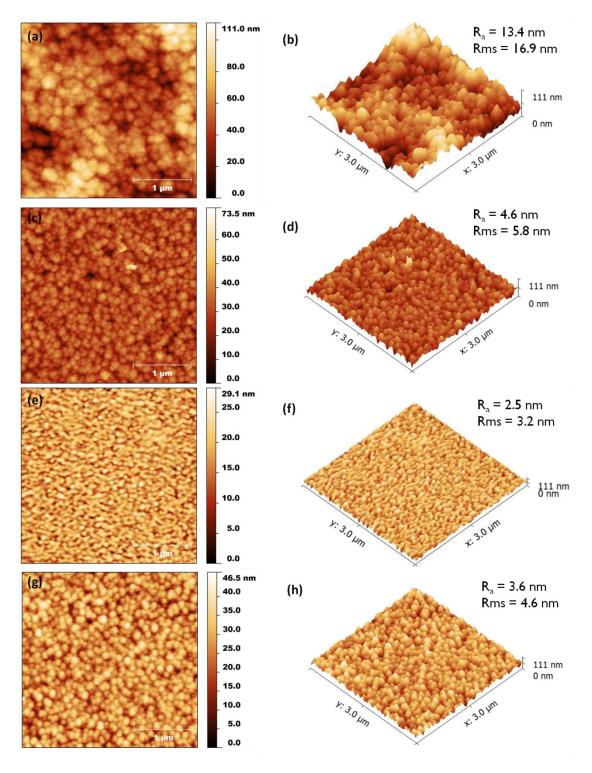


Figure III.2 AFM micrographs showing the topography of (a,b) a-C, (c,d) a-C\_Zr(14), (e,f) a-C:H and (g,h) a-C:H\_Zr(8) films deposited on Si wafers. The 3D in the right column corresponds to the 2D in the left column.

### 2.2 Structure and chemical bonding

The structure and chemical bonding of the coatings has been investigated using XRD, TEM, XPS and Raman. From XRD, pure a-C and a-C:H coatings were characterized as amorphous (Figure III.3); visible peaks were exclusively indexed as interlayer (Ti/TiN). When the coatings were alloyed with Zr, a broad peak typical of a nanocrystalline structure appeared near ZrC(111) and ZrC(200) phase. Figure III.4 shows the a-C\_Zr(14) film diffractogram with the fitted peaks. The fitting procedure enabled us to find the exact position of the ZrC peaks and the calculation of the particle size by applying the Scherrer formula. The crystallite size slightly increased from 1.0 to 2.2 nm for Zr content of 4 and 14 at.%, respectively. A similar trend was observed for hydrogenated coatings: increasing the Zr content increases the crystalline size (1.5 nm for 10 at.%).

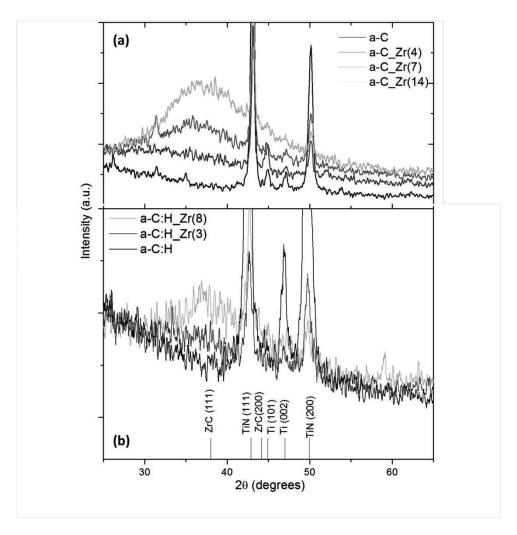


Figure III.3 Grazing-angle X-Ray diffractograms of (a) non-hydrogenated and (b) hydrogenated series of coatings with different Zr content.

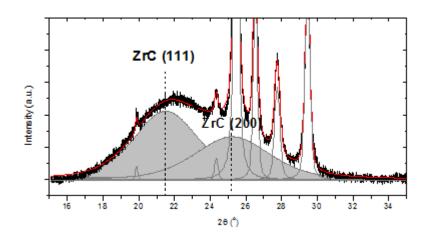


Figure III.4 Grazing-angle X-ray diffractogram of the coating a-C\_Zr(14). The measured data are displayed as black line and the fitted curves as red line.

XPS of C Is region (Figure III.5) allowed a clearl identification of C–C bonds (from the amorphous carbon phase). When Zr was incorporated in the C-matrix, a new peak appeared and grew at ~283.5 eV. Thus, it was expected that the interaction and/or bonding between zirconium and C would be different from the typical Zr-C bond of bulk ZrC (282.3 eV) [183]. A peak between typical C-Me and C-C bonds has been reported by many authors for nc-MeC/a-C systems (Me = Zr, Ti and Nb) [184-189] and it is suggested to originate from interfacial state at the carbide and amorphous matrix interface (Me-C\*). The presence of nanocrystallites can enhance charge transfer between the positive metal element and the C-matrix [128, 189] leading to an electron deficiency and, consequently, weakening the Me-C bond [188]. Moreover, since such interface region was predicted to be around I nm [187], most of the ZrC bonds in our case are expected to show the typical positive binding energy shifts in the core levels of both C1s and Zr3d (not shown), characteristic of small ZrC nanoparticles (C-Zr\*) [189, 190]. A detailed analysis of XPS data is given in Paper VII.

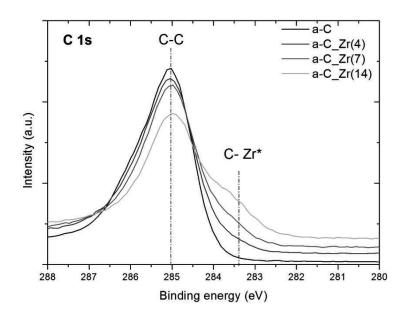


Figure III.5 XPS C1s spectra from the non-hydrogenated series of coatings with different Zr content.

TEM BF investigations were performed on selected coatings to support the results of XRD and XPS analyses. Figure III.6a highlights embedded ZrC nc-phase nanocrystals in a-C-Zr(14) film (blue line). Analysis by fast Fourier transform (FFT) (Figure III.6b) and diffraction images showed that the fringes corresponded to the lattice planes of the ZrC phase. The nc-ZrC crystals were randomly distributed in the C-matrix with a particle size of approximately 2 nm, as predicted by XRD. For lower Zr content (4 at.%), the nc-phase identification in TEM was very difficult due to the smaller grain size (~1nm). The matrix separation widths were estimated by using the Mathematica software to model the particle distribution taking into consideration the nanoparticle volume fraction (V<sub>F</sub>) and the ZrC nanoparticles size for a unit cell of 20 x 20 x 5 nm<sup>3</sup> (Paper VII). In all cases the matrix separation widths were high enough (>> particle size) to effectively isolate the nanoparticles and to limit the grain growth via diffusion and coalescence [185].

Zr-containing films could be then structurally characterized as nanocomposite structures consisting of ZrC nanoparticles embedded in an amorphous C-matrix.

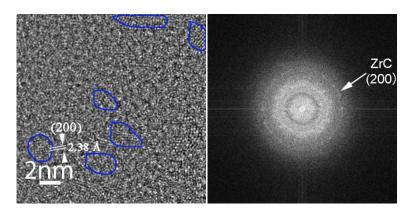


Figure III.6 (a) TEM BF images of the a-C\_Zr(14) film and (b) FFT diffraction fringes with phase identification.

Raman spectra were used to characterize the amorphous carbon matrix (Paper II). Figure III.7 shows  $I_D/I_G$  ratio and the G peak position as a function of Zr content. The presence of Zr and H played an important role in the formation of aromatic rings and, in the organization and size of the C-clusters, typically demonstrated by the increase of D peak intensity (i.e.  $I_D/I_G$  ratio) either when comparing films with and without H or for increasing Zr content. Unexpectedly, the trend of the shift of G peak position for higher wavenumbers (also typical of graphitization), which should follow  $I_D/I_G$  ratio, was not systematically observed. The interface bonding induced by the presence of nc-ZrC phase may influence the vibration modes of sp<sup>2</sup> C=C binding modes and the graphitization of the C-matrix [177, 191] can be responsible for that anomalous behaviour for higher Zr content.

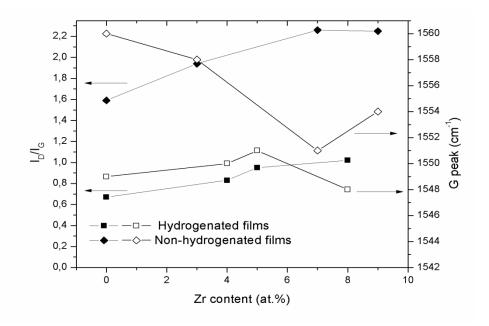


Figure III.7  $I_D/I_G$  ratio and G peak position as a function of Zr content.

## 2.3 Mechanical properties of the coatings

The properties of nanocomposites are largely controlled by their microstructure. In this thesis, the main differences were associated with the amount of Zr in the a-C matrix and, consequently, the content of nc-ZrC phase. Residual stress, hardness and Young modulus have been reported in Paper II and VII. Table III.2 shows the mechanical properties of the deposited coatings.

Samples	σ <sub>rs</sub> [GPa]	<b>H</b> [GPa]	<b>E</b> [GPa]	H/E	<b>H³/E</b> ² [GPa]
a-C_Zr(14)	0.3±0.06	11.9±0.5	137±2	0.09	0.09
a-C_Zr(7)	1.0±0.6	11.7±0.4	126±1	0.09	0.10
a-C_Zr(4)	1.6±0.3	10.7±0.9	4± 7	0.09	0.10
a-C	1.9±0.6	10.9±0.3	105±10	0.10	0.12
a-C:H_Zr(10)	1.4±0.4	12.7±0.6	120±1	0.11	0.14
a-C:H_Zr(8)	1.9±0.3	12.2±0.5	114±3	0.11	0.14
a-C:H_Zr(3)	1.9±0.05	10.0±2	87±3	0.12	0.13
a-C:H	2.5±0.8	8.9±0.3	70±1	0.13	0.14

Table III.2 Mechanical properties of the coatings.

Generally speaking, hydrogenated films exhibited higher compressive residual stress, higher hardness and better elastic behaviour than non-hydrogenated films. The incorporation of H is known to induce modification of the amorphous carbon matrix (decrease in the  $I_D/I_G$  ratio), which suggests a less graphitic matrix. Furthermore, the glassy microstructure made the relaxation of the films harder. Both factors contributed to a higher compressive residual stresses in hydrogenated films. On the other hand, alloying with Zr allowed the relaxation of the films. The presence of nc-ZrC helped breaking the C-C bonds (more graphite-like) and, consequently, facilitated the relaxation and decreased the stress. The hardness and, especially, the Young modulus varied as a function of composition. Globally, for both hydrogenated and nonhydrogenated coatings, the hardness increased as a function of Zr content; this trend was more pronounced for hydrogenated films. In fact, the hardness of nanocomposite films is typically influenced by size, orientation and shape of the embedded nanograins [129]. Since small nc-ZrC (< 3 nm) were dispersed in a thick C-matrix (see Section 2.2), hardness was only slightly different when compared to pure films [128, 129, 192]. On the other hand, the presence of H in the gas phase and the difference in the degree of ion bombardment during deposition may contribute to the observed differences in hydrogenated coatings: the higher the hydrogen/carbon ratio, the lower the hardness (pure a-C:H film exhibits the highest H content, see Table III.1) [126, 127]. A growing trend of the Young modulus could also be observed with the increase of Zr content. The embedded nc-ZrC phase, which has a higher Young modulus if compared to the a-C matrix [193], influenced the elastic properties of the whole system making it more rigid (Paper II and VII). Other important parameters for hard coatings performance is H/E and  $H^3/E^2$  ratios, which are closely related to the wear resistance [194]. Since the hardness was almost independent from the Zr content, both ratios have slightly decreased with the increase of Zr content. The presence of hydrogen resulted in a marked increase of H/E and  $H^3/E^2$ . As expected, the lack of columnar boundaries (i.e. glassy microstructure, see section 2.1) together with the modified nature of carbon matrix by hydrogen led to slightly higher hardness values. The wear resistance is then expected to be higher for hydrogenated films.

The cohesion/adhesion of the coatings was assessed by determination of the critical load (Lc) through scratch-testing. Since the adhesion of the coating is strongly influenced by the substrate mechanical properties, the coatings were first deposited onto hardened steel (M2) in order to effectively evaluate the influence of Zr and H. As a global trend, Zr-alloying improved the critical load values, in particular  $Lc_2$  (first observed spallation/flaking), which is related to adhesion failures (see Figure III.8). The observed decrease in the residual stresses and the improvement of the density of the columnar microstructure with the addition of Zr were the main factors to influence the increase of Lc values.

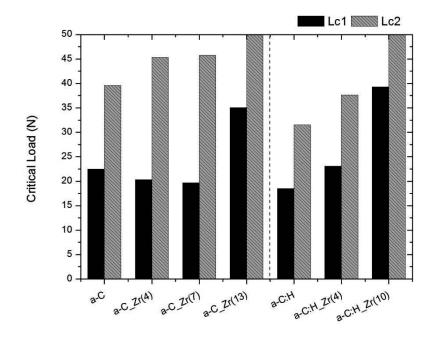


Figure III.8 Critical load of non-hydrogenated and hydrogenated coatings deposited on steel substrates.

As seen in Chapter II, the deposition of DLC on soft substrates is a continuous challenge. Once Ti-based substrates were chosen as final substrates, Zr-containing and pure films were also deposited on such substrates. Table III.3 shows the results achieved for scratch (critical loads:  $Lc_1$ ,  $Lc_2$  and  $Lc_3$ ) and Rockwell-C adhesion tests (HFI-HF6) performed on CP-Ti and Ti6Al4V substrates. The adhesion properties are explained in detail in Paper I and II.

Sample	Sc	Ti6Al4V Scratch test [N]		R-C	Sc	R-C		
Gampie	Lc	Lc <sub>2</sub>	Lc₃	test	Lc	Lc <sub>2</sub>	Lc <sup>3</sup>	test
a-C	7.0±2	12.7±2	35.8±16	HF3	4.5±0.2	7.9±3	33.8±3	HF3
a-C_Zr(4)	6.2±0.7	3. ±	36.3±5	HF3	4.4±0.4	8.0±2	35.5±3	HF3
a-C_Zr(7)	7.5±0.7	10.7±1	> 50	HF3	4.4±0.4	9.3±0.9	> 50	HF2
a-C:H	5.6±2	12.1±3	>50	HF3	4.8±2	6.3±3	12.0±6	HF5
a-C:H_Zr(3)	7.9±0.5	10.9±1	38.7±2	HF3	4.4±1	8.4±0.08	38.9±8	HF3
a-C:H_Zr(5)	6.7±0.7	12.2±2	37.8±0.6	HF3	4.6±0.2	10.2±0.07	35.7±1	HF3
a-C:H_Zr(8)	6.3±0.6	12.8±1	> 50	HF3	4.8±0.5	5.7±0.6	4.4±	HF5

Table III.3 Adhesive properties of the coatings on Ti6Al4V and CP-Ti.

Although no significant differences in adhesion were observed between Zralloyed and a-C(:H) films, the adhesion strength was substrate dependent. When deposited on Ti6Al4V, the coatings revealed higher Lc's than when deposited on CP-Ti. Nevertheless, the critical loads were always lower when compared to those on steel (see Figure III.8) showing the underlying substrate at relatively low loads (Lc<sub>3</sub> value). In fact, Ti6Al4V is 2 times harder than CP-Ti but softer than M2 steel (see Chapter II, Section 3.1). Plastic deformation was then higher on CP-Ti than on Ti6Al4V being the main reason for early failure of the coatings. Nevertheless, the elastic properties of the Ti-based substrates were strongly enhanced by the deposition of a protective coating layer delaying the onset plastic yielding and the subsequent failure *in service* (Paper II).

In order to estimate the capability of the coating to follow substrate plastic deformation, R-C indentation tests were performed (see Table III.3). Again, for Ti6Al4V there was no significant difference between coatings, with all of them showing acceptable failure (HF2-HF3). However, when deposited on CP-Ti, some differences were observed on hydrogenated coatings: a-C:H and a-C:H\_Zr(8) exhibit large-scale delamination at the indentation edges (unacceptable failure – HF5). Even though hydrogenated coatings exhibit higher elastic strain to failure (H/E ratio) and higher plastic deformation resistance ( $H^3/E^2$  ratio) than non-hydrogenated films, they present high residual stress.

Moreover, high pile-up could lead to higher wear of the counterbody; thus the use of CP-Ti is expected to provide inferior tribological properties and to increase the risk of counterbody damage during sliding tests as demonstrated in the next section.

The effect of the substrate hardness on the DLC coating performance was further highlighted through a collaborative project with the University of Southampton. By changing the mechanical properties of the substrate with surface pre-treatments the cohesion/adhesion could be strongly enhanced [195]. High pressure torsion (HPT)<sup>3</sup> was applied to CP-Ti substrates; the microstructure was refined from coarse-grained to ultrafine-grained (UFG) and, consequently, the hardness of the substrate was increased. In general, from these tests it could be concluded that the coatings deposited on either HPT-processed CP-Ti and Ti6Al4V substrates showed similar scratch-test and indentation behaviour (Paper III).

### 2.4 Tribological properties

The tribological properties of the coatings were primarily assessed by pin-on-disc experiments against Ti6Al4V balls. The tests were conducted in two different environments: ambient air (RH = 30%) and physiological solution (PS). Testing in PS aims not only to predict the wear behavior but also to evaluate the synergistic effect of wear and corrosion in the presence of water and salt ions, that can both contribute for the degradation of the coatings [116, 197]. Figure III.9 and III.10 shows the friction coefficient and wear rates for hydrogenated and non-hydrogenated coatings deposited on CP-Ti and Ti6Al4V substrates in ambient air and PS, respectively. The values for uncoated substrates are also shown for comparison. Paper II and IV described in detail the tribological properties of the coating in ambient air and PS, respectively.

Under both testing conditions, the coefficient of friction was found to be independent of the substrate material. In general, coated substrates presented better tribological properties when compared to uncoated ones. In ambient air, pure non-hydrogenated films showed the lowest COF (~0.10), which slightly increased when Zr was added. On the other hand, hydrogenated coatings did not show any significant difference in COF. The same behaviour was observed when tested in PS, i.e. no significant difference between coatings (COF ~0.13). Moreover, the wear of the coatings was also similar, except for a-C\_Zr(9) on CP-Ti; nevertheless, it was three orders of

<sup>&</sup>lt;sup>3</sup> *High pressure torsion (HPT)* is a method used in metal processing to tailor the microstructure and properties. The metal is subject to a compressive force and concurrent torsional straining which causes severe plastic deformation without changing the overall dimensions of the material [196]

magnitude lower than that of uncoated samples. The wear of the counterbody decreased with the incorporation of Zr into the C(:H)-matrix.

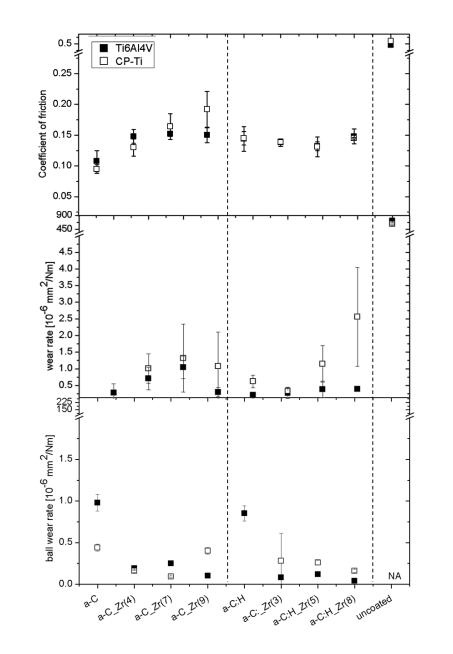


Figure III.9 Tribological properties of non-hydrogenated and hydrogenated coatings deposited on Ti grade 2 and 5, together with the uncoated values in ambient air. (NA stands for not analysed)

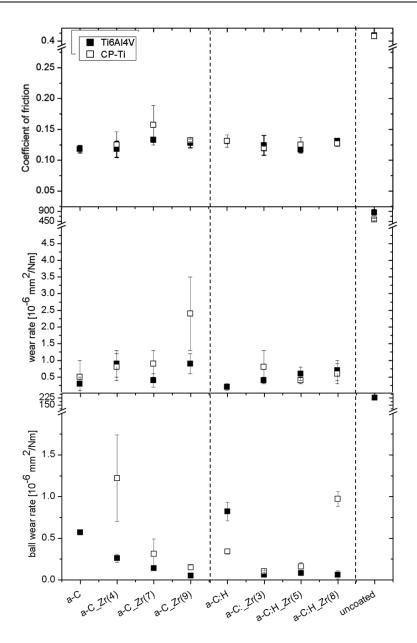


Figure III.10 Tribological properties of non-hydrogenated and hydrogenated coatings deposited on Ti grade 2 and 5, together with the uncoated values in PS.

Although the tribological properties (friction coefficient, wear rate, etc.) were only slightly dependent on the substrate, the wear track was found to be significantly different for the same film deposited on both Ti-based substrates. The use of a softer substrate (CP-Ti) led to a higher plastic deformation inducing large strains in the coating; its insufficient elastic deformation caused its fracture. In case of cyclic loading (pin-ondisc) fatigue cracks and/or blisters can occur, especially under corrosive media [198-200]. In fact, the highest deformation was observed for a-C\_Zr(9) (see Figure III.11); such severe deformation could not be accommodated by the coating and cracks parallel to the sliding direction were observed. Besides, the presence of such cracks led to local delamination which might also promote the wear of counterbody. Nonetheless, when using a load carrying capacity substrate (Ti6Al4V), Zr-containing films presented good wear-corrosion behaviour and, in general, lower ball wear rate (see Figure III.9 and III.10). Plastic deformation and its influence on the wear of the coatings is described in Paper II.

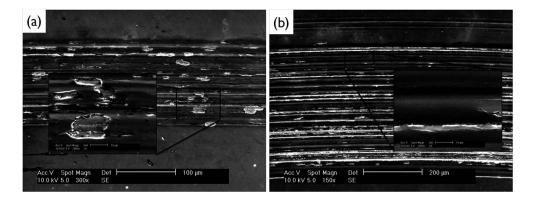


Figure III.11 SEM micrographic pictures from a-C\_Zr(9) wear track when deposited on Ti grade 2: (a) ambient air and (b) PS.

When sliding under non-lubricated conditions, DLC films are known to form a graphitic-like tribolayer that works as a solid lubricant. Friction and wear of the coating and counterbody are then dependent on the stability and thickness of such layer [66]. Figure III.12 shows the Raman spectra acquired on the ball wear scars after sliding against a-C and a-C\_Zr(9) deposited on Ti6Al4V in room conditions. Even though a-C\_Zr(9) wear track did not show any significant structural difference before and after sliding (no graphitization), the ball shows shallow scratches covered by a homogeneous and well-developed graphitic tribolayer (see typical G and D peaks from graphite featuring in the Raman spectrum, Figure III.12). On the contrary, when sliding against pure a-C films such layer could not be identified. The presence of nc-ZrC phase increased carbon structure disorder (see Section 2.2), which may help to develop a stable carbonaceous tribolayer that protected the counterbody from wear. An exhaustive study on the effect of the tribolayer formation under different environments of Zr-containing coatings was developed through a collaborative project with Czech Technical University; for more details see Paper V.

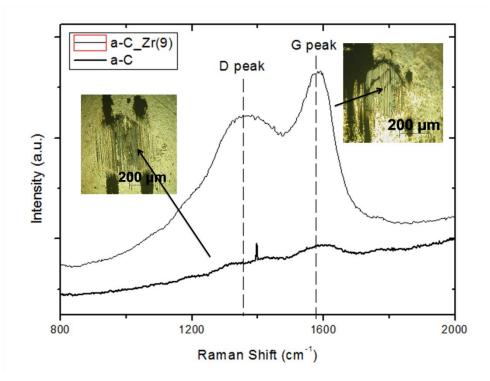


Figure III.12 Raman spectra and micrographs of the ball wear scar that slid against a-C and a-C\_Zr(9) films. The arrows show the acquisition zone.

In physiological solution, the formation of a protective transfer layer is typically suppressed [66]. The friction and wear behaviour is then influenced by the properties of the coating surface and the interaction with the environment. It is important to notice that when Zr was added to the C-matrix, the water contact angle increased. Next Chapter is dedicated to the physicochemical properties of the coatings. Nevertheless, lower surface energy is typically related with lower wear rates due to the decrease of the direct solid contacts between surfaces [201-203]. Moreover, it is expected that the interaction with H<sub>2</sub>O molecules will be similar with that of moisture (C-H, C-O and C=OOH bonds).

## 3. Conclusions

The formation of nanocomposite structures consisting in ZrC nanoparticles embedded in an a-C(:H) matrix was achieved when incorporating Zr in contents as low as 3 at.%. The grain size was estimated to be around 1-2.2 nm as a function of the Zr content. Compared to Zr-free, nanocomposite films were denser and smoother. The presence of nc-ZrC relaxed residual stresses in the films both by inducing modification of the C-matrix (graphite-like) and by changing the morphology of the coatings. When deposited in a reactive atmosphere, the films revealed a glassy/compact morphology and exhibited higher compressive residual stresses. In general, increasing Zr content led to higher elastic modulus, lower residual stress and better higher hardness, cohesion/adhesion to the substrate. However, an inverse trend was registered for H/E and  $H^3/E^2$  ratios. This featu re was particularly important when using low load carrying capacity substrates such as Ti-based materials: higher Zr content (more nc-ZrC dispersed in the C-matrix) lowered the adhesion and the tribological properties of the coatings deposited on softer CP-Ti substrates. Substrate plastic deformation resulted in the coatings fracture, cracking and/or blistering, especially when sliding in a corrosive medium. Under these circumstances, CP-Ti should be discarded as a potential substrate since it can increase the risk of either coating degradation or extensive counterbody wear. On the other hand, Ti6Al4V coated with Zr-containing films presented good wear-corrosion behaviour strongly reducing the counterbody wear when if compared with pure films.

# **CHAPTER IV**

The core objective of this chapter is the study of the surface interaction between a- $C:(H)_Zr$  films and biological fluids. The surface chemistry was assessed by means of surface energy and protein adsorption. The tribological properties under BS lubrication are also discussed with the focus on the role of albumin in the sliding process.

The main results of this chapter were presented in **Paper IV**.

## I. Introduction

In biomedical field, mechanical and tribological properties are important to assess the potential of materials for their application. Nevertheless, the key parameter is biocompatibility. The prediction of materials biocompatibility and their applicability is hard but not entirely impossible, being strongly dependent on the energy, morphology and chemistry of the biomedical surfaces [204]. Real surfaces are characterized by a certain degree of roughness and/or of chemical heterogeneity, thus inducing deviations from an ideal surface [205]. For instance, smooth surfaces (Ra < I  $\mu$ m) are known to limit the thrombogenicity [206] and bacterial adhesion [207] of the biomaterials related to cells flattening out, which prevents their nutrition [208]. The exact interaction between biomaterials and biological fluids is still under extensive study [209]. After implantation, the adsorption of protein is of great importance in understanding the interaction between the material and the human body, since it is one of the first events occurring after implantation and it drives the entire biological response to the implanted material [210-212].

In this Chapter, the physicochemical properties of the coating surface are described. Wettability measurements were used to evaluate the affinity of the coatings with water. Protein-adsorption studies were performed by focusing on albumin adsorption, since it is the most abundant constituent in synovial fluid and it has been recognized as critical for the lubrication mechanism [213,214]. Finally, the tribological behaviour of the coatings was also assessed in the presence of foetal bovine serum (BSA). The tests were performed by using pin-on-disc tests to investigate the wear mechanism, friction, and lubrication under protein containing solutions. The results shown here summarize the results presented in Paper IV.

## 2. Surface Chemistry – Physicochemical Properties

Immediately after implantation, water and ions are the first adsorbing on the implant surface, since water is the predominant substance in all tissues and fluids in the human body (50-80%) [1]. Thus, the way surfaces interact with water can be considered as one of the most important properties of the surface when it comes to predict its behaviour under biological environments. In fact, numerous physiological events at subcellular and cellular levels, such as cell adhesion and protein adsorption, are greatly affected by such property. Wettability is modulated by surface characteristics, such as

chemistry and surface topography [215], and it can be expressed in two different ways: (i) a less wettable surface (hydrophobic surface) which implies high contact angle values; and (ii) on the contrary, low contact angles values present a more wettable surface (hydrophilic surface) [216]. To study such property, water contact angles measurements were performed. Since surface roughness can strongly influence the results, the coatings were deposited on glass substrates. Moreover, the base radius of the drop was chosen bearing in mind that it had to be large enough in proportion to the scale of the sample roughness (larger than three orders of magnitude) to ensure accurate contact angle measurements [205]. Table IV.1 shows the roughness, the water contact angle and the calculated surface energy of the coatings. For non-hydrogenated coatings, the contact angle increased inversely with the surface roughness. However, hydrogenated coatings follow an inverse trend. Typically, the contact angle increases with decreasing surface roughness [217-219]; therefore, little or no correlation between roughness and surface wettability was found in this study. The surface topography cannot then be considered alone to explain the wettability behaviour of the coatings.

Table IV.1 Surface roughness ( $R_a$  and Rms), water contact angle ( $\theta$ ) and surface energy ( $\gamma_s$ ) of the coatings.

Samples	Roughness [nm]		$\Theta_{water}$	Surface energy [mJ.m <sup>-2</sup> ]			
	$R_{a}$	Rms	ĹĴ	$\gamma_s^P$	$\gamma_s^d$	γs	
a-C_Zr(14)	4.6	5.8	71±2	4.3	39.1	43.4	
a-C_Zr(7)	4.5	6.0	67±1	5.0	40.5	45.5	
a-C_Zr(4)	5.8	7.3	66±3	5.I	45.8	50.9	
a-C	13.4	16.9	50±7	11.2	51.2	62.3	
a-C:H_Zr(8)	6.4	8.0	73±2	3.9	36.7	40.6	
a-C:H_Zr(3)	3.6	4.6	65±2	12.8	27.2	40.0	
a-C:H	2.5	3.2	57±2	14.2	33.9	48. I	

Pure non-hydrogenated and hydrogenated carbon films were characterized as hydrophilic ( $\theta < 65^{\circ}$ ). When Zr was incorporated, the contact angle increased as a function of Zr content and became similar to uncoated samples. Air stored Ti-based substrates have typically water contact angles around 70° due to the naturally formed passive TiO<sub>2</sub> layer that is responsible for the good biological performance of Ti implants [205, 215]. The exact water-interaction mechanism of alloyed DLC coatings is still not clear and further investigation is needed. Nevertheless, similarly to the well-known influence of hydrogen on the decrease of unsaturated bonds C=C (compare pure a-C and a-C:H), the metallic element can further decrease the dipolar interaction with water, by decreasing the polar component and additionally the surface energy [220].

Protein adsorption is strongly influenced by surface chemistry, namely by surface wettability. For instance, albumin has three homologous domains (comparable amino acid sequences) assembled in a heart-shape structure which are sustained by hydrophobic interactions, hydrogen bonds and disulphide bridges [210, 221]. Once the adsorption is initiated, proteins tend to maximize the surface interaction by exposing either their hydrophobic domain (typically hidden toward the interior) or hydrophilic domains, depending on the surface wettability. Nevertheless, hydrophobic surfaces enhance a protein-surface interaction, when compared to hydrophilic due to the fact that they promote hydrophobic interactions. Actually, the adsorption on hydrophilic surfaces is thermodynamically unfavourable [216, 222]. On the other hand, proteins can undergo reversible/irreversible conformational changes when adsorbed on hydrophobic surfaces that may lead to unfavourable cell response if the proper binding domains are disrupted [204, 211]. Figure IV.1 shows the estimated amount of proteins adsorbed in relation to surface contact angles. As expected, Zr-containing coatings (hydrophobic surfaces) adsorbed more than pure films (hydrophilic surfaces). In fact, the incorporation of small Zr content (4 at.%) was able to increase 10 times the adsorption ratio of the films if compared to pure carbon films.

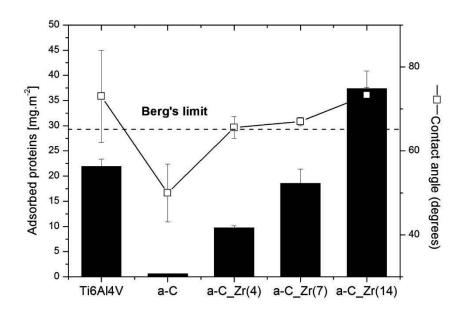


Figure IV.1 BSA adsorption vs contact angle measurements. The dash line represents the Berg's limit<sup>4</sup>  $(\theta = 65^{\circ})$ .

<sup>&</sup>lt;sup>4</sup> Berg's limit stands to the minimum detectable hydrophobic forces suggested by Berg et al. [216, 223].

Taking the albumin size, its molecular weight [224] and the surface area into account, there should be a maximum protein concentration allowed to be adsorbed; it is approximately 4 mg.m<sup>-2</sup> for monolayer adsorption (side-on and/or end-on) [225, 226]. However, it has been reported that multilayer adsorption is likely to occur on metallic substrates, especially for high concentration solutions [213, 225, 227]. Zr-containing samples revealed to have multilayer adsorption since the maximum amount for monolayer adsorption was highly exceeded. Moreover, after rinsing with water to remove any non-chemically bonded (chemisorbed) proteins on the surface, the amount of protein removed from the a-C surfaces was around 46% whereas no evidence of desorbed proteins was detected for Zr-alloyed samples. This may be attributed to the different conformation of the adsorbed layer that influences protein-protein and/or protein-surface interaction [213, 227]. After all, Zr-containing coatings (hydrophobic surfaces) showed not only a higher amount of adsorbed protein but also that they were more tightly bound than on a-C films.

In order to prove the existence of such adsorbed layer, XPS analyses were performed on the surfaces of the samples. Figure IV.2 shows the XPS survey spectra for uncoated substrate (Ti6Al4V) and non-hydrogenated coatings after immersion in BSA. The first main difference observed, when compared to as-deposited spectrum (Figure IV.2f), was the appearance of a well defined N Is peak (N Is 400 eV), usually assigned to the amino-acid of the protein [228, 229]. For the Zr-containing samples the intensity of Zr 3d peak decreased after immersion, meaning that the adsorbed protein layer is rather discontinued, leaving bare areas of the film/substrate surface. In fact, many authors [213, 227, 230] have reported that protein monomers preferably aggregate to surface clusters if compared to adsorbed on empty surfaces sites forming an island-like structure, whose thickness corresponds to several layers. Moreover, the S weak band can only belong to S-containing amino acids, i.e., methionine (Met) and cysteine (Cys) [231], which corroborates the presence of such layer (see the inset of Figure IV.2). Although XPS did not provide quantitative information about the total amount of adsorbed proteins, it was possible to monitor such tendency by analysing the N Is core level before and after immersion: the higher the adsorbed ratio, the higher the intensity [213]. Figure IV.3 shows the N Is core level spectra for uncoated and coated samples. The high intensity peak of a-C\_Zr(14) corroborated the adsorption results. A complete analysis of the XPS data (deconvolution of the peaks, bonding identification, etc.) is given in Paper IV.

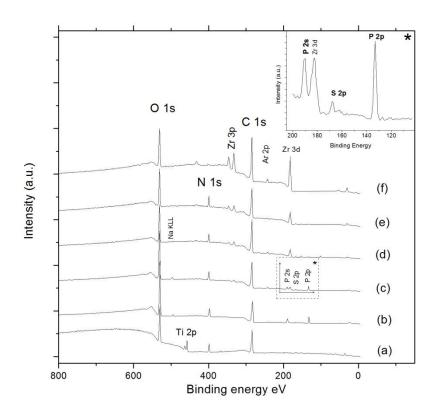


Figure IV.2 XPS spectra of the uncoated and coatings after immersion in BSA for 24h: (a) Ti6Al4V, (b) a-C, (c) a-C\_Zr(4), (d) a-C\_Zr(7) and (e) a-C\_Zr(14). a-C\_Zr(14) before immersion was also add for comparison (f). The inset (\*) shows a zoom in of the 200-400 eV region of the a-C\_Zr(4) film XPS spectra revealing the S 2p photo-peak.

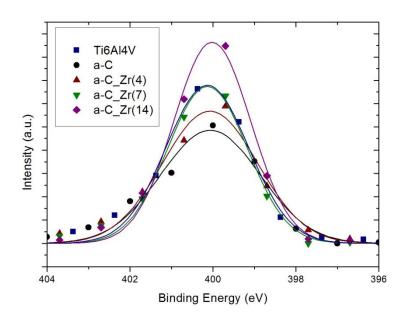


Figure IV.3 N Is core level spectra for coated and uncoated samples after immersion.

#### 3. Tribological properties under BSA

The natural lubricant present in the contact after arthroplasty is a complex mixture containing proteins. However, the research studies concerned with natural lubrication have focused on single-component fluid rather than on mixtures as the synovial fluid, mostly for reasons of availability and convenience. As albumin is by far the most abundant protein and readily adsorbs onto artificial materials, it has been widely considered the most important protein in lubrication studies. Thus, to identify a promising coating composition, pin-on-disc friction tests were performed under protein-containing solution (BSA). Figure IV.4 shows the COF and the ball wear rate of coated and uncoated samples. The presence of proteins in the lubricant protected the coatings from wear, since in all cases the worn surface did not show any measurable wear. Moreover, sliding against Zr-alloyed films resulted in a decrease of the wear of the Ti-alloy ball. On the other hand, despite the mechanical and structural differences all coated samples presented similar COF (~0.16).

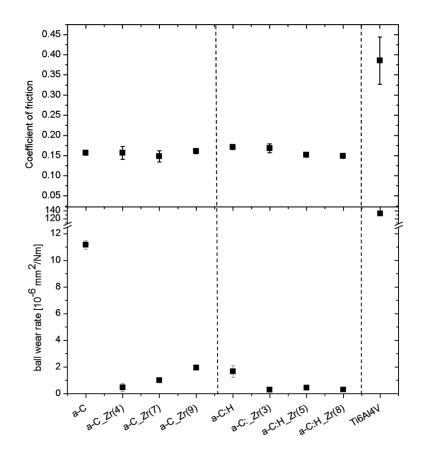


Figure IV.4 Average friction coefficient and ball wear rate of the coated and Ti6Al4V uncoated samples under BSA lubrication.

Upon adsorption, albumin created a layer more or less thick depending on the surface wettability, as mentioned in Section 2. Once adsorbed on hydrophobic surfaces, albumin typically restructures its conformation in order to maximize the inter protein and protein-surface interactions. On the other hand, on hydrophilic surfaces albumin maintains its native conformation being less tightly bound to the surface [227]. The interaction of protein with metallic surfaces is then complex and may play antagonist roles: acting as a lubricant [232] but also interfering with the sliding of the pin and providing resistive forces that lead to higher friction [201]. Thus, the ability to adsorb protein on the surface is of fundamental importance in the wear process: hydrophobic surfaces increase the protein adsorption that restrains the direct contact between mating materials and, consequently, decreases the counterbody wear. Figure IV.5 presents a schematic representation of the general wear process.

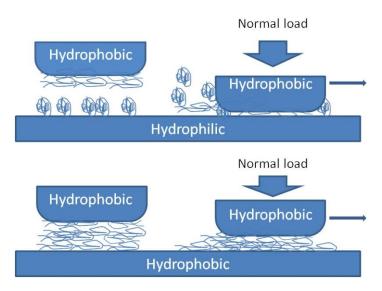


Figure IV.5 Schematic representation of the key mechanism of albumin-mediated lubrication on hydrophilic and hydrophobic DLC films.

Although this may be true, it is also important to notice that the wear of the ball was always higher when compared to tests in ambient air or PS, especially against Zr-free films (see Chapter III, section 2.3). Typically, protein adsorption increases the wear resistance of metallic alloys (ball) [233] but if the top surface of the material is rubbed and the natural passive film is destroyed, albumin plays an important role affecting the tribocorrosion behaviour of the alloy [234]. In fact,  $TiO_2$  produces two types of surface hydroxyl groups when interacting with water: one type derived from a protonated base and another type derived from the acidic hydroxyl groups [17]. When sliding in PS, the anions bind to the titanium ions in exchange reactions with those basic hydroxyl groups,

thus forming a new passive layer that protects the surface from rubbing. When proteins are added to the lubricant such process is delayed [234] and it depends on the stability of both the passive and the protein layer [235]. In addition, if a very thin protein layer is formed (such as on a-C films), direct contact between mating materials can still occur, further increasing the wear of the counterpart. The same behaviour was observed for uncoated samples; sliding in protein containing solutions resulted in slightly lower surface wear (5.7 x  $10^{-4}$  mm<sup>3</sup>/Nm) and counterbody wear (1.3x $10^{-4}$  mm<sup>3</sup>/Nm) compared to sliding in PS (7.2 x  $10^{-4}$  mm<sup>3</sup>/Nm and 2.2 x  $10^{-4}$  mm<sup>3</sup>/Nm, respectively, see Chapter III, section 2.3). It further confirms the protective effect of the protein layer. A closer look at Figure IV.4 bears it evident that, despite the reduction of wear of the counterbody when sliding against Zr-containing films if compared to pure films, the wear tendentiously increased with the increase of albumin absorption ratio. Such behaviour can be related to the mechanical properties of the coatings. It is expected that the local contact pressure of Zr-containing films is higher than Zr-free films, since they have higher Young modulus (see Chapter III). The protein adherent layer would be then easier squeezed out and contact between mating surfaces would occur [232]. Many authors [236-238] also refer to the thermal and shear induced denaturation and precipitation of proteins under sliding conditions that can change the rheological properties of the lubricant and, consequently, influence the lubrication properties and the wear mechanism. Nevertheless, this phenomenon is usually associated with a higher friction, which was not observed in this study (see Figure IV.4).

## 4. Conclusions

The physicochemical properties of the coatings were screened by evaluating the water- and protein-surface interactions. The incorporation of Zr decreased the surface energy (hydrophobic character), enhancing the albumin adsorption if compared to pure a-C(:H) films. XPS confirmed the presence of such adsorbed layer onto the surfaces and the adsorption tendency: the higher the Zr content, the higher the contact angle, and the higher the protein adsorption. Moreover, under sliding, the ability to effectively adsorb proteins was fundamental not only for protecting the coating from wear but also for protecting the surfaces from direct contact.

In general, Zr-containing films presented better tribological performance when compared to pure ones, since the wear of the counterbody decreased. Nevertheless, higher Zr content resulted in higher wear of the ball, even with higher protein adsorption rate. A synergetic effect is then expected between mechanical and chemical properties that influence the overall performance of the coating under biological conditions. Despite the excellent protein adsorption of coatings with high Zr content, the coatings that best fitted for further testing were the ones with lower Zr content. a- $C_Zr(4)$  was then selected for further testing as representative of the best Zr-alloyed films.

## **CHAPTER V**

This chapter describes the sliding behaviour of two selected coatings (pure a-C and a- $C_Zr(4)$ ) when tested against UHMWPE and PEEK. The tribological tests were carried out in a multidirectional pin-on-disc apparatus that replicated the typical cross-shear motion typical of endoprosthesis.

Paper VI fully reviews the results exposed herein.

## I. Introduction

One of the most commonly used material combinations in arthroplasty is UHMWPE vs metal (MoP). Despite the good mechanical and tribological properties of this tribopair, there is strong concern related to the production of wear debris in vivo, which eventually leads to the failure of the prosthesis. For instance, 10% of the knee implants failures were directly related with UHMWPE wear [57]. As a result, most studies on artificial joint materials concentrate on the wear analyses and the wearparticle characterization, especially when trying to implement new material combinations. This can be assessed in a first attempt by "bench tests" (e.g. linear reciprocating motion, ball-cup wear, delamination wear, etc.) and at the final stage by using joint simulators that replicate the real movement of the natural joints. In Chapter IV, the tribological properties of the coatings were evaluated by pin-on-disc measurements with linear reciprocating motion under physiological conditions in order to pre-select a promising set of coatings that could be tested in conditions closer to reality. To rank materials for polymer wear rates under simulated physiological conditions, the testing conditions should be representative of the wear rates and the mechanisms of those in a TJR [239-241]. More specifically, the wear of certain polymers is highly dependent on "crossing shear" sliding conditions. In fact, it was found that unidirectional motion strain-hardened the polyethylene, gradually increased its integrity along the established path and thus resulting in unreliable polymer wear [240]. Thus, the general requirement to evaluate the wear of the polymer is that the polymer surface must change direction in proportion to the counterface.

Unfortunately, our lab does not provide any other facility to effectively evaluate the wear against polymeric materials. Therefore, Professor Markus A. Wimmer, Head of the Section of Tribology at Rush University Medical Center, was contacted in order to use a multidirectional pin-on-disc (AMTI's Ortho-POD<sup>TM</sup>) equipment based in his laboratory. In this experiment, the pin is rubbed against a plate in an elliptical or square waveform to create the cross-shear motion conditions. This collaboration is summarized in Paper VI that describes in detail the wear behaviour of a-C and a-C\_Zr(4) coated Ti alloys against two different polymeric materials (UHMWPE and PEEK).

## 2. Tribological behaviour against UHMWPE and PEEK counterbodies

Contradictory results have been reported in literature when DLC slides against UHMWPE, ranging from strong improvement to drastically increased wear [103]; only one study, to our knowledge, of DLC against PEEK reported a significant increase in the life span of implants using such a combination [242]. Thus, there is enough space to explore such a combination and its use on load-bearing implant. The tribological properties were assessed by using multidirectional pin-on-disc test. In accordance to ASTM F-732 [174], this technique has been considered as the simplest and the most widely accepted model for the basic biotribological simulation of candidate biomaterials. The exact loading and motion patterns are not replicated with such machines though; the multidirectional motion generates "crossing shear" conditions, which are typical for TJR and strongly influence the polymer wear rate [240, 243-245], being closer to real contact conditions. The tests were performed for a maximum of 2 million cycles (Mc), which represents  $\sim 2$  years in service, under protein-containing solution (BS). The square motion pattern was chosen by taking the typical patient locomotion into account, since the trajectory of relative motion between a femoral head and an acetabular cup takes a general quasi elliptical or rectangular/square shape during a patient gait cycle [245, 246]. Moreover, the contact pressure used in the present study was ~1.5 MPa which is lower than values reported in literature (2 to 5 MPa) [59, 240, 243-245]. However, extremely loud squeaking due to high friction could be noticed when using higher contact pressures. Apart from coated and uncoated Ti6Al4V samples, CoCrMo alloy was also used as reference materials with prosthetic-quality surface finishing ( $R_a < 0.05 \mu m$ ) to avoid any influence of an abrasive wear mechanism [174].

Although there is no simple relationship between wear and friction, low friction materials can generally be expected to enhance the performance of an artificial hip; therefore, the friction is an important parameter to consider [247, 248]. Figure V.I shows the average coefficient of friction measured for all testing samples against both polymeric materials. The friction behaviour was strongly dependent on the polymer type. When testing against PEEK, the friction coefficient (COF) was almost 2 times higher than against UHMWPE. In fact, PEEK is known to present a relatively high friction coefficient ~0.3 against metals [243, 249]. The CoCr/polymer combination has always showed lower COF when compared to the other tribopairs. Moreover, statistically (p < 0.003), the metal/UHMWPE pairings presented lower COF when compared to coated samples. On the other hand, there was almost no difference in COF for different surfaces tested against PEEK (p > 0.057).

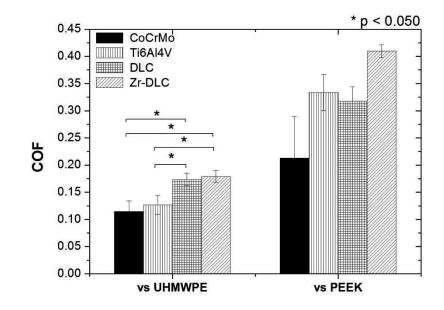


Figure V.I Average COF for all samples against UHMWPE and PEEK. (\*) represents statistically significant differences between samples in the same group (p < 0.05, ANOVA test).

Figure V.2 shows the volumetric wear loss versus the number of cycles for both UHMWPE and PEEK pins. The wear factor, calculated from linear regression analysis of the average of three measurements, is shown in Figure V.2 (see insets). Similarly to friction, wear was strongly dependent on the polymer counterbody. In nearly all cases, the wear was correlated with friction: PEEK wore more than UHMWPE (p < 0.049). The only exception was found for CoCr, where there were no significant differences in wear factor when sliding against PEEK and UHMWPE (p > 0.127). In fact, the wear factor for CoCr/UHMWPE ( $2.6\pm0.1 \times 10^{-6} \text{ mm}^3/\text{Nm}$ ) was at the range of those reported from clinically retrieved TJR couples ( $10^{-6} \text{ mm}^3/\text{Nm}$  [240, 250]), which demonstrates the reproduction of the *in vivo* wear mechanism. Thus, the behaviour of new materials combinations could be assessed by this method in a relative cost-effective way.

It is worth noting that the wear of the metallic alloys and the coatings could not be measured: (i) samples were too heavy for the microbalance and/or weight losses too small for the standard laboratory balance; and /or (ii) the wear track depth and shape could not be effectively detected by profilometry analysis to determine the wear rate. Therefore, Figure V.3 shows the discs surfaces for a better understanding of the wear phenomena. Although no significant differences were observed on the UHMVVPE wear (p > 0.66), the wear mechanism was dependent on the testing surface. Ti6Al4V revealed its poor tribological properties with signs of abrasive wear (grooving) when compared to CoCr (Figure V.3a and 3c). On the other hand, when Ti alloy was coated, no wear was initially observed; the coatings thus successfully protected Ti6Al4V from wear (Figure V.3e). Nevertheless, Zr-containing samples sliding against UHMWPE delaminated after 1.2 million cycles (Mc) (Figure V.3g). When sliding against PEEK, both coatings, a-C and a-C\_Zr(4), delaminated after 1.2 Mc and 0.9 Mc, respectively (Figure V.3f and 3h). The test with uncoated Ti6Al4V was also stopped after 300,000 cycles due to severe surface damage (see Figure V3d). Its poor shear strength led to a detachment of the oxide layer, which then acted as an abrasive [105, 158, 251, 252].

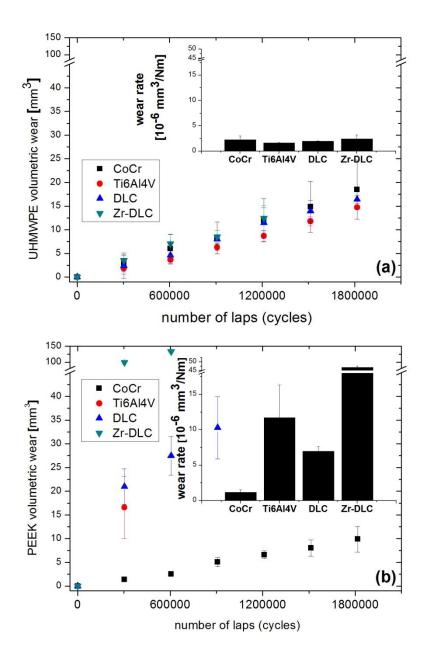


Figure V.2 (a) UHMWPE and (b) PEEK volumetric wear loss and wear rate (in the insets) against the different surfaces.

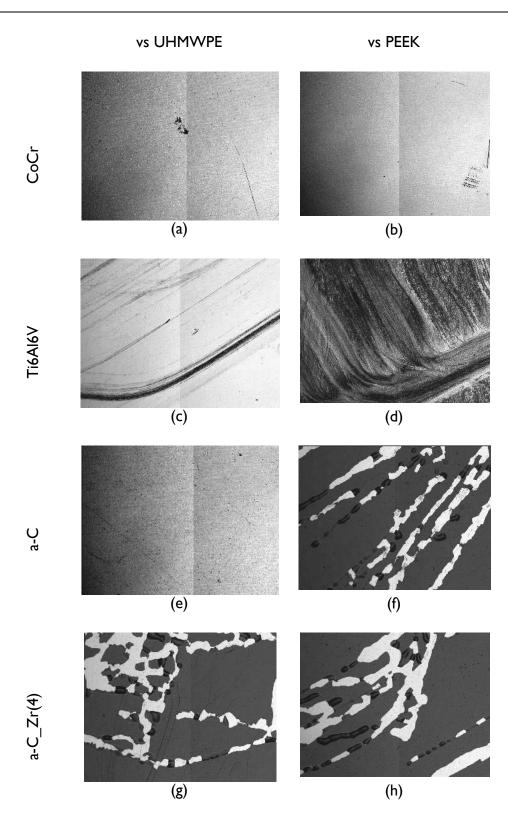


Figure V.3 Images from the wear track of the disc after testing against (a,c,e,g) UHMWPE (a,c,e,g) and (b,d,f,h) PEEK.

As seen in Chapter III, the incorporation of small amounts of Zr led to changes in the coating microstructure and mechanical properties. In fact, Zr-containing films exhibit higher adhesion strength and lower residual stress when compared to a-C pure films, especially for higher Zr content. However, low Zr content presented better tribological properties under BS solution, showing lower counterbody wear (see Chapter IV). Thus, one may think that a-C\_Zr(4)/Polymer would decrease the polymer wear and consequently the *in vivo* wear debris production and release. Instead, the coatings failed catastrophically when tested under multi-directional stress in body fluids.

Surface layer failure is the major concern when using coatings in the biomedical field and has quite frequently been reported *in vivo* and *in vitro* by many authors [103, 106, 253]. A graded interlayer scheme was applied in order to avoid such problem, resulting in a satisfactory composition gradient (see Chapter II). Moreover, such interlayer also resulted in a desired functional gradient of elastic properties at the coating/substrate preventing coating chipping and delamination. Despite the good mechanical adhesion achieved in ambient air, the presence of biological species in the body solution (ions, proteins, etc.) can boost the corrosion of the adhesion-promoting interlayer and consequently lead to the delamination of the coating [254]. In order to evaluate the effect of the body fluids on the adhesion strength of the coatings after testing, R-C indentation test was performed in the wear tracks and, when present, near the delamination zones (Figure V.4). The overall adhesion of the film was maintained (i.e. it was identical to that of the as-deposited samples), even when the film was clearly damaged after testing. Moreover, Raman spectra acquired in the wear tracks did not reveal any significant difference after testing (see Figure V.5).

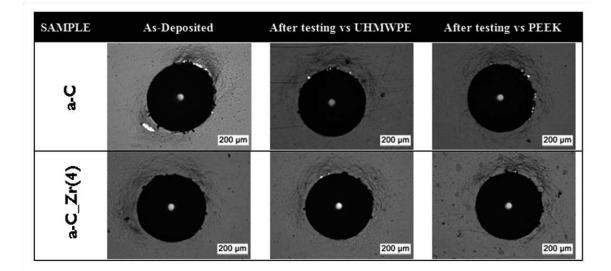


Figure V.4 Optical micrographs of the Rockwell-C indentation of a-C and a-C\_Zr(4) films before and after testing against UHMWPE and PEEK.

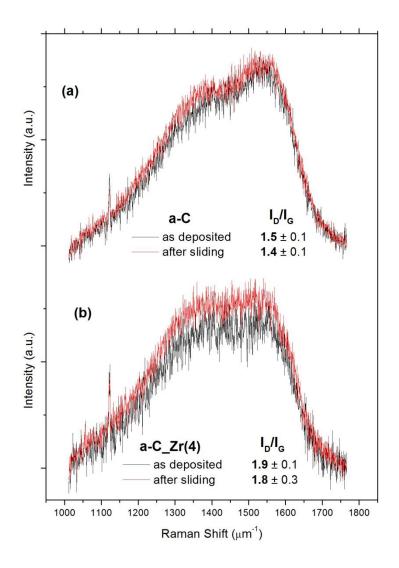


Figure V.5 Raman spectra for (a) a-C and (b) a-C\_Zr(4) acquired before and after sliding against UHMWPE under BS. Since a-C\_Zr(4) delaminate after 1.2 Mc cycles the acquisition was done near the delaminations zones. The I<sub>D</sub>/I<sub>G</sub> ratio is also displayed in the insets.

It has been suggested that the most common corrosion mechanisms hindering the cohesion/adhesion of the coatings in contact with body fluids are pitting, crevice corrosion (CC) and stress-corrosion cracking (SCC) [71]. CC and SCC are related to the slow crack advancement or interlayer dissolution resulting in a delayed delamination. On the other hand, pitting is related to surface defects. Since every coating present some surface defects (pinholes, scratches and/or loose grains) [255], the body fluids can penetrate through the coating and act as a corrosive electrolyte forming blisters that, if they exceed a certain dimension, they can enhance the local delamination [122]. Table V.1 shows the defects density on both coatings. The incorporation of Zr decreased the average surface coverage but the defects are larger than on pure a-C films. Thus, it is possible that under body fluid the presence of pinholes may promote SCC and initiate delamination.

Sample	Area coverage	Average size		
	[%]	[µm]		
a-C	7.6±0.7	4.1±0.3		
a-C_Zr(4)	2.9±0.4	6.8±1.3		

Table V.I Defects density on the coatings.

In order to investigate the possibility of SCC further, we followed a complementary method proposed by Hauert and co-workers [198, 254]: by inducing substrate deformation with Rockwell-C indentation and simultaneous immersion of the sample in a protein-containing solution (Figure V.6); SCC should be promoted at the mechanically loaded interface. However, even after 3 days at 37°C (the equivalent of approximately 300,000 testing cycles) the coatings did not reveal any changes in the indentation profile, thus suggesting a more complex failure mechanism than SCC alone. It is also important to point out that the average delamination speed in the body fluid lies on 145  $\mu$ m.year<sup>-1</sup> [70], which appears to be very slow to be solely considered as the failure mode. Moreover, frictional induced heating should be also taken into account due to the inevitable thermal unfolding of proteins which led to protein precipitation and rheological changes. In fact, during the test the serum bath changed from translucent reddish-brown tint to creamy yellow color, typical for this type of wear tests [256]. The samples were also immersed at 50°C and 70°C after R-C indentation (see Figure V.6) and again no significant differences were observed as far as the adhesion of the coatings at different temperature was concerned.

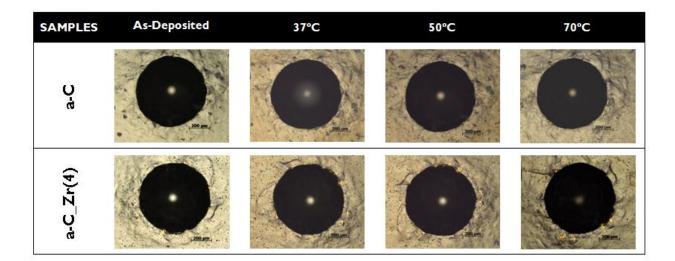


Figure V.6 Optical micrograph top-view of the Rockwell-C indentation test of the a-C and a-C\_Zr(4) films after immersion in BS for 72h at 37°C, 50°C and 70°C.

The exact failure mode has not been fully determined. However, it is expected that the presence of protein enhanced the delamination of the coatings [70, 107]. Although protein adsorption can act as a protecting barrier from wear by forming a *compact layer*, especially on Zr-containing films (see Chapter IV), it can entrap the liquid in the crevices. It limits exchange of liquid with the environment and, consequently, lowers the pH, which is known to enhance corrosion [70, 107, 122]. Thus, a less compact protein layer developed on the surface of pure carbon films may be beneficial to prevent the early coating spallation [257]. It is also important to take into account that UHMWPE and PEEK differs in surface energy: UHMWPE is more hydrophobic than PEEK ( $\theta_{UHMWPE} = 87^{\circ}$  and  $\theta_{PEEK} = 63^{\circ}$ , data from the manufacturers); the adsorption of proteins on UHMWPE will then be favourable (see Chapter IV), and therefore influence the whole frictional and wear behaviours [237]. Finally, a high friction induces an increased shear load in the coating, which reduces the resistance against the coating delamination, particularly in synergy with tribocorrosion. Such phenomena were particularly observed when sliding against PEEK.

### 3. Conclusions

The use of multidirectional testing was found to be an effective long-term "bench test" for polymer wear evaluation, which was proven by the testing reference materials (UHMWPE/CoCrMo); the results were comparable to those observed *in vivo*. Moreover, the poor tribological properties of Ti6Al4V were also confirmed with such technique and, in an early stage of sliding, the protective role of coatings was demonstrated as well.

Despite the use of an optimized gradient interlayer scheme and an improved adhesion by incorporation of Zr into the C-matrix, the Zr-containing film delaminated when sliding against UHMWPE and PEEK polymers. Moreover, the protein adsorption, which may lead to the decrease of the wear of metallic counterbody, was found to exhibit an opposite trend when sliding against polymers. The presence of proteins in solution plays an important role by acting as a corrosive electrolyte that could enhance the crevice condition in the surface defects. However, this effect could not be solely responsible for coating delamination, since there were no significant differences in the adhesion strength of tested coatings; Hauert's SCC evaluation did not show the decrease in the adhesion as well. The exact mechanism of coating delamination still has to be revealed. However, there is strong evidence that the failure mode is related to the corrosive effect of body fluids in combination with other factors such as cyclic failure.

## **OUTPUTS AND FUTURE RESEARCH**

The main and supporting idea of the research activity carried out in this thesis was that the application of ZrC/a-C(:H) nanocomposite coatings to biomedical implants could increase the life span of joint prosthesis by reducing the level of wear. To the date the use of protective coatings in the biomedical field has been limited due to adhesion problems under the body fluid environment.

In this thesis, hydrogenated and non-hydrogenated Zr-containing nanocomposite and pure carbon coatings were successfully deposited by DC magnetron sputtering onto Ti-based substrates. In light the literature review, several approaches were made to improve the adhesion of the coating to the substrate by introducing metal and/or gradient interlayers. A Ti/TiN/TiNC functional gradient layer was then chosen as an interface adhesive layer, which successfully prevented coating chipping and delamination. Subsequently, the Zr content was optimized in order to deposit nanostructured nc-ZrC/a-C(:H) coatings by co-sputtering a mixed graphite target with embedded Zr pellets. The incorporation of small amount of Zr (as low as 3 at.%) progressively led to the formation of nanocomposite structure consisting of small ZrC nanoparticles randomly dispersed in the a-C matrix. By XRD and TEM investigations, ZrC nanocrystals were found to vary in size in the range of I to 2.2 nm as a function of Zr content. The presence of such nanocrystals relaxed the films' stresses by inducing modifications on the C-matrix (graphite-like) and by changing the coatings morphology. When CH<sub>4</sub> was introduced into the chamber, pure carbon films (hydrogenated) revealed a glassy/compact morphology and high stress. The study also revealed that ZrC nanocrystallites strongly influenced the mechanical properties, specially increasing the elastic modulus and lowering the level of residual stress. Coatings with small crystallites revealed a more favourable elastic-plastic response (lower H/E and  $H^3/E^2$ ), which was particularly important when depositing on low load carrying capacity substrates such as titanium and titanium alloys. Zr alloyed films showed a similar performance in all sliding conditions (ambient air, PS and BSA solutions, sliding against Ti6Al4V), presenting good wear-corrosion behaviour together with strongly reduced wear of counterbody. The physicochemical properties of the coatings were also strongly related to the presence of ZrC nanocrystals and some  $ZrO_2$  that was detected on the surface. In fact, the wettability of the coatings increased with the increase of Zr content, enhancing albumin

adsorption. This was particularly important, since the ability to effectively adsorb proteins protected both surfaces from rubbing in direct contact.

Zr alloyed films developed in the frame of this thesis were superior to a-C films as far as the mechanical and tribological properties are concerned. Moreover, their adhesion was improved, by reducing the residual stress, and the wear of the counterbody during sliding test was lower, thanks to increased proteins adsorption. Our findings also indicate that the optimum properties were achieved by incorporation of small amounts of Zr, since smaller crystallites (low Zr content, ~4 at.%) led to better elastic properties and further decreased the counterbody wear. Nevertheless, the a- $C_Zr(4)$  film delaminated when sliding against UHMWPE and PEEK polymers under BSA. Moreover, the protein adsorption, which may lead to the decrease of the wear of metallic counterbody, was found to exhibit an opposite trend when sliding against polymers.

An area that has not been thoroughly explored within our framework is the study of the interface between the adhesion layer and the substrate, as well as the susceptibility of such interface to corrosion in body environment. It is known that an interface material can be formed by cross-contamination of the samples or from residual gas contamination. This interface will have an interfacial material with a different composition, with different mechanical properties and with particularly different corrosion properties, which may influence the overall adhesion of the coatings, especially under liquid media (protein-containing). In this study a rather solid emphasis was given to the optimization of the adhesion layer; however, typical adhesive tests only examine fracture toughness of the interface and do not address any corrosive deterioration. It is then certainly necessary to find an effective method of predicting long-term in vivo performance besides the standard adhesion tests and accelerating wear tests such as pin-on-disc or simulators. Therefore, additional investigations of all possible interfaces and interlayer deterioration effects should be done. If all these effects can be measured and predicted, DLC-based coatings could be further improved and coated implants could be eventually successfully applied in vivo as long term solution.

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# PAPER I

## DLC(H) Doped with Zr Coatings for Orthopaedics Applications: Adhesion Properties

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### DLC(H) Doped With Zr Coatings for Orthopaedics Applications: Adhesion Properties

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### ABSTRACT

DLC coatings doped with Zr were deposited by dc unbalanced magnetron sputtering in Ar (non-hydrogenated) and Ar+CH<sub>4</sub> (hydrogenated) atmosphere onto two different substrates: Ti grade 2 (pure titanium) and Ti grade 5 (Ti6Al4V). To improve adhesion between the films and substrates a functional Ti-based gradient layer was deposited (~300nm). Fundamental properties of coating were measured, such as chemical composition, structure, and mechanical properties. The main attention was paid to the adhesion testing evaluated using both Rockwell indentation and scratch test. It was shown that doping with Zr increased the adhesion between the coating and the substrate.

### INTRODUCTION

The production of wear particles from metal-on-metal (MoM) and metal-on-polyethylene (MoP) knee joint replacement and their adverse effects on the body has been a continual problem since the first implants were applied. Surface modification of the metallic bearing surfaces is expected to have the potential to reduce the generation of wear particles and, subsequently, to minimize aseptic loosening of the prosthesis. DLC ("diamondlike" carbon) coatings have been subject to a number of studies over the past years due to their excellent properties, such as wear, corrosion resistance, and biocompatibility. However, despite promising lab testing, these films have so far limited success in vivo. The major issue associated to these coatings is poor adhesion to the metallic substrates as a result of high residual stresses and poor chemical bonding between the film and substrate. Thus, partial delamination and consequent failure of the film cause failure of the implant [1, 2].

Among the metallic materials use for implants, Co-Cr alloy has been the standard material for hard bearing joint replacement components due to the superior tribological performances. However, it has been shown that Co- and Cr-ions released during wear and tribocorrosion processes are toxic or even carcinogenic [3, 4]. Compared to Co-Cr alloys, Ti grade 2 and 5 ELI show higher fatigue resistance and particularly excellent *in vivo* corrosion resistance [5, 6]. On the other hand, Ti-alloys exhibit low wear resistance due to low hardness (Ti grade 2 - 145 HV; Ti grade 5 ELI - 349 HV) and Young modulus (103-120 GPa) [7, 8]. Thus, to use Ti-alloys in demanding implant applications, surface modification is required to enhance the surface hardness and thus

wear resistance. Obviously the low friction coefficient is one of the requirements. The core objective of this work is the evaluation of the adhesion strength of zirconium-doped DLC coatings deposited by magnetron sputtering on two different Ti-based substrates.

### MATERIALS AND METHODS

The coatings were deposited using dc dual magnetron sputtering in an Ar atmosphere (non-reactive process) and Ar+CH<sub>4</sub> to produce non-hydrogenated and hydrogenated coatings, respectively. Two targets were used: Ti, to deposit an interface layer, and graphite, to prepare functional film. Zirconium was introduced using pellets embedded into the erosion zone of graphite target. The substrates (Ti grade 2 and Ti grade 5 ELI) were polished to a roughness of Ra≤50 nm and cleaned in an ultrasonic bath in acetone, ethanol and distilled water for 15 min each. The substrates were then etched via Ar ion bombardment at a substrate bias voltage of -650V. Table 1 summarizes the deposition parameters. The work pressure was kept constant at approximately 0.38 Pa; a pulsed bias voltage of -50V and frequency of 250 kHz was applied. A pure Ti/TiN bilayer and a grade TiNC layer, with a total thickness of about 550 nm, were deposited in order to improve the adhesion. The thickness of the coating was set to approximately 1.4 µm by adjusting the deposition time. To facilitate the reading, the coatings are denominated as DLC-Zr(X) and DLC:H-Zr(X) for doped films (X is the Zr content) and DLC and DLC:H the pure ones, the H is related to the hydrogenated films.

Table 1: Deposition parameters of DLC-Zr(X) and DLC-Zr:H(X) coatings.

Parameters	DLC-Zr(X)	DLC- Zr:H(X)			
Reactive gas flow (sccm)		5			
Ar flow (sccm)	45	40			
Chamber pressure (Pa)	1.48x10 <sup>-3</sup>				
Working Pressure (Pa)	0.38				
Substrate bias (V)	-50				
Graphite doped target (W/mm <sup>2</sup> )	0.075				

The chemical composition and structure of the coatings were evaluated by electron probe microanalysis (EPMA) and Xray diffraction (XRD), respectively. The molecular structure characterization was carried out by Raman spectroscopy with a laser wavelength of 532 nm. The hardness and Young modulus were measured by a depth-sensing nanoindentation test using a Berkovich indenter. The compressive stress of the films was calculated from Stoney's equation using the curvature of the substrate (SS 304), measured by an optical profilometer.

The coatings adhesion was evaluated using both Rockwell C indentation and scratch test, standard techniques commonly used to quantify the interfacial strength of coating-substrate systems. During Rockwell C adhesion test, a cone-shaped diamond 120° tip was perpendicularly indented into the coating causing layer damage at the boundary of the indentation. The results of the test were qualitatively evaluated by comparing the optical microscope images of the crack network and the degree of delamination to an adhesion quality chart and classifying the images into six categories from HF1-HF6 [9]. The coating adhesion was also evaluated using scratch test using a spherical diamond 120° tip with a diameter of 0.2 mm. The indentations were performed perpendicularly to the coating surface at 10 N/mm with a range of 2-50N.

### **RESULTS AND DISCUSSION**

### **Deposition Process, Film Structure and Composition**

The coatings were produced by reactive (Ar+CH<sub>2</sub>) and non reactive sputtering (Ar) deposition in order to obtain hydrogenated and non-hydrogenated films, respectively. By increasing the number of Zr pellets embedded into carbon target, the zirconium content increased to a maximum of 9 at.% (Table 2). Due to higher deposition rate of Zr pellets compared to C target, the deposition rate of non-hydrogenated films increased with number of zirconium pellets. The deposition rate of DLC(H) film was almost two times higher than that of DLC due to additional source of carbon in the chamber. However, the deposition rate was almost independent of the pellets number when the sputtering was carried out in a reactive atmosphere. In the presence of methane a compound layer is formed on the pellets surface significantly decreasing deposition rate (so-called 'poisoning effect'). Moreover, direct deposition of hydrocarbon from the plasma [10-12] could be expected. The deposition rate increased only for the highest number of pellets probably due to higher Zr surface area; CH<sub>4</sub> flow is then not sufficient to completely switch the sputtering from metallic to compound regime.

Sample		DLC	DLC_ Zr(3)	DLC_ Zr(7)	DLC_ Zr(9)	DLC:H	DLC:H_ Zr(4)	DLC:H_ Zr(5)	DLC:H_ Zr(8)
Zr Conte	nt (at.%)	-	3	7	9	-	4	5	8
Dep. Rate (nm/min)		4.6	5.0	7.3	8.3	8.8	8.5	10.0	12.0
Hardness	s (GPa)	10.9±0.5	$10.4 \pm 0.4$	11.9±0.6	$10.7 \pm 0.5$	8.9±0.3	10.0±1.6	$11.4 \pm 0.4$	12.2±0.5
Young Modulus (GPa)		94.0±01	109.9±3.1	126.0±2.6	133.3±2.2	69.9±0.3	87.3±2.6	103.1±1.2	114.1±3.4
D peak (cm <sup>-1</sup> )		1374	1373	1371	1374	1378	1384	1389	1379
G peak (cm <sup>-1</sup> )		1560	1558	1551	1554	1549	1550	1551	1548
$I_{\rm D}/I_{\rm G}$		1.6	1.9	2.3	2.3	0.7	0.8	1.0	1.0
	$Lc_{1}(N)$	4.5±0.2	$4.4 \pm 0.4$	$4.4 \pm 0.4$	5.1±0.6	4.8±1.6	4.4±1.2	4.6±0.2	4.8±0.5
Ti	$Lc_{2}(N)$	7.9±2.5	$8.0{\pm}1.8$	9.3±0.9	7.5±2.4	6.3±2.5	8.4±0.1	10.2±0.1	5.7±0.6
Grade 2	Lc3 (N)	41.9±11.5	35.5±3.2	>50	13.5±2.1	12.0±6.3	$38.9 \pm 8.4$	35.7±1.1	14.4±1.4
	R-C test	HF3	HF3	HF2	HF2	HF5	HF3	HF3	HF5
Ti Grade 5 ELI	$Lc_{1}(N)$	7.9±2.1	$6.0 \pm 0.8$	7.4±0.7	11.4±0.7	5.6±1.6	7.9±0.5	6.7±0.7	6.3±0.6
	$Lc_{2}(N)$	13.7±1.8	13.2±1.4	10.7±1.1	13.2±2.4	12.1±3.3	10.9±1.3	12.2±1.6	12.8±1.3
	Lc3 (N)	>50	37.1±7.1	>50	19.0±8.0	>50	38.7±1.5	37.8±0.6	>50
	R-C test	HF3	HF3	HF3	HF2	HF3	HF3	HF3	HF3

Table 2: Summary of structural, chemical and mechanical properties of the coatings.

<sup>*a*</sup>  $Lc_i$ , lower critical load is defined as the load where the first cracks occur (cohesive failure) and  $Lc_2$ , upper critical load, is defined as the load where the first spallation/flaking on the film is observed (adhesive failure).  $L_{c_3}$  is then defined as the load where the coating is penetrated in the scratch [13].

<sup>b</sup>HF1-HF4 is typically considered as an acceptable failure, whereasHF5-

XRD diffractograms clearly show the amorphous structure of the Zr-free coatings, where the only visible peaks were from the interlayer/gradient layer (Figure 1). High Zr content films showed broad peak identified as ZrC and thus indicating the presence of ZrC nanocrystalline structures embedded into the C-matrix.

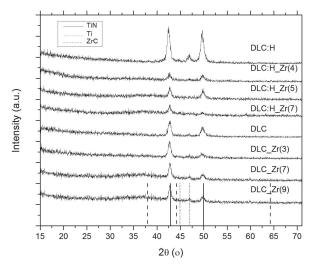


Figure 1: XRD diffractograms and phase labelling of the pure DLC and Zr-doped hydrogenated and non-hydrogenated films.

The deconvoluted positions of G (Graphite) and D (Disorder) peaks from the Raman spectra are shown in Table 2. The peaks were fitted using two Gaussian peaks [14]. For both hydrogenated and non-hydrogenated coatings the co-sputtering with Zr significantly influenced the structure of the carbon matrix.  $I_p/I_G$  ratio increased with the Zr content indication an increase of the carbon clustering (Table 2). XRD diffractograms showed a very broad peak at approx. 37 2 theta, i.e. close to the position of ZrC (111) (ICCD card n. 74-1221), as seen by Kao [15]. Coatings with high zirconium content are thus composed of

nanocrystalline ZrC embedded into the C-amorphous matrix. Bewilogua et al [12] reported the similar structures for several metal-doped DLC films. Furthermore, the injection of CH4 into the chamber resulted in saturation of sp<sup>2</sup>-C bonds with H converting than into sp<sup>3</sup> C-H sites, which could eventually form soft polymeric chains [16].

Measured Raman spectra were used to estimate hydrogen content in the coatings. The typical signature of hydrogen in visible Raman spectra is the presence of the photoluminescence background (PL); it increases with the H content. A good estimate of H content could be obtained from the ratio between the slope, m, of the fitted linear background and the intensity of the G peak, m/I(G) [17]. The analysis gave hydrogen content of metal-free hydrogenated coating (DLC:H) close to 28 at.%. We validate the method using similar films [18]; the H obtained from Raman spectra, ~34 at.%, was almost identical with direct measurements of H content by ERDA (~33 at.%).

### **Coating Adhesion and Cohesion**

Table 2 summarizes the main results of both adhesion tests performed. As expected the adhesion results were significant sensitive to the substrate used (Ti grade 2 and Ti grade 5 ELI).

Rockwell C indentation test is a qualitative adhesion method widely used in industry. The contact geometry in combination with the intense load transfer induces extreme shear stresses at the interface. Figure 2 shows the Rockwell indentations of the coatings (hydrogenated and non-hydrogenated). The adhesion decreased when H was incorporated into the C matrix. The compressive residual stress was similar for all coatings, about 2 GPa; therefore, the residual stress cannot be a key factor influencing coating adhesion. It is important to notice that hydrogenated films presented lower hardness, Young modulus and lower  $I_D/I_G$  ratio, which results in an inadequate load support of the coating particularly on soft

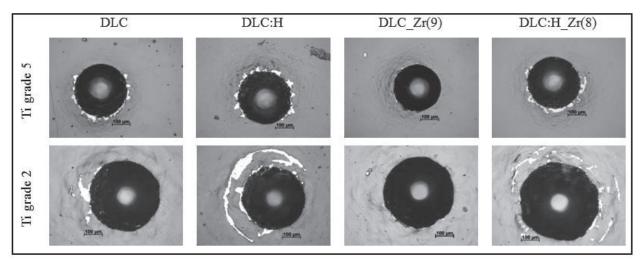


Figure 2: Coating Rockwell indentation.

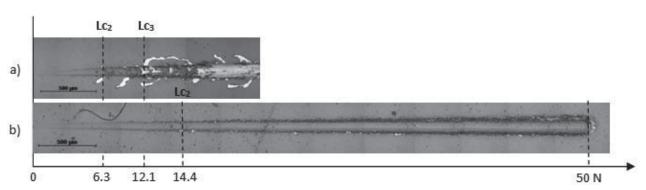


Figure 3: Scratching test at increasing load on DLC:H coated on both a) Ti grade 2 and b) Ti grade 5. The inset indicates the adhesion critical loads.

substrates [19]. Non-hydrogenated coatings doped with Zr showed sufficient adhesion increasing with Zr content. Moreover, there was no indication of delamination. Hydrogenated films exhibited large-scale delamination at the indentation edges (see Figure 2). Delamination and radial cracking could be related to substrate plastic deformation [20, 21]. HF6 is not accepted as a sufficient adhesion [9].

Scratch test, also known as 'short time tribological test' [22], is the only test method capable of generating stresses which exceed the interfacial bond strength of thin, well-adherent coatings [23]. Moreover, it has been considered as an adhesion assessment of new biocompatible materials [24]. As expected, the coatings deposited on Ti grade 2 exhibited lower adhesion than those on Ti grade 5; the adhesion is in general higher for non-hydrogenated films. Figure 3 demonstrates the influence of the substrate on the coating adhesion.

Cohesive failure of DLC:H coating occurred at the beginning of the test (Lc1) although with only few smooth cracks. However, the coating damage close to Lc2 depended on the substrate: (i) conformal cracks/chipping on Ti grade 5 and (ii) spallation/debonding on Ti grade 2. Moreover, Lc, was only observed when using Ti grade 2 as substrate, in such case a complete failure of the coating was observed. Doping with Zr improved the adhesion measured by Rockwell indentation on both substrates; metal doping typically leads to higher adhesion [25]. In general, the adhesion estimation obtained by both methods clearly showed the advantage of using harder Ti grade 5 as substrate. This finding corroborates the results shown by Kim et al, [26], where plastic deformation of CP Ti strongly influenced the fracture/fatigue cracks destroying the DLC layer and thus, limiting their tribological properties compared to Ti6Al4V/DLC combination

### CONCLUSION

DLC thin films doped by Zr were deposited by magnetron sputtering under reactive and non-reactive atmosphere. Increasing Zr content changed coating microstructure from amorphous (pure DLC film) to nanocomposite with ZrC nano-crystalline phase embedded into the amorphous carbon matrix. Pure DLC films show a very low adhesion; however, doping with Zr led to much higher critical load values obtained by scratch test and better qualitative adhesion estimated from Rockwell indents. When compared the two substrates, Ti grate 2 and Ti grade 5, it is clear that the later offer much better support for the coating.

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# PAPER II

a-C(:H) and a-C(:H)\_Zr coatings deposited on biomedical Ti-based substrates: Tribological properties

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## a-C(:H) and a-C(:H)\_Zr coatings deposited on biomedical Ti-based substrates: Tribological properties



## CrossMark

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### ABSTRACT

Amorphous carbon (a-C) based coatings are possible candidates as a surface treatment for various biocompatible materials used in medicine. In this study, the carbon coatings co-sputtered with Zr, deposited by dc unbalanced magnetron sputtering in Ar (non-hydrogenated, i.e. a-C/Zr) and  $Ar + CH_4$  (hydrogenated, i.e. a-C:H/Zr) discharges, were investigated and compared with pure carbon films. Polished pure commercial Ti grade 2 and Ti grade 5 ELI (Ti6Al4V) discs were used as substrates. To improve the coating/substrate adhesion, a gradient Ti-based interlayer was deposited (~450 nm). The coating structure was characterized by X-ray diffraction, X-ray photoelectron spectroscopy and Raman spectroscopy; the chemical composition was measured by electron probe microanalysis. The adhesion was evaluated by scratch-test and the hardness was measured by unanoindentation. Tribological testing of the coatings was carried out using unidirectional pin-on-disc experiments; pure titanium balls were used as counterparts. The wear rate of the coatings was negligible except for the films with the highest Zr content deposited on Ti grade 2 substrates.

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### 1. Introduction

Ti grades 2 and 5 ELI are among the most suitable metallic materials used for hard tissue replacement. These materials are an ideal choice due to the combination of low density, high strength to weight ratio, elastic modulus, sufficient failure resistance, corrosion resistance and excellent biocompatibility, particularly when compared to Co–Cr alloys [1,2]. However, Ti-based alloys are soft, their wear resistance is limited, particularly when adhesion or abrasion wear occurs, and they show high friction coefficients [3–5].

Amorphous (a-C) or hydrogenated amorphous (a-C:H) carbon has been considered as a potential protective layer of artificial joint materials due to excellent mechanical and tribological properties, corrosion resistance, chemical inertness, and biocompatibility [6–8]. The deposition of a-C films on soft substrates, such as Ti or Ti-based alloys, is a continuous challenge due to substrate plastic deformation, which could cause fracture and delamination [9–11]. Moreover, a-C films typically exhibit high internal compressive stress which significantly limits the adhesion of the coating and contributes to its tribological failure. The adhesion could be significantly improved by substrate surface pretreatments, such as nitriding or etching [2,12], by optimizing the deposition

\* Corresponding author. Tel.: + 351 239790745. E-mail address: ana.escudeiro@dem.uc.pt (A. Escudeiro). parameters, such as the substrate bias and Ar pressure [13–15], or by deposition of an adhesion improving interlayer [8]. The latter could be prepared as a gradient layer (Me/MeC, Me/MeN, or Me/MeN/MeNC, where Me is a metallic element) in order to avoid sharp interfaces with large mismatch of mechanical properties [9,16]. Co-sputtering of carbon films by metals (e.g. Ti, W, Cr) might improve the adhesion as well; however, these coatings then significantly differ from the pure carbon films [17–20].

Similar to titanium, zirconium is biocompatible and exhibits excellent tribological properties together with high corrosion resistance [21,22]. Therefore, zirconium is a potential candidate to improve properties of carbon-based films. The structure of carbon films co-sputtered with Zr is typically described as a nanocomposite, i.e., zirconium carbide nanograins embedded into an amorphous carbon matrix [23–25]. The core objective of this study is to evaluate the fundamental structural, mechanical and tribological properties of a-C and a-C:(H) coatings containing different zirconium contents deposited on soft Ti-based substrates.

### 2. Experimental details

### 2.1. Coating deposition

The coatings were deposited using dc dual magnetron sputtering in Ar (non-reactive) and Ar/CH<sub>4</sub> (reactive) atmospheres to produce non-hydrogenated and hydrogenated films, respectively. Two targets were used: titanium, to deposit the adhesion improving interlayer, and graphite to deposit the functional coating. To produce the Zr containing

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films, different numbers of Zr pellets (diameter 5 mm) were embedded into the erosion zone of the graphite target; thus, the Zr content could be controlled by the number of Zr pellets. Commercial pure Ti (Ti grade 2) and Ti6Al4V alloy (Ti grade 5 ELI) discs (diameter 50 mm; thickness 4 mm) were polished down to a roughness of  $R_a \leq 50$  nm and cleaned in an ultrasonic bath in acetone, ethanol, and distilled water for 15 min each. To measure residual stress and hardness, thin discs of 304 stainless steel and Si wafer were used, respectively. The substrates were mounted on the rotating sample holder (18 rpm) in the deposition chamber; the target-to-substrate distance was approximately 170 mm. The samples were etched via Ar ion bombardment for 1 h to remove all surface contaminants and oxides (bias voltage of -650 V). After etching, a Ti/TiN/TiNC(Zr) composite gradient layer with a total thickness of 450 nm was deposited to enhance the functional film adhesion, see scheme in Fig. 1. Ti target was first sputtered in pure Ar atmosphere (Ar flow 35 sccm); then, nitrogen was introduced (17 sccm) and, finally, the power of the graphite target (with or without embedded Zr pellets) was gradually increased. The top layer, approximately 1 µm thick, was deposited by sputtering the graphite (with or without Zr pellets) target in Ar or Ar/CH<sub>4</sub> atmosphere using a power density close to 7.5 W cm<sup>-2</sup>. The number of Zr pellets used in this study was 0 (referential coating), 8, 14, and 20 which corresponds to a relative erosion area  $(A_{Zr}/A_C)$  between 0 and 6%. The total pressure was kept constant at approx. 0.4 Pa by adjusting the gas flow and a pulsed substrate bias was applied (-50 V; frequency 250 kHz). The coating thickness was kept around 1.4 µm by adjusting the deposition time. The thickness was measured by 2D optical profilometer (Perthometer S4P) using a substrate/coating step.

### 2.2. Coating characterization

The chemical composition of the coatings was determined by electron probe microanalysis (EPMA) with a CAMECA Camebax SX 50 model. The applied voltage was 10 keV corresponding to a maximum penetration depth below 2 µm for carbon-based coatings. It should be noted that EPMA cannot detect hydrogen; therefore, it gives only Zr/C ratio in the case of hydrogenated coatings. The microstructure was analysed by X-ray diffraction (XRD) with a Philips, X'Pert diffractometer operating with Co K $\alpha$  radiation. To increase the signal from the film in relation to the substrate, grazing incidence with an angle of 2° was used. Chemical bonding was characterized by X-ray photoelectron spectroscopy (XPS) (ESCAprobe P, Omicron Ltd., Al K $\alpha$ , binding energy calibrated with Au 4f (84.1 eV)). The peak decomposition of C1s and Zr 3d were carried out using a GL(30) line shape, which stand for Gaussian/Lorentzian product formulas with 30% Lorentzian weight, and the contribution of the background was approximated by the Shirley method. The spin-orbital splitting in Zr 3d was assumed to be the same for all the phases and equal to 2.4 eV [26,27] and the integrated intensity of the Zr  $3d_{5/2}$  peak relative to that of the Zr  $3d_{3/2}$  peak

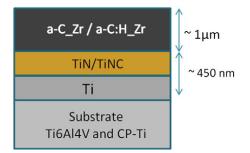


Fig. 1. Schematic design of a-C(:H)\_Zr coatings.

was considered equal to the spin-orbital multiplicity of 2/3. Moreover, to remove surface contamination, the samples were bombarded during 30 min by Ar<sup>+</sup> ions before analysis (beam energy 5 keV). Raman spectroscopy (Xplora, Horiba, DPSS laser with a wavelength of 532 nm) was used to analyse the carbon structure. The Raman spectra were fitted using two Gaussian peaks following the procedure described in Ref. [28]. The hardness was measured by depth-sensing indentation (Micromaterials Nanotest) using a Berkovich indenter. The normal stylus load was 5 mN (indentation depth approx. 150 nm); 32 independent indentations from two distinct areas on the sample were used to analyse the hardness data. Moreover, the reduced Young modulus was derived from the indentation measurements by the standard Oliver and Pharr method [29].

### 2.3. Adhesion and tribological properties

The coating cohesion and adhesion to the substrates were evaluated by scratch testing (CSM Instruments; spherical diamond 120° tip with a diameter of 200  $\mu$ m). The load was increased from 2 to 50 N (loading speed 10 N  $mm^{-1}$ ) and three critical loads were reported. Lc<sub>1</sub>, the lower critical load, is the load corresponding to the first cracks that occur (cohesive failure). Lc<sub>2</sub>, the upper critical load, is defined as the load where the first spallation/flaking of the film is observed (adhesive failure). Finally, Lc3 corresponds to the load where penetration of the coating is detected in the centre of the scratch. The tribological tests were carried out using a pin-on-disc set-up (CSM Instruments) in relative air humidity 30%; a normal load of 1 N, a linear speed of 20 cm s<sup>-1</sup> and test duration of 10,000 laps were used in this study. All coatings were tested with Ti6Al4V ELI counterbody machined as a ball with a diameter of 8 mm. The friction coefficient was measured during the sliding tests, whereas the wear rate was calculated as the worn volume per sliding distance per load. The worn volume was measured by 3D profilometer (Zygo NewView 7200); the wear of the ball was evaluated as well. Optical microscope and scanning electron microscopy (SEM) were used for the wear track evaluation. The influence of the plastic deformation of the substrate on the measured coating worn volume is discussed below. The mechanical properties of Ti grade 2 and Ti grade 5 ELI shown in Table 1 [4,5] differ significantly. However, the initial and mean Hertzian pressures for the ball-on-plate contact are almost identical.

### 3. Results and discussion

### 3.1. Coating deposition and chemical composition

Table 2 shows the zirconium content together with the mechanical properties of the deposited coatings. Low amounts of Ti and N originated from the interface were observed. The oxygen contamination arising from the residual atmosphere was lower than 3 at.%. Due to the higher sputtering yield of zirconium compared to the graphite target, the deposition rate increased with the number of Zr pellets. To keep the coating thickness constant, the deposition time was decreased when Zr

#### Table 1

Summary of the mechanical properties (hardness, H, Young modulus, E, elastic strain to failure, H/E, resistance to plastic deformation,  $H^3/E^2$ , and yield strength, Y) and the Hertzian contact conditions (maximum and medium contact pressure,  $P_0$  and  $P_m$ , and maximum shear stress,  $\tau_{max}$ ) for both substrates.

	H (GPa)	E (GPa)	H/E	H <sup>3</sup> /E <sup>2</sup> (GPa)	Y (MPa)		P <sub>m</sub> (MPa)	τ <sub>max</sub> (MPa)	
Ti grade 5 Ti grade 2		114 103		0.0023 0.0003	793 276	360 360	240 240	110 110	

Table 2

Deposition rate, chemical composition and mechanical properties (hardness, H, Young modulus, E, elastic strain to failure, H/E, resistance to plastic deformation, H<sup>3</sup>/E<sup>2</sup>, and elastic recovery, R) of the coatings.

	Sample	Dep. rate (nn/min)	Zr (at.%)	H (GPa)	E (GPa)	H/E	$H^3/E^2$ (GPa)	R (%)
Non-hydrogenated	a-C	4.6	0	$10.9\pm0.5$	$94.0\pm0.1$	0.116	0.147	71
	a-C_Zr(3)	5.0	3	$10.4\pm0.4$	$109.9\pm3.1$	0.094	0.092	62
	a-C_Zr(7)	7.3	7	$11.9\pm0.6$	$126.0 \pm 2.6$	0.094	0.105	68
	a-C_Zr(9)	8.3	9	$10.7\pm0.5$	$133.3 \pm 2.2$	0.080	0.069	59
Hydrogenated	a-C:H	8.8	0	$8.9\pm0.3$	$69.9\pm0.3$	0.128	0.146	74
	a-C:H_Zr(4)	8.5	4	$10.0\pm1.6$	$87.3\pm2.6$	0.115	0.132	71
	a-C:H_Zr(5)	10.0	5	$11.4\pm0.4$	$103.1\pm1.2$	0.111	0.140	73
	a-C:H_Zr(8)	12.0	8	$12.2\pm0.5$	$114.1\pm3.4$	0.107	0.140	67

pellets were added. The maximum Zr content was 9 at.%. When methane was introduced into the chamber, the deposition rate increased. Since Zr content was only slightly lower for hydrogenated films, it was probable that the poisoning of Zr pellets surface was negligible, since additional carbon was added through reactive gas [30,31].

To facilitate reading, the coatings are hereinafter denominated as  $a-C_Zr(X)$  and  $a-C:H_Zr(X)$ , where X is the Zr content and H indicates hydrogenated films.

### 3.2. Structure and chemical bonding

XRD diffractograms clearly show the amorphous structure of the carbon coatings, since the only visible peaks were originated from the gradient interlayer (Fig. 2). When Zr was added, it was possible to detect a very broad peak near 37.9° which was identified as the ZrC phase (111) (ICDD card n. 74-1221).

Fig. 3 shows XPS C1s and Zr 3d spectra of a-C:H\_Zr(8) coating after 30 min of sputter-etching. Three peaks were detected in C 1s spectrum; C-C bond, C-O bond, and a peak positioned at 283.2 eV. The binding energy of ZrC phase is typically in the range (281.8-282.3 eV) [25,26,32,33]. However, Smith et al. [34] reported the presence of such a peak near 283.6 eV, together with the typical ZrC peak deposited from single source organometallic precursors. The presence of such peak may correspond to the organometallic: Zr=C-, Zr-CHn-, or Zr=CH- bonds. Anderson et al. [25] also studied thin Zr containing carbon films. They observed a Zr-C peak in the standard position typical for bulk ZrC; however, their spectra revealed an additional peak at higher binding energy (282.7 eV). This peak was attributed to atoms at the matrix-carbide interface. Lewin [35,36] has previously found a similar peak in a nc-TiC/a-C nanocomposite; this peak was attributed to metal atoms in an interfacial state at the carbide/matrix interface. These works referred above showed significantly higher Zr(Ti) contents compared to the present work (close to 40 at.%). In our case, the Zr content is much lower and the presence of a broad peak in the XRD diffractograms may indicate the presence of a ZrC nanophase. Therefore, almost all Zr atoms in the Zr-C grains are positioned at the interface with the amorphous carbon matrix, which should give rise almost exclusively to the peak at higher binding energy. Zr 3d spectrum was deconvoluted considering the spin-orbit splitting equal to 2.4 eV [26] (Fig. 3). Two bonds were identified: Zr-O (probably from oxygen contamination from residual atmosphere) and Zr – C. The latter is positioned at ~180.4 eV, i.e. close to Zr - C (179.1-181.1 eV (Zr 3d<sub>5/2</sub>) [26,34,37]). The Zr metallic phase peak (178.7 eV Zr3d<sub>5/2</sub> [27]) was not observed.

Thus, XRD and XPS results could be considered as an indication of a nanocomposite structure with ZrC nanograins embedded in an amorphous carbon matrix. Similar results were reported by Adelhelm et al. [38] in the case of carbon films with approximately 7 at.% of Zr. Nevertheless, further analysis, namely transmission electron microscopy, is planned for future study. The typical Raman spectra of the deposited coatings shown in Fig. 4 are composed of two broad bands: the G (Graphite) peak and the D (Disorder) peak centred at around 1580 cm<sup>-1</sup> and 1350 cm<sup>-1</sup>, respectively [28]. Fig. 5 summarizes the  $I_D/I_G$  ratio (i.e. the ratio of integrated

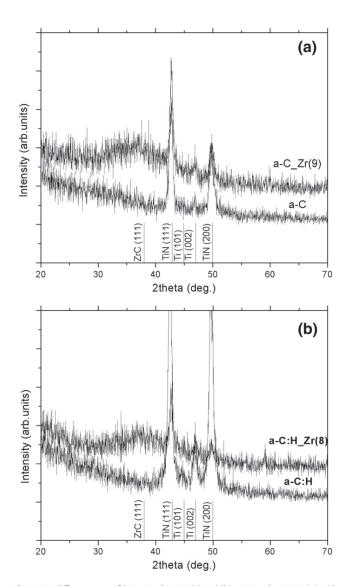


Fig. 2. XRD diffractograms of (a) a-C and a-C\_Zr(9) and (b) a-C:H and a-C:H\_Zr(8) with indexation patterns.

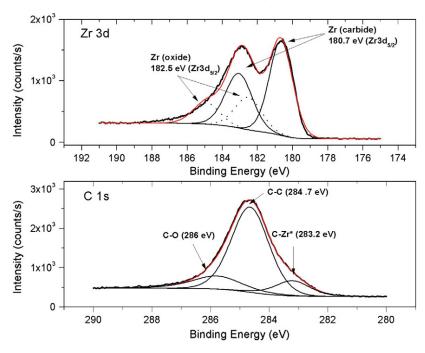


Fig. 3. XPS Zr 3d and C 1s spectra of a-C:H\_Zr(8) coating.

intensities of D and G band) and G peak position as a function of Zr content. Simultaneous, the presence of Zr and H in carbon films played an important role in promoting both the formation of aromatic rings and the increase of the organization of C clusters, as documented by the increase of either the D peak intensity (i.e. growing of  $I_D/I_G$  ratio) or the G peak position [28]. Although I<sub>D</sub>/I<sub>G</sub> ratio was also increasing with Zr content in the case of non-hydrogenated coatings, G peak position shifted to lower values (Fig. 5). These results corroborate those presented by Adelhelm et al. [23], who found a similar G peak shift for a-C:Me (Me = Ti or Zr). It can thus be concluded that: (i) there was higher bond disorder even if sp<sup>2</sup> clustering is higher and/or (ii) the presence of heavy metals can influence the vibration modes of adjacent C=C bond due to the slowing of the relative sp<sup>2</sup> movement, which is responsible for the G peak. Moreover, the presence of ZrC phase could act as a catalyst and promote the graphitization of the adjacent carbon matrix leading to higher sp<sup>2</sup> content in the films [23,31] and a slight increase of the G peak position for higher Zr contents. Nevertheless, further investigation will be carried out in order to better understand the influence of heavy atoms on the C-matrix structure.

### 3.3. Mechanical properties

The hardness and the Young modulus of the films are shown in Table 2. One of the main objectives of coating titanium-based substrates with Zr containing carbon-coatings was to improve the wear resistance. According to Archard model [39], there is a linear relationship between the volume worn during the sliding and the inverse of the hardness. Thus, a higher hardness should lead to a higher wear resistance. In general, the hardness of the deposited coatings was significantly higher than that of the substrates. Non-hydrogenated a-C\_Zr films showed a comparable hardness with that of pure a-C (~11 GPa). The lowest hardness value was observed in the case of pure hydrogenated carbon film (8.9 GPa); however, it progressively increased with Zr content. The Young modulus was calculated using the following parameters:  $E_i = 1140$  GPa and  $\nu_i = 0.07$  for the diamond indenter [29] and  $\nu = 0.3$  for the coating ( $\nu$  is the Poisson ratio) [40]. Co-sputtering with Zr decreased

the elastic strain to failure (H/E) and the resistance to plastic deformation (H<sup>3</sup>/E<sup>2</sup>) of the coatings; the effect was more pronounced for the non-hydrogenated coatings. Moreover, a higher Young modulus of hydrogen-free films implied that the elastic recovery (R) also decreased (Table 2). The slope of the linear section of the unload curve was higher for the pure film than that of a-C\_Zr(9); thus, the pure films presented a higher elasticity compared to the Zr containing films. The same trend was found in carbon films co-sputtered with Cr, where brittle CrC phase decreases the film toughness and elastic recovery [41]. In general, the elastic recovery of the hydrogenated films was higher compared to non-hydrogenated ones. In contrast to bulk Ti-based substrates (see Table 1), the H/E and the (H<sup>3</sup>/E<sup>2</sup>) values of the coatings were about one to two orders of magnitude higher.

Typical scratches produced by the adhesion test are shown in Fig. 6. Piling-up along the edges of the scratch track, which is clearly observed in both substrate materials, indicated plastic deformation (Fig. 6a and c). As expected, the plastic deformation was much more evident in the case of Ti grade 2 substrates, i.e. it was visible for lower loads. Fig. 6b and d shows the scratches of a-C:H coatings deposited on Ti grade 2 and Ti grade 5 ELI, respectively. The scratch tracks are much wider for the softer Ti grade 2 substrates when an identical load was applied. Fig. 7 shows the critical loads Lc<sub>1</sub>, Lc<sub>2</sub>, and Lc<sub>3</sub> as a function of Zr content. As expected, the critical loads of the coatings deposited on Ti grade 2 were in general inferior compared to those prepared on Ti grade 5 ELI substrates. This finding agrees with the results shown by Kim et al. [42], where the plastic deformation of pure titanium substrates strongly promoted the formation of cracks, leading to the destruction of a-C films; a harder substrate (Ti6Al4V/a-C(:H)) showed improved coating adhesion and wear resistance. All coatings presented in this study showed similar compressive residual stress, approx. 2 GPa, calculated by the Stoney's equation [43]. Thus, the residual stress was not expected to be a key factor determining the adhesion behaviour. a-C\_Zr(7) coating presented the best scratch behaviour for both substrates and delamination were not observed (i.e. Lc<sub>3</sub> > 50 N). Although the pure carbon coatings showed higher plastic deformation resistance  $(H^3/E^2 \text{ ratio})$  than the films containing Zr, there was no significant

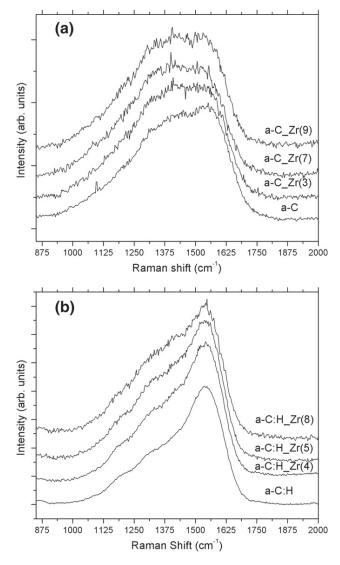


Fig. 4. Raman spectra of (a) non-hydrogenated films and (b) hydrogenated films.

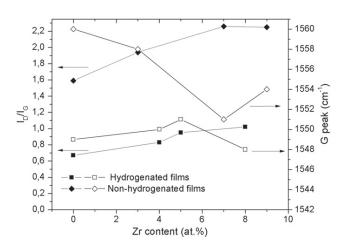


Fig. 5.  $I_D/I_G$  ratio and G peak position as a function of Zr content.

differences in the measured critical loads. Thus, the substrate mechanical properties were much more important than the difference among coatings. High pile-up could lead to higher wear of the counterbody; again, the Ti grade 2 substrates were inferior with increased risk of counterbody damage during the sliding tests.

#### 3.4. Friction and wear properties

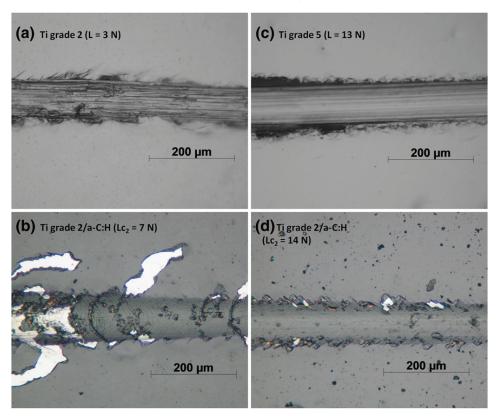
Table 3 summarizes the tribological results of the coatings; the values for the uncoated substrates are shown as a reference. As expected, the friction coefficient dropped significantly when pure carbon-based coatings (i.e. a-C and a-C:H) were applied. The friction, then, slightly increased when Zr was added. The wear rate of the films was about three orders of magnitude lower than that of uncoated substrates. Although the tribological parameters, such as the maximum depth of the wear track and the coating wear rate, were similar for both substrates (see Table 3), the appearance of the wear track was significantly different (Fig. 8). Fig. 9 shows a SEM micrograph from the wear track on softer Ti grade 2 substrates. Even though, the presence of areas lifted above the surface is a typical feature of the material transfer from the ball material onto the coating surface, there were traces neither from such transfer nor accumulation of the wear debris in the wear tracks. Therefore, only the plastic deformation of the substrate could lead to the formation of the wear tracks shown in Fig. 8. The highest deformation was observed in the coating a-C\_Zr(9), where the depth of the wear track oscillated between -1 (i.e. above the surface) and 2  $\mu$ m. Such severe deformation could not be accommodated by the coating and cracks parallel to the sliding direction were observed. Moreover, the presence of cracks led to local coating delamination (Fig. 9). It should be pointed out that  $a-C_Zr(9)$  was the only coating which failed during the sliding tests. The failure of the film corresponded to the results of scratch test (the lowest critical force Lc<sub>3</sub>); moreover, this film also exhibited the lowest elastic recovery (Table 2).

Since significant plastic deformation was observed, we had to consider it in the calculation of the wear rate. The wear rate calculated as the worn volume per sliding distance per load, which could be given in our case as the area of the wear track cross-section per number of cycles per load, would give misleading results. Therefore, we report here the worn volume as the difference between the volume of the wear track (i.e. missing material below original coating surface) and the volume of the material above the surface [44]. The wear rate calculated in this way is almost independent of the substrate material, although the maximum depth of the wear track was significantly higher for Ti grade 2 substrates. The wear of the counterbody calculated as the worn volume per sliding distance per load was the highest for pure carbon coatings; we can thus conclude that co-sputtering with Zr slightly increased the friction, but significantly diminished the wear of the counterpart.

Although the sliding tests were performed in humid air (relative humidity 30%) and the contact pressure was much higher than that in real biomedical applications, this study could indicate either the potential or the risks of applying a-C(:H)\_Zr coatings deposited on Ti-based substrates. Commercial pure titanium and Ti6Al4V alloy typically present abrasive and adhesive wear due to plastic deformation and material transfer [3]. Ti grade 2 is more likely to deform plastically than Ti grade 5 ELI. According to Tresca criterion, under normal load axisymmetric contact condition, the plastic yield occurs when the maximum contact pressure, P<sub>o</sub>, exceeds 1.6 the yield strength, Y, of the contacting material:

$$P_0^{Y} = 1.6Y_{Ti grade 5 ELI} = 1270 MPa and P_0^{Y} = 1.6Y_{Ti grade 2} = 442 MPa$$

Table 1 shows that the maximum contact pressure for both substrates was lower than the onset of plastic deformation. Moreover, the required load to initiate yield is related to the maximum contact

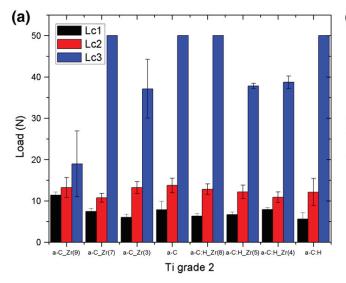


**Fig. 6.** Optical images of typical scratch tracks performed on (a) uncoated Ti grade 2 and (c) Ti grade 5 ELI and (b) a-C:H deposited on Ti grade 2 and (d) Ti grade 5 ELI. Scratch direction is from right to left. The arrows show the adhesive critical load Lc<sub>2</sub> and coating delamination (Lc<sub>3</sub>).

load [44]; it was 1.35 N for Ti grade 2, i.e. slightly higher than 1 N load used in this study. However, it is important to point out that the calculation was based on ideal smooth ball and disc surfaces. Since real surfaces in contact are rough the local contact pressure between asperities is much higher. Thus, plastic deformation can occur at lower loads [45].

#### 4. Conclusions

Zr doped a-C hydrogenated and non-hydrogenated coatings have been deposited by magnetron sputtering onto soft Ti-based biomaterials (pure titanium and Ti6Al4V alloy) in order to enhance their tribological properties. The coatings with the highest Zr content exhibited a



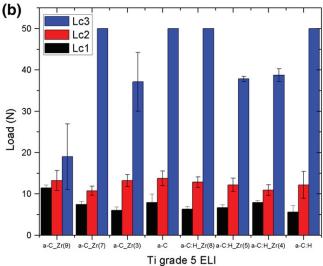


Fig. 7. Critical loads for the coating deposited on (a) Ti grade 2 and (b) Ti grade 5 ELI. Symbol (\*) is used when the delamination was not observed even at the highest applied load.

#### Table 3 Wear test results from pin-on-disc experiments. Symbol (\*) represents not measurable wear.

	Samples	Ti grade 2				Ti grade 5 ELI					
		$R_{a}\left(\mu m ight)$	Depth (µm)	Disc wear rate (10 <sup>-6</sup> mm <sup>3</sup> /Nm)	Ball wear rate (10 <sup>-6</sup> mm <sup>3</sup> /Nm)	COF	$R_a \left( \mu m \right)$	Depth (µm)	Disc wear rate (10 <sup>-6</sup> mm <sup>3</sup> /Nm)	Ball wear rate (10 <sup>-6</sup> mm <sup>3</sup> /Nm)	COF
Non-hydrogenated	a-C	0.086	$0.4 \pm 0.2$	$0.6\pm0.4$	$0.44\pm0.04$	0.09	0.055	$0.2\pm0.1$	$0.4 \pm 0.3$	$0.98 \pm 0.10$	0.10
	a-C_Zr(3)	0.086	$1.1 \pm 0.2$	$1.3 \pm 0.7$	$0.16 \pm 0.02$	0.12	0.085	$0.5 \pm 0.1$	$0.8 \pm 0.2$	$0.19\pm0.02$	0.16
	a-C_Zr(7)	0.086	$0.8 \pm 0.3$	$0.5 \pm 0.4$	$0.09 \pm 0.01$	0.15	0.088	$0.2 \pm 0.1$	$0.6 \pm 0.4$	$0.25\pm0.03$	0.15
	a-C_Zr(9)	0.085	$1.4 \pm 0.1$	$2.4 \pm 1.1$	$0.40\pm0.04$	0.19	0.103	$0.7 \pm 0.2$	$1.0 \pm 0.4$	$0.10\pm0.01$	0.14
Hydrogenated	a-C:H	0.101	*	*	$0.33 \pm 0.03$	0.14	0.076	$0.2 \pm 0.1$	$0.2 \pm 0.1$	$0.85 \pm 0.09$	0.16
	a-C:H_Zr(4)	0.064	$0.4 \pm 0.1$	$0.6 \pm 0.5$	$0.28 \pm 0.03$	0.14	0.051	$0.3 \pm 0.1$	$0.5 \pm 0.2$	$0.08 \pm 0.01$	0.13
	a-C:H_Zr(5)	0.083	$0.8 \pm 0.2$	$0.4 \pm 0.1$	$0.26 \pm 0.03$	0.13	0.055	$0.6 \pm 0.2$	$0.6 \pm 0.2$	$0.12\pm0.01$	0.13
	a-C:H_Zr(7)	0.073	$0.7\pm0.2$	$0.7 \pm 0.3$	$0.16 \pm 0.02$	0.15	0.076	$0.6 \pm 0.1$	$0.8 \pm 0.3$	$0.04 \pm 0.01$	0.16
Uncoated		< 0.050	$13.5 \pm 3.3$	$528.2 \pm 54.6$	-	0.52	< 0.050	$10.6 \pm 1.1$	$843.7 \pm 139.2$	-	0.49

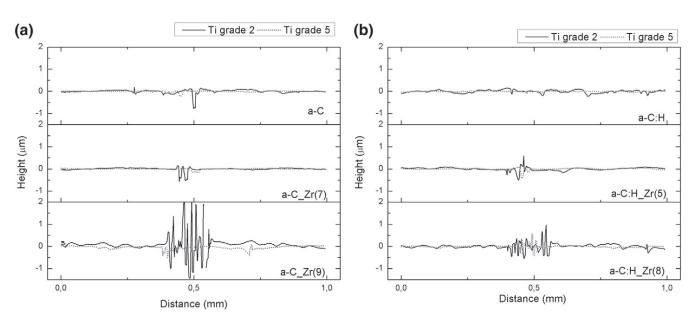


Fig. 8. Comparison of the wear track profiles (cross-section): (a) non-hydrogenated and (b) hydrogenated coatings.

nanocomposite structure with ZrC nanograins embedded into an amorphous carbon matrix. The incorporation of Zr into hydrogenated coatings increased the film hardness; however, such effect was not observed in the case of non-hydrogenated films. The adhesion of the carbon coatings on the substrates was improved by co-sputtering with Zr. Protection of Ti-based substrates against the wear, lower friction and significantly lower wear rates of the counterpart could be considered as major benefits of Zr containing a-C(:H) coatings.

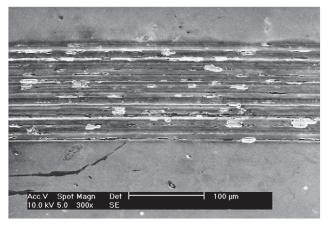


Fig. 9. SEM micrograph of the wear track, a-C\_Zr(9) coating deposited on Ti grade 2.

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96

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# Corrigendum

# Corrigendum to "a-C(:H) and a-C(:H)\_Zr coatings deposited on biomedical Ti-based substrates: Tribological properties" [Thin Solid Films 538 (2013) 89–96]

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The authors regret that the two plots in Fig. 7 are the same. Fig. 7a should be referred to the following plot.

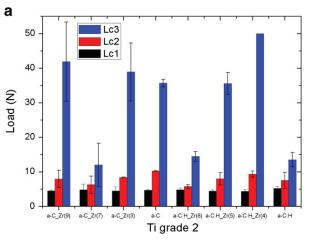


Fig. 7. a. Critical loads for the coating deposited on (a) Ti grade 2.

These corrections do not affect the conclusions in this paper. The authors would like to apologise for any inconvenience caused.

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# PAPER III

Indentation and scratch testing of DLC-Zr coatings on ultrafine-grained titanium processed by high-pressure torsion

C.T. Wang, <u>A. Escudeiro</u>, T. Polcar, A. Cavaleiro, R.J.K. Wood, N. Gao and T.G. Langdon Wear, 306, 1-2 **(2013)** 304-310

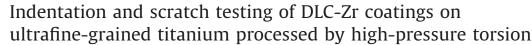
## Wear 306 (2013) 304-310



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#### 1. Introduction

Human joints might suffer from pain and functional loss due to degenerative diseases, aging and accidents. A total joint replacement is regarded as an effective procedure for treating joint diseases and fractures [1]. A technical survey reported that there is an increasing demand for new and improved implants because of a rapidly growing patient population and increasing numbers of younger patients [2]. The Ti-6Al-4V alloy was designed originally for the aerospace industry but it has been used widely for biomedical applications due to its high strength, good fatigue characteristics, bio-tolerance and excellent corrosion resistance [1]. Despite these attractive qualities, Ti-6Al-4V has significant drawbacks which limit its further capacity to be used as an orthopaedic and dental implant material. For example, the toxic Al and V ions released from Ti-6Al-4V may cause long-term health problems and adverse reactions with body tissues [3,4]. Therefore, there is now an urgent need for exploring the potential for developing other bio-metals.

Pure titanium has excellent properties including high corrosion resistance, low electronic conductivity, a low ion-formation tendency and very good biocompatibility. All these characteristics make it a very good candidate as an implant material.

#### ABSTRACT

High-pressure torsion was employed to refine the microstructure of grade 2 Ti under an imposed pressure of 3.0 GPa at room temperature. The microhardness of grade 2 Ti increased from 1.82 GPa for the coarse grain state to 3.05 GPa after high-pressure torsion processing, where this value is very close to the hardness of the Ti–6Al–4V alloy. Subsequently, several diamond-like carbon (DLC) coatings with thicknesses of ~1.4  $\mu$ m were deposited on as-received Ti, high-pressure torsion processed Ti and Ti–6Al–4V samples via physical vapour deposition. Both indentation and scratch tests showed a much improved adhesion of DLC-7Zr, DLC:H-7Zr and DLC-9Zr coatings with high-pressure torsion processed Ti as the substrate by comparison with the same coatings on coarse-grained Ti. The results suggest that commercial pure Ti processed by high-pressure torsion and coated with a diamond-like carbon coating provides a potential candidate material for bio-implant applications.

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Nevertheless, pure titanium has a relatively low strength and a very poor wear resistance when it is subjected to sliding and abrasion. Thus, it is generally not suitable for use in artificial joints which seek materials with high strength and good tribological properties. Recently, pure titanium with ultra-fine grain sizes, processed through the application of severe plastic deformation (SPD), appears to offer an alternative possibility for the production of implant materials [5,6]. In severe plastic deformation processing, a high level of straining is imposed on a metal using an extensive hydrostatic pressure and the strain is achieved without changing the overall dimensions of the material [7,8]. Thus, a very large strain may be imposed on materials via repeated severe plastic deformation processing. The ultrafinegrained (UFG) materials produced by severe plastic deformation often possess extraordinary properties, including both high strength and toughness [9], long fatigue life [10,11] and reasonable wear resistance [12,13]. To date, various severe plastic deformation processing methods have been applied successfully for the production of high strength pure Ti [6,14-17]. However, the evidence suggests that severe plastic deformation processing may lead to only a relatively minor improvement in the wear resistance of Ti [5,18]. As an alternative approach, it appears that surface treatments or coatings may be necessary to enhance the service durability of titanium as implant components [19].

The idea of improving wear resistance of Ti alloys via surface treatment has been reported extensively. Surface engineering technologies such as thermal oxidation, ion implantation and

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thin coatings can significantly improve the tribological properties of Ti alloys. The high strength of coatings were favourable to maintain the low surface roughness during sliding motion, which also led to less wear loss of the polyethylene counterpart [20,21].

In recent years, attention has moved to surface engineering of ultrafine-grained Ti. High-current-density nitrogen ion implantation was used to enhance the wear resistance of ultrafine-grained Ti and the results demonstrated the potential for forming a hard layer with good wear and corrosion resistance on the surface of ultrafine-grained Ti via ion implantation [22]. However, this method required that the samples were held at the relatively high temperature of 820 K and there is evidence this may be recrystallisation of the ultrafine-grained structure of Ti when the temperature is above  $\sim$ 620 K [23]. It appears, therefore, that chemical vapour deposition (CVD) and physical vapour deposition (PVD) may be better approaches for this purpose because this permits the use of a lower processing temperature and easier control over the surface roughness. It is important to note that PVD methods such as magnetron sputtering can be controlled at room temperature leading to a much lower internal stress at the interface. For example, a thin and hard TiN coating was deposited on both coarse-grained and ultrafine-grained Ti substrates using PVD at a temperature below 420 K in a recent study [19]. The subsequent wear tests showed that the TiN coating improved the wear resistance of Ti by nearly two orders of magnitude and the ultrafine-grained Ti was a better substrate than coarse-grained Ti due to its higher strength. Therefore, a PVD coating on ultrafinegrained Ti shows significant promise for further exploration in future bio-implant applications.

Diamond-like carbon (DLC) coatings are regarded as other good candidates for the purpose of wear protection because of their excellent mechanical and tribological properties such as high hardness, good wear and corrosion resistance. Moreover, DLC coatings also have a much lower coefficient of friction (COF) which can also reduce the wear of the counter surface during sliding [24]. To date, several studies have explored DLC coatings for bio-implant use [25-27]. Despite the excellent tribological properties of DLC coatings, delamination of the coatings was a major problem which appeared to limit their application as articulating joints [25]. It is reasonable to anticipate that, once removed from the substrate, these DLC coating particles would act as a third body causing severe damage to the substrate or even severe reactions with the tissue. In order to solve these issues, studies have been reported exploring the use of new interlayers to improve the bonding of DLC coatings [28]. Doping with metallic elements, such as Cr [21,29], Ti [30] and W [31], may also improve the adhesion and wear resistance. Recently, zirconium was used as a dopant metal to provide low-toxicity, good tribological behaviour and high corrosion resistance [32-34].

The present investigation was therefore initiated in order to explore the effect of substrate microstructure on the adhesion behaviour and scratch resistance of DLC-7Zr, DLC:H-7Zr and DLC-9Zr coatings deposited on grade 2 pure titanium substrates both with and without high-pressure torsion processing and, in addition, to make a direct comparison with deposition on a Ti–6Al–4V substrate.

#### 2. Experimental materials and procedures

#### 2.1. Materials and processing

The experiments were conducted using grade 2 pure titanium having a composition of 0.015 H, 0.1C, 0.25 O, 0.03 N, 0.3 Fe and balance Ti (wt%) and the Ti–6Al–4V alloy with extra low interstitials. Initially, some of the grade 2 Ti samples were subjected to

high-pressure torsion processing. The samples for high-pressure torsion were machined into disks with a diameter of 10 mm and thicknesses between 0.80 and 0.85 mm. During high-pressure torsion, the disk samples were held in shallow depressions on the faces of two massive anvils, a load was applied and then torsional straining was achieved through rotation of the lower anvil. In high-pressure torsion, the shear strain,  $\gamma$ , at different position of the disk can be estimated using the relationship [35]

$$\gamma = \frac{2\pi NR}{h} \tag{1}$$

where *N* is the number of rotation revolutions, *R* is the distance from the centre of the disk and *h* is the height (or thickness) of the sample. In this study, high-pressure torsion processing was conducted using an imposed pressure of 3.0 GPa for 10 revolutions under quasi-constrained conditions in which there is some restricted outflow of material around the periphery of the disk during the processing operation [36,37]. Additional details on the principles of high-pressure torsion processing are given in earlier reports [38–40].

Microhardness testing of the samples was performed under an indentation load of 1 kg for 15 s. As described earlier [19], tensile testing was conducted at room temperature after high-pressure torsion processing using an initial tensile strain rate of  $1.0 \times 10^{-2} \text{ s}^{-1}$ . The microstructures of the materials were examined using a JEM 3010 transmission electron microscope (TEM) operating under an accelerating voltage of 200 kV. A detailed description of these analytical procedures was given in earlier reports [5,19].

#### 2.2. Coating deposition and characterisation

Prior to deposition, the substrates were polished to a roughness less than 50 nm, cleaned in an ultrasonic bath in acetone, ethanol and distilled water for 15 min, and then etched using Ar<sup>+</sup> bombardment for 1 h at a substrate bias voltage of -650 V in order to remove all contaminants and oxides on the substrate surface. The coatings were deposited using dc dual magnetron sputtering in an Ar atmosphere (non-reactive process) and Ar+CH<sub>4</sub> (reactive process) to produce non-hydrogenated and hydrogenated coatings, respectively. Two targets were used: titanium was used to deposit an interface layer and a graphite target was used with embedded Zr pellets to prepare functional film. Table 1 summarises the deposition parameters. The work pressure was maintained constant for both reactive and nonreactive sputtering by adjusting the gas flow. A pulsed bias voltage of -50 V and a frequency of 250 kHz were applied. A TiN/TiCN interlayer with a varying composite gradient and a thickness of 450 nm was deposited in order to improve adhesion. The thickness of the functional coating was approximately  $1\,\mu m$ so that the total film thickness was of the order of  $\sim\!1.4\,\mu\text{m}.$  To facilitate a detailed description of the results, the coatings were denoted as DLC-XZr and DLC:H-XZr, where X represents the zirconium content.

Table 1
Deposition parameters of DLC-XZr and DLC:H-XZr coatings.

Parameters	DLC-XZr	DLC:H-XZr
Reactive gas flow (%)	-	5
Ar flow (%)	45	40
Base pressure (Pa)	$1.5 \times 10^{-3}$	
Work pressure (Pa)	0.4	
Substrate bias (V)	-50	
Graphite doped target (W/mm <sup>2</sup> )	0.075	

The chemical composition was determined by electron probe microanalysis (EPMA) applying a 10 keV voltage. The coating hardness was measured by depth-sensing indentation using a Berkovich indenter and a load of 5 mN. A series of 32 indentations from two distinct areas were carried out in order to critically evaluate the hardness of the coatings. In addition, the reduced Young modulus was derived from the indentation measurements using the standard method [41]. The Young modulus was calculated using the following parameters:  $E_i$ =1140 GPa and  $v_i$ =0.07 for the diamond indenter and v=0.3 for the coating [41].

#### 2.3. Rockwell C indentation and scratch test

The coating adhesion to the Ti substrates was evaluated using Rockwell C indentation and scratch testing where these are standard techniques commonly used to quantify the interfacial strength of coating-substrate systems. During the Rockwell C indentation test, a cone-shaped diamond 120° tip (200 µm in radius) was indented perpendicularly onto the coating applying a normal load of 200 N, thereby causing layer damage to the boundary of the indentation. The results of the test were qualitatively evaluated by comparing the optical microscope images of the crack network and the degree of delamination with an adhesion quality chart which classifies the images into six levels from HF1-HF6 [42]. In this classification, HF1 is featured with only a few minor cracks after indentation which indicates a good bonding while at the other extreme HF6 denotes extensive delamination of the coating and very poor adhesion. Adhesion levels from HF1 to HF4 are typically considered as acceptable coating adhesions for use in commercial applications.

The scratch test was performed by using a spherical Rockwell C diamond indenter (200 µm radius) according to standard testing methods [43]. The indenter was slid over the coating surface and the load increased from 2 to 50 N at a speed of 10 N/ mm. The lower critical load,  $L_{c1}$ , was defined as the load where the first cracks occurred (representing cohesive failure) and the upper critical load,  $L_{c2}$ , was defined as the load associated with the first delamination at the edge of the scratch track (representing adhesive failure). In addition, a load  $L_{c3}$  was defined as the load under which the coating was totally removed from the substrate. The scratch tests were repeated for three times in order to obtain reliable results. After scratch testing, the cracking or delamination of the coating was observed using optical microscopy along the scratch track. The cross-section profiles of the DLC coatings at the critical load were measured using an ultimate focus optical microscope operating in the line scanning mode.

#### 3. Experimental results

#### 3.1. Mechanical properties of Ti substrates

The results of optical microscopy and TEM showed that highpressure torsion processing significantly refined the grain size of pure Ti from ~8.6  $\mu$ m in the as-received state to ~130 nm after high-pressure torsion processing. Due to this grain refinement, the microhardness of pure Ti increased from an initial value of ~1.82 GPa in the coarse-grained state to ~3.05 GPa after highpressure torsion processing. This latter value for the ultrafinegrained pure Ti is comparable with the hardness of 3.09 GPa for the Ti–6Al–4V alloy used in this study [5].

The tensile testing demonstrated that the ultimate tensile strength (UTS) of Ti increased from  $\sim$ 660 MPa in the coarsegrained state to  $\sim$ 940 MPa after high-pressure torsion with some associated reduction in ductility [19]. This high strength and reasonable ductility of ultrafine-grained pure Ti is compatible with the Ti–6Al–4V alloy where the UTS is  $\sim\!980$  MPa and the elongation to failure is  $\sim\!14\%$ 

#### 3.2. Coating characterisation

Table 2 shows the zirconium content, the hardness and the values of the Young modulus of the deposited coatings. It should be noted that the low indentation depth for the indenter (approximately 10% of the coating thickness) made it difficult to ascertain the effect from the substrate material when obtaining these mechanical properties. Details of the coating structure were studied by X-ray diffraction, Raman spectroscopy and X-ray photoelectron spectroscopy and these results were reported elsewhere [33,44]. All of the coatings investigated in this study (with Zr contents of 7–9 at%) exhibited a nanocomposite structure with very small ZrC nanograins (up to  $\sim$ 2 nm) embedded within an amorphous carbon matrix.

#### 3.3. Adhesion performance of the DLC coatings

Table 3 summaries the adhesion results of DLC coating on all three substrates. The Rockwell C indentation tests showed HF3 and HF4 for all samples with only minor delamination and microcracking observed around the indentation marks, thereby demonstrating an acceptable adhesion for all tested coatings. The scratch test results showed that the DLC coatings with UFG Ti and Ti–6Al– 4V substrates had similar critical loads and these loads were much higher than with the coarse-grained Ti as substrates. Thus, the UFG Ti and Ti–6Al–4V substrates provided better support with their higher hardness and this produced a higher critical load of the DLC coating. These results demonstrate again the important role of substrates for the adhesion of thin DLC coatings.

Optical images of the Rockwell indentations are shown in Fig. 1. It can be seen that the load of 200 N led to radial plastic deformation of the coating which caused circumferential cracking of the film outside the indentation area. Through-thickness cracks were observed on all coatings which may be related to the elastic-plastic boundary of the substrates. The indentation marks had a diameter of  $\sim 380\,\mu{\rm m}$  on the DLC/coarse-grained Ti samples

#### Table 2

Summary of coatings chemical composition and mechanical properties.

System	Zr content (at%)	Hardness (GPa)	E (GPa)
DLC-7Zr DLC:H-7Zr DLC-9Zr	$7 \pm 1 \\ 8 \pm 1 \\ 10 \pm 1$	$\begin{array}{c} 11 \pm 2 \\ 13 \pm 1 \\ 12 \pm 2 \end{array}$	$\begin{array}{c} 120 \pm 3 \\ 117 \pm 4 \\ 131 \pm 3 \end{array}$

Table 3		
Summarv	of adhesion	results.

Substrate Coating		Rockwe	ell C	Scratch				
		Ø (µm)	Adhesion	L <sub>c1</sub> (N)	L <sub>c2</sub> (N)	L <sub>c3</sub> (N)	Track width at L <sub>c3</sub> (μm)	
CG Ti	DLC-7Zr	375	HF3	$7\pm 2$	$8\pm1$	$13\pm 2$	149.7	
	DLC:H-7Zr	391	HF4	$5\pm1$	$7\pm 2$	$13\pm1$	130.1	
	DLC-9Zr	384	HF4	$7\pm1$	$9\pm1$	$15\pm1$	130.7	
UFG Ti	DLC-7Zr	282	HF3	$8\pm1$	$12\pm1$	$22\pm 2$	113.3	
	DLC:H-7Zr	286	HF3	$7\pm 2$	$10\pm1$	$20\pm1$	120.9	
	DLC-9Zr	287	HF3	$10\pm 2$	$14\pm1$	$24\pm1$	123.8	
Ti-6Al-	DLC-7Zr	277	HF3	$9\pm 2$	$11\pm1$	$23\pm2$	139.7	
4V	DLC:H-7Zr	274	HF4	$5\pm1$	$10\pm 2$	$19\pm2$	132.1	
	DLC-9Zr	282	HF3	$11\pm 2$	$15\pm1$	$25\pm2$	126.2	

306

	CG Grade2 Ti	UFG Grade2 Ti	Ti -6Al -4V
DLC-7Zr		0	6
DLC:H- 7Zr			
DLC -9Zr		0	ОО 200 µm

Fig. 1. Rockwell indentation of DLC on different Ti substrates.

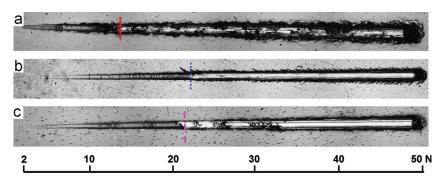


Fig. 2. Scratch tracks of DLC-7Zr coatings on (a) coarse-grained Ti, (b) ultrafine-grained Ti, and (c) Ti-6Al-4V substrates.

which would cause more deformation at the edge and more delamination compared to the DLC/ultrafine-grained Ti and DLC/ Ti–6Al–4V samples. This effect was mainly caused by the different hardness of the Ti substrates. Furthermore, the doping of H content led to a decrease in the adhesion strength because all the indents of the DLC:H-7Zr coating showed a higher delamination than DLC-7Zr. It is obvious that the DLC-9Zr coatings with higher Zr content presented the best adhesion behaviour with only small cracks and minor coating delamination. Therefore, it may be possible to further enhance the coating adhesion by increasing the zirconium content. A more comprehensive study of the adhesion of DLC coatings with different Zr content is given elsewhere [44]. The scratch test results matched well with observation from the Rockwell indentations. Fig. 2 shows the scratch tracks of the DLC-7Zr coating on the Ti substrates with increasing loads from 2 to 50 N. Generally, the plastic deformation of the Ti substrates was the main cause for the coating failure. As the diamond stylus was sliding with increasing normal load on the coating surface, the coating followed the deformation of the Ti substrate. The tensile stress both inside the coating and at the interface led to cracking and delamination of the coatings. As shown in Fig. 2(a), the DLC-7Zr coating on coarse-grained Ti failed at an early stage during the scratch test and the indenter caused extensive deformation of the substrate. On the other hand, the DLC-7Zr coatings deposited on ultrafine-grained Ti and Ti-6Al-4V failed at a higher

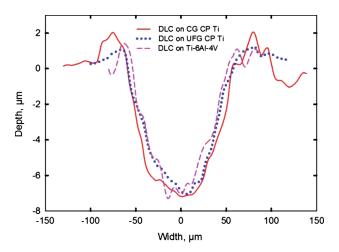


Fig. 3. Cross-section of the scratch tracks in the positions indicated in Fig. 2.

load and the adhesion of the films deposited on both substrates was almost identical (Fig. 2(b) and (c)). The cross-section of the scratch tracks where the substrate was revealed in the scratch track (see the short vertical lines in Fig. 2(a)-(c)) was measured using an ultimate focus optical microscope. Although the coating was deposited onto different substrates, these measurements showed that it always failed when the scratch track was approximately 7 µm deep and 130 µm wide as illustrated in Fig. 3.

#### 4. Discussion

In this study, a grade 2 pure Ti was processed using highpressure torsion to achieve significant grain refinement and then DLC coatings were successfully deposited on the different Ti substrates and their scratch behaviour was studied. In terms of future bio-medical applications, the results provide a clear demonstration that it is possible to replace the conventional Ti–6Al–4V alloy with ultrafine-grained pure Ti having preferable mechanical properties which may be achieved through highpressure torsion processing.

Various coating technologies are now available to provide wear resistance for bio-implants. Technologies such as plasma electrolytic oxidation (PEO) and internal oxidation (IO) have been used extensively to process Ti–6Al–4V and other bio-metals but these methods often involve a high processing temperature which may lead to recrystallisation of ultrafine-grained structures [45,46]. When ultrafine-grained materials are chosen as the substrates, the post processing temperature must not exceed the recrystallisation temperature in order to restrict any grain growth of the ultrafine-grained structures. Moreover, the oxidation layers produced through these methods are often very rough and this will entail additional polishing before their use in implants. Thus, PVD methods are regarded as one of the best choices for surface modifying ultrafine-grained materials.

A good adhesion between the thin coating and the substrate is essential for the coating/substrate system. The interposition of a gradient layer (Ti/TiN/TiCN) improves the interface shear strength and the load bearing capacity of the coating [28]. It was observed in this study that, when using ultrafine-grained Ti as the substrate, the load bearing capacity of DLC coatings was improved extensively compared to those with coarse-grained Ti as substrates. A similar trend was observed also on TiN thin coatings on different Ti substrates [19]. The TiN on ultrafine-grained Ti also had a higher critical failure load as it prevented the thin coating

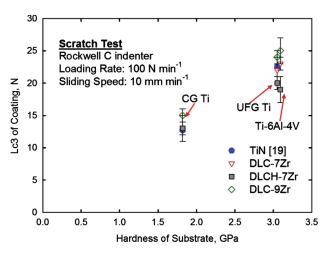


Fig. 4. Critical load of coatings versus the hardness of substrates.

from undergoing deformation. Fig. 4 shows the critical load of the DLC coatings and TiN coating on Ti substrates plotted against the hardness of the substrate. The trend is clear because a harder substrate after high-pressure torsion leads to a higher critical load for the coatings. Therefore, on one hand it is very important to explore better coating parameters such as new interlayer designs and coating compositions in order to achieve a good performance of the coatings. On the other hand, it is also very important to enhance the properties of substrates through surface hardening or grain refinement (i.e. increase the H/E ratio) and thereby give a better support to the thin coatings.

The results from this research showed that the DLC coating failures occurred at the same depth and width of penetration for all three substrates, as is clearly observed in Fig. 3. As the UFG Ti and Ti-6Al-4V had very similar hardness, therefore, the datum points were very close to each other. As shown in Fig. 3, the critical load of all coatings increased with increasing hardness of the substrates. A similar trend was also observed on various TiN coated substrates including stainless steel, high speed steel and WC [47]. An explanation was given by assuming the TiN coating adjusted to the elastic-plastic deformation of the substrates therefore the coating underwent a cohesive-adhesive failure that leads to film delamination. A critical indentation width of 60 µm was observed in that study, where the coating failed whenever it was bent to this width regardless of the substrate material.

By simplifying the coating-substrate system as a two-layer composite, the overall hardness of the coating-substrate system may be represented by the following relationship according to the Burnett–Rickerby model which is based on a volume law of mixtures [47–50]

$$H_{C,S} = \frac{V_C}{V_{total}} H_C + \frac{\chi^3 V_S}{V_{total}} H_S$$
<sup>(2)</sup>

where  $H_{C,S}$  is the apparent hardness of the coating-substrate system,  $H_S$  is the hardness of the substrate,  $H_C$  is the hardness of the coating,  $\chi$  is a factor by which the plastic zone changes, and the  $V_C$ ,  $V_S$  and  $V_{total}$  are the deformation volumes of the coating, substrate and total deformation volume, respectively.

The volumes of the deformation zones are expressed by the following equations:

$$V_{\rm C} \approx \pi R_{\rm C}^2 t \tag{3}$$

$$V_S \approx \chi^3 \frac{2}{3} \pi R_S^3 \tag{4}$$

where  $R_C$  and  $R_S$  are the radii of deformation zone of coating and substrate, respectively. The factor,  $\chi$  was determined via fitting experimental results by Burnett and Rickerby [48] and addressed as

$$\chi = \left(\frac{E_C H_S}{E_S H_C}\right)^{1/2} \tag{5}$$

where  $E_C$  and  $H_C$  are the Young's Modulus and hardness of the coating, and  $E_S$  and  $H_S$  are the Young's Modulus and hardness of the substrate, respectively.  $R_C$  and  $R_S$  can be obtained through the equation

$$R = r \left(\frac{E}{H}\right)^{1/2} \tag{6}$$

where *r* is the geometrical length of the indentation volume.

As observed in Fig. 3, the indentation depths of all samples were only around 7  $\mu$ m, and this depth was obtained after the diamond indenter was removed. In this case, the measured indentation depth was underestimated due to the elastic recovery of materials. Therefore, it is more accurate to recalculate the indentation depth using the measured indentation width considering the sphere shape of the indenter, according to the equation

$$r = \left[\pi \left(R_{ind}^2 - \frac{W^2}{4}\right) \left(R_{ind} - \frac{\sqrt{R_{ind}^2 - \frac{W^2}{4}}}{3}\right)\right]^{1/3}$$
(7)

where  $R_{ind}$  is the radius of diamond indenter and W is the track width at the critical load point.

Therefore, the overall hardness of this coating-substrate is achieved by rewriting Eqs. (2)-(6)

$$H_{C,S} = \frac{E_C t + \frac{2}{3} r H_S \left(\frac{E_C}{H_C}\right)^{3/2}}{\left(\frac{E_C}{H_C}\right) t + \left(\frac{2}{3}\right) r \left(\frac{E_C}{H_C}\right)^{3/2}}$$
(8)

During the scratch testing, the indenter was travelling with an increasing normal load and it was always the front half of the spherical cap in the coating-substrate system. Therefore, the critical load,  $L_{C3}$ , is expressed as

$$L_{C3} = H_{C,S} \pi R_{ind} \sqrt{R_{ind}^2 - \frac{W^2}{4}}$$
(9)

Thus, Eqs. (8) and (9) clearly emphasise the effect of substrate strength on the scratch behaviour of thin coatings. It further explains that the DLC coatings, with thicknesses around 1  $\mu$ m, often exhibit high critical loads with hard materials as substrates, such as more than 100 N on Si or glass [51].

The track width of each sample is listed in Table 3 and this can be applied to Eqs. (8) and (9). The comparison between experimental results and the Burnett–Rickerby model is plotted in Fig. 5. It worth noting that as the interlayers had varying compositions and properties and were much thinner than the coatings, thus this analysis simply neglects the properties of the interlayers although in practice the interlayer is important in improving the bonding. This may lead to an underestimation of the model. Moreover, errors may also be introduced due to the ridges formed at the edge of the scratch tracks which made it difficult to measure the track widths.

In this study, the strength and hardness of pure Ti were successfully improved by high-pressure torsion processing. Firstly, this improves the mechanical durability of pure Ti as the main body of bio-implants when they suffer fatigue and shear loadings. Secondly, ultrafine-grained pure Ti is a good substrate for thin coatings and provides improved load bearing capacity of the coating. Therefore, with good strength, fatigue life, excellent biocompatibility and no toxic release tendency of ultrafine-grained

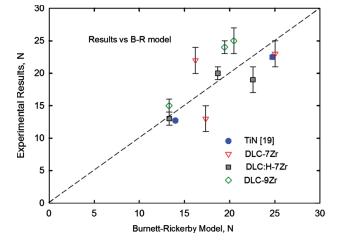


Fig. 5. Comparison of critical load between experimental results and calculation from Burnett and Rickerby model.

pure Ti, and with good wear resistance and an extremely low coefficient of friction of the DLC coatings, the DLC/ultrafine-grained Ti system appears to be an exceptionally strong candidate material for future bio-implant applications.

## 5. Conclusions

The adhesion behaviour and scratch resistance of three diamond-like carbon coatings deposited on grade 2 pure Ti substrates before and after high-pressure torsion processing were investigated and a comparison was made with a Ti-6Al-4Vsub-strate. The following conclusions are reached:

- 1. DLC coatings with a gradient TiN/TiCN interlayer show good adhesion on Ti substrates.
- 2. Hydrogen-free DLC-Zr coatings have better adhesion than hydrogen-doped DLC-Zr coatings, and increase in the percentage of Zr increases the adhesion.
- 3. The effect of substrate on the performance of the DLC coating under high load was highlighted, showing that coatings with ultrafine-grained pure Ti and Ti–6Al–4V substrates have similar scratch and indentation behaviour. Both are significantly better than the results obtained with coarse-grained Ti substrates.

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310

# PAPER IV

Adsorption of bovine serum albumin on Zr co-sputtered a-C(:H) films: implication on wear behavior

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# **Research Paper**

# Adsorption of bovine serum albumin on Zr co-sputtered a-C(:H) films: Implication on wear behaviour





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#### ABSTRACT

The use of protective coatings in biomedical field is an ongoing scientific challenge. Among different materials, carbon-based coatings are considered a potential surface treatment for orthopaedic implants. In this study, the effect of Zr incorporation in amorphous carbon coatings on the wear behaviour under protein containing lubrication was investigated. The coatings were deposited by dc unbalanced magnetron sputtering in Ar (non-hydrogenated) and Ar+CH4 (hydrogenated) discharges onto Ti based biomedical substrate. To improve the adhesion between the film and substrate a functional gradient Ti based layer was deposited ( $\sim$ 550 nm). The surface wettability was evaluated to assess the effect of the Zr and hydrogen content. The films with Zr were found to be hydrophobic enhancing the protein adsorption onto the surface; no significant differences were found when H was incorporated in the films. The adsorption layer characterized by X-ray photoelectron spectroscopy showed a well defined nitrogen peak originating from the organic layer. The tribological properties of the film were evaluated by unidirectional pin-on-disc testing with diluted bovine serum lubrication and physiological solution at  $37\pm3$  C°. The friction and the wear of the coatings were very low compared to uncoated substrates in both lubrication conditions. The ability of the surfaces to adsorb proteins was considered as the driving force for wear resistance acting as a protecting layer. In addition, the incorporation of Zr decreased the wear of the counterbody (Ti alloy) due to higher albumin adsorption.

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# 1. Introduction

Fatigue fracture and wear have been identified as the major problems associated with implant loosening, stress-shielding and ultimate implant failure. Demanding contact conditions coupled with the aggressive body environment lead to fatigue failure of almost all implant materials. The fatigue wear process causes the generation of wear debris which, by acute hosttissue reactions, tend to aggravate and speed up the failure of the biomaterial. Wear is a critical issue for prostheses, implants

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and other medical devices and its reduction is still an ongoing scientific and technological challenge. Diamond-like carbon (DLC) coatings have been widely studied to enhance implant performance due to its inertness, wear and corrosion resistance, hardness and excellent bio- and hemo-compatibility (Hauert 2003, 2004; Dearnaley and Arp, 2005; Love et al., 2013). In fact, DLC was already implanted as an artificial joint (Taeger et al., 2003; Joyce, 2007; Hauert et al., 2012a,b), surviving just 2-10 years due to delamination problems in-vivo. It is well known that DLC exhibits high internal stress which significantly limits the adhesion of the coating to metallic substrates. Thus, in order to overcome this problem, a metallic interlayer (Cr, Ti, Zr, Si, etc.) and/or functionally graded layers (Me/MeN/ MeNC or Me/MeC, Me corresponds to the metallic element) have been deposited between the metallic substrate and the DLC coatings avoiding abrupt changes in composition and diminishing the stress concentration (Choy and Felix, 2000; Thorwarth et al., 2010). The use of transition metal (Zr, Ti, Cr, etc.) co-sputtered amorphous carbon (a-C) films has also been widely studied as one possible solution to improve DLC performance (Chang et al., 2002; Corbella et al., 2005; Adelhelm et al., 2011). Moreover, the incorporation of hydrogen can further enhance structural changes by stabilizing the covalent bonding network (sp<sup>3</sup>) and playing a key role in the mechanical and tribological behaviour of the coating (Robertson, 2002). Even thought, under the highly corrosive human body environment the degeneration problem still persists promoting the failure of the coating (Hauert et al., 2012b). Actually, Hauert et al., (2012a) found that the main problem of DLC coatings is that, after some time in the body environment, the interface between substrate and functional coating can suffer a corrosion process by the penetration of body liquid through defects such as pinholes.

The exact interaction between biomaterials and natural fluids is still under extensive studies (Bauer et al., 2013). Immediately after implantation, water and ions from the body fluid are adsorbed and then a protein layer is formed onto the surface. Protein adsorption is indeed the first event which signalizes the overall biological response of the body to the implanted material (Andrade and Hlady, 1986; Roach et al., 2005; Rabe et al., 2011). A number of factors (surface chemistry, charge, topography, wetting behaviour, etc.) can alter protein conformation and/or orientation and consequently directly influence the cell response. For the joint implant point of view, proteins were also found to enhance lubrication through the adsorption of a protein layer on the joint materials surfaces (Heuberger et al., 2005; Serro et al., 2006). The natural lubrication typically minimizes the shearing damage and decreases the friction energy loss. However, when using artificial joint, the synovial fluid lubrication ability depends on many factors, such as surface treatment and the sliding conditions which could provoke catastrophic failure of the implant with partial or complete loss of functionality The effect of the synovial proteins on friction and lubrication is still unclear (Karimi et al., 2011; Runa et al., 2013; Myant and Cann, 2014), particularly when considering realistic surface engineering solutions. In fact, only few papers dealing with the influence of protein on (tribo)corrosion properties of DLC coatings (Hang et al., 2010; Liu et al., 2013; Wang et al., 2005) and even less for metal-containing DLC coatings (Maguire

et al., 2005; Escudeiro et al., 2011) can be found in literature. In the present study the effects of H and Zr incorporation in the a-C-matrix were analysed tribologicaly using physiological lubricants (0.9% NaCl and diluted foetal bovine serum) in order to predict its behaviour under such adverse environments. Additionally, the interaction of albumin with the surfaces was also accessed using X-ray photoelectron spectroscopy (XPS) together with wettability tests for surface chemistry characterization.

## 2. Materials and methods

#### 2.1. Coatings deposition

The coatings were deposited onto Ti grade 5 (Ti6Al4V) substrates and Si (111) wafers for the tribological testing and for coatings characterization, respectively. The Ti grade 5 substrates were mechanically polished with SiC paper (500, 600, 800, 1200 grits), and then finished by using diamond suspensions (6, 3 and  $1 \mu m$ ) and a colloidal silica suspension in order to obtain a mean substrate surface roughness (Ra) lower than 50 nm. Prior to deposition, the substrates were then cleaned in an ultrasonic bath in acetone, ethanol and deionised water for 15 min, and mounted on the rotating sample holder (18 rpm) in the deposition chamber. The coatings were deposited with a DC dual magnetron sputtering machine. A pure graphite target was used for the production of the a-C films in reactive (Ar/CH<sub>4</sub>) and nonreactive atmosphere (Ar), in order to produce hydrogenated and non-hydrogenated films, respectively. Zr pellets were added to the erosion zone of the graphite target (relative erosion area,  $A_{Zr}$ / A<sub>C</sub>, between 0 and 6%) to produce Zr containing films. All coatings were deposited with a constant applied bias voltage of -50 V and graphite target power density close to 7.5 W cm<sup>-2</sup>. Moreover, a pure Ti target was also sputtered for the deposition of a composite gradient interlayer (Ti/TiN/TiCN) to improve the film adhesion on the metallic substrates. For each deposition conditions, the deposition time was calculated to obtain films  $\sim$ 1.4  $\mu$ m thick. Further detailed deposition specification can be found elsewhere (Escudeiro et al., 2013). To facilitate reading, the coatings will be denominated as a-C\_Zr(X) and a-C:H\_Zr(X) for the non-hydrogenated and hydrogenated, respectively, where X is the Zr content.

# 2.2. Coating characterization

The chemical composition of the coatings was analysed by a full "Total Ion Beam Analysis (IBA)" (Jeynes et al., 2012) – Rutherford backscattering (RBS), non-Rutherford elastic backscattering (EBS), elastic recoil detection (ERD) and particleinduced X-ray emission (PIXE) self-consistently – used by the DataFurnace code (NDFv9.4f (Barradas and Jeynes, 2008)). This work was carried out at Surrey Ion Beam Centre, University of Surrey, UK. The analysis used an alpha particle beam of energies 3045 keV and 4315 keV with normal and tilted beam geometries, two backscattered particle detectors with different geometries, together with a forward recoil and an X-ray detector. Evaluated non-Rutherford particle scattering cross-sections (Gurbich, 2010) were used for H, C, N, O; those for H using the R-matrix parameters of the very Table 1 – Liquid's surface energy and their polar and dispersive components (Jie-Rong and Wakida, 1997; Żenkiewicz, 2007).

Liquid	$\gamma_{LV}$ (mJ m <sup>-2</sup> )	$\gamma_l^d$ (mJ m <sup>-2</sup> )	$\gamma_{l}^{p}$ (mJ m <sup>-2</sup> )
Water Glycerine (C <sub>3</sub> H <sub>8</sub> O <sub>2</sub> ) Formamide (CH <sub>3</sub> NO) Diiodomethane (CH <sub>3</sub> I <sub>2</sub> )	72.8 63.4 58.2 50.8	29.1 37.4 35.1 50.8	43.7 26.0 23.1 -

thorough treatment of Dodder et al. (1977) and those for C and N are described, respectively, by Gai and Gurbich (2013) and Gurbich et al. (2011). The He-PIXE used the LibCPIXE code of Pascual-Izarra et al. (2006) and the ionization crosssections of Taborda et al. (2011). SRIM2003 stopping (energy loss) cross-sections were used (Ziegler, 2004). For these samples all the information was in the 4315 keV data at  $15^{\circ}$ incidence angle, at which energy the 4263 keV  ${}^{12}C(\alpha,\alpha){}^{12}C$ resonance is excited giving high sensitivity at the surface to C in the presence of Zr. The ERD detector was at 30° scattering angle, and all four spectra (backscattering at 170° and 150°, ERD and PIXE at 120°) were interpreted self-consistently. The ERD detector relative solid angle was calibrated using a Kapton sample; the range foil thickness was 34 µm Kapton. The X-ray detector had a 146 µm Be filter to exclude backscattered particles. The PIXE showed the presence of Hf (assumed to be a contaminant at 0.85 at% relative to Zr), and of Ar (from the sputtering process: <5 at% relative to C).

The structure of the coatings was analysed by X-ray diffraction (XRD) (Philips, X'Pert diffractometer, Co K $\alpha$  radiation) whereas X-ray photoelectron spectroscopy (XPS) (ESCAprobe P, Omicron Ltd., Al K $\alpha$ , binding energy calibrated with Au 4f (84.1 eV)) was used to identify chemical bonding. The hardness was measured by depth-sensing indentation (Micromaterials Nanotest) using a Berkovich indenter. The normal stylus load was 5 mN (indentation depth approx. 150 nm); 32 independent indentations from two distinct areas on the sample were used to analyse the hardness data. Additional information is given in Escudeiro et al. (2013).

#### 2.3. Contact angle and surface free energy

The contact angle measurements were performed through the sessile drop method. Drops were generated with a Krüs GmbH G-23 goniometer at ~20 °C and room humidity (50%). A minimum of 5 drops were deposited on the surface and a sequence of images were acquired for the contact angle calculation. The surface energy was calculated by measuring the contact angle of various liquid solutions: water, glycerine ( $C_3H_8O_2$ ), formamide (CH<sub>3</sub>NO) and diiodomethane (CH<sub>3</sub>I<sub>2</sub>), see Table 1.

The surface free energy (SFE) was calculated using the Owens equation (Żenkiewicz, 2007). Wetting behaviour is governed by the Young equation:

where  $\theta$  is the measured solid–liquid contact angle and  $\gamma_s$  and  $\gamma_{SL}$  are the solid–liquid and liquid SFE. The surface energy ( $\gamma_s$ ) can be expressed as polar ( $\gamma^p$ ) and dispersive ( $\gamma^d$ ) components. The dispersive component is related to London interaction, arising from electron dipole fluctuation. Thus,

$$\gamma_{\rm SL} = \gamma_{\rm SV} - \gamma_{\rm LV} - 2\sqrt{\gamma_{\rm s}^d \gamma_{\rm l}^d} - 2\sqrt{\gamma_{\rm s}^p \gamma_{\rm l}^p}.$$
 (2)

Combining Eq. (1) with (2) we obtain:

$$\gamma_{\rm LV}(\cos\,\theta+1) = 2\sqrt{\gamma_{\rm s}^d \gamma_{\rm l}^d} + 2\sqrt{\gamma_{\rm s}^p \gamma_{\rm l}^p}.\tag{3}$$

Thus, the polar and dispersive component of the film surface energy can be calculated and, as a result, the total surface energy ( $\gamma_S$ ) is obtained.

#### 2.4. Protein adsorption

The interaction with bovine serum albumin (BSA) was studied by immersion of the samples in a 2 ml BSA containing solution  $(4 \text{ mg ml}^{-1})$  diluted with a basic solution (NaCl: 9 (g/l); EDTA: 0.2 (g/l); Tris: 27 (g/l); sodium azide: 0.3%, pH 7.6) for 24 h at 4 °C. The incubation time was chosen in order to take into account the Vroman effect and the tendency of the proteins to adjust its conformation with the surface (Andrade and Hlady, 1986). The amount of total immobilized protein was calculated using the Bradford reagent against a standard BSA calibration curve. This technique is a powerful technique to study protein-to-protein variability. However, it is influenced by the presence of interfering substances such as detergent and different ionic compounds (such as metallic ions) which prevents the precise quantification of proteins. Thus, a reliable qualitative result can be obtained by comparison between samples as soon as the experimental uncertainties are considered constant. After immersion, all the samples were washed with water Mili-Q for eventual detachment of non-chemisorbed proteins. The detached proteins were also taken into account for the protein quantification using the same method described above. The samples were then air-dried for 24 h before the XPS spectra were recorded. The XPS analysis was performed using a Kratos AXIS Ultra with VISION software for data acquisition and CASAXPS software for data analysis. The analysis was carried out with a monochromatic Al K $\alpha$  X-ray source (1486.7 eV), operating at 15 kV (90 W), in FAT mode (Fixed Analyser Transmission), with a pass energy of 40 eV for regions ROI and 80 eV for survey. Data acquisition was performed with a pressure lower than  $10^{-6}$  Pa, and a charge neutralisation system was used. To take into account shifts caused by charging of the sample surface, all spectra were adjusted taking the C 1s peak at 285.0 eV as a reference for the carbon contamination. The binding energy scale was charge referenced to the C 1s at 285 eV. The deconvolution of the spectra was performed using the CasaXPS programme, in which an adjustment of the peaks was performed using peak fitting with Gaussian-Lorentzian peak shape and Shirley type background subtraction. The spin-orbital splitting in Zr 3d was assumed to be the same for all phases and equal to 2.4 and the integrated intensity of the Zr 3d<sub>5/2</sub> peak relative to that of the Zr 3d<sub>3/2</sub> was considered equal to the spin-orbital multiplicity of 2/3 (Matsuoka et al., 2008).

#### 2.5. Tribological tests

The tribological tests were carried out using a pin-on-disc CSM tribometer in two different lubrication conditions: physiological solution (PS; 0.9% NaCl water solution), and foetal bovine serum (FBS), prepared according to the ASTM F732-00, (2006) standard test method. The temperature was maintained constant at 37  $\pm$  3 °C. A Ti6Al4V ball of 8 mm diameter was used as counterbody. An applied normal force of  $1\,N,$ linear speed of 20 cm  $\rm s^{-1}$ , and 10,000 cycles were employed. Tests on non-coated substrates were also performed under the same testing conditions for comparison purposes. The tribological behaviour was examined with respect to the friction coefficient and the wear rate; the latter was evaluated on the basis of 3D profile measurements on the wear track, whereas the wear rates of the balls were calculated from measurements of the spherical wear cap using optical microscopy.

# 3. Results and discussion

## 3.1. Coatings characterization

Table 2 shows the chemical composition obtained by IBA. The use of reactive (Ar+CH<sub>4</sub>) and non-reactive (Ar) sputtering led to the production of hydrogenated and non-hydrogenated coatings, respectively. The use of the reactive atmosphere promoted the incorporation of hydrogen in the coating in a range from 23 to 38 at% and almost doubled the deposition rate compared to non-hydrogenated coatings. The increase of the number of Zr pellets led to an increase of the Zr content and higher deposition rate, particularly for the nonhydrogenated coatings. The H content diminished with the increase of the Zr content. The density of the coatings was determined by IBA in  $\mu$ g/cm<sup>2</sup>; using the thickness of the coating measured by optical profilometer, it was possible to calculate the density of the coatings in g/cm<sup>3</sup>. The density varied from 2.6 to 3.9 for the non-hydrogenated coatings and from 1.9 to 3.1 for the hydrogenated coatings. The density is strongly related to the coordination defect content, H content, sp<sup>3</sup> bonding and lattice disorder, and, obviously, zirconium content (Charitidis, 2010). In general, the incorporation of H into the C-matrix led to the decrease of density compared to the non-hydrogenated films. Moreover, Zr co-sputtered films were harder and denser than pure films. Zr is a transition metal which present electrons at the outer shell loosely bound to their nuclei. Thus, the substitution of carbon atoms by Zr metal dopant in the rigid C–C and C–H network may distort the electron density distribution (Corbella et al., 2005) and, thus, decrease the coordination defect and increase the density. However, for Zr contents higher than 5 at% the density decreased. XRD diffractograms presented a weak and very broad peak close to ZrC (111) phase which indicates a nanocrystalline material with a grain size in the order of a few nanometres. Moreover, a nc-ZrC phase was also identified by XPS where the C 1s spectra showed a peak located at higher biding energy ( $\sim$  283.2 eV) compared to the typical Zr-C (281.8-282.3 eV) which is typical coatings with nanometric grain size. We can summarize here that, the increase

Table 2 – Chem	ical and mech	Table 2 – Chemical and mechanical properties of the coat	of the coatings.						
Name	Chemical composition	mposition			Dep. rate (nm/min)	Density (g/cm <sup>3</sup> )	Ra (nm)	H (GPa)	E (GPa)
	C (at%)	Ar (at%)	Zr (at%)	H (at%)					
a-C_Zr(9)	90.3	0.6	9.0	0.2	8.3	3.1	103	$10.7 \pm 0.5$	$133 \pm 2$
a-C_Zr(8)	88.1	3.0	7.5	1.5	7.3	3.9	88	$11.9\pm0.6$	$126\pm3$
a-C_Zr(4)	91.4	3.8	3.8	0.9	5.0	3.3	85	$10.4\pm0.4$	$110\pm3$
a-C	94.0	4.4	I	1.6	4.6	2.6	55	$10.7\pm0.5$	$94\pm1$
a-C:H_Zr(6)	67.6	1.9	5.7	24.8	12.0	2.8	76	$12.2 \pm 0.5$	$114\pm 3$
a-C:H_Zr(5)	71.3	1.0	4.8	22.9	10.0	2.3	55	$11.4\pm0.4$	$103\pm1$
a-C:H_Zr(3)	64.1	1.1	3.3	31.5	8.5	3.1	51	$10.0\pm1.6$	87±3
a-C:H	61.3	0.6	I	38.1	8.8	1.9	76	$8.9\pm0.3$	70土1

of Zr content led to the formation of nanostructure with ZrC nanocrystals embedded in the C-matrix. A detail study of the structural characterization of the Zr co-sputtered a-C films by XPS can be found in Escudeiro et al. (2013). The coordination number of C network can be then reduced by binding C atoms into carbide and, in accordance with thermodynamical models of DLC formation, resulting in the decrease of the local atom density in DLC film (Li et al., 2004; Wang et al., 2007).

#### 3.2. Contact angle and surface free energy

The interaction of the films with water is very important from the biomedical point of view. Numerous physiological events at subcellular and cellular levels, such as cell adhesion and protein adsorption, are greatly affected by such a property. The surface energy has been related to the adsorption ratio of diverse proteins and, consequently, cell spreading. Albumin is considered as a multifunctional transporter protein and the most abundant protein found in the plasma (approx.  $50 \text{ mg ml}^{-1}$ ) (Roach et al., 2005; Fanali et al., 2012). Moreover, its adsorption has been found to be related to the inhibition of the coagulation cascade and, consequently, platelet adsorption (Vogler, 1998; Lackner and Waldhauser, 2010). Albumin has three homologous domains (comparable amino acid sequences) assembled in a heart-shape structure which are sustained by mainly hydrophobic interactions, hydrogen bonds and disulphide bridges (Fanali et al., 2012). Thus, surface-protein interaction should be strongly related to surface chemistry. Table 3 shows the wettability characteristics assessed by contact angle measurements for selected coated samples. The wetting character of a surface can be obtained using water. Thus, high contact angle values imply a less wettable surface (hydrophobic surface) and, on the contrary, low contact angles values indicate a more wettable surface (hydrophilic surface). Pure non-hydrogenated and hydrogenated carbon films were characterized as hydrophilic coatings ( $\theta < 65^{\circ}$ ), which was in good agreement with the proposed contact angles found in literature (Robertson, 2002; Zhou et al., 2006). When zirconium was added into the C-matrix in increasing contents, the water contact angle increased suggesting hydrophobic surfaces. The exact water-interaction mechanism of alloyed DLC coatings is still not clear and furth er investigation is needed. However, albumin is known to have a higher binding affinity to hydrophobic surfaces due to hydrophobic interactions between the protein and the surface (Roach et al., 2005). Hence, Zr containing samples are expected to bond more proteins compared to pure films.

As expected, pure hydrogenated and non-hydrogenated coatings presented higher surface energy than the respective co-sputtered films. The incorporation of Zr led to a decrease of the surface energy due mainly to the reduction of the polar component. The metallic element can decrease the presence of unsaturated bonds and consequently decrease the dipolar interaction with water (Chen et al., 2001). Additionally, the presence of non-polar C-H bonds on the surface of hydrogenated coatings further decreased the interaction of the surface with polar molecules such as water and thus, for the same Zr content, the SFE was found to be lower compared to non-hydrogenated films. This fact is highlighted by the low polar component found in the Zr co-sputtered films, which strongly contributes to the hydrophilic functional chemical groups on the surface (Table 3). Each protein has a hydrophobic peptide backbone where the basis of polarity of R group emphasizes the possibility of functional role. The protein interfacial tension,  $\gamma_{BSA}$ , was calculated by the following equation (Paul and Sharma, 1981):

$$\gamma_{\rm BSA} = \left(\sqrt{\gamma_{\rm BSA}^d} - \sqrt{\gamma_{\rm s}^d}\right)^2 + \left(\sqrt{\gamma_{\rm BSA}^p} - \sqrt{\gamma_{\rm s}^p}\right)^2 + \Delta_{\rm BSAs},\tag{4}$$

where the value  $\Delta_{BSAs}$  describes the interdiffusion of ioniccovalent interactions which can be considered negligible. Further, if the interfacial tension approaches zero the interactions protein-surface are supposed to be lower. Albumin interfacial energy parameters are  $\gamma^d_{BSA}\!=\!31.4\,mJ\,m^{-2}$  and  $\gamma_{\rm BSA}^p$  = 33.6 mJ m<sup>-2</sup> (Paul and Sharma, 1981). Table 2 shows the calculated values for the coated surfaces. Low polar component and low polar/dispersive ratio presented higher interfacial tensions. In fact, many authors related a low ratio of polar to dispersive components with plasma protein adsorption (Baszkin and Lyman, 1980; Birdi, 1981). Additionally, Vogler (1998) found that the water adhesion tension,  $\tau_0$ , was the key parameter for biological reactivity of a biomaterial. The wettability is then measured by  $\tau_0$  and calculated as a product of water tension by the cosine of the measured water contact angle. Hydrophobic and hydrophilic surfaces are separated by Berg's limit ( $\tau_0$ =30 mJ m<sup>-2</sup>) which also limits the protein attraction/repulsion characteristic, respectively. Actually, the adsorption on hydrophilic surfaces was found to be thermodynamically unfavourable, i.e., the competition between water and proteins adsorption is an endothermic process (Paul and

Table 3 – Water contact angle ( $\theta$ ), surface energy ( $\gamma_s$ ), water adhesion tension values ( $\tau_0$ ) and BSA interfacial tension ( $\gamma_{BSA}$ ) of the deposited films and uncoated substrate (Azevedo et al., 2005; Pacha-Olivenza et al., 2008; Braic et al., 2011).

Samples	$\Theta_{ ext{water}}$ (°)	Surface ene	ergy (mJ m <sup>-2</sup> )		$\tau_0 ({\rm mJ}~{\rm m}^{-2})$	$\gamma_{BSA}$ (mJ m <sup>-2</sup> )
		γsp	γsd	γs		
a-C_Zr(8)	67±1	5.0	40.5	45.5	28.0	13.3
a-C_Zr(4)	66±3	5.1	45.8	50.9	30.1	13.8
a-C	$50\pm7$	11.2	51.2	62.4	46.8	8.4
a-C:H_Zr(6)	73±2	3.9	36.7	40.6	20.9	14.7
a-C:H_Zr(3)	65±2	12.8	27.2	40.0	31.0	5.1
a-C:H	57 <u>±</u> 2	14.2	33.9	48.1	40.2	4.2
Ti6Al4V	73±11	8.1	31.1	39.2	21.3	8.7

Sharma, 1981; Vogler, 1998). Once the adsorption is initiated, proteins tend maximize the surface interaction by exposure either hydrophobic domain (typically hidden toward the interior) or hydrophilic domains trough the surface. Therefore, the exclusion of water from the hydrophobic surface potentiates protein–surface interaction and, consequently, the hydrophobic interactions (except for the case when the protein in question has hydrophobic regions on its surface). Thus, the incorporation of Zr content led to lower  $\tau_0$ . Besides, the films with higher Zr content were found to have  $\tau_0 < 30 \text{ mJ m}^{-2}$  further highlighting the ability to adsorb protein. On the other hand, the adsorbed molecules through hydrophobic interaction can undergo reversible/irreversible conformational changes which may lead to unfavourable cell response if the proper binding domain is disrupted.

#### 3.3. Protein adsorption

In order to confirm the wettability results, the protein adsorption phenomenon was evaluated for the nonhydrogenated samples. Coated and un-coated samples were immersed for 24 h in a BSA containing solution and rinsed several times with ultra pure water in order to leave only the irreversibly bound proteins on top of the surfaces. The amount of protein absorbed was estimated using the Bradford protein assay and is shown in Fig. 1. As expected, pure amorphous coatings showed lower protein affinity compared to co-sputtered and uncoated surfaces. Although, Ti6Al4V presented the highest affinity to protein adsorption, the incorporation of Zr co-sputtered films significantly improved protein affinity when compared to "inert" a-C surface. Even though proteins often adsorbed as monolayers on metallic substrates (side-on and/or end-on), multilayer adsorption is not uncommon, particularly for high concentration solutions (Sousa et al., 2004; Serro et al., 2006). Taking into account the albumin size and molecular weight (Puskas et al., 2004), a close pack monolayer can be formed by approximately  $4 \text{ mg/m}^{-2}$  (Soderquist and Walton, 1980; Sousa et al., 2004). Thus, it is expected that in the case of Zr-containing samples albumin adsorbed as a multilayer coverage. This result corroborates the observation shown above: hydrophobic surfaces tend to bind more protein through "hydrophobic interactions" (Fig. 1). In general, the driving force for protein

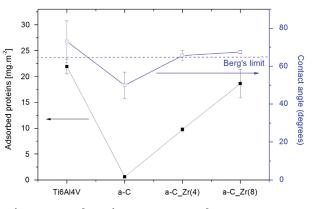


Fig. 1 - BSA adsorption vs contact angle measurements.

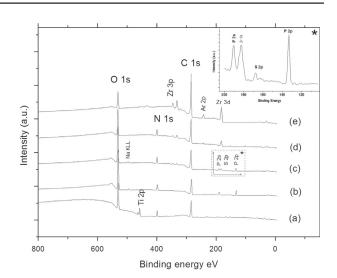


Fig. 2 – XPS spectra of the uncoated and coatings after immersion in BSA: (a) Ti6Al4V, (b) a-C, (c) a-C\_Zr(4), and (d) a-C\_Zr(8). a-C\_Zr(8) spectrum (e) before immersion was also added for comparison. The inset (\*) shows a zoom in of the 200–100 eV region of the a-C\_Zr(4) film XPS spectra revealing the S 2p photo-peak.

adsorption is the entropy gain resulting from dehydration of parts of the sorbent and the protein surface. a-C coating presented the highest surface energy together with the highest dispersive component; then, it should be expected to bond less protein compared to Zr co-sputtered films (hydrophobic surfaces). Moreover, the amount of protein desorbed after washing (i.e. not chemically adsorbed on the surface) was around 46% compared to 4% found for uncoated samples. It shows that the proteins are less tightly bound to the hydrophilic surface (Rabe et al., 2011).

XPS analysis was performed in order to prove the presence of the adsorbed albumin layer. Fig. 2 shows the XPS survey spectra for uncoated samples and non-hydrogenated coatings after immersion in BSA. In order to take into account the typical contamination layer, a-C\_Zr(7) XPS spectrum before immersion is also shown as a representative of all the samples in as-deposited conditions. In fact, after removing such contamination layer by argon sputtering (spectrum not shown here) the only visible changes were an increase in carbide bonds (for the Zr-containing samples) and a decrease in O-containing bonds, confirming the presence of a thin contamination oxide layer. The spectra show the peaks corresponding to oxygen (O 1s 532 eV), nitrogen (N 1s 400 eV), carbon (C 1s 285 eV) and sulphur (S 2p 164 eV, see Fig. 2 inset). The appearance of the well defined N 1s peak for all samples after immersion is usually attributed to the amino acids of the protein (Vanea and Simon, 2011; Gruian et al., 2012). However, it can be also related to the basic solution used for dilution. On the other hand, the presence of the S weak band can only belong to S-containing amino acids, i.e., methionine (Met) and cysteine (Cys). Indeed such amino acids correspond to around 7% of the total 538 amino acids residues that compose BSA (Hirayama et al., 1990). For Zrcontaining films another important feature was observed; the

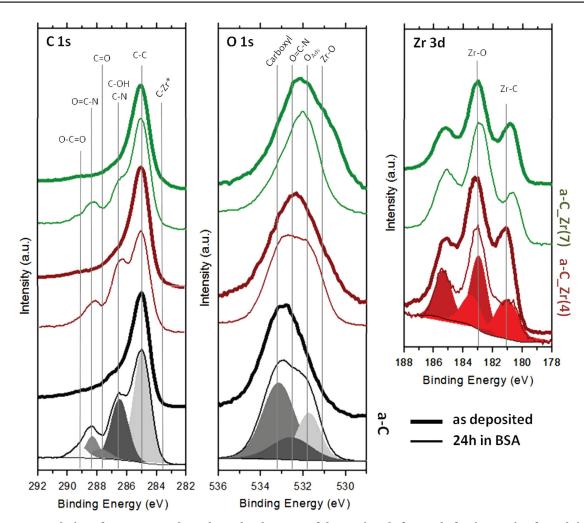


Fig. 3 - Deconvolution of C 1s, O 1s and Zr 3d core level spectra of the coatings before and after immersion for 24 h in BSA.

decrease in the intensity of the Zr 3d core level peak due to the presence of the organic adherent layer on the top of the sample (compare Fig. 2(d and e)). Sodium and phosphorus were also detected and considered as contamination from the basic solution used to dilute BSA.

The deconvolution of the peaks in C 1s, O 1s and Zr 3d core level spectra of the films, before and after BSA immersion, is shown in Fig. 3. Before immersion, all C 1s spectra were fitted using the alkyl type carbon (C-C, C-H) at 285 eV as charge reference; a second peak at 286.5 eV was added with the same FWHM as the main peak, indicating the alcohol (C-OH) and/or ester (C-O-C) functionality. Two other components can also be detected corresponding to the C=O and O-C=O at 2.8-3.0 eV and 3.6-4.3 eV, respectively (Kaufmann et al., 1988). When Zr was incorporated in the matrix an extra peak was also observed close to  $283.4\pm0.2\,\text{eV}$  attributed to C–Zr\*, i.e. Zr-C bond in the nanocrystal as referred to above (Escudeiro et al., 2013; Meng et al., 2013). After immersion in BSA an additional peak appeared at  $288.3\pm0.1\,\text{eV}$  attributed to the O=C-N groups from the peptide backbone (Serro et al., 2006; Premathilaka et al., 2007; Vanea and Simon, 2011; Gruian et al., 2012). O 1s band also revealed an extra peak at 532.6 eV after protein adsorption confirming the presence of such bond together with two other peaks revealing oxygen contamination (531.7 eV) and carboxyl groups (533.1 eV) on the film surface (Premathilaka et al., 2007). Again, for Zrcontaining films a shoulder around 532.1 eV is observed and identified as Zr-O bond. The Zr 3d spectra show the presence of both Zr-C (181.2 eV) and Zr-O (183 eV) bonds typically found for Zr-conatining DLC films (Escudeiro et al., 2013; Meng et al., 2013). After immersion in BSA, a decrease in intensity of the Zr-C bonding component is well visible indirectly supporting existence of protein layer.

Although XPS does not provide quantitative information about the total amount of adsorbed protein, it is commonly monitored by the intensity of N peak before and after protein adsorption (Serro et al., 2006). The presence of N was imperceptible before the immersion in BSA. Fig. 4 shows the N 1s core-level spectra for all measured samples after 24 h immersion in BSA. As expected, the intensity of the N 1s peak varies in a similar way as protein adsorption. The deconvolution of the peaks (not shown) revealed one component close to 400.2 eV characteristic of O=C-N groups from the peptide bonds (Serro et al., 2006; Gispert et al., 2006; Premathilaka et al., 2007; Vanea and Simon, 2011; Gruian et al., 2012) and an additional peak near to 401.8 eV which can be attributed to a protonate amine group (-NH<sup>+</sup>) of the terminal amino groups (Auditore et al., 2002; Ahmed et al., 2013; Lubambo et al.,

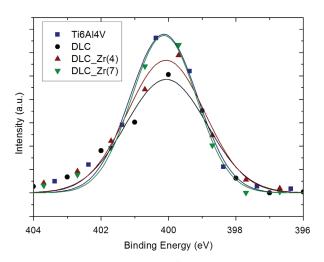


Fig. 4 – N 1s core level spectra of Ti6Al4V and nonhydrogenated coatings after immersion in BSA.

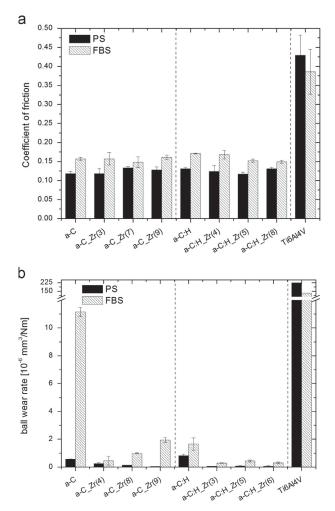


Fig. 5 – Average friction coefficients (a) and ball wear rate (b) of the coated and uncoated samples under both lubrication conditions.

2013). This may indicate that the BSA will bind through carboxyl acid group rather than amino group (Ueda et al., 1976).

#### 3.4. Friction and wear

To identify a promising coating composition for the articulating joints, forensic tribological screening tests (unidirectional pin-on-disc tests) were performed using a corrosive lubricants (0.9% NaCl, physiological solution – PS) and a protein containing lubricant (Fetal Bovine Serum – FBS). The use of physiological solution was chosen in order to create a synergistic effect between the wear and the corrosion due to the presence of water and ions that may accelerate material degradation (Wang et al., 2005; Kim et al., 2008). On the other hand, FBS was used in order to approach the physiological conditions.

Fig. 5 presents the friction data from the tribological tests in PS and FBS. Surprisingly, the friction coefficient of coated samples tested in PS was found to be similar to that of dry sliding (Escudeiro et al., 2013) and lower than that measured in FBS. Moreover, the incorporation of Zr did not lead to any statistically significant difference in friction among the coatings. In all cases, the worn surface did not show any signs of film failure (Fig. 6). In general, all coatings present very low wear rate (~ $0.5 \times 10^{-6}$  mm<sup>3</sup>/Nm) compared to the uncoated surface under both lubrication conditions (7.2 ×  $10^{-4}$  mm<sup>3</sup>/Nm

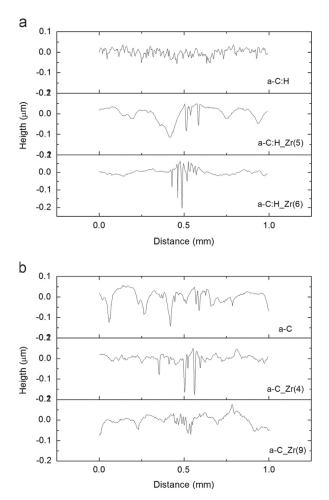


Fig. 6 – Comparison between the wear tracks (cross-section) of hydrogenated and nonhydrogenated films under PS lubrication.

and  $5.7 \times 10^{-4} \text{ mm}^3/\text{Nm}$ ) for PS and FBS, respectively, see Fig. 7). The incorporation of Zr did not significantly improve the wear compared to pure carbon coatings. Nevertheless, the wear of the counterbody was strongly reduced when testing against doped films (Fig. 5).

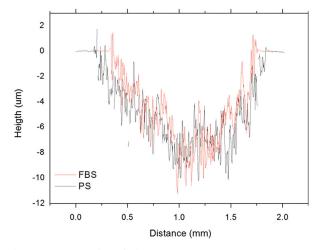


Fig. 7 – Wear profile of Ti6Al4V sample tested under PS and FBS.

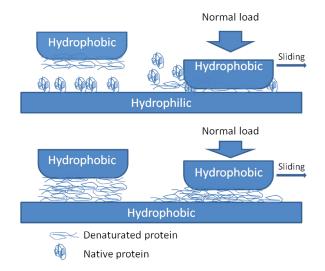


Fig. 8 – Schematic representation of the key mechanisms of the albumin-mediated lubrication on DLC films.

When tested in highly corrosive medium (PS), it is expected that the production of the wear products and their accumulation in the wear track precipitate abrasion wear and delay the repassivation (Kim et al., 2008). The co-sputtered Zr films are known to present good wear-corrosion behaviour. Kumar et al. (2013) showed similar or better corrosion properties of Zrdoped DLC coatings tested in Hank's solution compared to the Ti alloy substrate. Also, Wang et al. (2005) reported an excellent crevice and pitting corrosion resistance of Zr-DLC films. It is thus expected that the incorporation of Zr enhances the corrosion resistance compared to unalloyed films due to its ability to form an oxide layer as observed for Ti and Cr by Wang et al. (2013). The decrease of both wear debris amount and accumulation of corrosion products between the mating materials can consequently decrease the wear of the counterpart. On the other hand, when testing under FBS, the presence of proteins can play two roles in the sliding systems: (i) they act as a lubricant and (ii) they decrease the degradation process by forming a complex adsorbed film (Wimmer et al., 2010). Moreover, protein also can interact with the metallic debris/ions forming metalloprotein complexes that may be processed or eliminated in vivo (Hallab et al., 2001). Proteinrich lubricants are known to improve the stability of the passive film on metallic substrates (SS 316L, Ti6Al4V alloy and CoCrMo alloy) acting as a corrosion barrier layer and minimizing the surface degradation (Karimi et al., 2011; Runa et al., 2013). The presence of adsorbed proteins protected not only the coated surface, decreasing significantly its wear rate (negligible worn volume, see Fig. 9), but also the counterbody (Fig. 5), particularly when rubbing against Zr-containg coatings. Adding a metallic element to the C-matrix led to higher protein adsorption compared to a-C pure films (see Section 3.3), which could increase (tribo) corrosion resistance of coating-substrate system (Karimi et al., 2011; Runa et al., 2013).

For all coatings the friction coefficient using FBS as lubricant was higher compared to PS; nevertheless, it was still significantly lower compared to that of uncoated substrates (~0.38). The friction coefficient increased up to ~2000 cycles and then oscillated around an average value (~0.16). The same tendency was observed for DLC films co-sputtered with Ti (Escudeiro et al., 2011) and Si (Anil et al., 2010). Fig. 8 shows a schematic representation of the albumin-mediate lubrication on DLC-based films. Although albumin can undergo conformational change due to adsorption on hydrophobic surfaces, the adsorbed layer is more tightly bound

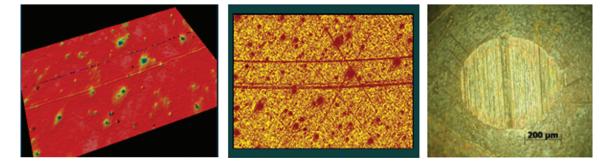


Fig. 9 – Wear track profile and ball micrograph of a-C:H film where only few scratches on the surface could be observed, which contrasted with the wear damage of the ball counterparts showing clear abrasive marks.

compared to hydrophilic surfaces (Hang and Qi, 2010). Thus the wear of the counterbody was particularly decreased by application of Zr co-sputtered coatings. The hydrophobic character of such films led to the adsorption of a robust protein layer onto the surface (high adsorption rate), which prevented the surfaces to rub in direct contact. However, it was also noticed from the wear scars inspection (Fig. 9) that higher surface roughness (Table 2) can be disruptive for such layer resulting in accelerated wear of the counterbody.

## 4. Conclusions

Zr co-sputtered amorphous films were deposited by DC magnetron sputtering under reactive (Ar+CH<sub>4</sub>) and non-reactive (Ar) atmosphere and tested in lubricated contact. Zr was added in small amounts (3-9 at.%) forming dense nanostructured coatings composed of ZrC nanocrystalls embedded into an amorphous C-matrix. The incorporation of H did not show any significant differences compared to non-hydrogenated coatings. Zr-alloyed coatings showed higher contact angle (and therefore lower surface free energy) than a-C(:H) ones which enhanced protein adsorption onto the surface. XPS measurements further indicated that albumin adsorbs better on the surface of Zr-doped coatings. When tribologically tested in PS, Zr-doped films behaved similarly to pure carbon films. However, the wear behaviour in FBS lubrication clearly indicated strong dependence on the ability of the surface to adsorb proteins. Although the wear of all coatings was negligible, higher protein adsorption rate of hydrophobic surfaces led to lower counterbody wear due to the presence of a robust protein layer.

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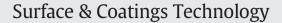
# PAPER V

# Sliding Properties of Zr-DLC Coatings: The effect of Tribolayer Formation

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# Sliding properties of Zr-DLC coatings: The effect of tribolayer formation



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## ABSTRACT

Amorphous carbon-based coatings (DLC) exhibit excellent mechanical and tribological properties such as high hardness, high elastic modulus, low friction and low wear. Reduced friction is often related to the formation of a low-friction tribolayer, which is formed during sliding and transferred to the counterpart. Here, we investigate the sliding of hydrogenated and non-hydrogenated DLC coatings alloyed with zirconium; pure DLC films are used as reference. The coatings were deposited by magnetron sputtering in Ar (non-hydrogenated) and Ar/methane atmosphere (hydrogenated) onto steel substrates and silicon wafers. The total thickness of the coatings was around 1.5  $\mu$ m including a complex Ti/TiN/TiCN adhesion-improving interlayer with a thickness of 450 nm. All deposited coatings were amorphous, Zr/C ratio was approx. 0.05. The hardness was in the range of 9–13 GPa. Tribological tests were carried out in humid air at room temperature, at 100 °C and in nitrogen environment using pin-on-disk. Intermittent tribological test analysis has been performed to understand running-in behaviour. The worn surfaces and wear debris were analysed by Raman spectroscopy. Coatings alloyed with Zr showed lower friction and wear at room temperature compared to pure DLC. In general, Zr-doped hydrogenated coating outperformed the non-hydrogenated one when tested in an inert nitrogen atmosphere or at elevated temperature (100 °C), exhibiting almost super-low friction ( $\mu = 0.03$  in the steady-state regime) due to the formation of a homogenous, thick and stable tribolayer.

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## 1. Introduction

It is well known that amorphous carbon-based coatings (typically called as diamond-like coatings, DLC) exhibit unusual mechanical and tribological properties such as high hardness, high elastic modulus, low friction and low wear [1–3]. DLC is used in a wide range of industrial applications in different environmental conditions [4–10]. However, the coating applicability might be limited due to their high level of residual stress. Co-deposition of an appropriate adhesion interlayer, respecting the chemical and mechanical properties of the substrate and coating, and introduction of metallic elements (Cr, Ti, W, etc.) into the C-matrix can successfully reduce the residual stress and enhance the adhesion [3,6,11,12]. Such coatings are often denoted in literature as Me-C(:H) or Me-DLC coatings, where Me stands for the metallic element.

The tribological properties of pure DLC coatings are strongly dependent on their structure and testing environmental conditions. Basically, the DLC coatings deposited on temperature sensitive substrates, such as steel, can be divided into two main groups — hydrogenated films deposited either by PECVD or by reactive magnetron sputtering (a-C:H) [1,2] and non-hydrogenated films prepared by non-reactive magnetron sputtering (a-C). Non-hydrogenated coatings generally perform better in humid environments, while hydrogenated coatings are excellent in dry sliding and inert gas conditions reaching friction values in the range of 0.001–0.2 [3,13,14]. At elevated temperatures ( $\geq$  300 °C), hydrogenated DLC films may gradually degrade into graphite and hence wear through quickly, while non-hydrogenated coatings are more resistant to oxidation or phase transformation [2.15–17]. The formation of a carbon-rich tribolayer between the surface of DLC coating and the counterpart is a typical product of dry sliding positively affecting the tribological properties [14,18]. A stable tribolayer suppresses friction fluctuations and lowers both the friction and the wear [19]. On the other hand, hard abrasive wear debris particles present in the tribolayer could lead to abrasive wear; consequently, the friction process becomes instable with high fluctuations of the friction coefficient [14,18,20]. The tribolayer composition and mechanisms of generation and growth are a function of the coating and counterpart properties and, particularly, strongly depend on the testing conditions, namely temperature, moisture, and presence of the environmental reactants (ambient air gases or other chemical compounds) [21,22].

The introduction of additional element (e.g. Ti [4,23–28], W [4–6,29], Si [30–34]) into both types of carbon matrix referred to above leads to profound changes of coating properties, such as structure, morphology, surface free energy, and mechanical properties. Doped or alloyed DLC

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coatings often exhibit improved tribological properties not only by lowering the friction and wear but also by the stabilisation of the tribological properties in different testing environments [35,36]. Scharf and Singer [14] reported that for metal-doped DLC films the tribolayer formation and growth was dependent on the metal dopant and the atmosphere humidity. DLC films co-sputtered with zirconium have been recently studied by several authors [37-39] showing that low content of Zr (<10 at.%) can effectively reduce the friction and wear of the coating compared to un-doped films. In this paper, the structural, mechanical and tribological properties of both non-hydrogenated and hydrogenated Zr-doped DLC coatings were evaluated. The main attention is paid to the mechanisms of surface tribolayer formation, its composition, compactness, and other properties affected by tribological testing conditions. The tribological measurements were carried out at room temperature (RT) and at 100 °C and in nitrogen or humid air atmosphere in order to identify the environment effect on the tribological properties.

### 2. Experimental details

The Zr-DLC(-H) films were deposited using dc dual magnetron sputtering. Two targets were used for the deposition of pure and cosputtered Zr films: a titanium target, for the deposition of the adhesion layer, and a graphite target with selected number of Zr pellets embedded in the target erosion zone. The Zr content in the films was controlled by the number of Zr pellets. The Zr content was selected with regard to the results of our previous study [40], where the best mechanical and tribological properties were observed in the range of 3-5 at.% of Zr. All the substrates (mirror polished AISI D2 steel coupons) and targets were sputter-cleaned before the deposition process started. To deposit hydrogenated films, CH<sub>4</sub> was introduced into the chamber. The flows of argon and methane were controlled with the aim of reaching the same total working pressure (0.4 Pa). The deposition parameters are summarized in Table 1. A Ti/TiN/TiCN functional (~450 nm thick) adhesion interlayer was deposited first to improve the adhesion of the coatings to the substrate.

The coating composition was analysed by Electron Probe Microanalysis (EPMA-Cameca SX 50). The structure of the coatings was determined in glancing-mode X-ray diffraction (Phillips diffractometer-Co  $K_{\alpha}$  radiation). The as-deposited carbon matrix bonds as well as the tribology-induced surface changes were studied by Raman spectroscopy (Horiba Jobin Yvon microscope, 532 nm). The Raman spectra have been fitted using Lorentzian peaks; total peak areas have been used to calculate the peak intensity ratio.

The coating hardness was evaluated by Micro Materials nano-test platform. To avoid the influence of both substrate and TiN/TiCN adhesion interlayer a normal load of 5 mN was applied reaching a maximum indentation depth of 130–160 nm. The adhesion of the coatings was

#### Table 1

(Zr-)DLC(:H) coating deposition parameters, composition and mechanical properties.

determined using a conventional scratch-test (CSEM Revetest); the critical load was assessed by optical-microscope observation.

The surface roughness  $R_a$  of the samples was determined using 3D profilometer Zygo NV7200; 20 surface profiles were used and the standard deviation was calculated.

Wettability tests were performed using the sessile drop method. The drops were generated by a micro syringe and deposited on the surface for at least 20 times. A sequence of images was captured using a DMK-41BU02 digital camera mounted horizontally on a robust stand. The tribological measurements were carried out using a pin-on-disk tribometer (CSM Instruments). The coatings were tested at ambient air at room temperature (RT) (25 °C, relative humidity  $40 \pm 5\%$ ), at 100  $^{\circ}$ C and N<sub>2</sub> environment. The normal load was 5 N, the duration of the test was 1000 cycles (N2 environment), 5000 (RT and 100 °C) and 30,000 (for lifetime testing), and the relative ball vs. coating velocity was 0.15 m $\cdot$ s<sup>-1</sup>. The testing counterparts were 100Cr6 steel balls with a diameter of 6 mm. The coating wear tracks and the ball wear scars were analysed by 3D non-contacting profilometer Zygo NV7200. The coating wear volumes were calculated from the average of four wear track cross-sections obtained in different positions along the wear track. The difference between the maximum and the minimum measured coating wear volume in one wear track did not exceed 25%, which was in a good agreement with several statistic models in tribology [41,42]. The ball wear volume was calculated from the wear scar 3D analyses using geometrical idealization of the original ball surface. The wear rates were calculated as the worn volume per load per sliding distance [42]. To understand wear mechanisms, particularly running-in stage, the contact pressure evolution was studied by ball contact area determined during the intermittent tribological test. Furthermore, the wear tracks, the ball wear scars and the generated wear debris were analysed by optical microscopy and Raman spectroscopy. The Raman analyses were performed in selected points within and around the coating wear track and counter-part contact area (Fig. 1). During the intermittent testing, Raman inspection was also performed at RT in order to evaluate the running-in behaviour.

#### 3. Results

#### 3.1. Coating characterization

The Zr/C ratio representing the chemical composition of the coatings is shown in Table 1. The oxygen content that originated from residual atmosphere in the deposition chamber was negligible; therefore we report only Zr/C ratio.

Fig. 2 shows the XRD diffractograms of the amorphous carbon films where the only visible peaks were associated to TiN/TiCN adhesion interlayer. The peaks corresponding to the positions of Ti (002), Ti (101), TiN (111), TiN (200) and TiN (220) are clearly visible in Fig. 2.

Parameter\sample			DLC	DLC-H	Zr-DLC	Zr-DLC-H
Deposition parameters	Ar flow	(sccm)	45	40	45	40
	CH <sub>4</sub> flow	(sccm)	-	17	-	17
	Bias voltage	(V)	-50	- 50	-50	- 50
	C/Zr target power	$(W \cdot cm^{-3})$	1500	1500	1500	1500
	A <sub>Zr</sub> /A <sub>C</sub> erosion area	(%)	0	0	3	6
	Deposition rate	(nm/s)	5.8	12.7	7.4	13.1
Composition	Zr/C ratio	(-)	0	0	0.05	0.05
Mechanical properties	Thickness	(µm)	1.1	0.9	1.1	0.9
	Critical load L <sub>c2</sub>	(N)	$43 \pm 3$	$22 \pm 2$	$42 \pm 9$	35 + 3
	Hardness	(GPa)	$12.3 \pm 0.5$	$10.1 \pm 0.3$	$8.8 \pm 0.4$	$13.3 \pm 0.6$
	Elastic modulus	(GPa)	$120 \pm 3$	$89 \pm 1$	$102 \pm 1$	$116 \pm 1$
	Surface roughness R <sub>a</sub>	(nm)	$16.9 \pm 3.3$	$14.3 \pm 4.6$	$18.4 \pm 5.6$	$18.9 \pm 4.5$
Wettability	Water drop contact angle	(deg)	$61.5 \pm 3.4$	$70.3 \pm 2.5$	$66.2 \pm 2.4$	79.3 ± 1.8
Tribological properties	Initial contact pressure	(MPa)	930	810	870	920
I	Contact pressure at 5000 cycles	(MPa)	80	140	110	160

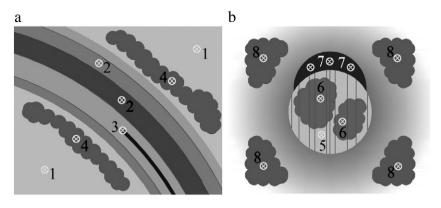


Fig. 1. Schematic plot of the points selected for Raman analyses: a) the wear track, b) the counter-part wear trace. In the text the points are described as: as-deposited surface (1), wear track area (2), wear track embedded particles (3), wear track free wear debris (4), ball contact area (5), tribolayer (6), leading edge wear debris (7), ball free wear debris (8).

As expected for such low Zr content, there was no evidence of ZrC crystalline phase. However, compared to the spectra of non-doped films, the Zr-doped coatings exhibited a very broad peak in the range of 30 to 40°. We have shown recently that the coatings with 9 at.% already contain ZrC nanograins [40], which corroborate the results of Adelhelm et al., who showed existence of ZrC in coatings with approx. 7 at.% of Zr [43]. It is very probable that Zr content presented in this study is close to the limit where the first ZrC nanograins (or sub-nanograins) are formed. Nevertheless, for the purpose of this particular study the structure of Zr-DLC(-H) coatings could be considered as amorphous.

The Raman spectra of as-deposited Zr-DLC(-H) coatings are shown in Fig. 3. The carbon matrix exhibited the typical D and G bands located in the range of 1300–1600 cm<sup>-1</sup>. When using visible Raman excitation (532 nm), the carbon D and G peaks are results of vibration of  $sp^2$  sites; in other words, the sp2/sp3 ratio cannot be determined. The hydrogenated structures showed a more pronounced G peak corresponding to the prevailing axial vibrations of carbon chains and rings rather than "breathing" mode of the rings that corresponds to the D peak [44,45]. Slight background photoluminescence (PL) effect was also observed. This effect is typical for hydrogenated DLC structures due to the existence of recombination centres in the structure by the presence of hydrogen [46]. The H content can be thus estimated by the ratio between the slope of the linear background and the intensity of the G peak [47]. The calculated H content was approximately 30 at.% which was in a good agreement with the ERDA/RBS analysis of DLC-H coating (not shown). When the Zr-dopant was introduced, the G peak positions decreased in conjunction with the  $I_D/I_G$  ratio growth (Fig. 4). Both effects are closely associated with the carbon bond disorder [44,45].

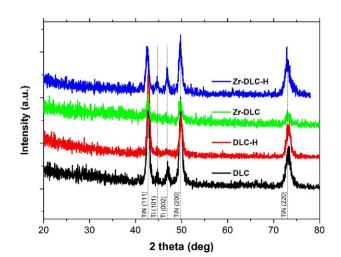


Fig. 2. X-ray diffraction spectra of Zr-DLC(-H) coatings.

Similar results were obtained by Adelhelm et al. [48] who correlated this behaviour with the presence of heavy Zr atoms.

The Zr-DLC(-H) coating hardness and elastic modulus results are shown in Table 1. The hardness values are placed in a relative narrow range between 9 and 13 GPa and, thus, any significant effects of the coatings composition on the mechanical properties could be hardly defined. Nevertheless, the highest hardness was obtained for the Zrdoped hydrogenated DLC structure suggesting a low proportion of non-bonded hydrogen and a denser carbon matrix [49,50]. Narrow range of the elastic moduli was similar to that of hardness. Apart from the hardness and elastic modulus, the coating surface roughness is considered as one of the important properties from the tribological point of view. Erdemir et al. [13] showed that higher roughness led to more extensive ploughing effect and thus increasing the friction coefficient. In our study there were no significant differences in roughness among the coatings (Table 1). However, higher water contact angle and thus lower surface energy was observed for the hydrogenated DLC surfaces. The same trend was obtained when the Zr atoms were introduced into the carbon structure - the surface energy decreased.

#### 3.2. Tribological performance

#### 3.2.1. Friction and wear

Fig. 5 shows typical friction curves of the deposited coatings tested at RT (Fig. 5a) and 100 °C (Fig. 5b). The friction at RT was relatively stable showing a clearly distinguishable running-in period of about 1500 cycles. The evolution of contact pressure during the tests is shown in Fig. 6. Intermittent tribological test analysis of the wear of the ball contact area clearly proved that the running-in period was closely related with rapid decrease of contact pressure. The shortest

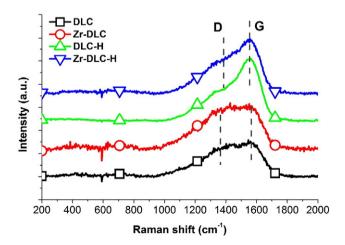


Fig. 3. Raman spectra of as-deposited Zr-DLC(-H) coatings.

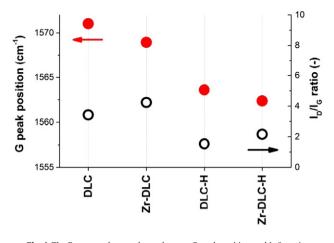


Fig. 4. The Raman carbon peaks analyses – G peak position and  $I_D/I_G$  ratio.

running-in phase was observed in the case of Zr-DLC-H coating — about 50 cycles. Then the contact area remained almost constant till the end of the sliding test.

Both hydrogenated and non-hydrogenated Zr-DLC coatings exhibited lower friction coefficients with much lower fluctuations than the coatings without Zr (Fig. 5a). At 100 °C, the friction of the nonhydrogenated coatings was rather instable with high fluctuations. In this case, there was significant difference between the films with and without hydrogen (Fig. 5b). Hydrogenated coatings showed much lower friction at the elevated temperature than at room temperature. Zr-DLC-H coating exhibited remarkably low and stable friction coefficient close to 0.03 in steady-state wear regime (Fig. 5b). For a more precise running-in phase study and determination of the role of the temperature and reaction with atmosphere, the coatings were tested in an inert atmosphere (N<sub>2</sub> environment). The friction evolution is shown in Fig. 7. As expected, the non-hydrogenated coatings did not show any significant difference compared to the tests performed in humid air at room temperature. On the other hand, Zr-DLC-H coating clearly demonstrated the frictional behaviour similar to the test at 100 °C, i.e. very low friction in the steady state regime. However, strong fluctuations were recorded during the Zr-DLC-H running-in phase.

The wear results for both coatings and counter-parts tested in ambient air are shown in Fig. 8. At RT the incorporation of metal and H did not significantly influence the wear performance, although slightly lower coating wear rates were obtained for Zr-doped structures. At elevated temperature the non-hydrogenated Zr coating showed the highest wear rate and the highest effect of testing environment on the wear. It is important to notice here that the depth of the wear tracks

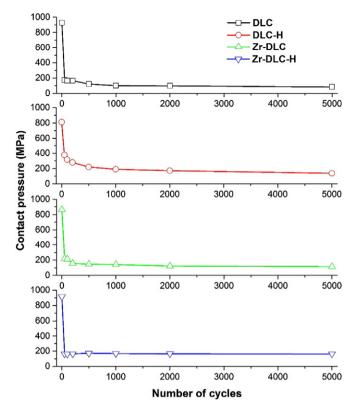


Fig. 6. Evolution of contact pressure with number of cycles calculated according to the ball contacting area – tribology at RT and ambient environment.

did not exceed 500 nm, i.e. less than a half of the total coating thickness. Furthermore, the tests in the  $N_2$  atmosphere did not show any measurable coating wear; only shallow scratches could be observed. The worn surfaces were analysed by Raman spectroscopy; the results are discussed in the following section.

The ball wear scars were characterized by shallow abrasive scratches parallel to the relative movement and a well-adhered surface tribolayer. The ball wear rates exhibited relatively similar results for both temperatures, except when sliding against DLC and Zr-DLC at elevated temperature. The high wear rates of Zr-DLC coating and corresponding counterpart were clearly related to the highest friction, see Fig. 5b.

In order to evaluate the lifetime of the hydrogenated structures, i.e. DLC-H and Zr-DLC-H, test with 30,000 cycles has been performed as well, see friction curves in Fig. 9. It is evident that Zr alloying had a positive effect on the friction coefficient stabilisation compared to

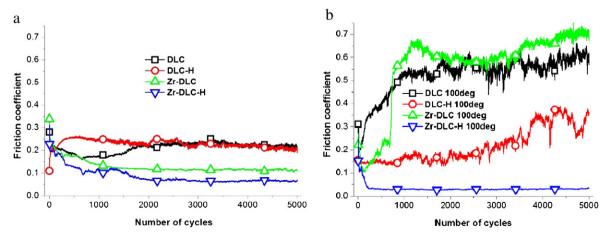


Fig. 5. Friction evolution of Zr-DLC(-H) a) at room temperature and, b) at 100 °C.

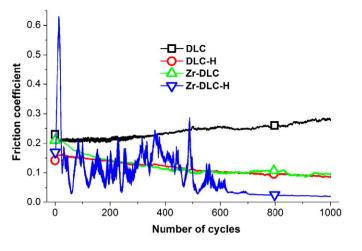


Fig. 7. Friction evolution of the Zr-DLC(-H) coatings in the N<sub>2</sub> atmosphere.

DLC-H coating. We should stress that the coatings were not worn through (maximum wear depth was about 700 nm).

#### 3.2.2. Analysis of worn surfaces

To identify the wear mechanisms of the coatings, optical microscopy, 3D surface analysis and Raman spectroscopy were used to analyse the worn surfaces. The analyses have been predominantly focused on the hydrogenated films, since these coatings represent the majority of DLC-based industrial applications. Nevertheless, the main differences in the hydrogenated and non-hydrogenated coating behaviour are discussed as well.

The running-in behaviour study showed that the surface adaptation (i.e. wear process sufficient to remove asperities from original roughness) was finished at the point when contact pressure steady-state regime was reached. Thus, the balls and coatings were mostly worn during the running-in stage (typically several hundreds to 1500 cycles). The structural changes of the tribolayer adhered on the ball, namely change in carbon structure and iron oxide production, were clearly related to the friction coefficient. The first Raman peaks of iron oxides (if any) appeared within the first 200 cycles. The carbon-related Raman peaks gradually changed positions and intensities till considerably high number of cycles: DLC-H at 500 cycles and all other coatings at 2000 cycles. Thus, the analyses described in the following text are focused on the worn surfaces after 5000 cycle tests when the steady-state regime was fully developed.

The analyses of the wear tracks, ball wear scars and debris adhered on the ball of the hydrogenated DLC coating sliding at room temperature are shown in Fig. 10. The ball wear scar exhibited shallow scratches covered by relatively homogeneous and well-developed tribolayer. The Raman spectra analysis of the tribolayer showed number of peaks in the range of 200–800 cm<sup>-1</sup> together with the typical carbon bands.

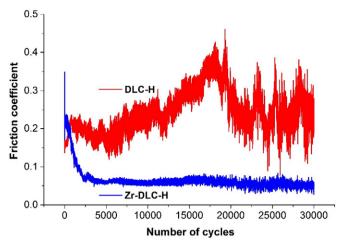
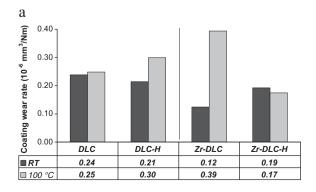


Fig. 9. Friction curves of DLC-H and Zr-DLC-H samples in the lifetime test.

The Raman peaks positioned at lower Raman shifts were identified as iron oxides, mainly hematite  $Fe_2O_3$  (peaks at 228 and 294 cm<sup>-1</sup>) and magnetite Fe<sub>3</sub>O<sub>4</sub> (strongest peak at 665 cm<sup>-1</sup>) [51]. The shape and position of the carbon peaks significantly differed from that of asdeposited coating and the wear track on the coating. The splitting of D and G bands was evident with D band down-ward shift (1391 to 1354 cm<sup>-1</sup>) and up-ward G band shift (1564 to 1580 cm<sup>-1</sup>). The Raman spectrum acquired in the centre of the ball wear scar exhibited almost completely separated D and G peaks; the intensity of the latter was very low. In some cases, the  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> and Fe<sub>3</sub>O<sub>4</sub> phases could exhibit an additional indecisive second harmonic vibration at 1320-1322 cm<sup>-1</sup> [52,53]. Since we do not have any result indicating the presence of such peak, it was not taken into account during spectra deconvolution. Nevertheless, the existence of a weak additional peak in the position referred to above would not significantly change the shift of carbon peaks. The leading edge and the wear debris attached to the ball scar edges showed carbonaceous material with vestiges of iron oxides (Fig. 10, pos. 7, 8). Finally, a weak background photoluminescence (PL) was observed in all spectra taken from various positions at the ball wear scar, which indicated the presence of hydrogen saturating the recombination centres in the carbon matrix [47].

The coating wear track appeared extremely smooth with only low number of very shallow scratches parallel to the ball-coating relative movement. The maximum wear depth did not exceed 100 nm. Thus, the polishing wear was the predominant wear mechanism. The Raman spectra did not show any vestiges of iron oxides or any other contaminants in the wear track. Moreover, no significant changes in carbon structure were observed when compared to as-deposited coatings (Fig. 10, pos. 1–3). Hence, the friction behaviour is mainly related to properties of the tribolayer.



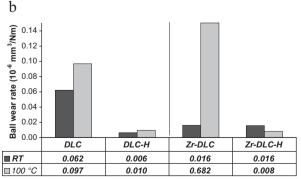


Fig. 8. The values of a) Zr-DLC(-H) coating wear rates and b) ball wear rates.

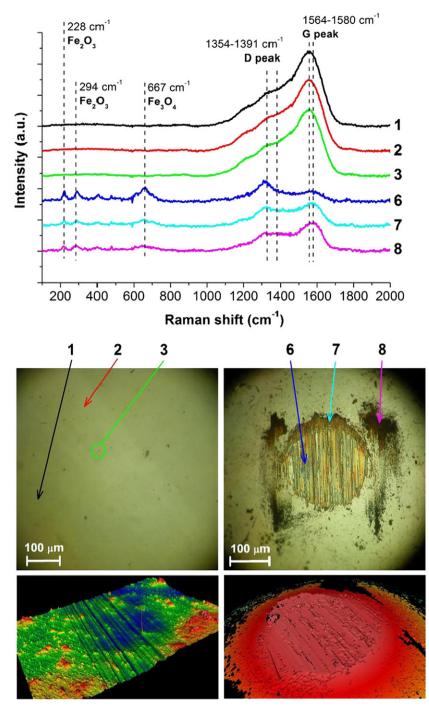


Fig. 10. Raman analysis of the ball wear scar and coating wear track - DLC-H at room temperature. The numbers correspond to the scheme shown in Fig. 1.

Increasing the testing temperature resulted in a higher amount of attached wear debris at the ball leading edge. The tribolayer formed on the ball wear scar was less compact, showing Raman peaks related to iron oxides (magnetite) and carbon (Fig. 11, pos. 6). Interestingly, the peaks of hematite were observed together with those related to magnetite and carbon in the wear debris adhered on the side of the ball wear scar (Fig. 11, pos. 8). As expected, photoluminescence effect was not observed at elevated temperature. The coating surface showed shallow scratches together with a groove in the centre of the wear track (wear depth about 110 nm), which was partially responsible for the increased wear rate when compared with the corresponding test at RT. Again, the Raman spectra from different parts of the wear track were remarkably similar (Fig. 11, pos. 1 – 3).

The results of Raman analyses of the non-hydrogenated coatings sliding at room temperature were comparable with those of hydrogenated films, i.e., the shift of carbon peaks, the decrease of carbon peaks intensities and the presence of iron oxides. At 100 °C, the magnetite and hematite peaks observed on the ball wear scar were more pronounced suggesting a higher steel ball oxidation [51]. Even though the tribolayer was found to be less compact, the production of free wear debris was higher being directly connected to higher friction coefficient [2].

Lower coverage of the ball wear scar by tribolayer was also observed when the hydrogenated Zr-DLC-H coating was tested at room temperature (Fig. 12). The Raman spectra showed sharp separated carbon peaks both shifted in opposite directions (Fig. 12, pos. 3, 6, 7). These shifts were more significantly pronounced than in the case of

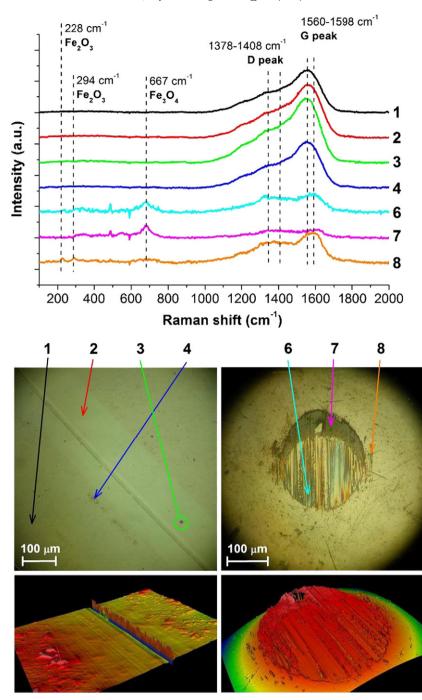


Fig. 11. Raman analysis of the ball wear scar and coating wear track – DLC-H at 100 °C. The numbers correspond to the scheme shown in Fig. 1.

Zr-DLC (not shown) suggesting stronger bond order effects. Slight background photoluminescence was observed indicating the hydrogen presence in the tribolayer. The Raman analysis did not exhibit any vestiges of iron oxides; thus, the formation of the tribolayer inhibited the ball oxidation in the later stages of sliding process due to complete separation of the ball and coating surfaces. The analyses of the wear debris adhered to the ball leading edge and free wear debris produced during the sliding showed again only carbon-related peaks (Fig. 12, pos. 7, 8). The coating wear track was very smooth with shallow scratches (up to 80 nm); no significant changes in the Raman spectra of the wear track and the as-deposited coating were observed. In fact, we identified very small spots covered by the wear debris giving rise to Raman spectra identical to those of the wear debris found at the ball wear scar (Fig. 12, pos. 3). Nevertheless, the area of such spots is negligible compared to the area of the wear track.

Sliding tests at elevated temperature resulted in an almost immeasurable coating wear rate; only few shallow scratches could be observed on the coating surface (maximum wear track depth of 50–70 nm), see Fig. 13. The ball wear scar was fully covered by a tribolayer; the amount of the adhered wear debris at the leading edge of the ball was enormous. The Raman spectra analysis did not show any changes in the wear track when compared to the as-deposited coating (Fig. 13, pos. 1, 2).

The friction coefficient of Zr-DLC and Zr-DLC:H at elevated temperature was 0.7 and 0.03, respectively. Surprisingly, the corresponding Raman spectra of the worn surfaces were similar for both coatings (Figs. 13 and 14). The only difference was the presence of hematite

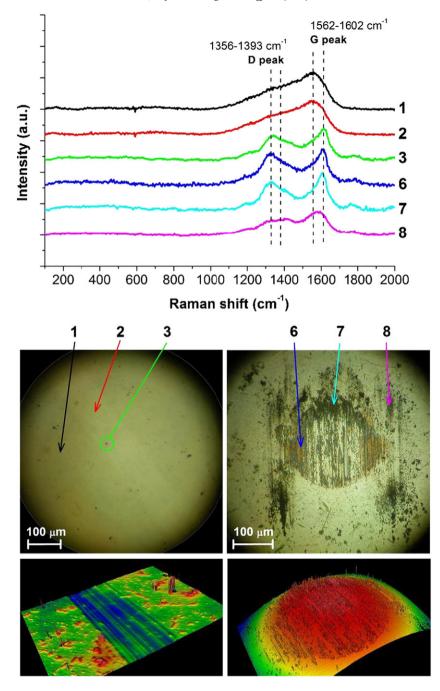


Fig. 12. Raman analysis of the ball wear scar and coating wear track – Zr-DLC-H at RT. The numbers correspond to the scheme shown in Fig. 1.

peaks (228 and 294 cm<sup>-1</sup>) in the wear debris on the side of the ball wear scar after the test of Zr-DLC (see Fig. 14, pos. 8). Notably lower tribolayer formation on the ball wear scar has been observed for Zr-DLC coating as well.

The coatings were also tested in the N<sub>2</sub> atmosphere to clarify the role of humid air. Iron oxides peaks were not found in corresponding Raman spectra of the ball worn surfaces. Carbon peaks remained at the same positions (when compared to test in humid air); however, the  $I_D/I_G$  ratio increased significantly in the case of hydrogenated coatings.

#### 4. Discussion

The effect of environment on sliding properties of DLC coatings is well studied [54]; nevertheless, it is quite complex phenomena depending on sliding conditions and physical and chemical properties of DLC films. Typically we have to consider three mechanisms acting simultaneously: i) friction at nanoscale (passivation of surfaces, adhesion, etc.), ii) tribochemical changes and reactions with atmosphere, and iii) formation of a tribolayer from wear debris.

In order to relate our results – mechanical properties, Raman of asdeposited as well as worn surfaces, and tribology – we will consider the effects of hydrogen and Zr-dopant separately.

The effects of hydrogen in the DLC structure have been described by many authors [1–3,5,11,29]. Increase hydrogen proportion leads to the structural changes of carbon matrix of the DLC that could be observed by Raman spectroscopy. Coatings studied in this study followed the same trend, i.e.  $I_D/I_G$  ratio was lower when the hydrogen was present. This is directly proportional to the prevailing axial vibrations of shorter carbon chains and rings [44,45]. Such variations in the structure typically result in lower hardness and improved tribological properties due to

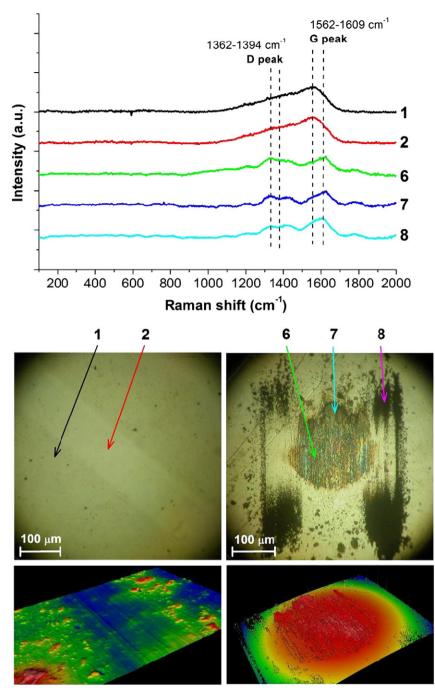


Fig. 13. Raman analysis of the ball wear scar and coating wear track - Zr-DLC-H at 100 °C. The numbers correspond to the scheme shown in Fig. 1.

the formation of low-shear carbonaceous tribolayer. The ability to form low-friction tribolayer depends on the environmental conditions, and proportion of hydrogen in the film [3,8,9,11]. In our case, the effects of hydrogen in the structure in the sliding process could be characterized by three fundamental phenomena - i) higher contact pressure in the steady-state friction regime (due to lower ball wear and narrower wear track), ii) generally lower friction coefficient in the RT ambient environment test and, iii) higher coating sensitivity to the environmental conditions. Hydrogenated DLC coatings are known to exhibit low friction coefficient in vacuum and inert atmospheres due to the formation of low-shear surface tribolayer adhering predominantly onto the counterpart that consequently results in improved coating wear resistance [2,9,11,36,55]. Some authors explained such behaviour by the saturation of the free carbon bonds by hydrogen that predominantly allows formation of weak van der Waals forces between the mating surfaces resulting in low friction [11,48]. The macroscopic effect of this phenomenon could be seen in our results as well, namely in the Zr-DLC-H tests at 100 °C and in the N<sub>2</sub> environment. However, the tribolayer formation in the DLC-H coating test was partially suppressed leading to the increase of friction coefficient and its instability. Deteriorated tribological properties could be an indirect proof of strong bonding established on the surface during sliding [11]. Although there was no significant difference between the Raman spectra of the coatings before and after testing (i.e. no surface graphitization), the presence of iron oxides clearly indicated the detrimental effect on the carbon tribolayer (Fig. 10). It should be also noted that the adhered layer on the leading ball edge contained a mixture of carbon and iron oxides when sliding in humid air, whereas iron oxides were predominantly detected at elevated temperature

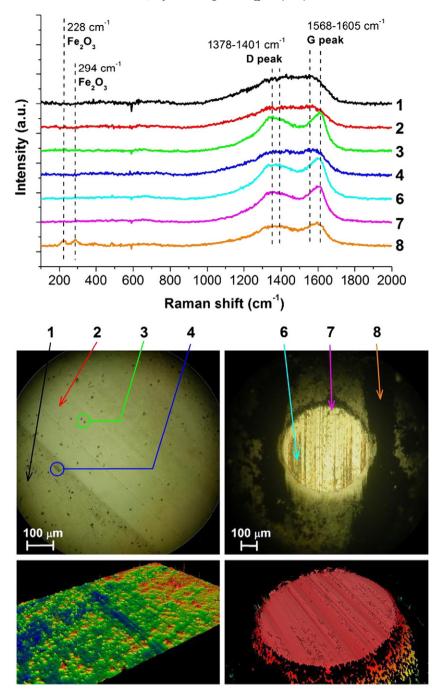


Fig. 14. The Raman analysis of the ball wear scar and coating wear track – Zr-DLC at 100 °C. The numbers correspond to the scheme shown in Fig. 1.

(carbon peaks were very weak). The results of as-deposited surface wettability corroborate the afore-mentioned discussion: the higher the contact angle, the lower the surface chemical activity. Rabinowicz [56] noted that the wear of the surface was directly proportional to surface energy. Moreover, lower surface energy is related to smaller debris formation and thus higher wear, since agglomerated and compacted wear debris acts as a protective layer for both surfaces in the contact [57].

Typically, the non-hydrogenated DLC films exhibit an opposite trend compared to hydrogenated films showing higher friction in inert gases and high temperature [1,3,8,11,15,17]. In our study, the non-hydrogenated coatings reached almost the same mean friction coefficient in  $N_2$  as well as in ambient environment. However, the effect of elevated temperature (suppression of moisture and acceleration of

ball oxidation) results in a sharp increase in the friction coefficient (Fig. 5), which is at least partially the result of higher coating surface reactivity and, thus, higher tendency to form contaminant compounds [2,3,9,11].

Zr as doping element was chosen due to its corrosion properties and high chemical stability compared with many other metallic dopants [37,38,43,48]. Alloying the DLC and DLC-H coatings with such element did not significantly change their mechanical properties; nevertheless, the surface energy showed notably lower surface energy for Zr-doped coatings (see Table 1). This is in full agreement with several works dealing with heavy metallic dopants alloying the DLC published in recent years [2,50]. The Raman spectroscopy of our Zr-doped coatings proved growing D peak intensity associated with down-ward G peak shift. Both phenomena are closely connected with carbon structure disorder, which is a typical effect of metallic dopants introduced into DLC [14,29,48]. The tribology of these structures showed significantly shortened running-in phase and stable steady-state friction evolution at RT. However, the non-hydrogenated Zr-DLC coating sliding at 100 °C showed similar features to that of pure DLC coating. We observed a production of enormous amount of the wear debris, a stable carbonaceous tribolayer did not form and severe adhesive wear occurred (note the limited adhered material at the leading edge of the ball wear scar in Fig. 14). Such behaviour indicates the main result of our observations – the protective role of the tribolayer. Basically, we can conclude that the main role of surface third-body tribolayer is the coating surface protection resulting in lower wear and, in most cases, lower friction as well. Moreover, formation of thick and stable tribolayer suppressed the oxidation of the ball surface, which was observed when tribolayer was not well developed. Quick formation of the tribolayer leads to shorter running-in phase and elimination of friction fluctuations. This effect was evident mainly in the case of the Zr-DLC-H coating. The surface low-shear tribolayer was well developed at 100 °C (Fig. 13). The excellent tribological properties achieved by this coating at elevated temperature, as well as in the N<sub>2</sub> environment can be attributed to its low-shear protective tribolayer formation ability that effectively suppresses coating/counterpart oxidation. This result was supported by surface wettability measurements showing the as-deposited coating surface chemical activity. Lower number of active surface bonds should be expected for Zr-DLC-H coating that

Finally, the intermittent tribo-tests at RT shed the light on the running-in process of Zr-DLC(-H) coatings. Initial sliding process is characterized by three interconnected phases: i) initial ball and coating wear leading to surface adaptation and effective contact pressure reduction, ii) production of surface oxides (in case of the non-hydrogenated coatings), and iii) production of carbonaceous surface layer and debris. Our results suggest that the structural changes of the ball-adhered tribolayer closely follow the evolution of friction coefficient, whereas actual wear is more related to the contact pressure. The formation of the carbonaceous layer could effectively act as a protection from surface oxidation at RT decreasing both the friction and the wear.

implicates decreased adherence of contaminants and other products

of the tribological process [2,11,58].

#### 5. Conclusions

In this study, Zr-DLC(-H) coatings and pure DLC(-H) films were produced by magnetron sputtering and studied in relation to their tribological behaviour in three different environmental conditions: RT, 100 °C and N<sub>2</sub>. Doping with Zr was found to be insignificant from the film's mechanical properties point of view. The main results can be divided into three basic aspects:

- Tribological properties of the coatings are mostly influenced by the formation and composition of tribolayer and the evolution of contact pressure. Although both these properties are related, the analysis of the running-in phase indicated that the friction evolution could be related to tribolayer formation, whereas the wear rate corresponds to actual contact pressure. Our results clearly showed that the most promising tribological behaviour at all tested conditions was achieved for the Zr-DLC-H coating. The other coating structures demonstrated higher sensitivity to the tribological test environment, thus, the applicability of these coatings is limited.
- Production of dense and homogeneous low-shear carbonaceous tribolayer is the key factor to achieve low friction and avoid counterpart oxidation. The composition and structure of such layer are significantly affected by as-deposited coating properties and friction test conditions.
- Doping of DLC(-H) coating with Zr led to lowered surface energy together with increased ability to form a stable tribolayer on the ball surface. Consequently, the friction coefficient of Zr containing

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# PAPER VI

Tribological behavior of uncoated and DLC-coated CoCr and Ti-alloy in contact with UHMWPE and PEEK counterbodies

<u>A. Escudeiro</u>, M.A. Wimmer, T. Polcar and A. Cavaleiro *Tribology International*, under review

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Abstract: In this study, DLC and Zr-DLC coatings were deposited by magnetron sputtering onto Ti alloy substrates. Coated and un-coated Ti alloy samples were tribologically tested in multidirectional pin-ondisk equipment. CoCr alloy was also tested as control. Each of the materials was tested against UHMWPE and PEEK for 2 million cycles (Mc) in bovine serum lubricant (37°C). Gravimetrical weight loss and changes in surface profile were similar when tested against UHMWPE. However, Zr-DLC samples delaminated after 1.2 Mc. When tested against PEEK, both coated samples failed, despite that the coatings maintained their integrity. The synergetic effect of stress-induced corrosion through biological fluid and high cyclic shear stress may have caused interface fatigue and subsequent delamination of both coatings. Ana Escudeiro SEG-CEMUC - Department of Mechanical Engineering Rua Luis Reis Santos, 3030-788 Coimbra Portugal Telef. 351 239 790745 / 00 FAX 351 239 790701 e-mail -<u>ana.escudeiro@dem.uc.pt</u>

Dear Philippa Cann,

Tribology International, Editor-in-chief

Coimbra, June 28<sup>th</sup> 2014

The authors would like to state that the paper submitted with the title:

Adsorption of bovine serum albumin on Zr co-sputtered a-C(:H) films: implication on wear Tribological behavior of uncoated and DLC-coated CoCr and Ti-alloys in contact with UHMWPE and PEEK counterbodies

A. Escudeiro, M.A. Wimmer, T. Polcar, A. Cavaleiro

The authors declare that the work submitted has not been published previously, and it is not under consideration for publication elsewhere. Its publication is approved by all authors and, if accepted, it will not be published elsewhere in the same form, in English or in any other language, without the written consent of the Publisher.

Yours sincerely,

Ana Escudeiro

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Dear Philippa Cann,

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Yours sincerely,

Ana Escudeiro Markus Wimmer Tomas Polcar Albano Cavaleiro

## Highlights

- Zr-DLC presented better adhesion and low residual stress compared to DLC coatings.
- No significant differences were found in the wear of UHMWPE for all material combinations
- Zr-DLC and DLC coatings delaminated when tested against PEEK counterbodies
- The presence of biological fluids together with high cyclic shear stress caused interface fatigue

## Tribological behavior of uncoated and DLC-coated CoCr and Ti-alloys in contact with UHMWPE and PEEK counterbodies

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## Abstract

In this study, DLC and Zr-DLC coatings were deposited by magnetron sputtering onto Ti alloy substrates. Coated and un-coated Ti alloy samples were tribologically tested in multidirectional pin-on-disk equipment. CoCr alloy was also tested as control. Each of the materials was tested against UHMWPE and PEEK for 2 million cycles (Mc) in bovine serum lubricant (37°C). Gravimetrical weight loss and changes in surface profile were similar when tested against UHMWPE. However, Zr-DLC samples delaminated after 1.2 Mc. When tested against PEEK, both coated samples failed, despite that the coatings maintained their integrity. The synergetic effect of stress-induced corrosion through biological fluid and high cyclic shear stress may have caused interface fatigue and subsequent delamination of both coatings.

#### Introduction

The generation of wear particles from metal-on-metal (MoM) and metal-on-polymer (MoP) total joint replacements (TJRs) and related adverse local tissue effects have been an ongoing problem since TJRs were first implanted. In fact, in the case of MoP, 10% of the knee implant failure was directly related to UHMWPE wear [1]. Phagocytosis of wear debris by macrophages induces an inflammatory reaction associated with the release of cytokines and other inflammatory mediators. Once osteolysis has occurred, it tends to progress and may ultimately lead to failure of the implant. On the other hand, despite an overall reduced wear rate, MoM bearings are known to elicit an immune response involving lymphocytes in addition to a macrophage-mediated inflammatory reaction. Nevertheless, the biological effects of metal (nano)particles and ion release is still under investigation and of great clinical concern. Thus, advancements are still desired to improve reliability and long-term implant performance. Among available polymers for *in vivo* application, PEEK and its composites have been used as an alternative bearing surface for UHMWPE due to its biocompatibility, high strength, stiffness, fatigue and wear resistance [2, 3]. Moreover, PEEK particles were found to be less cytotoxic in vitro than UHMWPE [4]. However, a recent in vivo study found increased cytokine expressions for carbon-fiber reinforced PEEK compared to UHMWPE [5]. PEEK increases the wear of the metallic component (SS 316, Ti6Al4V and CoCr) compared to UHMWPE couples [6-8]. Particularly Ti6Al4V may suffer due to its poor tribological properties which are attributed to the low resistance of the oxide layer to plastic shearing [9-11]. On the other hand, titanium and its alloys are interesting candidate metals for biomedical applications because of the known milder tissue reactions of their degradation products compared with CoCr-alloy or Ni-containing steels [12, 13]. Therefore, surface engineering of titanium bearing surfaces and/or the adoption of new coupling materials could be a promising approach to reduce both metal and polymer wear debris.

Diamond-like carbon (DLC) coatings have been widely studied as a possible solution for implant wear due to their excellent tribological properties and performance in a wide range of industrial applications (e.g. car engines, hard disks, etc.) [14]. DLC is biocompatible [15] and presents low friction and extremely low wear rates [14, 16]. In fact, Hauert [17] tested DLC coatings in an intervertebral spinal disk simulator for 101 million cycles with extremely low material loss being reported. However, contradictory results have been found in literature when testing DLC against UHMWPE, ranging from strong improvement to drastically increased wear [18]. There are few studies of DLC against PEEK. Franta [6] reported a significant increase in the life span of implants using such a combination. In general, one of the main issues of DLC coatings have been associated with the adhesion of the coating to the substrate, especially in a corrosive environment such as the human body [17]. High internal compressive stresss imposed by the deposition conditions can further reduce the adhesion strength and promote delamination of the coating. Thus, the interposition of a carefully designed layer between substrate and coating avoiding abrupt compositional changes (Me/MeC,

Me/MeN or Me/MeN/MeNC, where Me stands for metallic element), and/or alloying the DLC coating with metallic elements can help to prevent such issues.

In this study, DLC and Zr-DLC coatings were deposited using magnetron sputtering and tested against UHMWPE and PEEK. The incorporation of Zr in the amorphous carbon coating should improve the adhesion of the coating [19, 20] which may beneficially impact its tribological performance. In order to assess the potential of such DLC/polymer combinations for the use in orthopedic implants, the wear behavior of the coatings was studied using a multidirectional pin-on-disc apparatus under physiological load and lubricant conditions. For comparison, CoCr alloy and Ti-alloy were also tested against UHMWPE and PEEK.

#### **Materials and Methods**

#### 1. Materials and substrates

UHMWPE (GUR 1050) and PEEK (TECAPEEK natural) wear pins were manufactured and provided by Zimmer, Inc. (Warsaw, IN, USA) and Ensinger (Dobřany, Czech Republic), respectively. The cylindrical pins were 0.375 in. (9.5 mm) in diameter and 0.750 in. (19.1 mm) in length. The metallic counterbodies, disks with a diameter of 40 mm, Ti grade 5 (Ti6Al4V) and CoCr (Co30Cr6Mo) alloy, were mechanically polished with SiC paper (500, 600, 800, 1200 grits), and then finished by using diamond suspensions (6, 3 and 1  $\mu$ m) and a colloidal silica suspension. Table 1 shows the average and quadratic mean roughness, R<sub>a</sub> and R<sub>q</sub>, for both metallic and polymeric materials measured by scanning white light interferometry.

#### 2. Coating deposition and characterization

The coatings were deposited onto Ti6Al4V disk substrates and Si(111) wafers for tribological testing and for coating characterization, respectively. Prior to deposition, the substrates were cleaned in an ultrasonic bath containing acetone, ethanol and deionized water for 15 min, and then mounted on the rotating sample holder (18 rpm) in the deposition chamber. The coatings were deposited with a DC dual magnetron sputtering machine. A pure graphite target was used for the production of the a-C films in reactive (Ar/CH<sub>4</sub>) and non-reactive atmosphere (Ar), in order to produce hydrogenated and non-hydrogenated films, respectively. Zr pellets were added to the erosion zone of the graphite target (relative erosion area,  $A_{Zr}/A_C$ , 1%) to produce Zr containing films. All coatings were deposited with a constant applied bias voltage of -50 V and graphite target power density close to 7.5 W/cm<sup>2</sup>. Moreover, a pure Ti target was also sputtered for the deposition of a composite gradient interlayer (Ti/TiN/TiCN) to improve the film adhesion to the metallic substrates. For each deposition condition, the deposition time was calculated to obtain films of ~1.4 µm thickness. Further detailed deposition specification can be found elsewhere [19].

The chemical composition was evaluated by electron probe microanalysis (EPMA, CAMECA Camebax SX 50). The structure of the coatings was analyzed by X-ray diffraction (XRD) (Philips, X'Pert diffractometer, Co K $\alpha$  radiation) whereas X-ray photoelectron spectroscopy (XPS) (ESCAprobe P, Omicron Ltd., Al K $\alpha$ , binding energy calibrated with Au 4f (84,1eV)) was used to identify chemical bonding. Raman spectroscopy was used to evaluate the C structure. The analyses were performed at Northwestern University, Evanston, USA with an Acton TriVista CRS Confocal Raman System (Princeton Instruments, Trenton NJ, USA), with excitation radiation of an Ar-Kr gas laser (514.5 nm wavelength). Raman spectra were recorded in the range of 1000-1800 cm<sup>-1</sup> and fitted using Gaussian fitting following a procedure describe elsewhere [21]. The hardness was measured by depth-sensing indentation (Micromaterials Nanotest) using a Berkovich indenter. The normal stylus load was 5 mN (indentation

depth approx. 150 nm); 32 independent indentations from two distinct areas on the sample were used to analyze the hardness data. Rockwell-C indentation and scratch test were used to evaluate coating adhesion and cohesion. Additional information can be found in [19, 20]. In order to quantify any superficial defects in the coatings, optical microscopy analysis was performed for DLC coatings on Ti6Al4V. The quantification was performed by determining the area covered and average size by surface defects. For each coating 6 different areas were analyzed, with a magnification of 50X.

#### 3. Wear test

The wear tests were performed on a six- station pin-on-disc apparatus (OrthoPod® AMTI Inc., Boston MA, USA), with each test containing pins of UHMWPE and/or PEEK. Test and control pins where pre-soaked in the same lubricant used for testing in order to account for any lubricant absorption. The saturation curves were monitored and testing began once equilibrium weight was attained. The pins were subject to a 15 mm x 15 mm square motion pattern in order to generate crossing motion trajectories. All wear tests were conducted at a cycle frequency of 1 Hz and with a constant applied load in order to generate a nominal contact pressure of 1.5 MPa. Each articulation was submersed in a solution containing new born calf serum (NBCS Gibco®) and deionized water with sodium-chloride to achieve 30 g/l protein concentration at physiological salt levels. The pH was adjusted to 7.6. In addition, the solution contained 20mM EDTA (to prevent precipitation of calcium phosphate), Tris-hydroxymethylaminomethane (buffer), and 0.3% sodium azide (anti-microbial agent). The temperature of the lubricant was maintained at 37°C using a recirculating water bath. A total of up to 2 million cycles (Mc) were conducted. Prior to testing and at regular 0.3 Mc intervals during testing, all metal samples (coated and un-coated) and all polymer components were cleaned in ultrasonic bath with, consecutively, distilled water, Terg-A-Zyme<sup>®</sup> and propanol for 10 min each and dried with a N<sub>2</sub> gas jet. After cleaning, the pins were weighed to obtain a gravimetric measurement. The gravimetric weight loss per pin was determined by taking the average weight loss of six measurements (according to ASTM 2025 and F732). The total weight loss of each polymer pin was corrected by subtracting the weight increase of soak control specimens during the same test interval. Volumetric wear loss was determined by dividing (corrected) weight loss and density of the polymer (UHMWPE: 0.94 g/cm<sup>2</sup>; PEEK: 1.35 g/cm<sup>2</sup>, data from the manufacturer). The volumetric wear rates were determined using a linear regression. The tests were then continued with completely fresh lubricant.

Surface topography measurements were performed using a Zygo New Vision 6300 (Middlefield CT, USA) optical 3D profilometer. Ten measurements of  $R_a$  and  $R_q$  were taken of each pin and plate for every material combination prior and after testing. The surfaces were also inspected by scanning electron microscope (SEM, JSM-5600, JOEL Co., Japan).

#### 4. Statistics

Differences in the wear rates between UHMWPE and PEEK articulating against the same material (i.e. Ti6Al4V, CoCr, DLC, Zr-DLC) were statistically evaluated using independent student's t-tests. Comparisons within each polymer group were performed using one-way analysis of variance (ANOVA) followed by Student-Newman-Keuls (SNK) method to determine pairwise differences. Regression analyses were conducted to determine the correlation between friction and gravimetric wear of each material group. The level of significance was set to p = 0.05.

#### Results

#### 1. Characterization of DLC Coatings

Table 2 shows the coating chemical and mechanical properties. In general, the mechanical properties of the a-C films were maintained when small amounts of Zr (4 at.%) were incorporated. As expected, the residual compressive stress decreased leading to a slight improvement of the adhesion of the coating (see Figure 1). The incorporation of small amounts of a transition metal element into the C-matrix is known to create Me-C or Me-C nanocrystalline phase which has a catalysis effect to the C-matrix, i.e., it breaks the continuity of the carbon network, which results in a decrease of the coordination factor of the C-network [24]. In fact, these films were characterized as ZrC nanocrystalls embedded in an amorphous C-matrix by means of X-ray photoelectron spectroscopy and diffraction analysis [19]. The G position and the  $I_D/I_G$  ratio (integrated intensity ratio) can give important clues about the structural (dis)order of the coating [21]. The incorporation of Zr led to an increase of  $I_D/I_G$  ratio (see Table 2) suggesting higher organization of the C-clusters within the matrix.

#### 2. Tribological Behavior

#### 2.1. Friction and wear

Figure 2 and 3 show the evolution of the friction coefficient (COF) with respect to the number of cycles. The average friction coefficient for all samples against both polymer materials is presented. The friction behavior was strongly dependent on the polymer type. UHMWPE demonstrated approximately 2 times lower and a more stable COF compared to PEEK. Note that every 300,000 cycles the experiment was stopped to carry out gravimetric measurements and then replenished with new lubricant. In general, the CoCr/Polymer combination had the lowest coefficient of friction. When testing UHMWPE against metal (CoCr and Ti6Al4V) the COF tendentiously increased after the running period (300,000 cycles). The inverse trend was observed for coated samples (DLC and Zr - DLC). However, Zr-containing films did not survive more than 1.2 Mc

against UHMWPE due to delamination. Statistically (p < 0.003), the metal/UHMWPE pairings presented lower COF compared with coated samples.

All experimental surfaces showed a higher COF for PEEK than for UHMWPE (p < 0.01) (Figure 2). In general, the COF was much more variable for PEEK, and particularly for CoCr, showing high periodicity with the cleaning cycles. There were no statistical differences (p > 0.057) in the friction coefficient between samples when tested against PEEK. Metal/PEEK showed a tendency for increasing tangential force with increasing number of cycles for the both metal surfaces (from ~0.20 to ~0.35), i.e. not reaching the friction "steady state" during the experiment. Specifically, Ti6Al4V was only tested for 300,000 cycles due to severe surface damage that will be discussed later. Also coated samples failed and delaminated against PEEK after 1.2 Mc and 0.9 Mc for DLC and Zr - DLC, respectively. Only CoCr endured a full 2 Mc testing sequence.

Figure 4 shows the volumetric wear loss versus the number of cycles for both UHMWPE and PEEK pins. The wear factor, calculated from linear regression analysis of the average of three measurements, is shown in Figure 4 (see insets). In general, PEEK wore more than UHMWPE (p < 0.049). The only exception was found for CoCr, where the difference was not significant (p > 0.127). It is worth noting that the wear of the metallic alloys and the coatings could not be measured (samples were too heavy for the microbalance and/or weight losses too small for the standard laboratory balance). However, extensive abrasive wear patterns (scratches and groves) were found on top of Ti6A14V after sliding against PEEK. For a better understanding of the wear phenomena, pin and disks surfaces were analyzed and results will be shown in the next section.

#### 2.2 Surface analysis

Figure 5 and 6 show optical images of the disk and pin surfaces after testing. As expected, CoCr did not present signs of wear when sliding against either UHMWPE or PEEK (Figure 5a and 6a). On the contrary, Ti6Al4V showed signs of abrasive wear (grooving), particularly when tested against PEEK (Figure 5c and 6c). Ti6Al4V is known to behave worse than CoCr under tribological stress conditions. In fact, Ti6Al4V can lose its passive (oxide) layer due to sliding and the generated oxide particles may act as a third-bodies aggravating abrasive wear. When a coating (DLC and Zr-DLC) was applied, no wear was observed initially, thus successfully protecting the Ti-alloy from wear. However, Zr-DLC delaminated after a few hundred thousand cycles irrespectively of counterbody material. The exposed areas of the Ti6Al4V surface of DLC-coated samples maintained their integrity though (i.e. no signs of abrasive wear in exposed areas - see Figure 5e and g and 6e and g). Both disc surfaces and pins were analyzed by 3D optical interferometry. There were no significant differences before and after testing for the discs, except for Ti6Al4V. Figure 7 shows the roughness values of the pins for all material combinations. The measurements were taken at the center of pin. In general, UHMWPE pins were found to become randomly scratched during

testing. When hard DLC coatings were used, the UHMWPE surface became smoother. The same findings held up for PEEK pins. The Ti6Al4V/PEEK couple showed increased abrasive wear due to released oxide debris (Figure 6d).

#### Discussion

This study investigated DLC and Zr-containing DLC coatings which were deposited on Ti-based substrates. Both showed beneficial coating characteristics obtained through standard laboratory testing and the incorporation of Zr into the amorphous carbon coating further improved adhesion to the substrate. However, the obtained characteristics did not translate into sustainable biotribological performance. In fact, the novel Zr-DLC coating delaminated already after a few hundred thousand cycles against either UHMWPE or PEEK, which would prohibit its use in patients. This unexpected result highlights the importance of a proper testing sequence to evaluate new materials for TJR. In this section we will discuss the obtained results and make an attempt to elucidate the failure mechanism of the DLC coatings.

Pre-clinical friction and wear measurements are important to assess potentially beneficial combinations of materials for use in TJR. Excessive friction can cause loosening of the prosthetic device at the bone-implant interface and consequently instability and pain. Also wear reduces the prosthesis' life span [25, 26]. Phagocytosis of the wear particles due to macrophages can lead to a cascade of inflammatory reactions that causing osteolysis at the prosthetic interface with subsequent loosening and failure [12]. Multidirectional pin-on-disc machines, that provide both reciprocating and rotational motion, represent the simplest and the most widely accepted model for basic biotribological simulation of candidate biomaterials for joint prosthesis [27]. The exact loading and motion patterns are not replicated with such machines though; the multidirectional motion generates "crossing shear" conditions, which are typical for TJR [28, 29]. In fact, the wear factor found for CoCr/UHMWPE ( $k = 2.6 \pm 0.1 \times 10^{-6}$ mm<sup>3</sup>/Nm) is in the range of those reported from clinically retrieved TJR couples (k  $\sim$  $10^{-6}$  mm<sup>3</sup>/Nm [30, 31]). Thus, the behavior of new materials combinations can be assessed in a relatively cost effective way. We will now discuss the results attained for coated and un-coated samples against PEEK and UHMWPE.

For all materials, both friction and wear results were strongly dependent on the polymer counterbody. The friction coefficient was found 2 times higher when tested against PEEK. Also the wear was higher for PEEK, although - in general - the wear mechanism was found to be similar for both polymer types showing typical signs of micro-abrasion [3, 32]. In nearly all cases, wear correlated with the friction coefficient, except for CoCr when tested against PEEK. In fact, PEEK is known to present a relatively high friction coefficient of ~0.3 against metals [8, 33].

There were no significant differences between PEEK and UHMWPE when tested against CoCr. In contrast, in the literature, six to eight times higher wear rates of PEEK

compared to UHMWPE have been reported [3, 8, 26]. However, the contact pressure applied in the present study was at the low end (~1.5 MPa) compared to other studies. Davim et al. [25] and Laux and Schwartz [34] have reported that PEEK wear is strongly related to contact pressure. In fact, rising the contact pressure from 1.8 to 3.9 MPa led to a three- to six-fold wear increase of PEEK (dependent on supplier and molecular weight) [34]. The effect between contact pressure and wear is opposite for UHMWPE: both Wang et al. [35] for the hip and Galvin et al. [36] for the knee found that an increase in contact pressure will lead to a decrease in wear. Hence, the relatively low pressure in our study may have diminished PEEK wear but increased UHMWPE wear per load increment, thus causing similar wear factors. Ti6Al4V showed poor tribological properties against polymer surfaces in this study (see Figure 5c and 6c). Its poor shear strength led to disruption of the oxide layer, which then acted as abrasive [7, 10, 11, 37]. The effect was even more pronounced against PEEK. The use of protective coatings was meant to protect the Ti-based substrate from abrasive wear. However, except for DLC against UHMWPE, they all failed catastrophically. In other words, coatings failure was linked to the delamination or spallation of the coating in the sliding region (see Figure 5g, 6e and g).

The incorporation of Zr enhanced the adhesion and reduced the residual stress of the coatings compared to DLC coatings (Table 2). Thus, one would think Zr-DLC should behave superiorly compared to DLC. Despite the good mechanical adhesion that was achieved in ambient air, the coatings failed when tested under multi-directional stress in body fluids. Surface layer failure is the biggest concern for the use of coatings in the biomedical field and has been reported by many other authors for both in vitro and in vivo [18, 38, 39]. The presence of biological species in solution (ions, proteins, etc.) can encourage corrosion of the adhesion-promoting interlayer and consequently lead to delamination of the coating. The most common corrosion mechanism found in coatings when in contact to body fluids are pitting, crevice corrosion (CC) and stress-induced corrosion (SCC) [17]. CC and SCC are related to slow crack advancement or interlayer dissolution resulting in a delayed delamination. On the other hand, since even a perfect coating will always present some surface defects (pinholes, scratch and/or loose grain that can traverse their entire thickness down to the substrate) pitting may occur. The body fluids can then penetrate through the film and act as a corrosive electrolyte forming coating blisters. If the blister exceeds a certain dimension local delamination can occur [40]. In vivo, the average delamination speed lies at 145  $\mu$ m/year [41]. This process appears too slow to be solely responsible for the observed delamination failure of this study. Also, the incorporation of Zr led to a decrease of surface defect density, with a defect size slightly higher than that of pure carbon films (see Table 3). It might be possible though that these defects promote SCC cracks, which then grow underneath the coating due to the high induced shear stresses at the contact. Figure 8 shows SEM micrograph of Zr-DLC wear track after testing against UHMWPE. The presence of pinholes on the wear track may promote stress-induced corrosion and initiate delamination.

In order to evaluate the effect of body fluids on the adhesion strength of the coatings after testing, Rockwell indentations were performed on the wear track and, when existing, near to the delamination zones. Figure 9 demonstrates that the film adhesion was maintained after testing. Moreover, Raman spectra of the wear track (data not shown) did not reveal any significant changes between pre- and post-testing suggesting that the structural integrity of the coatings was maintained. In order to investigate the possibility of SCC further, we followed a complementary method proposed by Hauert and co-workers [42, 43]: by inducing substrate deformation with Rockwell indentation and simultaneous immersion of the sample in a protein-containing solution, SCC should be promoted at the mechanically loaded interface. However, even after 3 days (the equivalent of approximately 300,000 testing cycles) the coatings did not reveal any differences on the indentation profile, suggesting a more complex failure mechanism than SCC alone.

In summary, the body fluids' corrosive effect cannot be solely responsible but a combination of factors, including cyclic fatigue, boosted by the presence of huge shear stresses, may have caused failure of the coating.

#### Conclusions

In this study, DLC and Zr-DLC coatings were deposited on Ti-alloy substrates in order to study their tribological behavior against UHMWPE and PEEK for potential *in vivo* applications. Using a multi-directional pin-on-disk wear testing device and a lubricant mimicking body fluids, coated and un-coated samples were tested against the polymers and compared to CoCr-alloy which served as reference. No significant differences were found in the wear of UHMWPE for all material combinations; however, Zr-containing DLC failed early in the test. Although the incorporation of Zr in the C-matrix increased the adhesion strength and reduced the residual stress of the coating, delamination was observed after a few hundred thousand cycles of testing. This study suggests that the presence of surface defects in the presence of body fluids and high shear stress can boost the interface failure and consequently lead to delamination. In the future, it will be important to understand the synergetic effect between stress-induced corrosion and cyclic fatigue on the interpositional layer and/or substrate in order to eliminate such problems.

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SFRH/BD/75071/2010 which was co-funded by FSE and MSTES. Additional funding was obtained through the Rush Arthritis and Orthopedics Institute. The authors would like to thank Dr. Robin Pourzal Rush University, Chicago for technical assistance with wear testing and imaging, Dr. Yifeng Liao from Northwestern University, Evanston for the Raman spectra acquisitions, and Noora Manninen from University of Coimbra for EPMA analysis.

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TITANIUM AND STAINLESS STEEL. Journal of Bone & Joint Surgery, British Volume. 2003;85-B:133-41.

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## **List of Figures**

**Figure 1** Scratch test with increasing load on a) DLC and b) Zr - DLC coated Ti grade 5. The inset indicates critical coating adhesion loads as described in Table 2.

**Figure 2** Friction coefficient vs number of cycles for all experimental surfaces against UHMWPE and PEEK.

**Figure 3** Average coefficient of friction (COF) for all surfaces against UHMWPE and PEEK. (\*) represents statistically significant differences between samples in the same group (p < 0.05 ANOVA test).

Figure 4 (a) UHMWPE and (b) PEEK volumetric wear loss and wear rate (in the inserts) against the different surfaces.

**Figure 5** Images from the wear track on the disc and corresponding UHMWPE pin surface after testing: (a,b) CoCr, (c,d) Ti6Al4V, (e,f) DLC and (g,h) Zr-DLC.

**Figure 6** Images from the wear track of the disc after testing against PEEK: (a,b) CoCr, (c,d) Ti6Al4V, (e,f) DLC and (g,h) Zr-DLC.

Figure 7 Average roughness measured at the centre of each polymer pin after sliding.

Figure 8 (a) DLC and (b) Zr-DLC SEM image of the wear track after testing against UHMWPE.

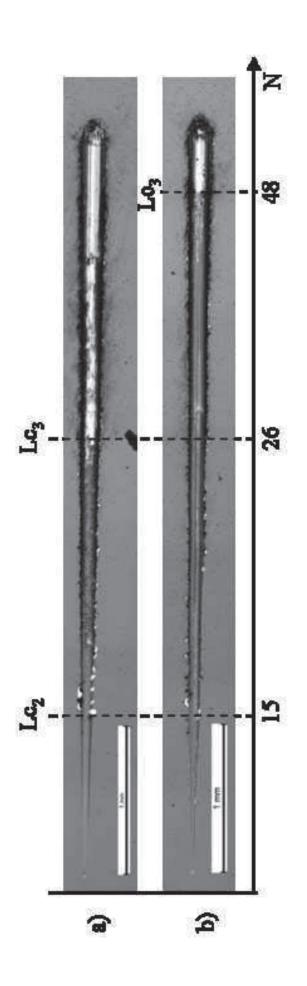
**Figure 9** Rockwell-C indentation test of the coated films before and after testing against UHMWPE and PEEK.

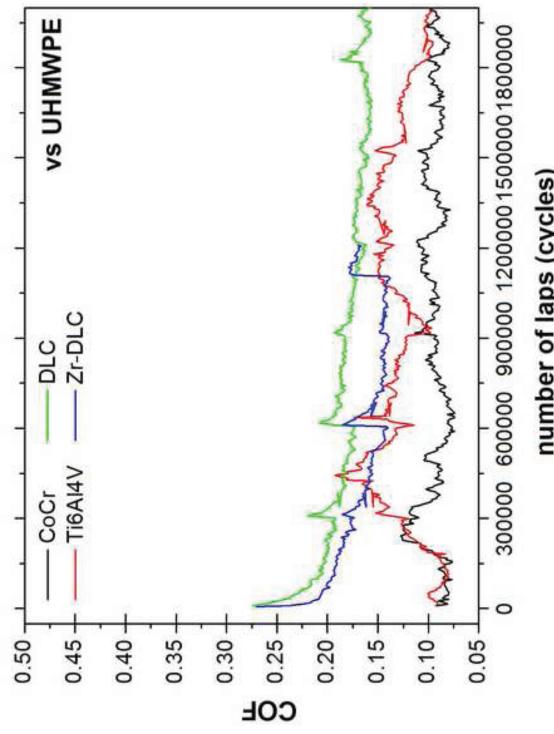
### List of Tables

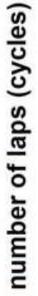
Table 1 Roughness values [µm] for uncoated discs and polymeric pins.

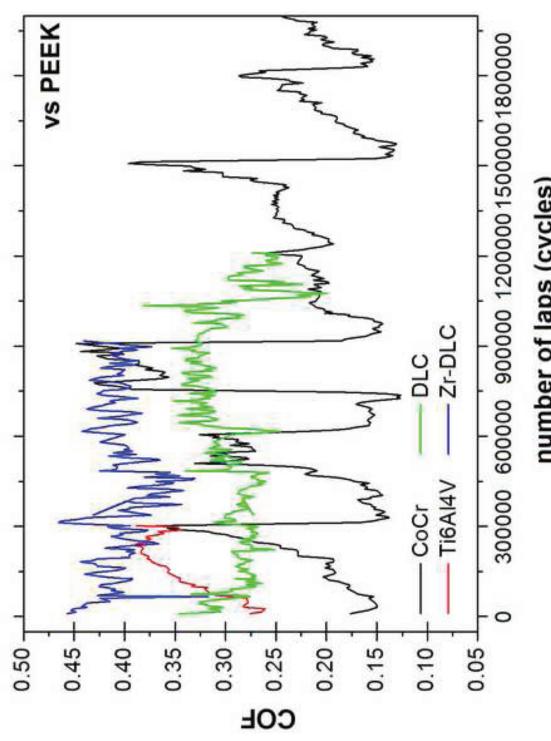
**Table 2** Summary of DLC coating properties: Zr/C ratio,  $I_D/I_G$  ratio, G peak position, hardness (H), Young modulus (E), Residual Stress ( $\sigma$ ), adhesion critical loads ( $Lc_x^a$ ) and Rockwell-C<sup>b</sup> adhesion test label.

 Table 3 Defects density on the coatings.









number of laps (cycles)

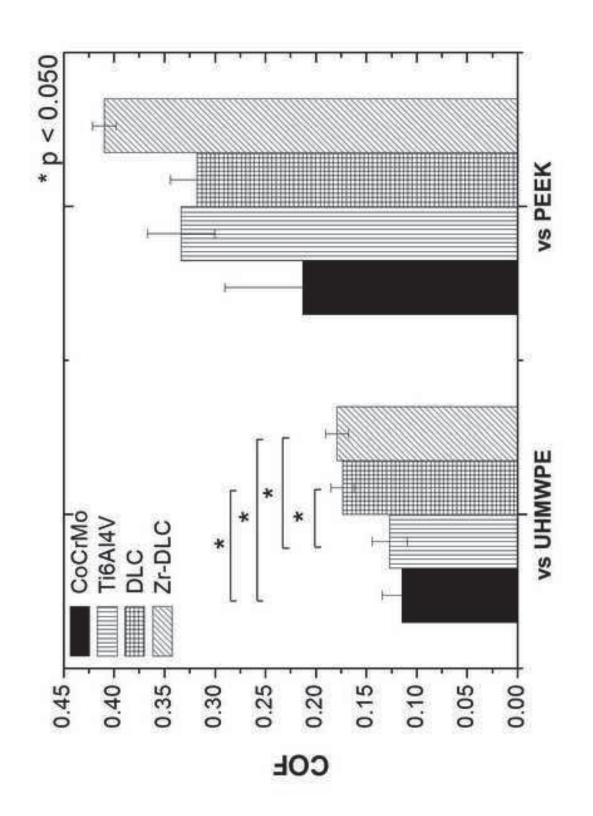
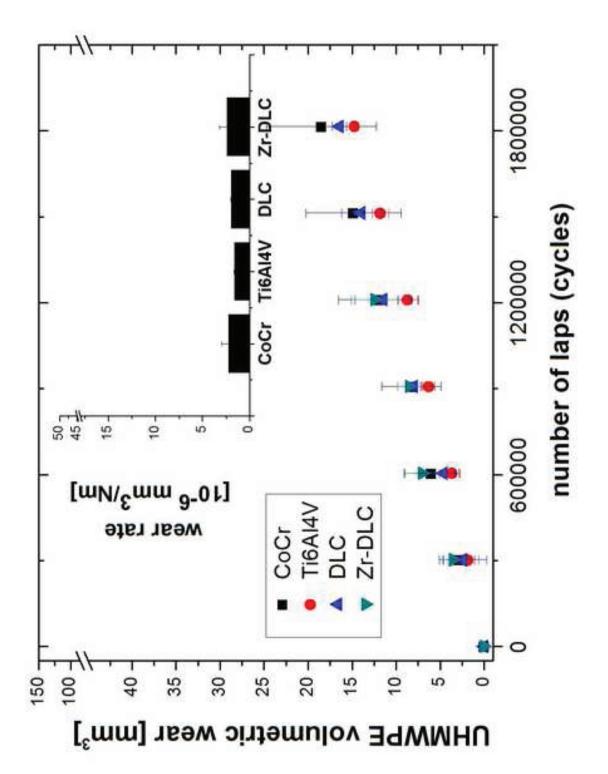


Figure3





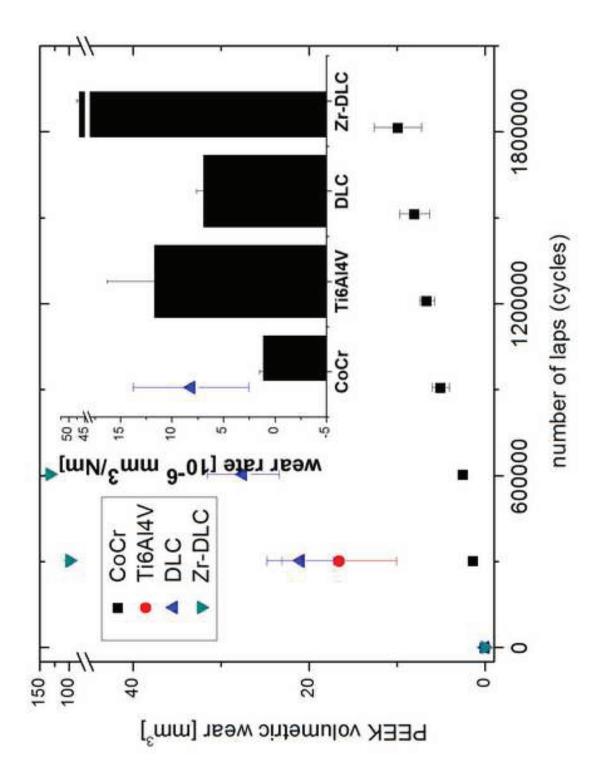




Figure5

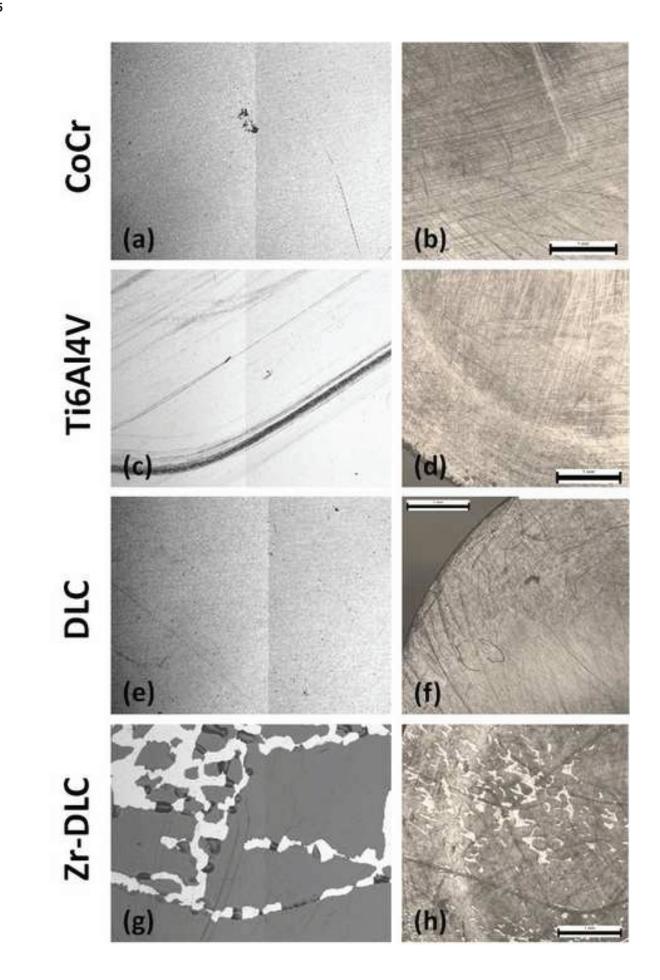
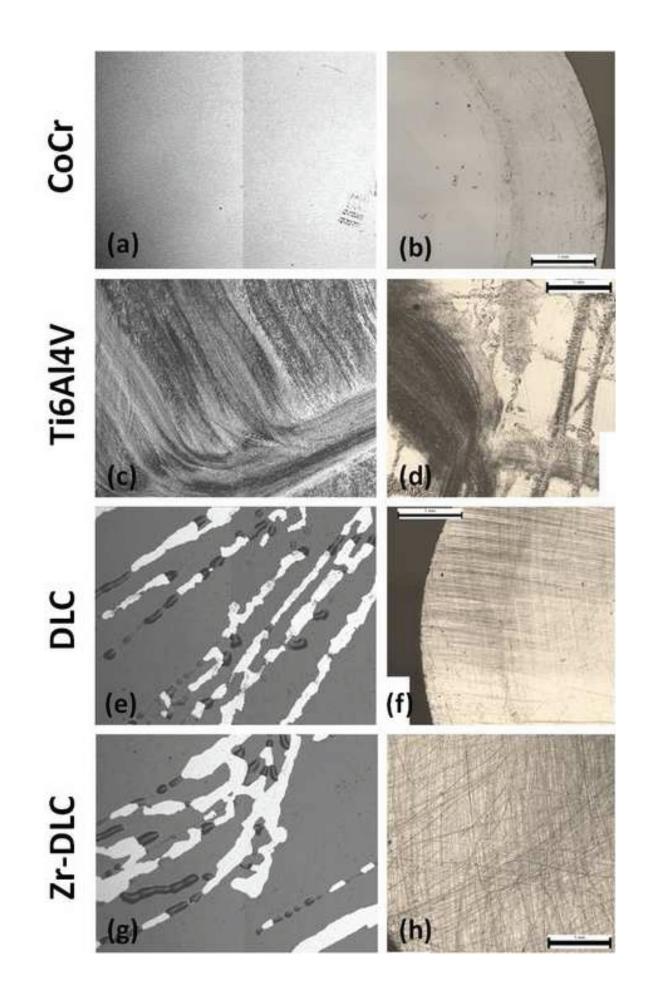
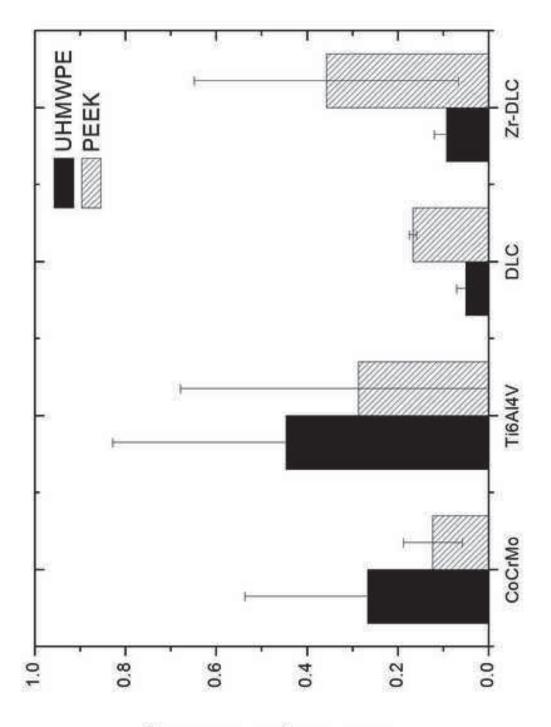
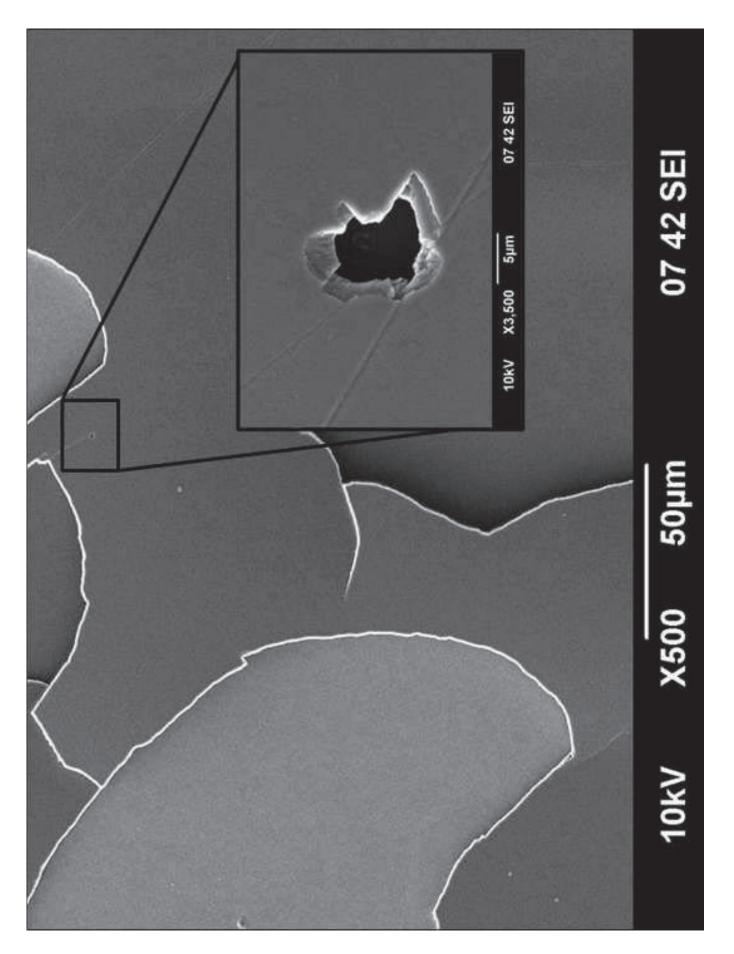


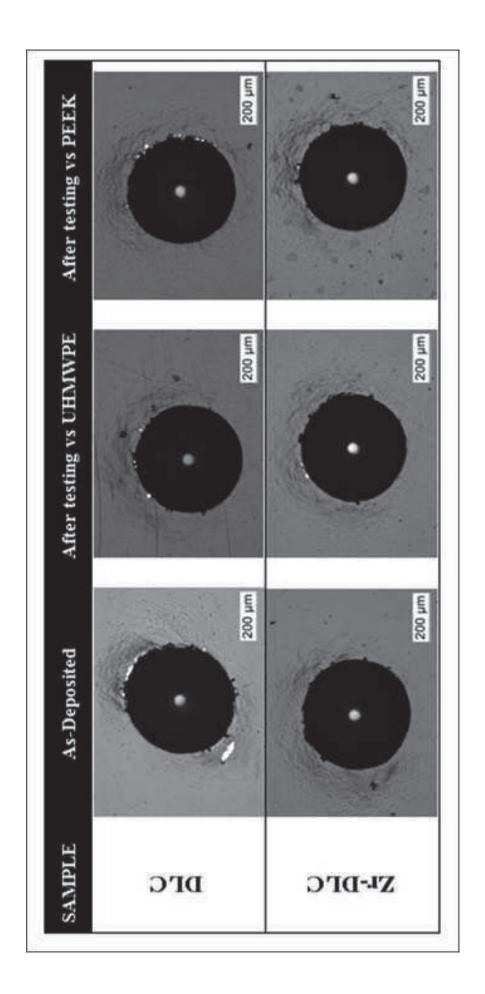
Figure6





(mV\<sup>6</sup>mm <sup>3</sup>.01) ətsi nsəw





	Ra	Rq
CoCr	$0.008 \pm 0.001$	$0.011 \pm 0.003$
Ti6Al4V	$0.020 \pm 0.013$	$0.029 \pm 0.017$
PEEK	$1.000 \pm 0.070$	$1.288 \pm 0.551$
UHMWPE	$1.852 \pm 0.242$	2.159±0.830

Samples Zr/C I <sub>D</sub> /I <sub>G</sub>	Zr/C	$I_{\rm D}/I_{\rm G}$	G peak [μm <sup>-1</sup> ]	H [GPa]	E [GPa]	<sup>G</sup> G peak H [GPa] E [GPa] σ [GPa] Lc <sub>1</sub> [N] Lc <sub>2</sub> [N]	Le <sub>1</sub> [N]	Lc <sub>2</sub> [N]	Le <sub>3</sub> [N] R-C test	R-C test
DLC	0	0 1.6	1564	$10.7{\pm}1$	$111.1{\pm}20$	1564 10.7±1 111.1±20 -1.92±0.1 5.4±1 10.7±3	$5.4{\pm}1$	$10.7{\pm}3$	$21.5{\pm}10$	HR3
<b>Zr-DLC</b> 0.04 1.9	0.04	1.9	1564	$10.0{\pm}1$	$143.5{\pm}4$	$1564   10.0\pm1 143.5\pm4 -1.46\pm0.1   6.6\pm3 13.0\pm4$	$6.6\pm3$	$13.0{\pm}4$	$34.7{\pm}18$	HR2
<sup>a</sup> Lc <sub>1</sub> , lowe	r critical	load is d	efined as th	le load wher	e the first cr	Lc <sub>1</sub> , lower critical load is defined as the load where the first cracks occur (cohesive failure) and Lc.	hesive failu	(re) and $Lc_2$	$b_2$ , upper critical load	cal load, is
defined as	the load	where the	tirst spalla hv the scr	defined as the load where the first spallation/flaking on the where the coating is penetrated by the scratch indenter [77]	on the film i	defined as the load where the first spallation/flaking on the film is observed (adhesive failure). $Lc_3$ is twice the coating is penetrated by the scratch indenter [22]	hesive tailu	re). Lc <sub>3</sub> is t	then defined as the load	as the load

where the coating is penetrated by the scratch indenter [22] <sup>b</sup>HF1-HF4 is typically considered as an acceptable failure, whereas HF5-HF6 is not accepted as a sufficient adhesion [23].

Sample	Area coverage [%]	Average size [µm]
DLC	7.6±0.7	4.1±0.3
Zr-LC	2.9±0.4	6.8±1.3

# PAPER VII

Structural and mechanical properties of nanocrystalline Zr co-sputtered a-C(:H) amorphous films

<u>A. Escudeiro</u>, N.M. Figueiredo, T. Polcar and A. Cavaleiro Applied Surface Science, under review

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# Structural and mechanical properties of nanocrystalline Zr co-sputtered a-C(:H) amorphous films

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# ABSTRACT

The aim of this study was to investigate the effect of Zr as alloying element to carbon films, particularly in respect to film structure and mechanical properties. The films were deposited by magnetron sputtering in reactive ( $Ar + CH_4$ ) and non-reactive (Ar) atmosphere with different Zr contents (from 0 to 14 at.%) in order to achieve a nanocomposite based films. With an increase of Zr content a broad peak was observed in X-ray diffraction spectra suggesting the presence of nanocrystalline (nc) ZrC phase for the coatings with Zr content higher than 4 at.%. The application of Scherrer formula yielded a grain sizes with a dimension of 1.0–2.2 nm. These results were supported by X-ray photoelectron spectroscopy showing typical charge transfer at Zr—C nanograins and carbon matrix interface. The nc-ZrC phase was also observed by transmission electron microscopy. The hardness of the coatings was approximately independent of Zr content. However, the Young modulus increased linearly. The residual stress of the coatings was strongly improved by the presence of nc-ZrC phase embedded in the a-C matrix. Finally, the incorporation of H into the matrix led to denser and harder films.

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# 1. Introduction

Diamond-like carbon (DLC) coatings are considered an attractive protective surface treatment which combines high hardness and wear resistance with low coefficient of friction, helping to reduce friction losses and to increase load carrying capacity. At the same time, hard DLC coatings are generally characterized by high compressive stress which limits their adhesion to the metallic substrate and makes them brittle under the applied loading. Fortunately, their doping or alloying with non-metallic (N, H, O) and/or metallic (Ti, Cr, W, Ag) elements helps to relax the amorphous carbon (a-C) network by forming a nanocomposite structure [1–4]. Moreover, transition metal-carbides show unique combination of materials of solid-state properties such as very high hardness and excellent electrical and thermal conductivity [5]. If transition metals are introduced into the carbon matrix above a certain concentration, nanocrystalline (nc) particles are formed acting as a reinforcing phase. In fact, Meng et al. [6,7] reported a maximum limit of dissolution of Ti atoms into an a-C:H (hydrogenated) matrix between 0.9 and 2.5 at.% and for a-C matrix between 4 and 8 at.%; for

http://dx.doi.org/10.1016/j.apsusc.2014.11.015 0169-4332/© 2014 Elsevier B.V. All rights reserved. higher Ti concentration, nanocomposite materials could be formed. Mechanical, thermal and electrical properties of the coatings can be then enhanced controlling the size and the volume fraction of the nc-phase [8,9]. Voevodin and co-workers pioneered nanostructured materials [10,11]. For instance, when alloying with Ti, the TiC/DLC systems exhibit a large amount of grain boundaries between the nc-TiC phase and the amorphous DLC matrix, which effectively enhances the hardness and toughness of the coatings, limiting crack initiation and/or terminating further crack growth and film delamination [8-10,12-14]. Moreover, such nanocomposite structure could also lower the friction and wear and significantly improve load carrying capacity [11]. Apparently, encapsulation of finer grains of hard nc-TiC within the amorphous DLC matrix restricts dislocation activity, diverts and arrests macrocracks development, and maintains the high level of hardness [9,13]. The interface between both phases also plays an important role, since interfacial fraction in nanocomposites is very high [15]. It was found that the charge transfer from the Ti 3d to C 2p orbitals may contribute to the mechanical stability of the material by enhancing the inter-phase bonding [15–17]. Therefore, nanocomposite coatings are expected to have high hardness and fracture toughness since the carbide crystals are small and help to block (micro-)crack propagation together with the amorphous carbon phase which provides self-lubrication properties [8,9,13,18,19].



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#### Table 1

Coatings deposition parameters ( $A_{Zt}/A_C$  is the relative Zr pellets coverage of the erosion area,), chemical composition measured by XPS, deposition rate (Dep rate), thickness (t) and average roughness (Ra).

Sample	$A_{\rm Zr}/A_{\rm C}$ (%)	[O] (at.%)	[C] (at. %)	[Zr] (at. %)	Dep rate (nm s <sup>-1</sup> )	<i>t</i> (μm)	Ra (nm)
a-C	-	1	99	-	7.9	1.7	13.4 ± 1
$a-C_Zr(4)$	2	2	94	4	8.4	1.7	$7.3 \pm 1$
a-C_Zr(7)	3	4	89	7	10.5	1.5	$5.9 \pm 1$
a-C_Zr(14)	6	8	78	14	11.4	1.5	$5.8 \pm 1$
a-C:H_Zr(10) <sup>a</sup>	6	3	87	10	13.2	1.6	$4.4\pm1$

<sup>a</sup> Chemical composition not considering H content;  $H \sim 30$  at.% [24].

In this study, Zr, as a strong carbide forming element, was chosen as doping element and incorporated into amorphous carbon matrix in order to form nanocomposite coatings. Coating microstructure is studied in detail and related with mechanical properties.

# 2. Materials and experimental details

### 2.1. Coatings deposition

The coatings were deposited using dc magnetron sputtering under reactive (Ar+CH<sub>4</sub>,) and non-reactive atmosphere (Ar) in order to deposit hydrogenated and non-hydrogenated coatings, respectively. Two targets were used: Ti, to deposit the gradient interface layer (Ti/TiN/TiCN), and graphite, to deposit the functional coating. Zr-containing films were produced through the incorporation of Zr pellets into the erosion area of the graphite target (see Table 1). Prior to deposition the substrates (Si wafers, mirror polished thin 304 SS and M2 steel discs) were ultrasonic cleaned in acetone, alcohol and distilled water. The substrates were then placed into the vacuum chamber and sputtered etched, via Ar<sup>+</sup> bombardment (-650V bias voltage), for cleaning any additional oxides and surface contaminants. The Ti-based gradient layer (~650 nm) was then deposited in order to improve adhesion. The deposition parameters were maintained constant for all depositions: 7.5 W cm<sup>-2</sup> target power density, total pressure 0.4 Pa, sample rotation 18 rpm, bias voltage applied to substrates -50 V. In order to produce hydrogenated coatings, CH<sub>4</sub> was introduced into the deposition chamber while the number of Zr pellets was maintained. The Ar/CH<sub>4</sub> flux was adjusted in order to maintain the total pressure constant. The deposition time was set to deposit coatings approximately 1.6 µm thick. Further deposition details can be found elsewhere [20]. To facilitate reading, the coatings were denominated as a-C.Zr(X) and a-C:H.Zr(X) for non-hydrogenated and hydrogenated coatings, respectively, where X is the approximate Zr content.

# 2.2. Microstructure, composition and morphology

The phase identification was performed using synchrotron radiation at the Materials Research station of the Rossendorf beam line (ROBL-MRH) located at the bending magnet 20 (BM20) at the European Synchrotron Radiation Facility (ESRF) in Grenoble. The incident X-ray beam was monochromatized to an energy of 11.5 keV ( $\lambda$  = 0.1078 nm). To increase the signal from the coating and in order to identify any crystalline phase, grazing angle incidence X-ray diffraction was performed; hereby the sample surface was tilted  $2^{\circ}$  in relation to the synchrotron beam. The chemical bonding of the films was evaluated by XPS analysis using a Kratos AXIS Ultra HSA, with VISION software for data acquisition. The analysis was carried out with a monochromatic Al K $\alpha$  X-ray source (1486.7 eV), operating at 15 kV (90 W), in FAT mode (Fixed Analyser Transmission), with a pass energy of 40 eV for regions ROI and 80 eV for survey. Data acquisition was performed at a pressure lower than  $1 \times 10^{-6}$  Pa and charge neutralization system was used. The samples were sputter-cleaned with Ar<sup>+</sup> ion gun (4 keV) prior to spectra acquisition (30 min) in order to remove any residual contamination, corresponding to a sputter depth of 0.01 nm/min (~0.3 nm). The modeling of the spectra was performed using the CasaXPS program, in which an adjustment of the peaks was performed using peak fitting with Gaussian–Lorentzian peak shape and Shirley type background subtraction. The charging effects were corrected at the surface by assuming C–C bond binding energy (BE) at 285 eV in the C 1s spectrum. For the deconvolution of Zr 3d peaks the spin–orbital splitting was assumed to be constant for all phases (2.4 eV) and the integrated area relatively to that of the Zr  $3d_{5/2}$  peak was considered as equal to the spin–orbit multiplicity of 2/3 [21]. The chemical composition was also evaluated by XPS from the relative peak intensity, corrected by experimentally determined sensitivity factors.

The surface morphology and cross-section of the coatings were investigated using a scanning electron microscope (SEM, FEI Quanta 400FEG E SEM–EDAX Genesis X4M) and atomic force microscopy (AFM, AFM–Bruker Innova). The average roughness (Ra) was evaluated by AFM on a 5  $\mu$ m × 5  $\mu$ m line scan. The investigation of the nanostructure was carried out by transmission electron microscopy (TEM) with a FEI Tecnai G2 T20 microscope with a LaB<sub>6</sub> filament operated at 200 kV and a FEI Titan 80-300 microscope equipped with field emission gun (FEG) and Gatan image filter (GIF) Tridium and operated at 300 kV. Both bright field (BF) imaging and selected area electron diffraction (SAED) were carried out. SAED was performed with an aperture size of ~100 nm in diameter. TEM specimens were prepared using an FEI Versa3D focus ion beam–scanning electron microscope (FIB-SEM).

# 2.3. Mechanical properties

The hardness of the coatings was evaluated by depth-sensing indentation (Micromaterials Nanotest) using a Berkovich indenter. The normal stylus load was 5 mN (indentation depth approx. 150 nm); 32 indentations from 2 distinct areas on the sample were carried out to evaluate hardness. Moreover, the reduced Young modulus was derived from the indentation measurements by standard Oliver and Pharr method [22]. The residual stress was calculated using the corrected Stoney equation [23] by measuring the curvature of the SS304 thin disk substrate radii by 2D profilometer (Perthometer S4P) before and after coating deposition.

# 3. Results and discussion

# 3.1. Coating deposition and microstructure

Table 1 summarizes the deposition parameters and the corresponding chemical composition of the films (H excluded). The increase of the number of Zr pellets led to higher Zr content in the coatings and, consequently, the deposition rate also increased due to the higher sputtering rate of Zr in relation to C [20,24]. The introduction of  $CH_4$  further increased the deposition rate; however, Zr content was lower due to the presence of a second C source. The

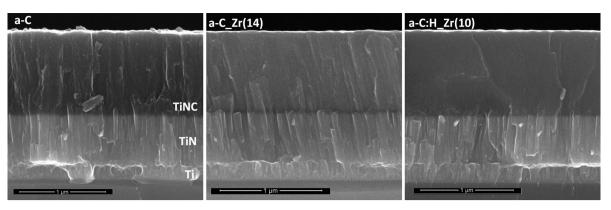


Fig. 1. SEM micrographs showing the fractured cross-section of a-C, a-C.Zr(13) and a-C:H.Zr(10) films.

oxygen contamination originated from the chamber atmosphere was very low (2 at.% data from EDX). The oxygen content measured by XPS was found to be dependent on the Zr content (see Table 1). Prior to XPS signal acquisition, the surfaces were sputterclean by ion bombardment to clean the surface from contamination. For high Zr contents (14 at.%), the presence of oxygen was detected even after very long ion bombardment (40 min). We assume that the increase of metal/carbon ratio caused by preferential sputtering led to a higher oxygen adsorption and the generation of oxide films on the exposed ZrC nanocrystallines, as will be discuss later [25–27].

The SEM micrographs presented in Fig. 1 show fractured crosssection of selected coatings. The columnar intermediate layer (Ti/TiN/TiCN) is visible between substrate and functional top coating. The same morphology was observed when TiN was deposited by reactive magnetron sputtering at low temperature and low bias voltage [28,29]. a-C films revealed a columnar morphology, which led to a cauliflower-like patterned surface resulting from

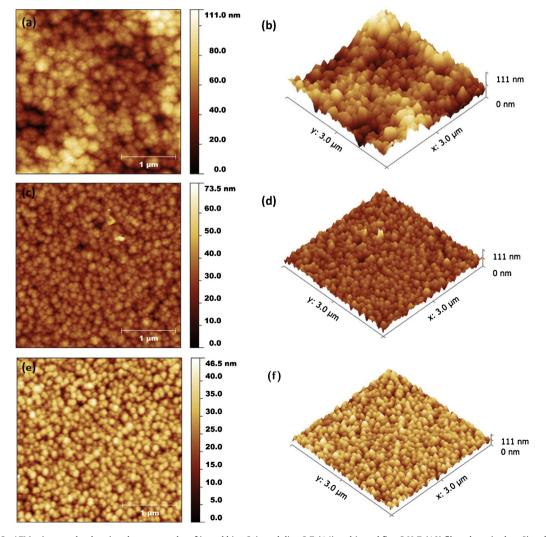


Fig. 2. AFM micrographs showing the topography of (a and b) a-C, (c and d) a-C.Zr(14) and (e and f) a-C:H.Zr(10) films deposited on Si wafers.

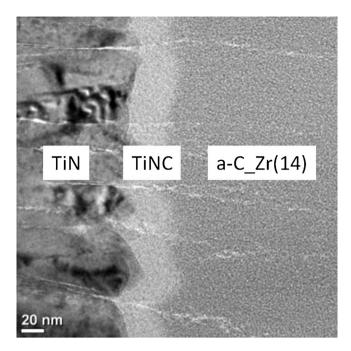
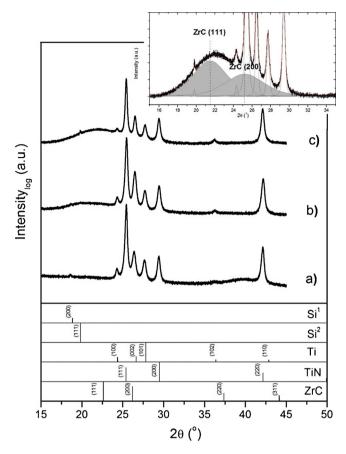


Fig. 3. Cross-sectional TEM BF images of a-C\_Zr(14) coating.

the shadow effects and low adatom mobility (see Fig. 2). Columnar growth is commonly associated with growth defects and low toughness coatings which may led to crack initiation/propagation through the columnar boundaries after high load contact [9]. The incorporation of Zr did not produce any significant morphological change; however, the films became denser and smoother. In our previous study, we found that the density was higher with the increase of Zr content. Fig. 3 shows a TEM BF image from the interface between adhesion interlayer and a-C\_Zr(14) functional coating. The pronounced columnar structure from the Ti-based interlayer quickly transforms to a rather featureless structure, where only thin and discontinuous columnar boundaries can be seen along the coating growth direction. The presence of heavier atoms, such as Zr, causes preferential growth of ZrC islands while carbon, as lighter atom with expected higher mobility, will be knocked-down to fill the voids reaching the shadowing portions of the coatings and thus promoting the structure rearrangement [13]. The introduction of CH<sub>4</sub> enhances the compact microstructure and eliminated the columnar structure of the films due to two factors: (i) the incorporation of H in the carbon coatings prevent formation of the column boundaries due to high mobility of hydrocarbon species [14] and (ii) a quick fill up of the shadowed areas, since the carbonaceous phases can land from any angle smoothing the growing interface and thus restraining the columnar growth [13].

# 3.2. Structure and chemical bonding

Fig. 4 shows the X-ray diffractograms of the alloyed nonhydrogenated films. The presence of Si, Ti and TiN peaks originated in a graded Ti/TiN/TiCN interlayer and Si substrate. With the increase of Zr content, a broad peak appeared at ZrC phase position (ICDD card no. 74-1221). The presence of broad peaks is common in nanocomposite and nanostructured films, where the limited size of the diffracting crystals causes broadening of the peak. The peaks were fitted using a pseudo-Voigt function (see Fig. 4 inset) and by fixing the position the interlayer correspondent peaks and allowing the background and the position areas of ZrC peaks to vary. For low Zr content (4 at.%), the deconvolution was found to be very difficult due to the extremely low intensity peak. The particle size estimated by the Scherrer formula [30] slightly increased from 1.0 to 2.2 nm



**Fig. 4.** XRD diffractograms of (a) a-C.Zr(4), (b) a-C.Zr(7) and (c) a-C.Zr(14) films with selected patterns. The inset shows the fitted curve s of the peak phases present in the a-C.Zr(14) diffractogram, highlighting the ZrC phase (gray filling).

for Zr content 4 and 14 at.%, respectively. The same was observed for the hydrogenated coating (not shown). Raman spectra was used to characterized the amorphous C-matrix. The incorporation of Zr and H led to the formation of aromatic rings and the organization and size of the C-clusters in the C-matrix as seen in our previous study [20].

TEM investigations were performed on selected samples to verify the XRD results. Fig. 5 shows the TEM micrographs and the schematic representation of the particle distribution of a-C.Zr(14) and a-C.Zr(4). The schematic representation was obtained by using the Mathematica software to model the particle distribution taking into consideration the nanoparticle volume fraction  $(V_F)$  and the ZrC nanoparticles size for a unit cell of  $20 \times 20 \times 5$  nm<sup>3</sup>. TEM of a-C.Zr(14) coating clearly shows ~2 nm nanocrystalline ZrC grains randomly distributed in an amorphous carbon matrix (see Fig. 5a). Other smaller nanoparticles are also expected to be present; however, their very small size (<1.5 nm) makes them almost imperceptible to the BF TEM. For a-C\_Zr(4) coating (see Fig. 5b) only smaller nanoparticles are expected to be dispersed in the a-C matrix. Again, such small sizes (<1 nm) should be rather difficult to identify in TEM micrographs. Fast Fourier transform (FFT) analysis of the a-C\_Zr(14) revealed sharp and discrete rings between the lattice planes corresponding to that of ZrC (not shown). Using the model, the average inter-particle distance was found to be around 9 nm, irrespectively of the Zr content. Thus, in all cases the separation of ZrC nanograins by an amorphous carbon phase is high enough (>particle size) to isolate the nanoparticles effectively and limits the grain growth via diffusion and coalescence [31].

Information about the chemical state and chemical environment of C and Zr in the coatings was evaluated by XPS. Fig. 6 shows

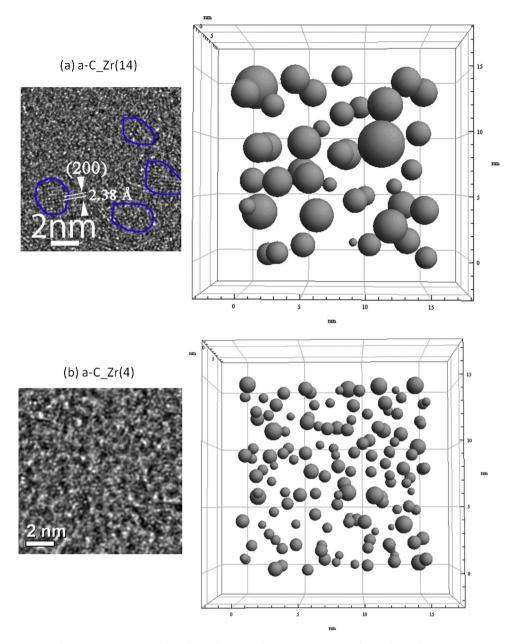


Fig. 5. TEM BF images and particle distribution schematic representation of a-C\_Zr(14) and a-C\_Zr(4).

the core-level spectra of the analyzed elements: C 1s, Zr 3d and O 1s after Ar<sup>+</sup> sputter cleaning. Tables 2 and 3 show, respectively, the relevant BE found in literature and the experimental BE values obtained after fitting the XPS spectra according to the different types of bonds expected in the coatings.

Fig. 6a shows the carbon C 1s spectral fitting for nonhydrogenated samples. The carbon peak was fitted using the main peak (285 eV) ascribed as alkyl type carbon (C–C, C–H). A second peak was added (286.3 eV) with the same FWHM as the main peak ascribing the alcohol (C–OH) and/or ester (C–O–C) functionality. Two other components were also added corresponding to the C=O and O–C=O at 287.8 eV and 289.3 eV, respectively. When Zr was incorporated in the matrix an extra peak appeared with BE close to  $283.5 \pm 0.1$  eV (see Table 3). The position of this peak (BE value) stayed approximately constant with Zr content in the films. A peak between typical C–Me and C–C bonds has been reported by many authors for nc-MeC/a-C systems (Me=Zr, Ti and Nb) [15–17,31,38,42]; it was suggested to originate from an interfacial state at the carbide and amorphous matrix interface (Me–C\*). The presence of nanocrystallites can enhance the charge transfer between the positive metal element and the C-matrix through adsorption of stable building units of a-C(:H) (e.g. extended sp<sup>2</sup>bonded islands with their stabilizing  $\pi$ -system) [42,44]. In fact, Magnuson et al. [17] suggested that, for nc-TiC/a-C films, electron transfer occurs between Ti 3d and C 2p. The electron deficiency, caused by such net charge transfer from Ti to the C-matrix, can be then spread out across the entire carbide grain [17,45]. For larger grains such deficiency is negligible; however, for nanograins the electron deficiency will weaken the Me—C bond within carbide. These interface regions were predicted to be around 1 nm thick [16]. A higher total charge transfer from the Zr nearest neighbors to interfacial C atoms is expected leading to higher chemical shift [42,45]. Thus, for the present case, the ZrC bonds are expected to be characteristic of small ZrC nanoparticles (C—Zr\*).

Zr 3d spectra are shown in Fig. 6c. The Zr metallic phase (178.0–179.1 eV) was not observed in any sample. Likewise in the C 1s spectra, Zr 3d peak also shifts approximately 1.0 eV to HBE when compared to typical ZrC peak of a bulk material (179.9 $\pm$ 1.8 eV).

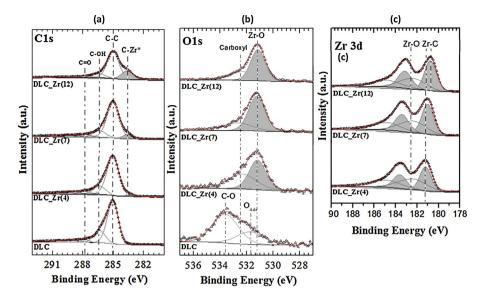


Fig. 6. XPS (a) C1s, (b) O1s and (c) Zr 3d spectra for the non-hydrogenated films with different Zr contents after Ar<sup>+</sup> etching of the surface oxides.

Many authors [38,44,46,47] reported similar shifts for small grain sizes. Contrarily to the C 1s, for the Zr 3d case it was observed a reduction of the BE shift with increasing Zr content (Fig. 6c and Table 2). For smaller cluster sizes the electron deficiency in Zr atoms is expected to be higher, leading to HBE shifts [45].

The presence of Zr–O bonds in the Zr 3d spectra (182.4eV) was also observed, even after ion bombardment. Despite the low oxygen content in the film, the O–Zr contribution in the Zr 3d spectra was around 50% for higher Zr contents. The ion bombardment may induce carbon preferential sputtering causing ZrC nanocrystals exposure to the environment resulting in ZrO<sub>2</sub> formation [27]. Zr is easily oxidized (ZrO<sub>2</sub>) showing its high ability to passivate, which might be a beneficial property for corrosion resistance. These bonds were also confirmed in O 1s range of energies revealing a peak at 531.2 eV previously reported as Zr–O bond (Fig. 6b).

#### 3.3. Mechanical properties

DLC films are known to have high compressive residual stress due to their peculiar amorphous structure (>3 GPa) [48]. Table 4 shows the compressive residual stress for Zr-free and Zr-containing films. In our case, thermal mismatch is minimized since magnetron sputtering is characterized by a low deposition temperature [49] and no external heating was used. Furthermore, the use of graded layers (Ti/TiN/TiNC) helped to decrease the abrupt composition changes between the film and the substrate and thus decreasing the residual stress and improving the adhesion. The intrinsic stress was strongly dependent on the chemical composition, morphology and microstructure of the films. The incorporation of Zr relaxed the stresses in the films: the higher the Zr content, the lower the stress. Zr acted as a catalyst forming a nanophase of ZrC and reacting with the C-matrix by overlapping Zr d-orbitals with C p-orbitals. The

#### Table 2

Literature values on BE of the C1s, O1s and Zr 3d core-levels relevant for this study.

Bond type	Spectra	Referenced BE (eV)	FWHM (eV)	Ref.
Zr (metal)	Zr 3d <sub>5/2</sub>	178.5 178–179.1	1.8	[32] [21,33]
Zr—C	Zr 3d <sub>5/2</sub>	179.8 181.1 178.6-180.2	~1.2 ~ 2 _	[34] [35] [33,36–39]
Zr—0 (Zr <sup>4+</sup> )	Zr 3d <sub>5/2</sub>	182.3–182.4 185.1 182.9 (Zr <sup>4+</sup> ) 181.9–183	~2.6 ~2.5 1.8	[34] [35] [32] [21,33,37]
C—C C—O—C, C—OH C=O O—C=O	C 1s C 1s C 1s C 1s	284.2-285 285-286.5 287.8-288 288.6-289.3	- - -	[21,37,40,41] [21,37,40,41] [21,37,40] [37,40,41]
Zr—C	C 1s	282.5 282.9 281.7–282.3	~2 ~1.5	[34] [35] [33,36–38]
Zr-C* OAds	C 1s O 1s	282.6-283.3 531.9 531.5	- - 2.2	[20,38,42] [37] [41]
Zr—O	O 1s	531.2 530.2-530.6	1.7	[32] [21,33,37]
C—O, C=O, O—C=O (carbonyl)	0 1s	533.2-533.6 532.4-532.7	_	[41] [34,43]

Table 3	
XPS BE for the deconvolution of the peaks after Ar <sup>+</sup> s	sputtering.

Sample	Spectra	C—C		C—0, C=0, 0	C=O(carbonyl)	Zr—C		Zr—O (Zr	5/2)	O (adsorbed moisture)	
		BE (eV)	FWHM (eV)	BE (eV)	FWHM (eV)	BE (eV)	FWHM (eV)	BE (eV)	FWHM (eV)	BE (eV)	FWHM (eV)
DLC	C 1s	285.0	1.3	286.3	1.3	-	_				
				287.8	1.0						
				289.3	2.0						
	O 1s			533.6	2.3			-	-	531.8	2.4
DLC_Zr(4)	C 1s	285.0	1.4	286.3	1.4	283.4	1.0				
				287.8	1.4						
				289.3	2.0						
	O 1s			532.3	2.5			531.2	1.4	-	-
	Zr 3d					181.2	1.2	182.4	3.1		
$DLC_Zr(7)$	C 1s	285.0	1.4	286.3	1.4	283.6	1.1				
				287.8	1.3						
				289.3	2.0						
	O 1s			532.4	2.7			531.2	1.4	-	-
	Zr 3d <sub>5/2</sub>					181.0	1.3	182.4	2.9		
DLC_Zr(12)	C 1s	285.0	1.3	286.4	1.3	283.6	1.4				
				287.8	1.4						
				289.3	2.0						
	O 1s			532.3	2.5			531.2	1.3	-	-
	Zr 3d <sub>5/2</sub>					180.8	1.3	182.4	2.8		
DLC:H_Zr(8)	C 1s	285.0	1.4	286.3	1.4	283.6	1.3				
				287.8	1.5						
				289.3	2.0						
	O 1s			532.4	2.8			531.2	1.5	-	-
	Zr 3d <sub>5/2</sub>					180.7	1.2	182.4	2.6		

break of the C bonds may release the stress through the formation of a more graphite-like film as seen in our previous study [20]. The same tendency was also observed by Zou et al. [50] for DLC coatings doped with low Cr contents. On the other hand, simultaneous incorporation of both H and Zr did not promote such relaxation. The presence of a reactive gas (CH<sub>4</sub>) enhanced the bombardment of the film by the hydrocarbon species and their consequent penetration in the structure. Thus, a denser network structure was formed (see Section 3.1) which led to an increase in the residual stress [13,48].

The presence of ZrC nanograins did not significantly change the hardness of the coatings compared to Zr-free ones. The hardness of nanocomposite films is typically influenced by the size, orientation and shape of the embedded nanograins [11]. For hard coatings the key challenge is to avoid grain boundary sliding leaving the grain rotation as the deformation mechanism: small interparticle separation (few atomic layers) and higher volume fraction  $(V_F)$  of nano particles (3–5 nm) is required. On the other hand, toughness is achieved by larger interparticle separation and thus, lower  $V_F$ , which allows shear delocalization [11,44,51]. Since small nanograins of ZrC (<3 nm) were separated by a thick carbon layer (see Section 3.2) the classical Hall-Petch effect cannot be considered; moreover, two factors contribution to high hardness could be well balanced in our system - higher amount of carbides results in lower compressive stress, so the final hardness could be almost identical for different Zr contents and similar to pure carbon films.

### Table 4

Mechanical properties of the coating: compressive residual stress ( $\sigma_{compressive}$ ), hardness (H), young modulus (E), H/E and  $H^3/E^2$  ratios.

Samples	$\sigma_{ m compressive}$ (GPa)	H(GPa)	E(GPa)	H/E	H <sup>3</sup> /E <sup>2</sup> (GPa)
a-C	$1.9\pm0.6$	$11.2 \pm 0.6$	$110.4\pm6.5$	0.10	0.12
$a-C_Zr(4)$	$1.6 \pm 0.3$	$11.7\pm0.5$	$118.2\pm0.9$	0.10	0.12
a-C_Zr(7)	$1.0\pm0.6$	$11.4\pm0.6$	$126.6\pm4.4$	0.09	0.09
a-C_Zr(13)	$0.3\pm0.1$	$11.9\pm0.5$	$137.2\pm1.6$	0.09	0.09
a-C:H_Zr(10)	$1.7 \pm 0.4$	$12.7\pm0.6$	$120.2\pm1.0$	0.11	0.14

On the other hand, the Young modulus increased with the increase of Zr-content (see Table 4), since the embedded ZrC phase presents higher elastic modulus compared to a-C matrix [5].

Other important parameters for hard coatings are H/E and  $H^3/E^2$  which are closely related to the wear resistance and the plastic behavior under Hertzian contact, respectively [52]. Since the hardness was almost independent of Zr content, both parameters slightly decreased with the increase of Zr content. The methane flux during the deposition resulted in an increase of H/E and  $H^3/E^2$  when compared to non-hydrogenated films. As expected, the lack of columnar boundaries (i.e. glassy microstructure, see Fig. 1) together with modified nature of carbon matrix by hydrogen led to slightly higher hardness value. The wear resistance is thus expected to be higher for Zr-containing hydrogenated samples.

# 4. Conclusion

The formation of nanocomposite structures consisting of ZrC nano-particles in an amorphous carbon matrix has been analyzed in detail by SEM, XRD, XPS and TEM. The incorporation of 4 at.% of Zr resulted in the presence of ZrC nanophase embedded in a carbon matrix. The grain size estimated by XRD peak fitting and directly measured by TEM was between 1.0 and 2.5 nm. The presence of such nanophase ZrC was also confirmed by XPS where additional bonding states were identified due to the small grain size of the carbide phase. The interaction of ZrC with the matrix led to significant charge transfer, which shifted ZrC XPS peak to HBE. Mechanical properties were related to chemical bonding and coating microstructure. With the increase of Zr content the columnar microstructure of the coatings became featureless. The hardness was almost independent of Zr content; however, the change in Young modulus led to different H/E and  $H^3/E^2$ , which decreased with the increase of Zr content. Higher amount of Zr decreased residual stress. The introduction of the reactive gas into the chamber further enhanced the properties of the coatings. The high mobility of the hydrocarbon species led to a more compact/glassy microstructure, constraining the typical columnar microstructure and improving the mechanical properties.

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